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Supporting information Differences in heterocycle basicity distinguish homocysteine from cysteine using aldehyde-bearing fluorophores

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Materials

All chemicals were purchased from Sigma Aldrich and used as received. Ultrapure water obtained from a Milli-Q direct water purification system was used to prepare all aqueous solutions. All spectroscopic measurements were carried out in DMSO:buffer (1:99) solutions. **1** and **2** were synthesized from the procedure reported in the literature.¹

Methods

¹H-NMR and ¹³C-NMR were recorded on a ARX-400 Advance Bruker spectrometer. All chemical shifts (δ) are reported in ppm relative to DMSO- d_6 (2.50 ppm, ¹H; 39.52 ppm, ¹³C) unless otherwise indicated. ESI-HRMS (high resolution mass spectrometry) were recorded at the PSU Bioanalytical Mass Spectrometry Facility on a ThermoElectron LTQ-Orbitrap high resolution mass spectrometer.

Spectrophotometric measurement

Fluorescence spectra were obtained on a Cary Eclipse fluorescence spectrophotometer (Agilent Technologies) with slit widths set at 5 nm for both excitation and emission, respectively. Fluorescence spectra were corrected for the wavelength dependent response of the R928 photomultiplier tube with the help of a manufacturer generated correction file.

pK_a measurements

pKa values of the amine of 2-substituted thiazolidines/thiazinanes-4-carboxylic acids, **4a**, **4b**, **5a** and **5b** were obtained by the titration of their aqueous solutions with 0.01M NaOH. In the case of **3a** and **3b** 0.1 M NaOH was used. The pH measurements were conducted using an Orion 410A pH meter, which was standardized with Scientific Products potassium hydrogen phthalate pH 4.00 buffer, BDH pH 7.00 and Scientific Products boric acid/potassium hydroxide pH 10.00 buffer. All pH titrations were done in triplicates and their average is reported.

Deproteinization of plasma

Deproteinization of plasma was carried out by reconstituting the lyophilized bovine plasma with ultrapure water to 1/3 of the reconstitution volume. Two equivalents of acetonitrile (2/3 of the reconstitution volume) were added. This solution was vortexed for 10 min. after letting the solution stand for 10 minutes, it was centrifuged at 4000 RPM for 30 minutes to pellet the proteins. The decanted supernatant was aliquoted into sample vials and re-lyophilized. The vials were then sealed and stored in a refrigerator (4 °C).

Synthesis of compounds 3a, 3b, 4a, 4b, 5a and 5b

Scheme S1. Synthetic route of 3a, 4a and 5a



Scheme S2. Synthetic route of 3b, 4b and 5b



(2S,4S) and (2R,4S)-2-(2-hydroxy phenyl) thiazolidine-4-carboxylic acid (3a)

L-cysteine (1.67 mmol) and salicylaldehyde (1.67 mmol) in 5 mL of distilled H₂O are stirred at RT for 2 h. The precipitates are filtered, washed with iced cold H₂O and dried under vacuum to yield **3a** as a white solid (307 mg, 83%).¹H-NMR (400 MHz, DMSO-*d*₆); δ (ppm): 9.29 (m, 1H), 7.06 (m, 1H), 6.80 (m, 1H), 5.84 (s, 1H), 5.65 (s,1H), 4.21 (t, *J* = 4.6 Hz 1H), 3.85 (m, 1H), 3.18 (m, 1H), 3.03 (m, 1H). ¹³CNMR (101 MHz, DMSO-*d*₆); δ (ppm): 172.88, 172.40, 155.13, 154.54, 128.99, 128.06, 127.85, 127.56, 126.04, 124.17, 118.99, 118.70, 116.62, 115.03, 67.61, 65.56, 65.13, 64.73, 38.10, 37.01. EI-MS, *m/z* 226.05602 [M+H]⁺, calculated for C₁₀H₁₁NO₃S, *m/z* 225.04596.

(2S,4R), (2S,4S), (2R,4R) and (2R,4S)-2-(2-hydroxyphenyl)-1,3-thiazinane-4-carboxylic acid (3b)

DL-homocysteine (1.88 mmol) and salicylaldehyde (1.88 mmol) in 6 mL of distilled water and 0.25 mL of ethanol are stirred at RT for two days. The resultant precipitate is filtered, washed with iced cold H₂O and dried under vacuum to yield **3b** as a white solid. (322 mg, 82%). ¹H-NMR (400 MHz, DMSO-*d*₆); δ (ppm): 7.31 (m, 1H), 7.11 (m, 1H), 6.80 (m, 2H), 3.43 (s, 1H), 3.57 (d, *J* = 3.60, 1H), 3.19 (t, *J* = 11.64 Hz, 1H), 2.86 (m, 1H), 2.05 (m, 1H), 1.49 (m, 1H). ¹³CNMR (101 MHz, DMSO-*d*₆); δ (ppm): 173.48, 154.12, 128.71, 127.81, 126.42, 118.96, 115.63, 59.45, 59.26, 28.77, 27.93. EI-MS, *m/z* 240.07182 [M+H]⁺, calculated for C₁₁H₁₃NO₃S, *m/z* 239.06161.

(2S,4S) and (2R,4S)-2-phenylthiazolidine-4-carboxylic acid (4a)

Compound **4a** was prepared according to the procedure described for **3a** but using benzaldehyde instead of salicylaldehyde. Compound **4a** was obtained as white solid (488 mg, 99%). ¹H-NMR (400 MHz, DMSO- d_6) δ (ppm) 7.4 (m, 5H) 5.76 (s, 1H), 4.24 (q, *J* = 4.53 Hz, 1H), 3.28 (m, 1H), 3.15 (m, 1H) . ¹³CNMR (101 MHz, DMSO- d_6) δ (ppm) 172.95, 172.19, 141.18, 139.91, 128.31, 127.25, 127.57, 126.91, 71.73 71.06, 6540, 64.85, 38.39, 37.94. EI-MS, *m/z* 210.06017 [M+H]⁺, calculated for C₁₀H₁₁NO₂S, *m/z* 209.05105.

(2S,4R), (2S,4S), (2R,4R) and (2R,4S)-2-phenyl-1,3-thiazinane-4-carboxylic acid (4b)

Compound **4b** was prepared according to the procedure described for **3b** but using benzaldehyde instead of salicylaldehyde in 6 mL of distilled H₂O and stirred at RT for three days. Compound **4b** was obtained as white solid (273 mg, 65%). ¹H-NMR (400 MHz, DMSO-*d*₆) δ (ppm) 7.35 (m, 5H), 5.27 (s, 1H), 3.62 (dd, *J* = 11.99, 2.65 Hz, 1H), 3.23 (dt, *J* = 2.64, 13.80 Hz, 1H), 2.91 (td, *J* = 13.55, 3.43 Hz, 1H), 2.07 (qd, *J* = 13.5, 2.58, 3.58 Hz, 1H), 1.52 (dq, *J* = 3.48, 3.21, 11.71 Hz, 1H). ¹³CNMR (101 MHz, DMSO-*d*₆) δ (ppm) 173.50, 140.76, 128.32, 127.79, 126.53, 64.06, 59.19, 28.87, 27.72. EI-MS, *m/z* 224.07707 [M+H]⁺, calculated for C₁₁H₁₃NO₂S, *m/z* 223.06670.

(S)-thiazolidine-4-carboxylic acid (5a)

Compound **5a** was prepared according to the procedure described for **3a** but using formaldehyde instead of salicylaldehyde. Compound **5a** was obtained as a white solid. (152 mg, 68.5 %). ¹H-NMR (400 MHz, D₂O) δ (ppm) 4.43 (m,2H), 4.36 (d, *J* = 10.2 Hz, 1H), 3.43 (m, 1H), 3.33 (m, 1H). ¹³CNMR (101 MHz, D₂O) δ (ppm) 171.87, 63.89, 48.55, 32.91. EI-MS, *m*/*z* 134.0252 [M+H]⁺, calculated for C₄H₇NO₂S, *m*/*z* 133.01975.

(R) and (S)-1,3-thiazinane-4-carboxylic acid (5b)

DL-homocysteine (1.67 mmol) and formaldehyde (1.67 mmol) in 6 mL of distilled water were stirred at RT for 2 days. The precipitates were filtered. Ethanol was added to the filtrate and left the solution standing overnight. After one day, transparent shiny crystal were obtained. Crystals were washed with iced cold ethanol and dried under vacuum to yield **5b**. (135 mg, 55%) ¹H-NMR (400 MHz, D₂O) δ (ppm) 4.26 (dd, J = 32.0, 13.2 Hz, 2H), 3.68 (dd, J = 11.9, 2.4 Hz 1H), 2.96 (m, 1H), 2.85 (m, 1H), 2.53 (dd, J = 14.8, 3.2 Hz, 1H), 2.04 (qd, J = 1H). ¹³CNMR (101 MHZ, D₂O) δ (ppm) 173.27, 59.42, 45. 12, 28.10, 26.00. EI-MS, *m/z* 148.04422 [M+H]⁺, calculated for C₅H₉NO₂S, *m/z* 147.03540.



Figure S1. ¹HNMR spectrum of 3a (top) and 3b (bottom)



Figure S2. ¹³CNMR spectrum of 3a (top) and 3b (bottom)



Figure S3. Mass spectrum of 3a (top) and 3b (bottom)



Figure S4. ¹HNMR spectrum of 4a (top) and 4b (bottom)



Figure S5. ¹³CNMR spectrum of 4a (top) and 4b (bottom)



Figure S6. Mass spectrum of 4a (top) and 4b (bottom)



Figure S7. ¹HNMR spectrum of 5a (top) and 5b (bottom)



Figure S8. ¹³CNMR spectrum of 5a (top) and 5b (bottom)



Figure S9. Mass spectrum of 5a (top) and 5b (bottom)



Figure S10. Absorbance spectra (top) and absorbance as a function of pH (bottom) of 1 (left) and 2 (right), $(4.0 \times 10^{-6} \text{ M})$ in 0.1M phosphate and carbonate buffer at pH 4.5-9.0, λ_{abs} at 480 nm and 500 nm



Figure S11. Optical sensing behavior of 1 towards Cys and Hcy at pH 6.0, λ_{abs} and $\lambda_{ex} = 480 \text{ nm}$ - (a,b) Time dependent absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, λ_{abs} and $\lambda_{ex} = 480 \text{ nm}$. (c,d) Time course absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, λ_{abs} and $\lambda_{ex} = 480 \text{ nm}$. (c,d) Time course absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, $\lambda_{ex} = 480 \text{ nm}$. Solutions are composed of 4 μ M of 1 with 1 mM of analyte in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{em} = 515 \text{ nm}$.



Figure S12. Optical sensing behavior of 1 towards Cys and Hcy at pH 6.0, λ_{abs} and $\lambda_{ex} = 495 \text{ nm} - (a,b)$ Time dependent absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, λ_{abs} and $\lambda_{ex} = 495 \text{ nm}$. (c,d) Time course absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, λ_{abs} and $\lambda_{ex} = 495 \text{ nm}$. (c,d) Time course absorbance (left) and fluorescence (right) spectral changes of 1 with Cys and Hcy, $\lambda_{ex} = 495 \text{ nm}$. Solutions are composed of 4 μ M of 1 with 1 mM of analyte in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{em} = 515 \text{ nm}$.



Figure S13. Optical sensing behavior of **2** towards Cys and Hcy at pH 6.0 - (a,b) Time dependent absorbance (left) and fluorescence (right) spectral changes of **2** with Cys and Hcy, λ_{abs} and $\lambda_{ex} = 495$ nm. (c,d) Time course absorbance (left) and fluorescence (right) spectral changes of **2** with Cys and Hcy, $\lambda_{ex} = 495$ nm. Solutions are composed of 4 μ M of **2** with 1 mM of analyte in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{em} = 515$ nm.



Figure S14. H¹ NMR of 1,1 in the presence of Hcy (3 equiv) and 1 in presence of Cys (3 equiv) at pH 6.



Figure S15. ¹H NMR of salicylaldehyde-Cys derived thiazolidines 3a and salicylaldehyde-Hcy derived thiazinanes 3b in D₂O:H₂O.



Figure S16. Average of triplicates of pH titration curve and derivatives 3a, 3b, 4a, 4b, 5a and 5b obtained from pH titration- (a) salicylaldehyde-Cys derived thiazolidine 3a and salicylaldehyde -Hcy derived thiazinane 3b, (b) benzaldehyde-Cys derived thiazolidine 4a and benzaldehyde -Hcy derived thiazinane 4b, (c) HCHO-Cys derived thiazolidine 5a and HCHO-Hcy derived thiazinane 5b. For (a) 0.1 M NaOH was used and for (b and c) 0.01 M of NaOH was used.



Figure S17. Time dependent fluorescence spectral changes (left) and enhancement factor (right) of fluorescence of **1** with Cys and Hcy, (a₁ and a₂) pH 5.5, (b₁ and b₂) pH 6.0, (c₁and c₂) pH 6.5, (d₁and d₂) pH 7.0, (e₁and e₂) pH 7.4. Solutions are composed of 4 μ M of **1** with 1 mM analyte in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{ex} = 495$ nm and $\lambda_{em} = 515$ nm. For Time dependent fluorescence spectral changes, $\lambda_{ex} = 495$ and $\lambda_{em} = 515$. Enhancement factor (EF) is the response of **1** + analyte divided by 1. When EF < 1 quenching is observed, while when EF > 1 enhancement is observed. The vertical line in the enhancement factor plots is the experimentally determined optimal excitation wavelength of 495 nm. At pH 6.0 (b2), excitation at 495 results in enhancement for Cys in response to **1**.



Figure S18. Optical sensing behavior of 1 towards Cys and Hcy as compared to other biologically relevant species. Fluorescence response of 1 upon addition of different analytes (Cys, Hcy, GSH,Asn, Ala, Gln, Thr, Asp, Arg, His, Glu, 1 mM). All Solutions are composed of 4 μ M of 1 with 1 mM analyte in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{ex} = 495$ nm $\lambda_{em} = 515$ nm.



Figure S19. Spectral response of 2 towards increasing levels of Hcy in phosphate buffer (100 mM, pH 6.0). Solutions are composed of 4 μ M of 2 with 0-80 μ M of Hcy in phosphate buffer (100 mM, pH 6.0):DMSO 99:1at 20 °C, $\lambda_{ex} = 500$ nm and $\lambda_{em} = 515$ nm.



Figure S20. Optical sensing behavior of **2** towards Cys and Hey at their biological concentrations at pH 6.0 - (a) Time dependent fluorescence spectral changes of **2** with Cys and Hey. (b) fluorescence spectra of **2** with Cys and Hey after two hours. Solutions are composed of 30 μ M of **2** with 250 μ M of Cys and 15 μ M of Hey in phosphate buffer (100 mM, pH 6.0):DMSO 99:1 at 20 °C, $\lambda_{ex} = 500$ nm and $\lambda_{em} = 515$ nm.



Figure S21. Optical sensing behavior of **2** towards Cys and Hcy in deproteinized plasma at pH 6.0. Time dependent fluorescence spectral changes of **2** with Cys and Hcy. Solutions are composed of 25 μ M of **2** with 250 μ M of Cys and 100 μ M of Hcy in phosphate buffer (100 mM, pH 6.0):DMSO 99:1at 20 °C, $\lambda_{ex} = 495$ nm and $\lambda_{em} = 515$ nm.

Reference

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