1	Supplementary Information
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3	Accumulation of Supramolecular Nanoparticles
4	Self-Assembled from a Bola-Shaped Cytidylic Acid-Appended
5	Fluorescein Dye into Cell Nucleus
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Materials and general methods: Fluorescein was purchased from Kishida Chemical 1819 Co. (Osaka, Japan), and 5-bromo-1-pentanol, potassium carbonate, and sodium borohydride were purchased from Tokyo Chemical Industry (Tokyo, Japan). 20N-Benzovl-3'-O-[2-cvanoethoxy(N,N-diisopropylamino)phosphino]-5'-O-(4,4'-dimetho 21 22xytrityl)cytidine was purchased from Applied Biosystems (Tokyo, Japan). The 23structures of the intermediates and the final products were confirmed by Fourier transform infrared (FTIR) spectroscopy, proton nuclear magnetic resonance $\mathbf{24}$ spectroscopy (¹H NMR), and electrospray ionization Fourier-transform ion cyclotron 2526resonance mass spectrometry (ESI FTICR-MS) or electrospray ionization Fourier-transform orbitrap mass spectrometry (ESI FTMS). ¹H NMR spectra were 2728recorded on a Bruker Advance 400 spectrometer (Bruker BioSpin, MA, U.S.A.). FTIR measurements were performed with a Spectrum One spectrometer (Perkin-Elmer, MA, 2930 U.S.A.). Mass experiments were performed with an Apex II 70e mass spectrometer 31(Bruker Daltonics, MA, U.S.A.) or an Orbitrap Veros Pro ETD mass spectrometer 32(Thermo Fisher Scientific, MA, U.S.A.). Fluorescence spectra were recorded on an RF-5300 spectrofluorometer (Shimadzu, Kyoto, Japan). 33

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Synthesis of bis-(5-hydroxypentyloxy)fluorescein: A mixture of fluorescein (3.2 g, 10 mmol) and potassium carbonate (6.9 g) in 200 mL of dimethyl formamide was warmed until complete dissolution had occurred. 5-Bromo-1-pentanol (7.4 g, 45 mmol) was then added over 20 min. The reaction mixture was refluxed for 12 h and cooled to room temperature, and the solvent was evaporated under reduced pressure. Water was added to the remaining solid, and the aqueous layer was extracted with dichloromethane. The collected dichloromethane extract was dried with magnesium sulfate overnight and 42filtered. Fractionation by gel permeation chromatography with chloroform and 43subsequent evaporation of the solvent gave bis-(5-hydroxypentyloxy)fluorescein as an orange oil (2.5 g, 50% pure, 5 mmol). Thin layer chromatography (TLC): (silica gel, 44chloroform/methanol = 4/1, v/v) $R_f = 0.78$. ¹H NMR: (400 MHz in CDCl₃ at 25 °C) δ 451.5-1.6 (m, 4 H, c), 1.6-1.7 (m, 4 H, d), 1.8-1.9 (m, 4 H, b), 3.6-3.7 (t, 4 H, e), 6.6 (dd, 46J = 9.7 and 1.9 Hz, 1 H, g), 6.75 (d, J = 1.8 Hz, 1 H, f), 6.9 (d, J = 6.6 Hz, 1 H, h), 7.3 47(dd, J = 7.6 and 1.1, 1 H, 1), 7.7 (dt, J = 7.6 and 1.4 Hz, 1 H, j), 7.7 (dt, J = 7.5 and 1.5)48 Hz, 1 H, k), 8.3 (dd, J = 7.7 and 1.4 Hz, 1 H, i). ESI-FTMS (m/z): Calcd for C₃₀H₃₃O₇⁺ 49[M+H]⁺, 505.2221; Found, 505.2217. 50

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Synthesis of bis[5-(3'-cytidylyloxy)pentyloxy]fluorescein (C-FLU-C): C-FLU-C was
synthesized as previously described with slight modifications (R. Iwaura et al. *Chem. Mater.* 2002, *14*, 3047.). Dried tetrahydrofuran (THF, 60 mL) was added to 1.0 g (1.2
mmol) of

56N-benzoyl-3'-O-[2-cyanoethoxy(N,N-diisopropylamino)phosphino]-5'-O-(4,4'-dimetho and solution 57xytrityl)cytidine, the was mixed with 583,6-bis-(5-hydroxypentyloxy)fluorescein (250 mg, 0.5 mmol) and 1H-tetrazole (0.2 g, 2.8 mmol) and stirred for 3 days. To this reaction mixture, 0.3 mL of 70% 5960 t-butylhydroperoxide aqueous solution was added and the mixture was stirred for 1 h. 61The solution in the reaction mixture was evaporated under reduced pressure, the residue 62 was purified by gel permeation chromatography with chloroform as the eluent followed 63 by the evaporation of the solvent. The protecting cyanoethyl group of the residue was 64 removed by mixing with 4 mL of 25% aqueous ammonia and 20 mL of methanol for 1.5 h at 50 °C and 24 h at room temperature and then evaporating the solvent. The 65

66 resulting residue was treated with a chloroform solution containing 5% trifluoroacetic 67 acid to remove the protecting dimethoxytrityl group. After the removal of the solvent, the residual solid was reprecipitated from acetone, and the precipitate was filtered and 68 dried. The final compound, C-FLU-C, was obtained as a yellow powder (110 mg, 20%) 69 pure). TLC: (silica gel, chloroform/methanol = 4/1, v/v) R_f = 0.1. ¹H NMR: (400 MHz) 70in dimethyl sulfoxide-d₆ at 25 °C) δ 1.5–1.6 (m, 4 H, c), 1.6–1.7 (m, 4 H, d), 1.8–1.9 (m, 714 H, b), 2.0–2.3 (br, 4 H, 2'-H), 3.5 (d, J = 3.8 Hz, 4 H, 5'H), 3.6–3.7 (br, 4 H, e), 723.8-3.9 (br, 4H, a) 4.5 (br, 2 H, 4'), 5.7 (br, 2 H, 5-H), 6.1 (t, J = 6.5 Hz, 2 H, 1'-H), 6.573(dd, J = 9.7 and 1.9 Hz, 2 H, g), 6.6 (d, J = 1.8 Hz, 2 H, f), 6.7 (br, 2 H, h), 6.8 (br, 1 H, f)741), 7.0 (br, 1 H, j), 7.2 (br, 1 H, k), 7.4 (m, 1 H, i) 7.7 (d, J = 7.8 Hz, 2 H, 6-H), 7.8 (m, 754-NH₂). IR: v (cm⁻¹) = 3191, 3069, 2945, 1701, 1645, 1600, 1496, 1471, 1427, 1372, 76771330, 1281, 1177, 1103, 1056, 1001, 951, 827, 782, 759. ESI FTICR-MS (m/z): Calcd for $C_{48}H_{55}N_6O_{19}P_2^{-}$ [M-H]⁻, 1081.3003; Found, 1081.3015. mp = 177 °C. 78



85 (b) ¹H NMR spectrum of **C-FLU-C**.

Atomic force and transmission electron microscopic (AFM and TEM) observations 86 of the nanoparticles: For the AFM observation, 1 μ L of the aqueous solution 87 containing C-FLU-C ([C] = 2×10^{-3} mol/L) was placed on a sheet of highly ordered 88 pyrolytic graphite and dried in air for a few minutes. The specimen was then washed 89 90 with excess cold Milli-Q water and blotted with filter paper. Tapping-mode AFM was 91carried out with a NanoNavi station and an S-image system (1024×1024 pixels; SII 92Nanotechnology, Tokyo, Japan) and a silicon microcantilever (spring constant = 40 Nm^{-1} , frequency = 120 kHz, aluminum coated, SII Nanotechnology). For the TEM 93 observation, the aqueous solution containing C-FLU-C was diluted 100 times. A drop 94of the diluted solution was put on an elastic carbon-supporting film, dried at room 95temperature, and then subjected to TEM observation with an H-9500 300 kV 96 transmission electron microscope (Hitachi High Technologies, Tokyo, Japan) and a 97 CCD camera (AMT Corp. ER-B, Woburn, MA). 98



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Fig. S2 (a,b) Transmission electron microscopic images and (c) atomic force
 microscopic images of self-assembled C-FLU-C nanoparticles in aqueous solution.

102 **Particle size distribution measurement:** The particle size distribution of 103 self-assembled **C-FLU-C** nanoparticles in aqueous solution ($[C] = 4 \times 10^{-4} \text{ mol/L}$) was 104 analyzed with a Zetasizer (Malvern Instruments, UK) at 20 °C.



106 Fig. S3 Particle size distribution of self-assembled C-FLU-C nanoparticles in aqueous

107 solution.

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109 Fig. S4 Normalized emission spectra of fluorescein (---) and C-FLU-C (---) in Tris-HCl

110 buffer (pH = 8) at 25 °C ([C] = 2×10^{-6} M, Ex. = 480 nm).

Cell cultures: Caco-2 cells (human colon carcinoma cell line) and BALB/3T3 cells
(contact-inhibited seminormal cell line: mouse embryo cell line) were obtained from

114RIKEN Cell Bank (Tsukuba, Japan) and were used to examine the uptake characteristics of self-assembled C-FLU-C nanoparticles. An aqueous solution of 115C-FLU-C was added to both cell lines after 1 day of culture, and the concentration was 116adjusted to 2×10^{-6} mol/L. Both cell lines were cultured in culture medium containing 117 antibiotics (Wako Pure Chemical Industries, Tokyo, Japan), non-essential amino acids 118 119 (GIBCO Life Technologies, CA, U.S.A.), and 10% fetal bovine serum (FBS, Sigma 120Aldrich, MO, U.S.A.). Dulbecco's modified Eagle's medium and Eagle's minimal 121essential medium (both GIBCO Life Technologies) were used as the culture medium for the Caco-2 and BALB/3T3 cells, respectively. The culture medium was changed at 3, 7, 122123and 10 days of cell culture. Cells were grown in T25 tissue culture flasks (Falcon, 124Corning Incorporated Life Sciences, MA, U.S.A.) at 37 °C in 100% humidity and 5% CO₂ and were passaged when they reached near confluency. 125

As references, the experiments were repeated by using fluorescein instead of C-FLU-C. Fluorescein was dissolved in a trace amount of dimethyl sulfoxide, and the concentration of fluorescein was adjusted to 5×10^{-6} mol/L by using water.

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Measurements of cell growth curves: Caco-2 or BALB/3T3 cells were cultured in C-FLU-C-containing medium in 12-well plates and were treated with 0.025% trypsin solution after 1–14 days incubation. Cultured cell suspensions were collected by centrifugation ($100 \times g$, 3 min) and washed with phosphate buffered saline (PBS) several times before the cells were counted.

As references, both cell lines were cultured, and the cells were counted without the
addition of C-FLU-C. Cytotoxicity and the median lethal dose (LD₅₀) of C-FLU-C on

137 Caco-2 cells were determined with a WST-1 cell proliferation assay system (Takara,138 Shiga, Japan).

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141Fig. S5 Cell growth curves of Caco-2 cells (\circ), Caco-2 cells + C-FLU-C (\bullet),142BALB/3T3 cells (\Box), and BALB/3T3 cells + C-FLU-C (\bullet). [C-FLU-C] = 2 × 10⁻⁶143mol/L. Median lethal dose (LD₅₀) = 3.5 × 10⁻³ mol/L.

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145 **Optical and laser-scanning confocal microscopic observations and flow cytometric** 146 **analysis of cells:** Cells cultured with **C-FLU-C** nanoparticles were washed with PBS 147 and harvested by adding a trypsin/ethylenediaminetetraacetic acid mixture (Wako). 148 Trypsin was inactivated with a mixture of culture medium and FBS, and cells were 149 resuspended in PBS and collected by centrifugation ($100 \times g$, 3 min). The collected cells 150 were washed three times with PBS and resuspended in PBS, and the suspension was 151 subjected to optical microscopic measurements and flow cytometric analysis. 152Fluorescence microscopic images were obtained with an optical microscope (BX51; 153Olympus, Tokyo, Japan) equipped with a $40 \times$ objective lens, a $10 \times$ ocular lens (Olympus), band-pass filters (510-550 nm Olympus), and a 3-CCD video camera 154(CS520MD; Olympus). Laser-scanning confocal microscopic observations of cells were 155performed on a TCS SP8 system (Leica Microsystems, Tokyo, Japan) with a visible 156157semiconductor laser operating at a wavelength of 488 nm and a fluorescence filter cube (I3). Flow cytometry experiments were carried out on an EC800 flow cytometry 158analyzer (Sony, Tokyo, Japan). Cells containing C-FLU-C were excited at 488 nm, and 159fluorescence signals from individual cells were collected through a band-pass filter 160(500–550 nm). Flow cytometric histograms were constructed for 1×10^4 cells in each 161 162experiment.



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Fig. S6 (a) Composite image of fluorescence and bright field images of a Caco-2 cell
cultured for 4 days with C-FLU-C obtained by using a laser-scanning confocal
microscope. Arrows indicate fluorescent particles. Microscopic images of a Caco-2 cell
cultured for 7 days with C-FLU-C for (b, c) a light field image, and (d) a fluorescence
microscopic image. Bar = 20 μm.

Adsorption of C-FLU-C on nucleosomes: Cell cultures were treated with colcemid (Sigma; final concentration = 0.04 mg/mL) for 2 h. After colcemid treatment, the cell cultures were trypsinized and suspended in culture medium. The cell suspensions were then collected by centrifugation ($100 \times g$, 5 min), washed several times with PBS, and

then, as a hypotonic treatment, stirred in 0.075 mol/L KCl solution containing 2×10^{-6} mol/L C-FLU-C for 25 min. Sediments were collected by centrifugation (150 ×*g*, 10 min, at 4 °C), fixed in methanol–acetic acid (3/1 = v/v, on ice), and dried on a slide glass. Fluorescence microscopic images of the treated nucleosomes were obtained by using the method described above.

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187 Fig. S7 Light (left) and fluorescence (right) microscopic images of chromosomes of

188 Caco-2 cells (a, b) and BALB/3T3 cells (c, d) cultured for 2 days in the presence of

¹⁸⁹ **C-FLU-C**. Prior to observation, cells were treated with colcemid.