# **Supporting Information**

# Efficient optical resolution of water-soluble self-assembled tetrahedral $M_4L_6$ cages with 1,1'-bi-2-naphthol

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#### 1. General

The NMR spectra were measured with a Bruker Unity 500 MHz spectrometer using TMS as an internal standard for <sup>1</sup>H NMR and a Bruker Unity 850 MHz spectrometer using 2-methyl-2-propanol as an internal standard for <sup>13</sup>C NMR. Circular dichroism (CD) spectra were measured with a Jasco J-810 spectropolarimeter. The slit band width of CD spectra measurements was set at 2 nm. Mass spectral data were acquired using a Bruker Fourier transform ICR spectrometer. Fluorescence spectra were recorded on a F-7000 fluorescence spectrophotometer at an excitation wavelength of 280 nm with both of excitation and emission monochromators set at 2.5 nm. Elemental analyses for C, H, and N were performed on an Elementar Vario EL III elemental analyzer. Racemic  $(Me_4N)_4[Fe_4L_6]$  and  $(Me_4N)_4[Ni_4L_6]$  were synthesized according to literature.<sup>[S1,S2]</sup> The  $[Ru(phen)_3](PF_6)_2$  was synthesized and resolved according to published procedures.<sup>[S3,S4]</sup> Solvents and commercially available reagents were used without further purification.

#### 2. Resolution of cage T with Leu<sup>+</sup> and Bcic<sup>+</sup>.

*Rac*-(Me<sub>4</sub>N)<sub>4</sub>[Fe<sub>4</sub>L<sub>6</sub>] 36 mg and Leu<sup>+</sup> 12 mg (Bcic<sup>+</sup> 17 mg) were added to a 50 mL flask containing methanol 15 mL. The reaction was stirred for 1 h at 50 °C. The less soluble salt ( $\Lambda\Lambda\Lambda\Lambda$ -T) was collected by centrifugation, washed with a small portion of methanol, and dried under vacuum. The remaining methanolic solution was concentrated on a rotational evaporator and the more soluble salt was obtained. The Leu<sup>+</sup> (Bcic<sup>+</sup>) ions were exchanged against Me<sub>4</sub>N<sup>+</sup> on a cation exchange column. (Me<sub>4</sub>N)<sub>4</sub>[Fe<sub>4</sub>L<sub>6</sub>] with enantiomer excess were isolated by slow vapor diffusion of acetone into their aqueous solutions.

#### **3.** Resolution of cage T with [Ru(phen)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub>

*Rac*-(Me<sub>4</sub>N)<sub>4</sub>[Fe<sub>4</sub>L<sub>6</sub>] 36 mg and [Ru(phen)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub> 9 mg were added to a 50 mL flask containing methanol 15 mL. The reaction was stirred for 1 h at 50 °C. The less soluble salt ( $\Lambda\Lambda\Lambda\Lambda$ -T) was collected by centrifugation, washed with a small portion of methanol, and dried under vacuum. The remaining methanolic solution was concentrated on a rotational evaporator and the more soluble salt was obtained. The [Ru(phen)<sub>3</sub>]<sup>2+</sup> ions were exchanged against Me<sub>4</sub>N<sup>+</sup> on a cation exchange column. (Me<sub>4</sub>N)<sub>4</sub>[Fe<sub>4</sub>L<sub>6</sub>] with enantiomer excess were isolated by slow vapor diffusion of acetone into their aqueous solutions.

#### 4. Resolution of cage T with BINOL.

*Rac*-(Me<sub>4</sub>N)<sub>4</sub>[Fe<sub>4</sub>L<sub>6</sub>] 108 mg and (*S*)-BINOL (or (*R*)-BINOL) 36 mg were added to a 25 mL flask containing 6 mL water/methanol (v/v = 1:1). The reaction was stirred for 1 h at 50 °C. The less soluble diastereoisomer was collected by centrifugation and washed with water ( $3 \times 2$ mL) and then with acetone, after that dried in vacuum, and last dissolved in 4 mL water/methanol (v/v = 1:1).  $\Delta\Delta\Delta\Delta$ -T was obtained by slow vapor diffusion of acetone into the water/methanol solution

for several days. Yield: 35%. AAAA-T was obtained from slow vapor diffusion of acetone into the filtrate. Yield: 45%. <sup>1</sup>H NMR (500 MHz, 298 K, D<sub>2</sub>O):  $\delta$  = 9.29 (s, 12H, imine), 8.66 (d, 12H, 3pyridine), 8.34 (t, 12H, 4-pyridine), 7.72 (t, 12H, 5-pyridine), 7.50 (d, 12H, 6-pyridine), 7.09 (d, 12H, 6,6'-benzidine), 6.38 (s, 12H, 3,3'-benzidine), 5.78 (d, 12H, 5,5'-benzidine), 3.30 ppm (s, [NMe<sub>4</sub>]<sup>+</sup>). <sup>13</sup>C NMR (212.5 MHz, 298 K, D<sub>2</sub>O, referenced to 2-methyl-2-prop-anol at 29.5 ppm as internal standard):  $\delta = 176.0, 157.9, 155.7, 150.0, 143.1, 139.7, 135.9, 132.0, 131.9, 129.8, 121.7,$ 120.8. HR-ESI-MS: m/z: 1123.0083 ( $[Fe_4L_6]^{4-} + Na^+$ ), 548.0199 ( $[FeL_2]^{2-}$ ), 836.0117 ( $[Fe_2L_3]^{2-}$ ), 1412.9976  $([Fe_4L_5]^{2-}).$ Elemental 1124.5048  $([Fe_3L_4]^{2-}),$ analysis calcd for C<sub>160</sub>H<sub>144</sub>N<sub>28</sub>Fe<sub>4</sub>O<sub>36</sub>S<sub>12</sub>·Me<sub>2</sub>CO·20(H<sub>2</sub>O): C 48.20, H 4.72, N 9.66; found C 48.31, H 4.80, N 9.50.

#### 5. Resolution of cage $[Ni_4L_6]^4$ with BINOL.

*Rac*-(Me<sub>4</sub>N)<sub>4</sub>[Ni<sub>4</sub>L<sub>6</sub>] 108 mg and (*S*)-BINOL (or (*R*)-BINOL) 36 mg were added to a 25 mL flask containing 6 mL water/methanol (v/v = 1:1). The reaction was stirred for 1 h at 50°C. The less soluble diastereoisomer was collected by centrifugation and washed with water (3 × 2 mL) and then with acetone, after that dried in vacuum, and last dissolved in 4 mL water/methanol (v/v = 1:1).  $\Delta\Delta\Delta\Delta$ -(Me<sub>4</sub>N)<sub>4</sub>[Ni<sub>4</sub>L<sub>6</sub>] was obtained by slow vapor diffusion of acetone into the water/methanol solution for several days. Yield: 30%.  $\Lambda\Lambda\Lambda\Lambda$ -(Me<sub>4</sub>N)<sub>4</sub>[Ni<sub>4</sub>L<sub>6</sub>] was obtained from slow vapor diffusion of acetone into the filtrate. Yield: 40%. HR-ESI-MS: m/z: 1143.7192 ([Ni<sub>4</sub>L<sub>6</sub>]<sup>4-</sup>+Me<sub>4</sub>N<sup>+</sup>), 839.0130 ([Ni<sub>2</sub>L<sub>3</sub>]<sup>2-</sup>), 1128.0081 ([Ni<sub>3</sub>L<sub>4</sub>]<sup>2-</sup>), 1418.0025 ([Ni<sub>4</sub>L<sub>5</sub>]<sup>2-</sup>).

#### 6. The chiral stability of resolved cages upon complexation of cyclohexane.

Aqueous solution of  $\Delta\Delta\Delta\Delta$ -T (2.5 mL,  $1.4 \times 10^{-5}$  M<sup>-1</sup>) and excess cyclohexane (0.1 mL, 1.52 mmol) were added to a cuvette. The sealed cuvette was heated in a water bath to 323 K for 6 h, and reaction progress was monitored by CD spectra.

7. *Figure S1*. CD spectra of cage T resolved by Leu<sup>+</sup> and Bcic<sup>+</sup>.



### 8. *Figure S2.* <sup>1</sup>H NMR spectrum of the resolved cage **T**.



9. *Figure S3*. <sup>13</sup>C NMR spectrum of the resolved cage T.







11. Figure S5. HR-ESI-MS of the resolved cage (Me<sub>4</sub>N)<sub>4</sub>[Ni<sub>4</sub>L<sub>6</sub>].



**12.** *Figure S6.* CD spectra of the resolved tetranuclear cluster  $[Ni_4L_6]^{4-}$ :  $\Delta\Delta\Delta\Delta$  (a) and  $\Lambda\Lambda\Lambda\Lambda$  (b) forms.



**13.** *Figure S7.* Dynamic CD spectra of  $\Delta\Delta\Delta\Delta$ -T and  $\Lambda\Lambda\Lambda\Lambda$ -T taken immediately after preparation of the 0.01 mM solution and after 30 days.



**14.** *Figure S8.* CD spectra of  $\Delta\Delta\Delta\Delta$ -T varied with time at 323 K: a) in the presence of cyclohexane (C<sub>6</sub>H<sub>12</sub>); b) in the absence of C<sub>6</sub>H<sub>12</sub>.



**15.** *Figure S9*. The <sup>1</sup>H NMR spectrum of  $C_6H_{12} \subset \Delta\Delta\Delta\Delta$ -T.



**16.** *Figure S10*. Fluorescence responses of (a) (*R*)-BINOL ( $2 \times 10^{-5}$  M) and (b) (*S*)-BINOL ( $2 \times 10^{-5}$  M) upon addition of  $\Lambda\Lambda\Lambda\Lambda$ -T at  $1.25 \times 10^{-6}$  M intervals; (c) the Benesi-Hildebrand plots.



17. Scheme S1. Optical resolution of T using (S)-BINOL in methanol.



**18.** *Scheme S2*. Optical resolution of **T** using (*R*)-BINOL in 1:1 water-methanol solution followed by removal of (*R*)-BINOL during crystallization.



#### 19. Reference

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