

Rapid, selective, sensitive fluorometric detection of cyanide anion in aqueous media by cyanine dyes with indolium–coumarin linkage

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Electronic Supplementary Information (ESI†)

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Experimental methods

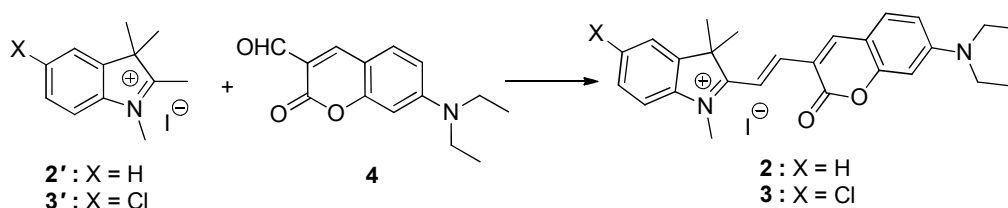
General

Absorption spectra were measured on an UV-visible photodiode-array spectrophotometer (Shimadzu; Multispec-1500) equipped with a temperature controller (S-1700). Fluorescence spectra were measured on a JASCO FP-6500 fluorescence spectrophotometer. Light irradiations were carried out with Xe lamp (300 W; Asahi Spectra Co. Ltd.; MAX-302) equipped with 334 nm (light intensity: 69.1 mW m⁻²) band-pass filter. ¹H and ¹³C NMR spectra were obtained by a JEOL JNM-ECS400 spectrometer. FAB-MS analysis was performed by a JEOL JMS 700 Mass Spectrometer. The fluorescence quantum yield (Φ_F) was determined according to literature method using Rhodamine B (in ethanol) as a reference.^[1]

Calculation details

Calculations were carried out with tight convergence criteria at the DFT level with the Gaussian 03 package,^[2] using the B3LYP/6-31G* basis set for all atoms. The excitation energies and the oscillator strength of each structure were calculated by time-dependent density-functional theory (TD-DFT)^[3] at the same level of optimization using the polarizable continuum model (PCM)^[4] with water as a solvent. Cartesian coordinates for the merocyanine form of **2** and **3** are summarized in the end of this Supporting Information.

Synthesis



Compound 2 **2'** (1,2,3,3-tetramethyl-3*H*-indolium iodide) and **4** (7-diethylaminocoumarin-3-aldehyde) were prepared according to literature procedure,^[5,6] **2'** (123 mg, 0.41 mmol) and **4** (143 mg, 0.58 mmol) were stirred in EtOH (20 mL) for 20 h at 60 °C. The resultant was concentrated by evaporation, and purified by recrystallization with MeOH (3 mL), affording **2** as a green solid (101.6 mg, 47 %). ¹H-NMR (400 MHz, DMSO-d₆, TMS) δ (ppm): 8.81 (1H, s), 8.24 (1H, d, *J* = 15.6 Hz), 7.83 (3H, dq, *J* = 9.3, 2.1 Hz), 7.58 (3H, dq, *J* = 9.3, 2.5 Hz), 6.92 (1H, dd, *J* = 9.2, 2.3 Hz), 6.72 (1H, d, *J* = 2.3 Hz), 3.97 (3H, s), 3.57 (4H, q, *J* = 7.0 Hz), 1.76 (6H, s), 1.19 (6H, t, *J* = 7.1 Hz). ¹³C NMR (100 MHz, DMSO-d₆, TMS) δ (ppm): 180.7, 159.2, 157.4, 153.8, 150.0, 149.1, 142.9, 141.8, 132.1, 128.7, 128.4, 122.6, 114.3, 112.1, 111.2, 110.0, 109.3, 96.4, 51.2, 44.7, 33.6, 25.8, 12.3. FAB-MS: *m/z*: calcd for C₂₆H₂₉N₂O₂⁺ ([**2** - I]⁺) 401.2; found: 401.2; HRMS (FAB⁺): *m/z*: calcd for C₂₆H₂₉N₂O₂⁺ ([**2** - I]⁺): 401.2224; found: 401.2227.

Compound 3 **3'** (5-chloro-1,2,3,3-tetramethyl-3*H*-indolium iodide) was prepared according to literature procedure.^[5] **3'** (130 mg, 0.39 mmol) and **4** (77 mg, 0.31 mmol) were stirred in EtOH (10 mL) for 3 h at 80 °C. The resultant was concentrated by evaporation, and purified by recrystallization with a mixture of EtOH and ethyl acetate (1/1, 6 mL), affording **3** as a green solid (68.7 mg, 39 %). ¹H-NMR (400 MHz, DMSO-d₆, TMS) δ (ppm): 8.81 (1H, s), 8.26 (1H, d, *J* = 16.0 Hz), 8.03 (1H, d, *J* = 1.8 Hz), 7.82 (2H, dd, *J* = 22.2, 12.1 Hz), 7.67 (1H, dd, *J* = 8.7, 2.3 Hz), 7.59 (1H, d, *J* = 9.2 Hz), 6.93 (1H, dd, *J* = 9.2, 2.3 Hz), 6.72 (1H, d, *J* =

1.8 Hz), 3.94 (3H, s), 3.58 (4H, q, $J = 6.9$ Hz), 1.77 (6H, s), 1.19 (6H, t, $J = 7.1$ Hz). ^{13}C NMR (100 MHz, DMSO- d_6 , TMS) δ (ppm): 180.8, 159.2, 157.6, 154.0, 150.4, 149.8, 144.8, 140.8, 133.0, 132.3, 128.8, 123.2, 115.9, 112.1, 111.3, 109.8, 109.4, 96.5, 51.4, 44.8, 33.7, 25.6, 12.4. FAB-MS: m/z : calcd for $\text{C}_{26}\text{H}_{28}\text{N}_2\text{O}_2\text{Cl}^+$ ($[\mathbf{3} - \text{I}]^+$); 435.2; found: 435.2; HRMS (FAB $^+$): m/z : calcd for $\text{C}_{26}\text{H}_{28}\text{N}_2\text{O}_2\text{Cl}^+$ ($[\mathbf{3} - \text{I}]^+$); 435.1834; found: 435.1836.

- [1] a) B. Bag, P. K. Bharadwaj, *J. Phys. Chem. B*, 2005, **109**, 4377–4390. b) A. Proutiere, E. Megnassan, H. Hucteau, *J. Phys. Chem.* 1992, **96**, 3485–3489.
- [2] a) M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, Jr., J. A. Montgomery, T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, J. A. Pople, *Gaussian 03 (Revision B.05)*, Gaussian, Inc., Wallingford CT, 2004. b) R. Dennington II, T. Keith, J. Millam, K. Eppinnett, W. L. Hovell, R. Gilliland, *GaussView, (Version 3.09)*, Semichem, Inc., Shawnee Mission, KS, 2003.
- [3] R. E. Stratmann, G. E. Scuseria, M. J. Frisch, *J. Chem. Phys.* 1998, **109**, 8218–8224.
- [4] M. Cossi, V. Barone, R. Cammi, J. Tomasi, *Chem. Phys. Lett.* 1996, **255**, 327–335.
- [5] Q. Li, J. Tan, B. Peng, *Molecules* 1997, **2**, 91–98.
- [6] W. Xuan, C. Chen, Y. Cao, W. He, W. Jiang, K. Liu, W. Wang, *Chem. Commun.*, 2012, **48**, 7292–7294.

Table S1 Calculated excitation energy (E), wavelength (λ), and oscillator strength (f) for low-lying singlet state (S_n) of **3** and **3**– CN^- species.

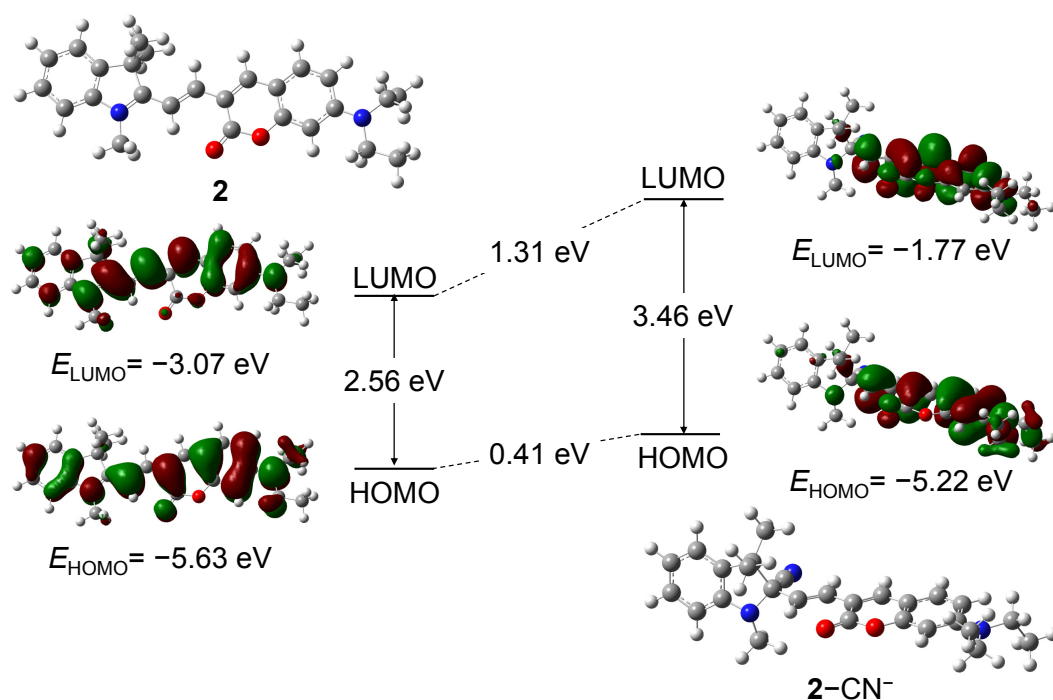
species		Main orbital transition (CIC ^a)	E (eV) [λ (nm)]	f
3	$S_0 \rightarrow S_1$	HOMO $\rightarrow$ LUMO (0.62264)	2.3706 [523.01]	1.8812
	$S_0 \rightarrow S_1$	HOMO $\rightarrow$ LUMO (0.63542)	3.1818 [389.67]	1.1388
3 – CN^-		HOMO–1 $\rightarrow$ LUMO+2 (–0.10646)		
	$S_0 \rightarrow S_4$	HOMO $\rightarrow$ LUMO+1 (0.55446)	4.3376 [285.84]	0.2380
		HOMO–1 $\rightarrow$ LUMO+1 (0.14735)		
		HOMO–2 $\rightarrow$ LUMO (0.30346)		

^a CI expansion coefficients for the main orbital transitions.

Table S2 Calculated excitation energy (E), wavelength (λ), and oscillator strength (f) for low-lying singlet state (S_n) of **2** and **2**-CN⁻ species.

species		Main orbital transition (CIC ^a)	E (eV) [λ (nm)]	f
2	$S_0 \rightarrow S_1$	HOMO $\rightarrow$ LUMO (0.62391)	2.4001 [516.58]	1.800
	$S_0 \rightarrow S_1$	HOMO-1 $\rightarrow$ LUMO (0.20711) HOMO $\rightarrow$ LUMO (0.61624)	3.1746 [390.56]	0.9701
2 -CN ⁻	$S_0 \rightarrow S_2$	HOMO-1 $\rightarrow$ LUMO (0.67061) HOMO $\rightarrow$ LUMO (-0.17037)	3.2855 [377.36]	0.1752
		HOMO-3 $\rightarrow$ LUMO (0.11621)		
	$S_0 \rightarrow S_4$	HOMO-2 $\rightarrow$ LUMO (0.33682) HOMO $\rightarrow$ LUMO+1 (0.5512)	4.3667 [283.93]	0.2789

^a CI expansion coefficients for the main orbital transitions.



Calculated energy diagrams for **2** and **2**-CN⁻

DFILE CIM Proton-1-1.jdf
 COMNT single_pulse
 DATIM 2013-11-29 21:11:36
 OBNUC 1H
 EXMOD proton.jxp
 OBFRQ 399.78 MHz
 OBSET 4.19 KHz
 OBFIN 7.29 Hz
 POINT 16384
 FREQU 7503.00 Hz
 SCANS 8
 ACQTM 2.1837 sec
 PD 15.0000 sec
 PW1 5.00 usec
 IRNUC 1H
 CTEMP 30.0 C
 SLVNT DMSO
 EXREF 0.00 ppm
 BF 0.20 Hz
 RGAIN 44

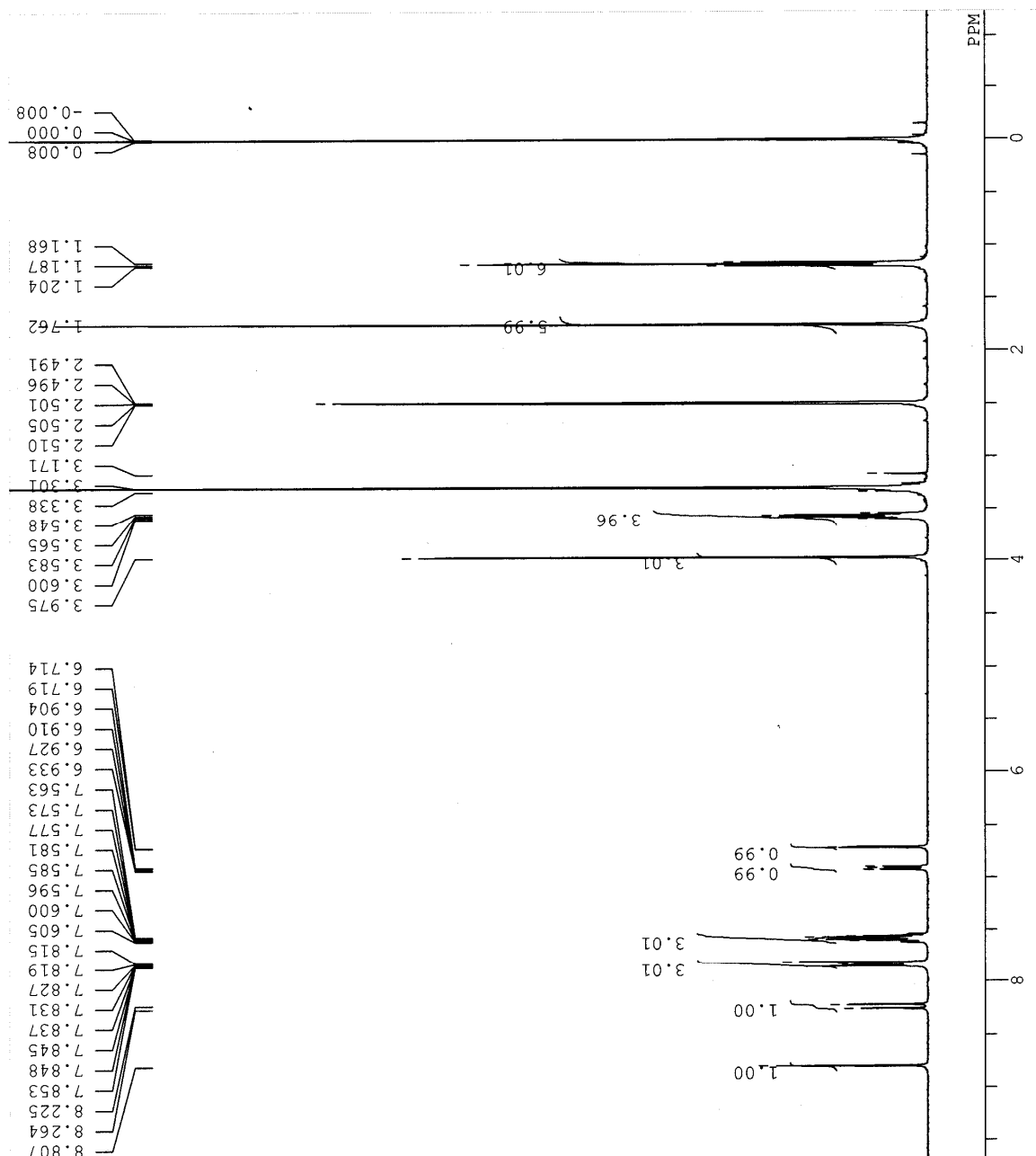
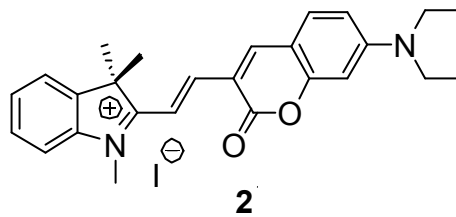


Fig. S1 ^1H NMR chart of **2** (DMSO- d_6 , 400 MHz).

DFILE CIM Carbon-1-1.jdf
 COMNT single pulse decoupled gate
 DATIM 2013-11-30 17:34:15
 OBNUC 13C
 EXMOD carbon.jxp
 OBFRQ 100.53 MHz
 OBSET 5.35 KHz
 OBFIN 5.86 Hz
 POINT 32767
 FREQU 31407.04 Hz
 SCANS 1024
 ACQTM 1.0433 sec
 PD 2.0000 sec
 PW1 2.87 usec
 IRNUC 1H
 CTEMP 30.0 c
 SLVNT DMSO
 EXREF 0.00 ppm
 BF 0.10 Hz
 RGAIN 50

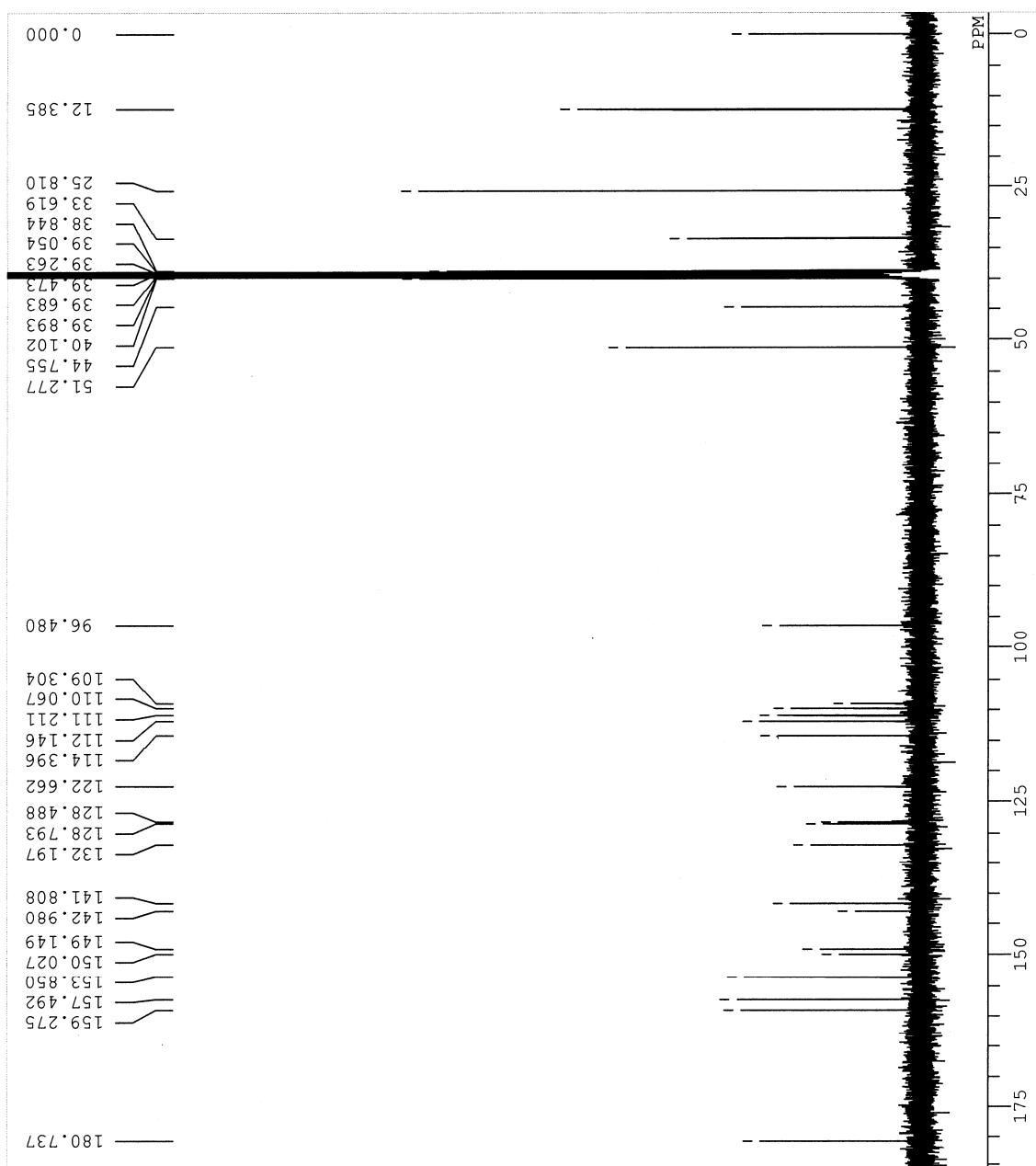
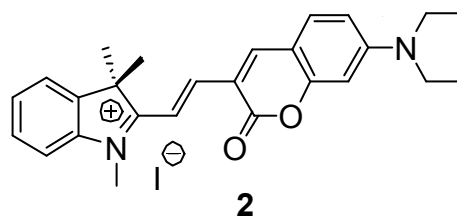


Fig. S2 ^{13}C NMR chart of **2** (DMSO- d_6 , 100 MHz).

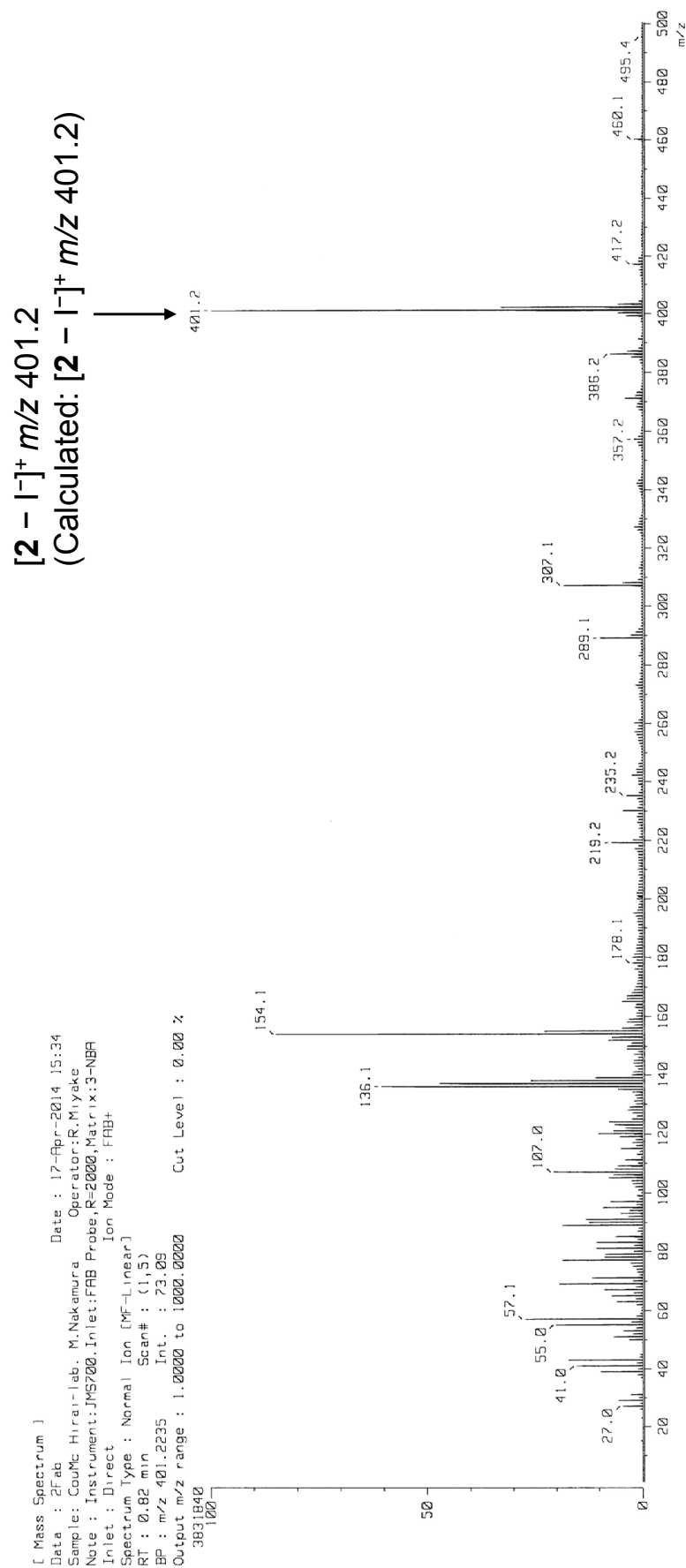


Fig. S3 FAB-MS chart of 2.

DFILE CouMC-Cl_Proton-1-1.jdf
 COMNT single_pulse
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 proton.jxp
 EXMOD 399.78 MHz
 OBFQ 4.19 KHz
 OBSE 7.29 Hz
 OBFN 16384
 POINT 7503.00 Hz
 FREQU 8
 SCANS 2.1837 sec
 ACQTM 10.0000 sec
 PD 5.00 usec
 PW1 1H
 IRNUC 30.0 C
 CTMP DMSO
 SLVT 0.00 ppm
 EXREF 1.20 Hz
 BF 44
 RGAIN

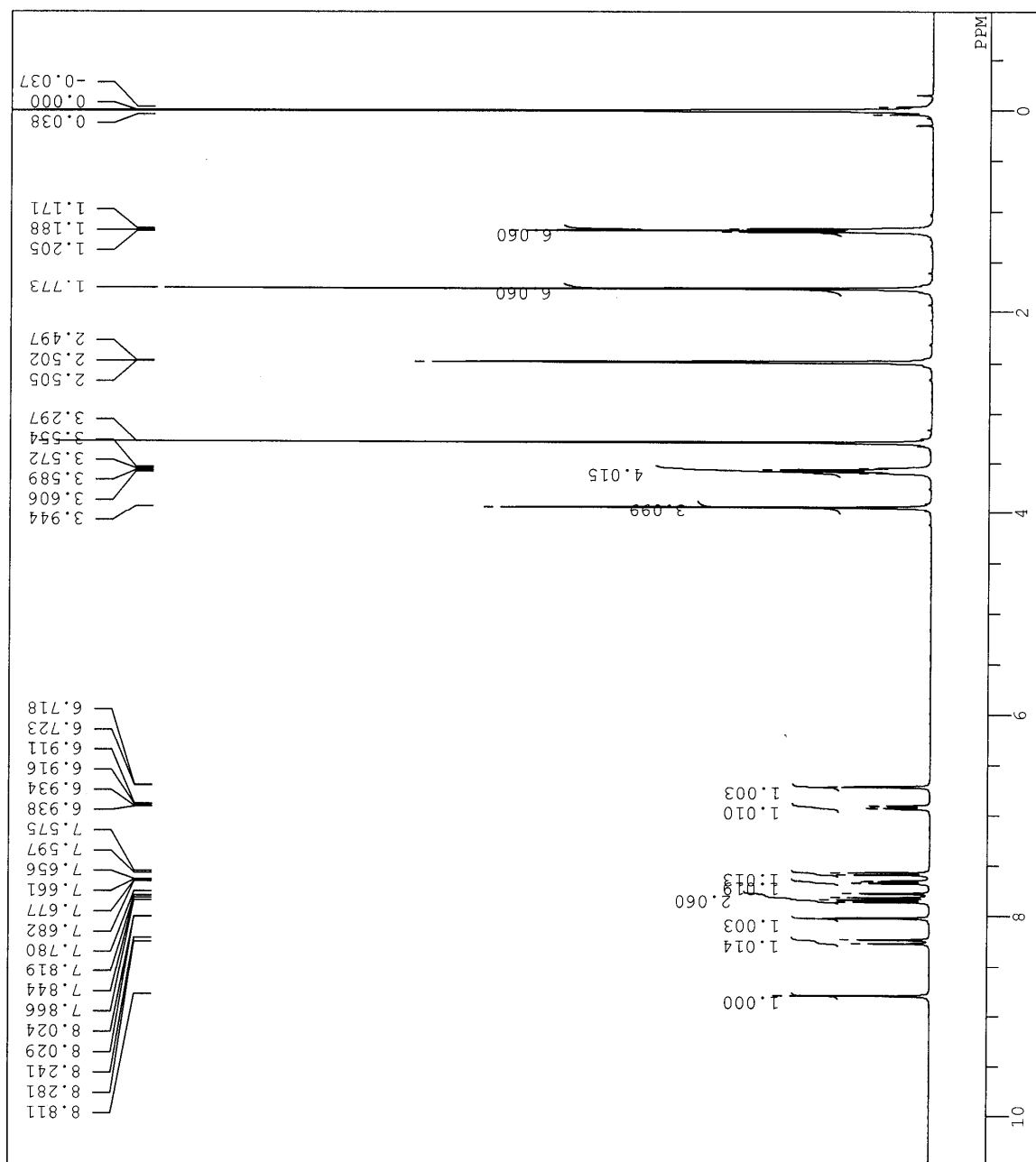
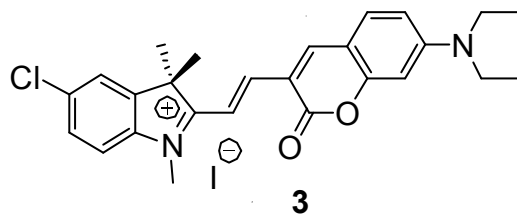


Fig. S4 ^1H NMR chart of **3** (DMSO- d_6 , 400 MHz).

DFILE CouNC-Cl_Carbon-1-1.jdf
 COMNT single pulse decoupled gate
 DATIM 2014-04-23 18:49:56
 OBNUC 13C
 EXMOD carbon.jxp
 OBFRQ 100.53 MHz
 OBSET 5.35 KHz
 OBFIN 5.86 Hz
 POINT 32767
 FREQU 31407.04 Hz
 SCANS 1024
 ACQTM 1.0433 sec
 PD 2.0000 sec
 PW1 2.87 usec
 IRNUC 1H
 CTEMP 30.0 c
 SLVNT DMSO
 EXREF 39.50 ppm
 BF 0.20 Hz
 RGAIN 50

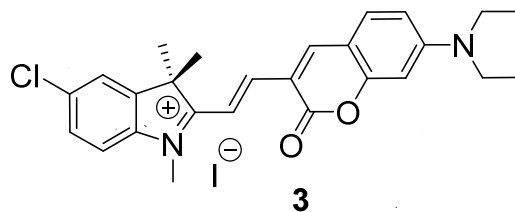
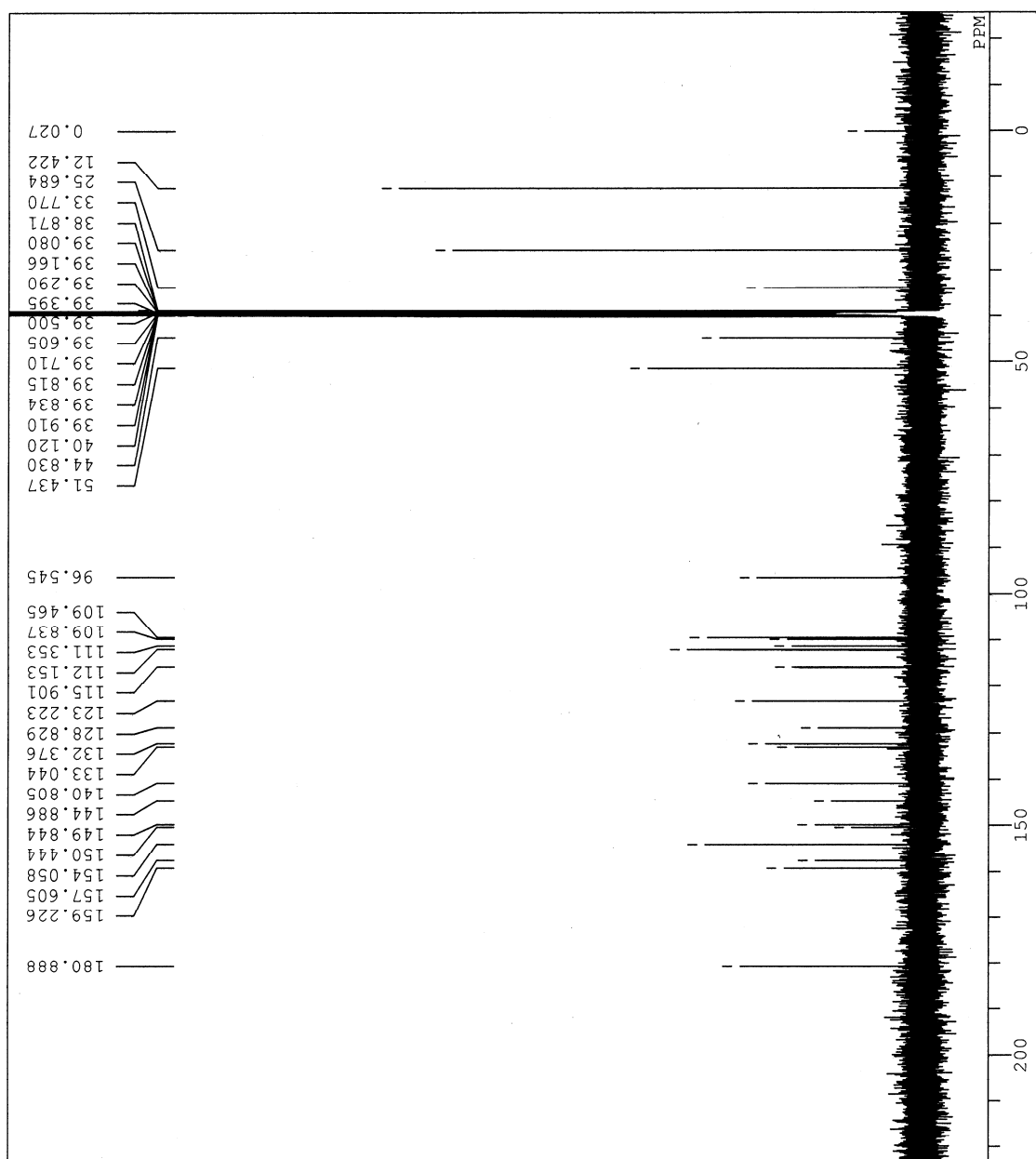


Fig. S5 ^{13}C NMR chart of **3** (DMSO- d_6 , 100 MHz).

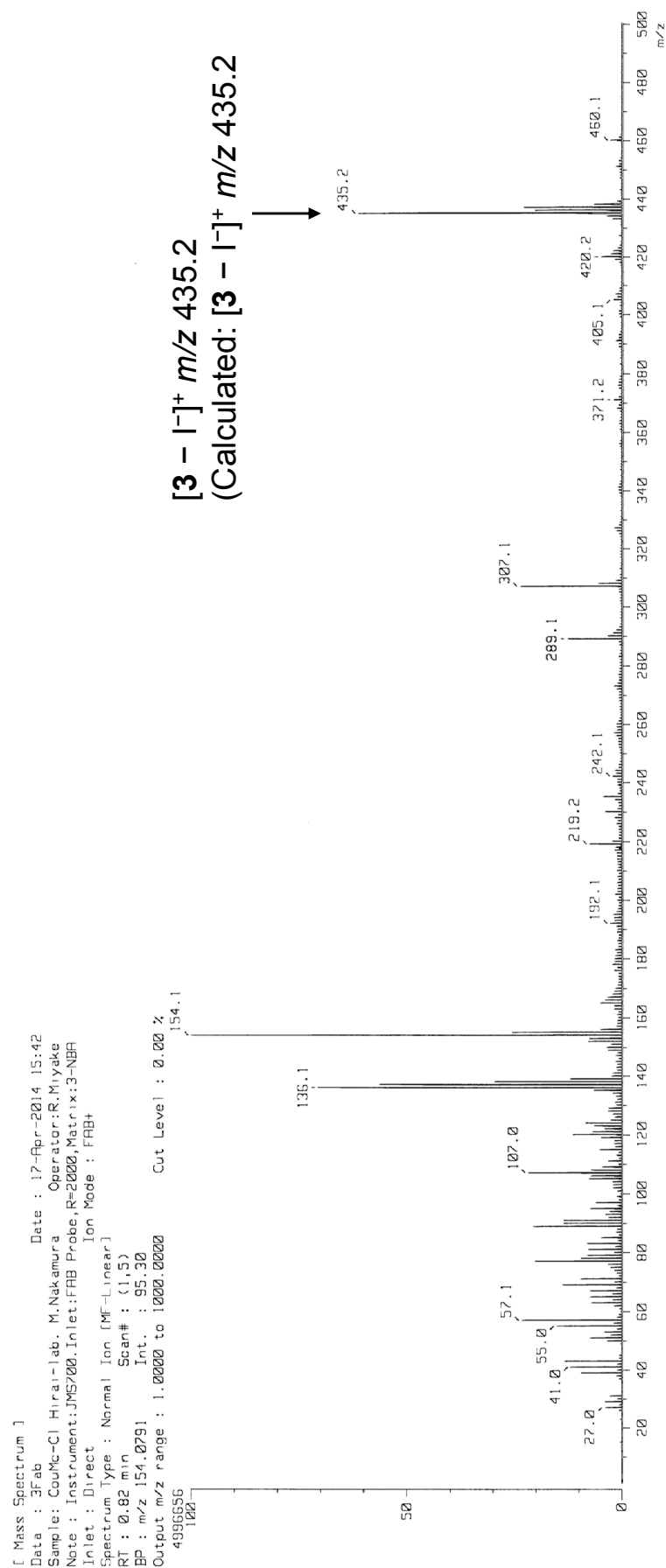


Fig. S6 FAB-MS chart of **3**.

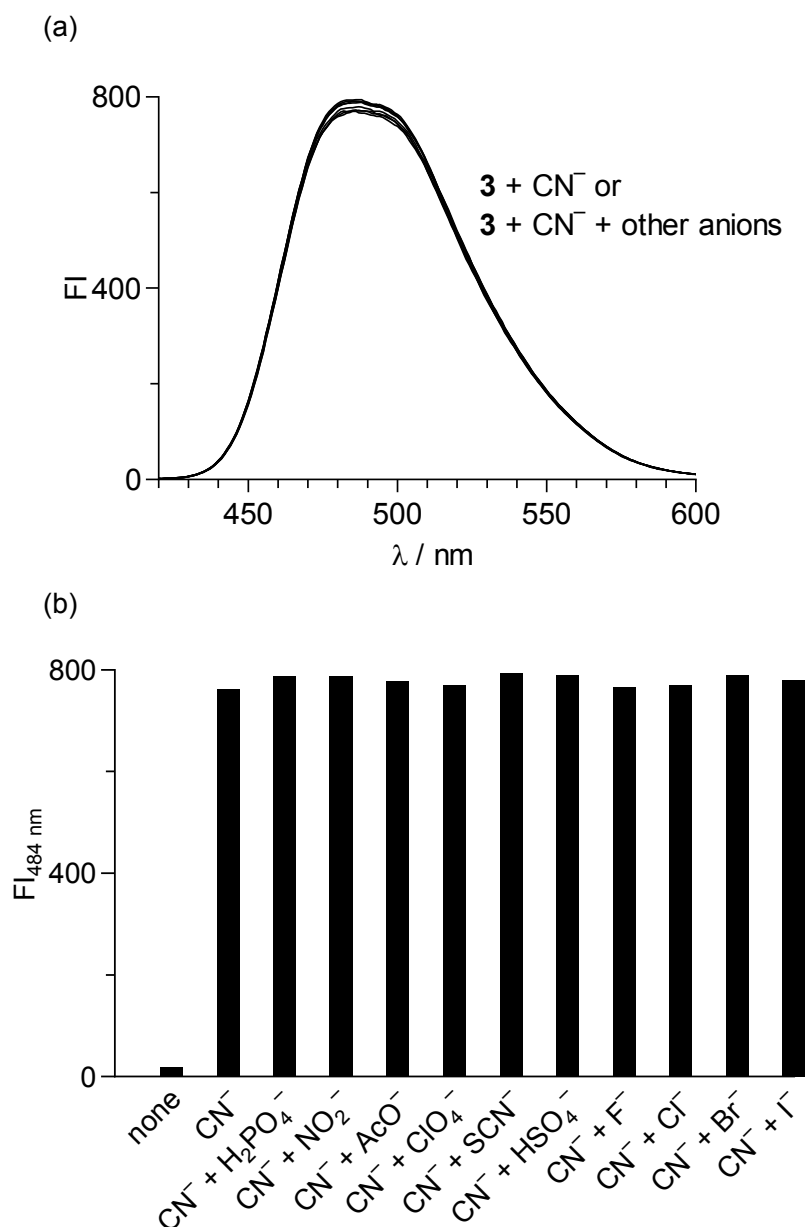


Fig. S7 (a) Fluorescence spectra ($\lambda_{\text{ex}} = 415 \text{ nm}$) of **3** ($5 \mu\text{M}$) measured with 50 equiv of CN^- together with 50 equiv of other respective anions in a buffered water/MeCN mixture (7/3 v/v; CHES 100 mM, pH 9.0) at 25°C . (b) Fluorescence intensity of the respective solutions.

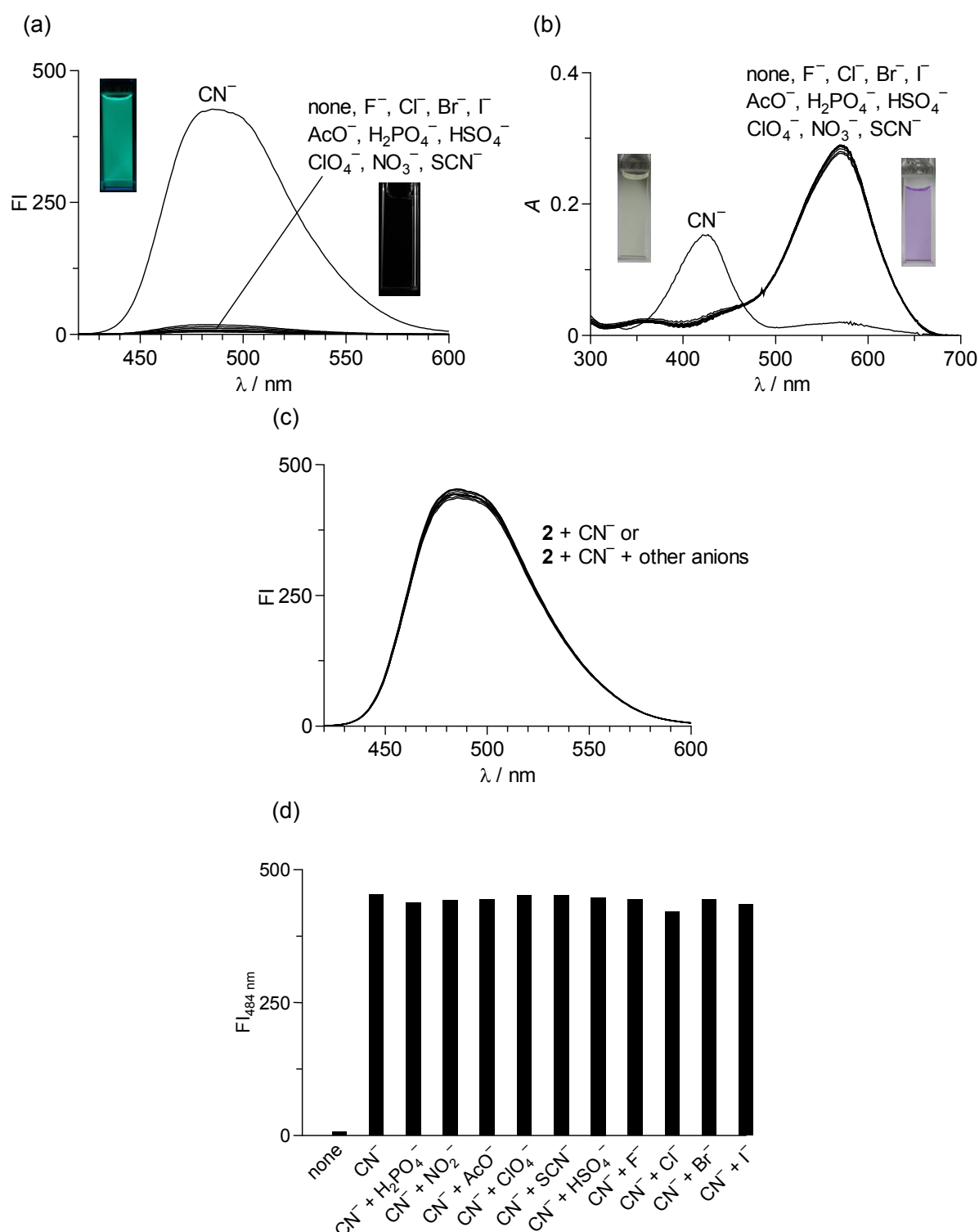


Fig. S8 (a) Fluorescence spectra ($\lambda_{\text{ex}} = 415 \text{ nm}$) of **2** (5 μM), measured with 20 equiv of each respective anion in a buffered water/MeCN mixture (7/3 v/v; CHES 100 mM, pH 9.0) at 25 $^\circ\text{C}$. (b) Change in absorption spectra. (c) Fluorescence spectra of **2** measured with 50 equiv of CN^- together with 50 equiv of other respective anions. (d) Fluorescence intensity of the respective solutions.

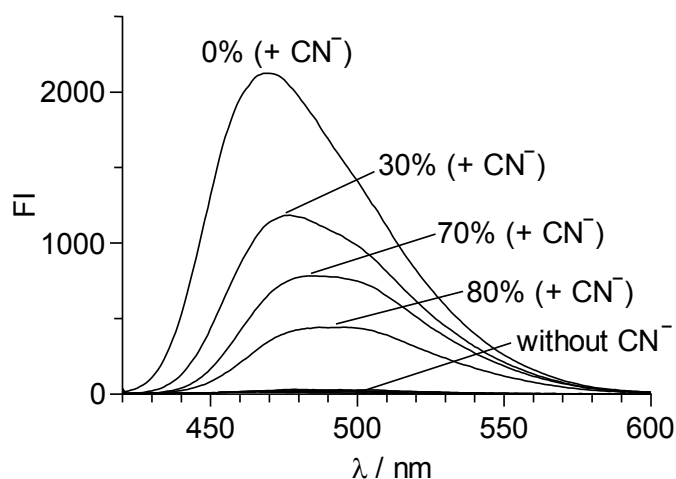
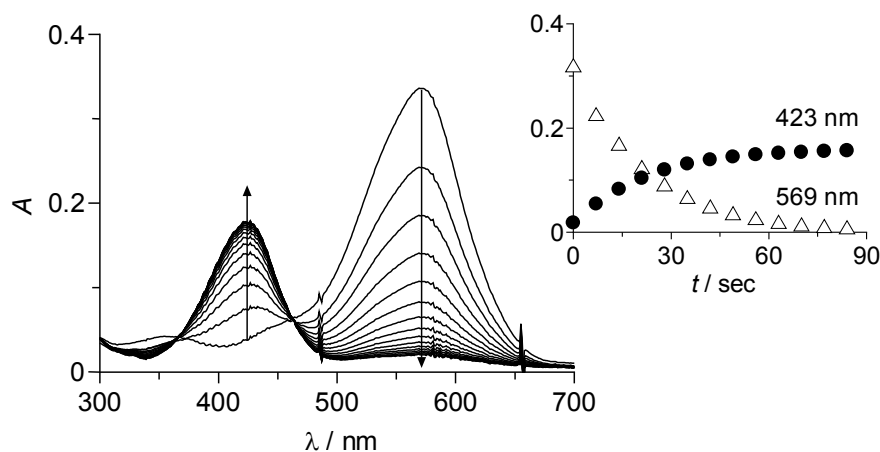


Fig. S9 Fluorescence spectra of **3** (5 μM) measured without or with CN^- (50 equiv) in MeCN solutions (CHES 100 mM, pH 9.0) with different water content (0, 30, 70, and 80%).

(a)



(b)

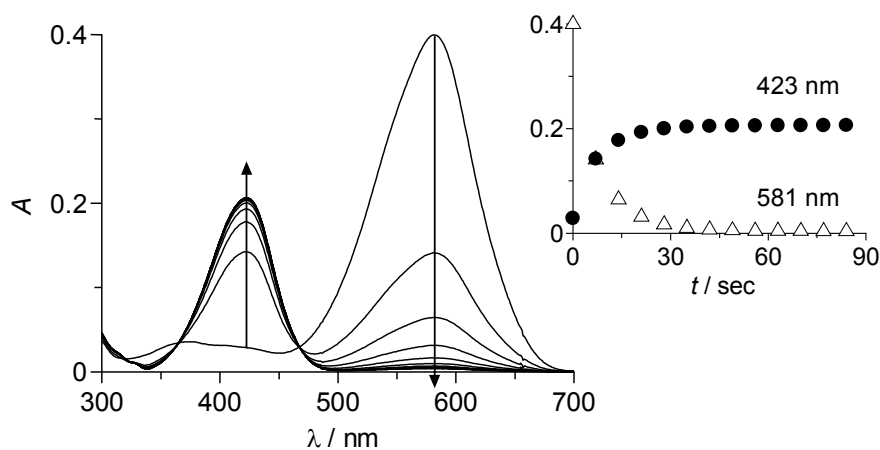


Fig. S10 Time-dependent change in absorption spectra of (a) **2** and (b) **3** (5 μM), measured with 50 equiv of CN^- in a buffered water/MeCN mixture (7/3 v/v; CHES 100 mM, pH 9.0) at 25 °C. (inset) Change in absorbance for respective bands.

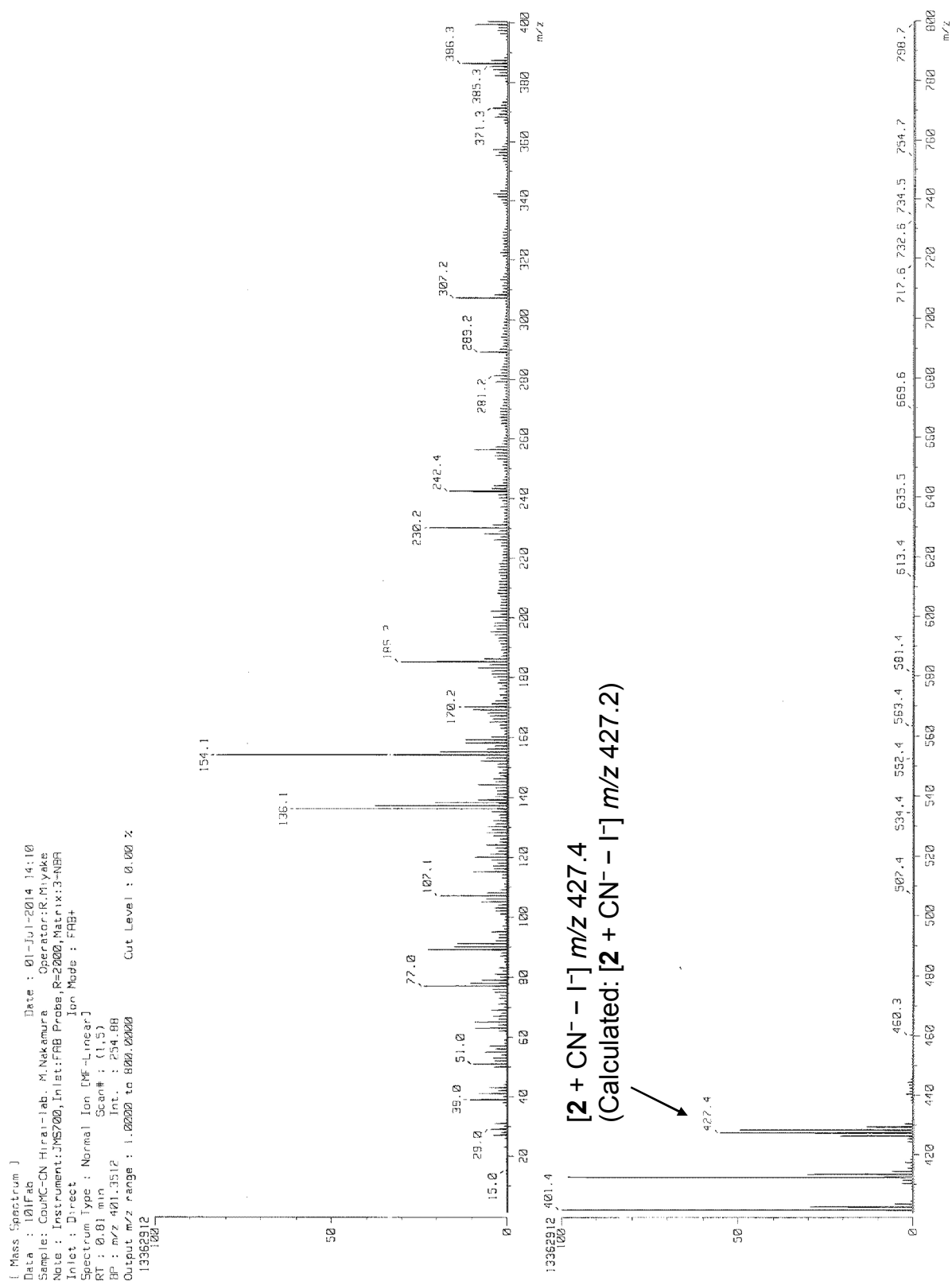


Fig. S11 FAB-MS chart of 2-CN⁻ species.

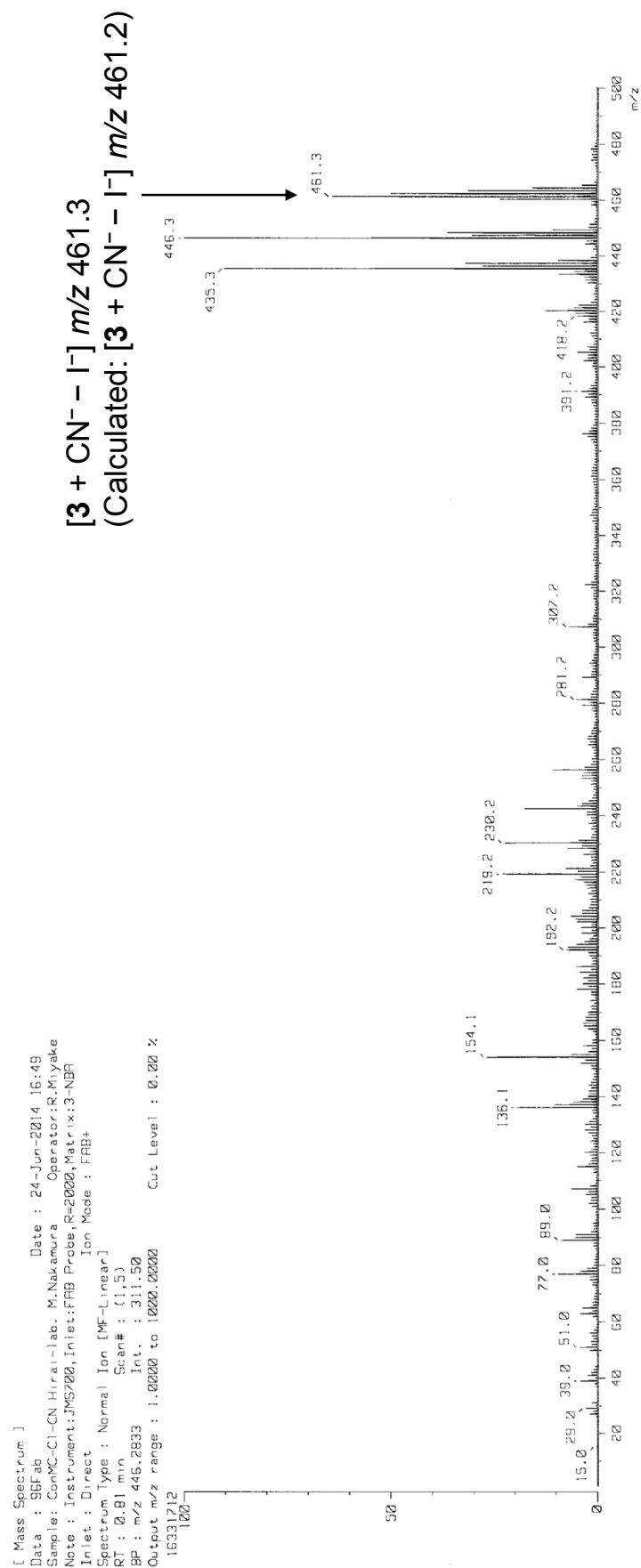


Fig. S12 FAB-MS chart of **3-CN⁻** species.

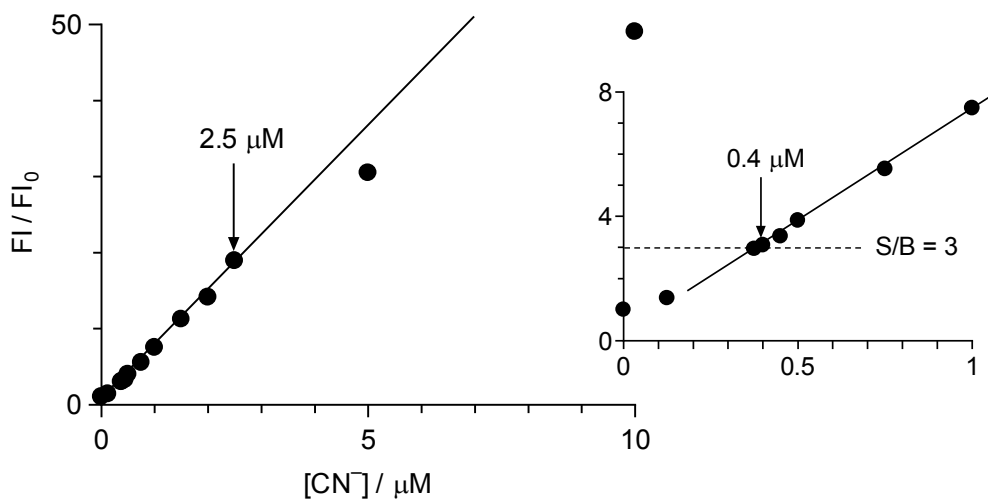


Fig. S13 Change in the ratio of fluorescence intensity ($\lambda_{ex} = 415$ nm; $\lambda_{em} = 484$ nm) of **2** ($5 \mu M$) as a function of CN^- concentration in a water/MeCN mixture (7/3 v/v; pH 9.0) at $25^\circ C$.

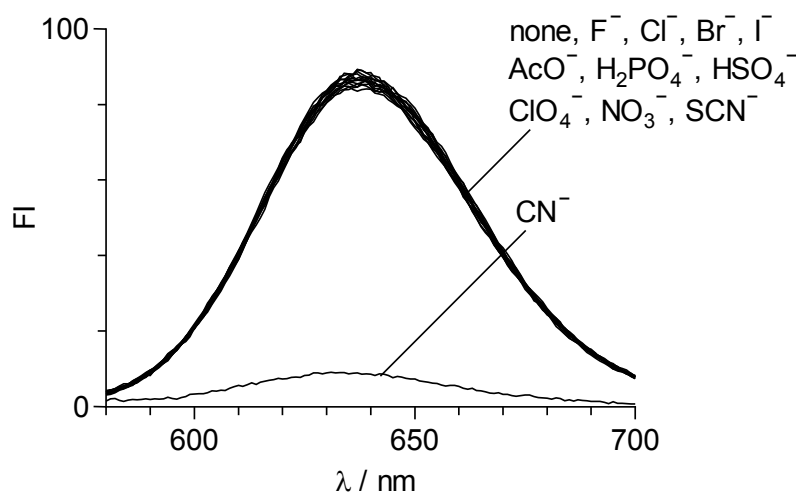
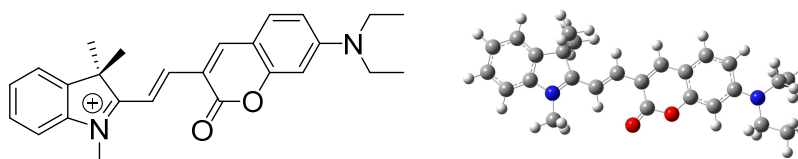


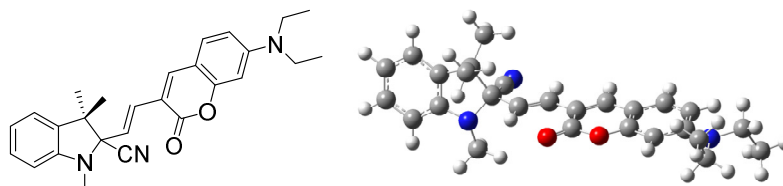
Fig. S14 Fluorescence spectra of **3** ($5 \mu M$) measured in a water/MeCN mixture (7/3 v/v; pH 9.0) by excitation at 570 nm with different anions (50 equiv).

Cartesian Coordinates (in Å) of 2



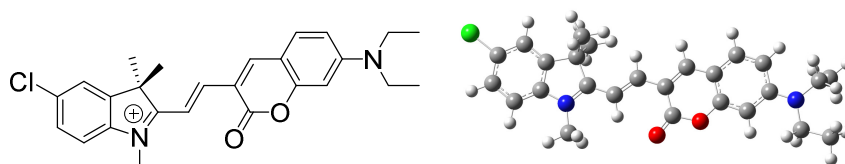
C	-8.154502	-0.848079	0.026544	H	-9.068165	-1.439185	0.005227
C	-8.233218	0.546181	0.117566	H	-9.206834	1.03034	0.166486
C	-7.079527	1.337209	0.147288	H	-7.159492	2.419427	0.217963
C	-5.858171	0.672666	0.08265	H	-6.857707	-2.578153	-0.107617
C	-5.759697	-0.717277	-0.008377	H	-3.698793	2.971211	-0.699274
C	-6.911579	-1.492348	-0.037392	H	-5.199426	3.195541	0.224411
N	-4.545823	1.213517	0.093678	H	-3.681533	2.858662	1.084592
C	-3.609933	0.253807	0.016626	H	-2.9304	-2.131163	-1.460376
C	-4.295723	-1.116085	-0.059865	H	-4.591738	-2.744151	-1.4618
C	-4.260635	2.646656	0.181208	H	-4.218343	-1.200195	-2.252059
C	-3.981705	-1.837755	-1.394472	H	-2.883349	-2.295408	1.152495
C	-3.936738	-2.002374	1.159231	H	-4.145881	-1.482497	2.099415
C	-2.235681	0.581708	0.015026	H	-4.542297	-2.913867	1.129259
C	-1.199315	-0.324029	-0.0608	H	-1.970463	1.628405	0.078538
C	0.199813	-0.038527	-0.065843	H	-1.424715	-1.384902	-0.12624
C	1.111658	-1.093166	-0.148109	H	0.731127	-2.115252	-0.204838
C	2.494592	-0.886818	-0.161139	H	3.134054	-2.953166	-0.28813
C	2.977908	0.443911	-0.087153	H	5.505979	-2.465795	-0.296162
O	2.095564	1.485637	-0.000858	H	4.591666	1.802242	-0.017296
C	0.71767	1.330411	0.014358	H	7.308434	-1.886905	-0.804248
C	3.470394	-1.91819	-0.237423	H	8.548767	-0.687507	-0.601253
C	4.810873	-1.636964	-0.24358	H	6.495293	2.003832	-0.820303
C	5.291831	-0.280727	-0.183793	H	6.748982	1.784063	0.911563
C	4.320313	0.756643	-0.091549	H	8.660185	-2.359809	1.240747
N	6.623139	0.005319	-0.218524	H	8.270179	-0.762075	1.906312
C	7.629056	-1.066939	-0.156843	H	6.994833	-1.980421	1.723653
C	7.050354	1.416676	-0.080082	H	8.694733	2.765102	-0.190123
C	7.900847	-1.570109	1.265345	H	9.168643	1.19518	0.462149
C	8.538267	1.685456	-0.286072	H	8.87761	1.389607	-1.284209
O	0.070005	2.358533	0.093976				

Cartesian Coordinates (in Å) of 2-CN⁻



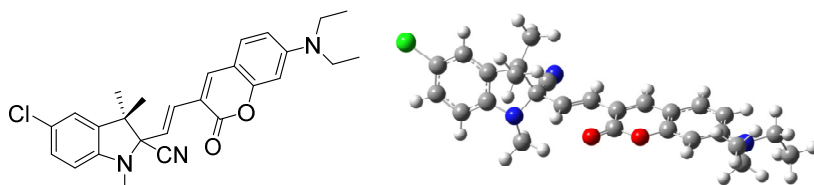
C	3.413382	0.329081	0.05296	N	3.620996	2.975486	-0.020414
C	1.975611	-0.112164	0.192622	H	1.85394	-1.142075	0.503898
C	0.900786	0.657359	-0.058713	H	1.055483	1.694849	-0.360577
C	-0.51156	0.292644	0.016059	H	-1.16085	2.231601	-0.592395
C	-1.47426	1.229445	-0.293954	H	-3.627233	2.871137	-0.854666
C	-2.860816	0.944313	-0.243089	H	-5.970065	2.247222	-0.710385
C	-3.257902	-0.350798	0.147719	H	-4.789168	-1.765106	0.525035
O	-2.307997	-1.28865	0.456899	H	-7.692542	1.308905	-1.284632
C	-0.940597	-1.044747	0.412768	H	-8.84553	0.107402	-0.751751
C	-3.894288	1.858536	-0.553855	H	-6.693376	-1.794546	1.314855
C	-5.221258	1.499867	-0.476011	H	-8.347276	-1.353257	0.954489
C	-5.614387	0.180325	-0.07822	H	-9.313231	2.294648	0.349168
C	-4.583069	-0.742507	0.232022	H	-8.854367	1.089493	1.567541
N	-6.934916	-0.172999	0.000988	H	-7.683877	2.318457	1.054253
C	-8.013407	0.732979	-0.411179	H	-7.769947	-3.510934	-0.154131
C	-7.358311	-1.48491	0.502264	H	-8.125194	-2.27669	-1.377982
C	-8.491349	1.664647	0.708392	H	-6.444793	-2.731785	-1.041978
C	-7.425586	-2.563781	-0.585002	H	6.882826	-0.528675	-2.361364
O	-0.228343	-1.988581	0.716498	H	8.980622	-0.553231	-1.01407
C	4.197492	-0.324251	-1.181602	H	8.875278	-0.352903	1.457984
C	5.616635	-0.305186	-0.62871	H	6.69952	-0.114226	2.623095
C	5.564558	-0.199057	0.771641	H	4.52475	-0.062782	-3.305971
N	4.232185	-0.106018	1.209299	H	2.939828	0.497409	-2.756039
C	3.509229	1.818998	-0.036544	H	4.392684	1.469784	-2.430326
C	6.835743	-0.441861	-1.276284	H	2.719538	-1.846114	-1.742645
C	8.017462	-0.458059	-0.516332	H	4.401596	-2.259447	-2.128924
C	7.955911	-0.346262	0.874059	H	3.830457	-2.365353	-0.454627
C	6.728738	-0.211253	1.540346	H	2.847192	0.372435	2.692205
C	4.002645	0.449593	-2.490515	H	4.41338	-0.19819	3.288559
C	3.752306	-1.790049	-1.382331	H	4.25938	1.464208	2.662543
C	3.926933	0.422151	2.53025				

Cartesian Coordinates (in Å) of 3



C	-7.491308	-0.422116	0.034142	H	-8.490496	1.485621	0.173079
C	-7.533677	0.971037	0.123812	H	-6.389331	2.791541	0.220522
C	-6.345867	1.707216	0.150697	H	-6.285501	-2.216923	-0.100871
C	-5.150496	0.997357	0.085527	H	-2.905846	3.204115	-0.704239
C	-5.110616	-0.395506	-0.004549	H	-4.388718	3.491673	0.230798
C	-6.286629	-1.130556	-0.031605	H	-2.880651	3.089604	1.079875
N	-3.820219	1.4835	0.094403	H	-2.34652	-1.915933	-1.469467
C	-2.923315	0.484107	0.015813	H	-4.028102	-2.470733	-1.464976
C	-3.665045	-0.85672	-0.060422	H	-3.604203	-0.938291	-2.253442
C	-3.474351	2.903266	0.180398	H	-2.300782	-2.09772	1.141556
C	-3.386496	-1.586186	-1.398681	H	-3.522425	-1.23435	2.097655
C	-3.340691	-1.760518	1.155334	H	-3.983918	-2.645961	1.126207
C	-1.539499	0.755584	0.011094	H	-1.233132	1.791116	0.071082
C	-0.536722	-0.190018	-0.063307	H	-0.801121	-1.242358	-0.12395
C	0.86972	0.045392	-0.070181	H	1.328626	-2.050005	-0.204428
C	1.744796	-1.041834	-0.149863	H	3.698763	-2.972068	-0.28753
C	3.132826	-0.883909	-0.162869	H	6.086391	-2.569171	-0.294338
C	3.66325	0.429683	-0.09067	H	5.324127	1.729767	-0.021958
O	2.818393	1.502127	-0.007943	H	7.906774	-2.049758	-0.814048
C	1.436038	1.395906	0.005953	H	9.188974	-0.896099	-0.603036
C	4.071865	-1.949738	-0.237461	H	7.240395	1.861684	-0.82231
C	5.420965	-1.716253	-0.242764	H	7.466992	1.63707	0.912719
C	5.949679	-0.377382	-0.182536	H	9.245436	-2.583113	1.226066
C	5.015345	0.694457	-0.09378	H	8.906257	-0.979477	1.905197
N	7.289353	-0.138878	-0.21297	H	7.592989	-2.154992	1.712332
C	8.256871	-1.246633	-0.160774	H	9.45922	2.54492	-0.163808
C	7.766204	1.257133	-0.074612	H	9.868143	0.958036	0.491448
C	8.511672	-1.769856	1.257392	H	9.605308	1.165434	-1.257999
C	9.265007	1.47175	-0.263411	Cl	-9.006359	-1.320477	0.002722
O	0.825274	2.44639	0.081129				

Cartesian Coordinates (in Å) of 3-CN⁻



C	2.773822	0.434981	0.158908	N	2.948443	3.055731	-0.229505
C	1.339243	-0.010551	0.312131	H	1.222571	-0.999748	0.736838
C	0.262241	0.708898	-0.053393	H	0.412443	1.70784	-0.467788
C	-1.146061	0.331149	0.027854	H	-1.808996	2.187123	-0.789221
C	-2.114861	1.215356	-0.397048	H	-4.277902	2.756209	-1.164692
C	-3.497671	0.913401	-0.346606	H	-6.613456	2.110742	-1.009628
C	-3.884653	-0.342533	0.164267	H	-5.40327	-1.737124	0.649332
O	-2.92859	-1.228372	0.586925	H	-8.31016	1.096049	-1.517619
C	-1.564453	-0.966044	0.54957	H	-9.455284	-0.071102	-0.899384
C	-4.537037	1.774474	-0.769483	H	-7.324235	-1.724457	1.391197
C	-5.859955	1.403261	-0.684358	H	-8.975729	-1.349021	0.953446
C	-6.24295	0.123538	-0.164911	H	-9.991242	2.195975	-0.02001
C	-5.20548	-0.745945	0.258972	H	-9.531501	1.125534	1.318019
N	-7.559381	-0.242641	-0.079601	H	-8.378059	2.326994	0.710135
C	-8.642352	0.5997	-0.600855	H	-8.339796	-3.593585	0.072877
C	-7.974536	-1.506768	0.53803	H	-8.685928	-2.489981	-1.272246
C	-9.163389	1.623647	0.414056	H	-7.006756	-2.880875	-0.858049
C	-8.000193	-2.686684	-0.440457	H	6.324512	-0.664911	-2.040504
O	-0.845472	-1.861731	0.963314	H	8.206371	-0.008694	1.78596
C	3.600794	-0.350677	-0.966115	H	5.996356	0.342758	2.854737
C	5.001701	-0.247096	-0.378036	H	3.988347	-0.340476	-3.096979
C	4.906809	0.02555	0.997845	H	2.383248	0.259676	-2.66091
N	3.565997	0.156132	1.381504	H	3.815812	1.283849	-2.413674
C	2.853083	1.904318	-0.107073	H	2.161283	-1.951169	-1.384207
C	6.234579	-0.447485	-0.97797	H	3.859283	-2.382659	-1.665766
C	7.378103	-0.355634	-0.172932	H	3.234082	-2.297099	-0.00939
C	7.299641	-0.080943	1.189538	H	2.129631	0.773627	2.759713
C	6.049908	0.117613	1.792411	H	3.68351	0.298393	3.462649
C	3.438312	0.258509	-2.36305	H	3.527701	1.878304	2.649084
C	3.180829	-1.837369	-1.001153	Cl	8.969001	-0.587899	-0.918818
C	3.213067	0.823064	2.626073				