## Supporting Information

# Cu-catalysed oxidative $\mathbf{C}-\mathbf{H} / \mathrm{C}-\mathrm{H}$ coupling polymerisation of benzodiimidazoles: an efficient approach to regioregular polybenzodiimidazoles for blue-emitting materials 

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## I. General remarks

Unless otherwise noted, all reagents were obtained from commercial suppliers and used without further purification. $\mathrm{Cu}(\mathrm{OAc})_{2}$ was washed with acetic anhydride using a Soxhlet extractor for 5 days prior to use. Benzodiimidazole derivatives were synthesized according to the literature procedures. ${ }^{1}$ NMR spectra were obtained on a Bruker AMX-400. The ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ) chemical shifts were reported relative to $\mathrm{CDCl}_{3}$ as the internal reference ( $\mathrm{CDCl}_{3}: \delta=7.26 \mathrm{ppm}$ ); The ${ }^{13} \mathrm{C}$ NMR ( 100 MHz ) chemical shifts were given using $\mathrm{CDCl}_{3}$ as the internal standard $\left(\mathrm{CDCl}_{3}: \delta=77.16\right.$ ppm). High-resolution mass spectra (HR-MS) were obtained with a Waters-Q-TOF-Premier (ESI). Melting points were determined with XRC-1 and are uncorrected. Gel permeation chromatography (GPC) measurements were performed on a LC-20AD system using $\mathrm{CHCl}_{3}$ as eluent and polystyrene as standards at a column temperature of $40{ }^{\circ} \mathrm{C}$. Thermogravimetric analysis (TGA) was carried out using NETZSCH TG 209F1 Iris at a heating rate of $10^{\circ} \mathrm{C} /$ min under $\mathrm{N}_{2}$ atmosphere. Absorption spectra were recorded on HITACHI U-2910 spectrophotometer. Fluorescence spectra and absolute quantum yields were collected on a Horiba Jobin Yvon-Edison Fluoromax-4 fluorescence spectrometer with a calibrated integrating sphere system. To reduce the fluctuation in the excitation intensity, the lamp was kept on for 1 hour prior to the experiment.

## II. Synthesis of the benzodiimidazole derivatives



To a well-stirred mixture of $7.31 \mathrm{~g}(50 \mathrm{mmol})$ of $m$-dichlorobenzene with 25 g ( 247.5 mmol ) of potassium nitrate was added 80 mL of concentrated sulfuric acid in one portion. The temperature of the reaction mixture rose during a few minutes to $80^{\circ} \mathrm{C}$ and kept the temperature for 30 minutes. And then rose the temperature to $130^{\circ} \mathrm{C}$ for 4 h . After the reaction mixture was cooled to room temperature, it was poured into
crushed ice. A yellow precipitated was collected and washed by water ( $3 \times 100 \mathrm{~mL}$ ). The resulting solid was recrystallized with ethyl alcohol, affording 1,5-dichloro-3,4-dinitrobenzene as a yellow solid (10.82 g, 91.1\% yield).


After 1,5-dichloro-3,4-dinitrobenzene ( $1.00 \mathrm{~g}, 4.22 \mathrm{mmol}$ ) was dissolved in EtOH ( 75 mL ), amine ( 16.9 mmol ) was added in a single portion. The mixture was then placed in an oil bath at $80^{\circ} \mathrm{C}$ and stirred for 48 h . The mixture was poured into $\mathrm{H}_{2} \mathrm{O}$ (200 mL) which caused solids to precipitate. The solids ( $N^{1}, N^{3}$-dialkyl-4,6-dinitrobenzene-1,3-diamine) were collected via vacuum filtration, rinsed with $\mathrm{H}_{2} \mathrm{O}$, and dried under vacuum.

$N^{1}, N^{3}$-Dialkyl-4,6-dinitrobenzene-1,3-diamine ( 2.44 mmol ) was suspended in $\mathrm{HCO}_{2} \mathrm{H}(88 \%, 100 \mathrm{~mL})$. To the suspension was added $\mathrm{HCO}_{2} \mathrm{Na}(1.99 \mathrm{~g}, 29.2 \mathrm{mmol})$ and $\mathrm{Pd} / \mathrm{C}(103 \mathrm{mg}, 3 \mathrm{wt} \%, 0.05 \mathrm{mmol} \mathrm{Pd})$. The mixture was heated in an oil bath at $140{ }^{\circ} \mathrm{C}$ for 48 h . Upon completion, the cooled reaction mixture was filtered through celite and the filtrate volume was reduced to ca. 20 mL under vacuum. The solution was then added slowly into a vigorously stirred saturated solution of aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ ( 150 mL ). Precipitated solids were collected via vacuum filtration, rinsed with $\mathrm{H}_{2} \mathrm{O}$. Purification via silica gel column chromatography (petroleum ether/acetone $=1 / 1, \mathrm{v} / \mathrm{v}$ ) afforded the desired product.

$N^{1}, N^{4}$-Dioctyl-3,6-dinitrobenzene-1,4-diamine ${ }^{2}(2.23 \mathrm{~g}, 5.29 \mathrm{mmol})$ was suspended
in $\mathrm{HCO}_{2} \mathrm{H}(88 \%, 80 \mathrm{~mL})$. To the suspension was added $\mathrm{HCO}_{2} \mathrm{Na}(8.0 \mathrm{~g}, 118 \mathrm{mmol})$ and $\mathrm{Pd} / \mathrm{C}(1.8 \mathrm{~g}, 5 \mathrm{wt} \%)$. The mixture was heated in an oil bath at $140^{\circ} \mathrm{C}$ for 40 h . Upon completion, the cooled reaction mixture was filtered through celite and the filtrate volume was reduced to ca. 20 mL under vacuum. The solution was then added slowly into a vigorously stirred saturated solution of aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ ( 150 mL ). Precipitated solids were collected via vacuum filtration, rinsed with $\mathrm{H}_{2} \mathrm{O}$. Purification via silica gel column chromatography (petroleum ether/acetone $=1 / 1, \mathrm{v} / \mathrm{v}$ ) afforded the desired product ( $800 \mathrm{mg}, 39.5 \%$ ).

## III. Optimization of the oxidative $\mathbf{C}-\mathbf{H} / \mathrm{C}-\mathrm{H}$ coupling polymerization

1,7-Dioctyl-1H,7H-benzo[1,2-d:4,5- $d$ ]diimidazole (BDI-8, 0.25 mmol ), $\mathrm{Cu}(\mathrm{OAc})_{2}$ ( 9 $\mathrm{mg}, 20 \mathrm{~mol} \%$ ) and oxidant were dissolved in solvent ( 1 mL ). The reaction container was purged with $\mathrm{O}_{2}$ for 30 min to remove air and then heated at $140^{\circ} \mathrm{C}$ for 24 h . The mixture was cooled to room temperature, and the polymer was precipitated by slowly adding the mixture of $\mathrm{CH}_{3} \mathrm{OH}$ and $\mathrm{H}_{2} \mathrm{O}$. The resulting solid was filtered and subjected to Soxhlet extraction in methanol, acetone and hexane for the removal of low molecular weight materials and impurities. The remaining polymer was extracted with chloroform, precipitated again from methanol, filtered, washed with methanol and dried under vacuum.

Table S1 Optimization of the Cu -catalysed oxidative $\mathrm{C}-\mathrm{H} / \mathrm{C}-\mathrm{H}$ coupling polymerisation ${ }^{a}$

|  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Oxidant ${ }^{\text {b }}$ | Solvent | Yield ${ }^{\text {c }}$ | $M \mathrm{n}^{\text {d }}$ | $\mathrm{PDI}^{d}$ |
| 1 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}(1.5) / \mathrm{O}_{2}$ | xylene | 63\% | 8100 | 1.78 |
| 2 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}(1.0) / \mathrm{O}_{2}$ | xylene | 70\% | 3300 | 3.08 |
| 3 | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | xylene | 66\% | 32000 | 1.52 |
| 4 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}(0.2) / \mathrm{O}_{2}$ | xylene | 67\% | 8200 | 3.55 |
| $5^{e}$ | - | xylene | 70\% | 2700 | 2.63 |
| $6^{e}$ | - | DMF | 64\% | 2200 | 2.21 |
| $7{ }^{f}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | xylene | 71\% | 6700 | 2.32 |
| 8 | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | dioxane | 67\% | 5800 | 2.47 |
| 9 | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | dioxane/DMF | 67\% | 4700 | 3.27 |
| 10 | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | toluene | 74\% | 5300 | 3.46 |


| 11 | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | toluene $/ \mathrm{CHCl}_{3}$ | $63 \%$ | 4200 | 1.93 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| $12^{g}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3} / \mathrm{O}_{2}$ | xylene | $68 \%$ | 31500 | 2.16 |

${ }^{a}$ Reaction conditions: BDI-8 ( 0.25 mmol ), $\mathrm{Cu}(\mathrm{OAc})_{2}$ ( $20 \mathrm{~mol} \%$ ), $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ ( 0.5 equiv), and solvent $(1.0 \mathrm{~mL})$ under $\mathrm{O}_{2}(1 \mathrm{~atm})$ at $140{ }^{\circ} \mathrm{C}$ for $24 \mathrm{~h} ;{ }^{b}$ The number in parentheses is the equivalent of $\mathrm{Ag}_{2} \mathrm{CO}_{3} ;{ }^{c}$ The products were obtained by reprecipitation from $\mathrm{CHCl}_{3} / \mathrm{MeOH}$ after Soxhlet extraction; ${ }^{d}$ Estimated by gel permeation chromatography (GPC) on polystyrene standards; ${ }^{e}$ $\mathrm{Cu}(\mathrm{OAc})_{2}\left(1.0\right.$ equiv); ${ }^{f} \mathrm{Cu}(\mathrm{OAc})_{2}{ }^{\cdot} \mathrm{H}_{2} \mathrm{O} ;{ }^{g} 48 \mathrm{~h}$.

## IV. General procedure for the oxidative $\mathrm{C}-\mathrm{H} / \mathrm{C}-\mathrm{H}$ coupling polymerization of benzodiimidazoles



The benzodiimidazole derivative ( 0.25 mmol ), $\mathrm{Cu}(\mathrm{OAc})_{2}(9 \mathrm{mg}, 20 \mathrm{~mol} \%)$ and $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ ( $34.5 \mathrm{mg}, 50 \mathrm{~mol} \%$ ) were dissolved in xylene ( 1 mL ). The reaction container was purged with $\mathrm{O}_{2}$ for 30 min to remove air and then heated at $140^{\circ} \mathrm{C}$ for 24 h . The mixture was cooled to room temperature, and the polymer was precipitated by slowly adding the mixture of $\mathrm{CH}_{3} \mathrm{OH}$ and $\mathrm{H}_{2} \mathrm{O}$. The resulting solid was filtered and subjected to Soxhlet extraction in methanol, acetone and hexane for the removal of low molecular weight materials and impurities. The remaining polymer was extracted with chloroform, precipitated again from methanol, filtered, washed with methanol and dried under vacuum.

## V. Characterization data of products


$N^{1}, N^{3}$-Dibutyl-4,6-dinitrobenzene-1,3-diamine

A yellow powder, m.p. $100-102{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.01(\mathrm{t}, J=7.4$ $\mathrm{Hz}, 6 \mathrm{H}), 1.46-1.55(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.79(\mathrm{~m}, 4 \mathrm{H}), 3.25-3.30(\mathrm{~m}, 4 \mathrm{H}), 5.64(\mathrm{~s}, 1 \mathrm{H}), 8.31$ (s, 2H), $9.22(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=13.9,20.4,30.6,43.2$, 90.2, 124.2, 129.7, 148.8 ppm . HRMS (ESI ${ }^{+}$): calculated for $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{~N}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 333.1539, found 333.1536.


## 4,6-Dinitro- $N^{1}, N^{3}$-dioctylbenzene-1,3-diamine

A yellow powder, m.p. $76-78{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.89(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $6 \mathrm{H}), 1.29-1.49(\mathrm{~m}, 20 \mathrm{H}), 1.72-1.80(\mathrm{~m}, 4 \mathrm{H}), 3.24-3.29(\mathrm{~m}, 4 \mathrm{H}), 5.63(\mathrm{~s}, 1 \mathrm{H}), 8.31(\mathrm{~s}$, 2H), 9.22 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.2,28.6,29.3$, 29.4, 31.9, 43.5, 90.2, 124.2, 129.7, 148.8 ppm. HRMS (ESI ${ }^{+}$): calculated for $\mathrm{C}_{22} \mathrm{H}_{38} \mathrm{O}_{4} \mathrm{~N}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 445.2791$, found 445.2788.

$N^{1}, N^{3}$-Didodecyl-4,6-dinitrobenzene-1,3-diamine
A yellow powder, m.p. $84-86{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $6 \mathrm{H}), 1.26-1.36(\mathrm{~m}, 32 \mathrm{H}), 1.42-1.49(\mathrm{~m}, 4 \mathrm{H}), 1.73-1.80(\mathrm{~m}, 4 \mathrm{H}), 3.24-3.29(\mathrm{~m}, 4 \mathrm{H})$, $5.63(\mathrm{~s}, 1 \mathrm{H}), 8.32(\mathrm{~s}, 2 \mathrm{H}), 9.23(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.3$, 22.8, 27.2, 28.6, 29.4, 29.5, 29.6, 29.7, 29.77, 29.79, 32.1, 43.5, 90.2, 124.2, 129.8, 148.8 ppm. HRMS (ESI ${ }^{+}$): calculated for $\mathrm{C}_{30} \mathrm{H}_{54} \mathrm{O}_{4} \mathrm{~N}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 557.4043$, found 557.4045 .


4,6-Dinitro- $N^{1}, N^{3}$-dioctadecylbenzene-1,3-diamine

A yellow powder, m.p. $82-84{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $6 \mathrm{H}), 1.25-1.36(\mathrm{~m}, 56 \mathrm{H}), 1.42-1.47(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.80(\mathrm{~m}, 4 \mathrm{H}), 3.24-3.29(\mathrm{~m}, 4 \mathrm{H})$, $5.63(\mathrm{~s}, 1 \mathrm{H}), 8.32(\mathrm{~s}, 2 \mathrm{H}), 9.23(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=14.3$, 22.8, 27.2, 28.6, 29.4, 29.5, 29.65, 29.73, 29.80, 29.81, 29.83, 29.9, 32.1, 43.5, 90.2 , 124.2, 129.8, 148.8 ppm . HRMS ( $\mathrm{ESI}^{+}$): calculated for $\mathrm{C}_{42} \mathrm{H}_{78} \mathrm{O}_{4} \mathrm{~N}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 725.5921, found 725.5919.


## 1,7-Dibutyl-1H,7H-benzo[1,2-d:4,5-d]diimidazole (BDI-4)

A white powder, m.p. $76-78{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.93-0.97(\mathrm{~m}, 6 \mathrm{H})$, $1.33-1.40(\mathrm{~m}, 4 \mathrm{H}), 1.85-1.90(\mathrm{~m}, 4 \mathrm{H}), 4.16-4.20(\mathrm{~m}, 4 \mathrm{H}), 7.19(\mathrm{~s}, 1 \mathrm{H}), 7.89(\mathrm{~s}, 2 \mathrm{H})$, 8.16 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=13.7,20.2,31.5,45.0,88.3$, 110.3, 132.1, 141.1, 143.7 ppm. HRMS ( $\mathrm{ESI}^{+}$): calculated for $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{4}[\mathrm{M}+\mathrm{H}]^{+}$ 271.1923, found 271.1920.


## 1,7-Dioctyl-1H,7H-benzo[1,2-d:4,5-d]diimidazole (BDI-8)

A white powder, m.p. $72-74{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=0.84(\mathrm{t}, J=6.6 \mathrm{~Hz}$, $6 \mathrm{H}), 1.22-1.32(\mathrm{~m}, 20 \mathrm{H}), 1.86-1.91(\mathrm{~m}, 4 \mathrm{H}), 4.17(\mathrm{t}, J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H})$, 7.89 (s, 2H), 8.17 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=14.1,22.7,27.0$,
29.2, 29.5, 31.8, 45.3, 88.3, 110.3, 132.0, 141.1, 143.7 ppm. HRMS (ESI $^{+}$): calculated for $\mathrm{C}_{24} \mathrm{H}_{39} \mathrm{~N}_{4}[\mathrm{M}+\mathrm{H}]^{+}$383.3175, found 383.3173.


## 1,5-Dioctyl-1,5-dihydrobenzo[1,2-d:4,5-d]diimidazole (BDI-8')

A white powder, m.p. $106-108{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.83(\mathrm{t}, J=6.8$ $\mathrm{Hz}, 6 \mathrm{H}), 1.22-1.30(\mathrm{~m}, 20 \mathrm{H}), 1.87-1.92(\mathrm{~m}, 4 \mathrm{H}), 4.19(\mathrm{t}, J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.72(\mathrm{~s}, 2 \mathrm{H})$, 7.93 (s, 2H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.1,22.7,26.9,29.2,29.5,31.8$, 45.4, 99.1, 131.6, 141.6, 144.2 ppm. HRMS (ESI ${ }^{+}$): calculated for $\mathrm{C}_{24} \mathrm{H}_{39} \mathrm{~N}_{4}[\mathrm{M}+\mathrm{H}]^{+}$ 383.3175, found 383.3173.


## 1,7-Didodecyl-1H,7H-benzo[1,2-d:4,5-d]diimidazole (BDI-12)

A white powder, m.p. $70-72{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.87(\mathrm{t}, J=6.6 \mathrm{~Hz}$, 6 H ), 1.24-1.35 (m, 36H), 1.87-1.96 (m, 4H), $4.20(\mathrm{t}, J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H})$, 7.92 (s, 2H), 8.19 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.1$, 29.3, 29.5, 29.57, 29.61, 29.68, 29.73, 29.74, 32.0, 45.4, 88.3, 110.5, 132.1, 141.2, 143.8 ppm. HRMS (ESI ${ }^{+}$): calculated for $\mathrm{C}_{32} \mathrm{H}_{55} \mathrm{~N}_{4}[\mathrm{M}+\mathrm{H}]^{+}$495.4427, found 495.4426.


A white powder, m.p. $68-70{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.87(\mathrm{t}, J=6.8 \mathrm{~Hz}$, 6 H ), 1.24-1.34 (m, 60H), 1.90-1.94 (m, 4H), 4.20 (t, $J=7.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.21 (s, 1H), 7.93 (s, 2H), 8.18 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.1$, 29.3, 29.5, 29.57, 29.61, 29.7, 29.76, 29.79, 29.83, 32.1, 45.4, 88.4, 110.3, 132.1, 141.1, 143.7 ppm. HRMS (ESI $)$ : calculated for $\mathrm{C}_{44} \mathrm{H}_{79} \mathrm{~N}_{4}[\mathrm{M}+\mathrm{H}]^{+}$663.6305, found 663.6308.


PBDI-4

According to the general procedure, PBDI-4 was obtained as a brown solid ( 42 mg , $63 \%$ yield $) . \mathrm{Mn}=22500, \mathrm{PDI}=2.09 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.94-0.99(\mathrm{~m}$, 6 H ), 1.43 ( $\mathrm{s}, 4 \mathrm{H}$ ), 1.97 ( $\mathrm{s}, 4 \mathrm{H}$ ), $5.11(\mathrm{~s}, 4 \mathrm{H}), 7.43(\mathrm{~s}, 1 \mathrm{H}), 8.30(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=13.9,20.3,31.9,45.3,89.4,110.2,134.8,140.7,144.0$.


PBDI-8
According to the general procedure, PBDI-8 was obtained as a brown solid ( 63 mg , $66 \%$ yield). $\mathrm{Mn}=32000$, $\mathrm{PDI}=1.52 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.83(\mathrm{~s}, 6 \mathrm{H})$, $1.23(\mathrm{~s}, 20 \mathrm{H}), 1.98(\mathrm{~s}, 4 \mathrm{H}), 5.09(\mathrm{~s}, 4 \mathrm{H}), 7.43(\mathrm{~s}, 1 \mathrm{H}), 8.32(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.1,29.3,29.77,29.84,31.9,45.6,89.3,110.4$, 134.8, 140.8, 144.0. The structure of PBDI-8 was further confirmed by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY spectrum and HMQC ( ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ COSY) spectrum.


PBDI-8'

According to the general procedure, PBDI-8' was obtained as a brown solid ( 67 mg , $71 \%$ yield). $\mathrm{Mn}=5800$, PDI $=1.33 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.85(\mathrm{~s}, 6 \mathrm{H})$, 1.24-1.28 (m, 20H), $2.01(\mathrm{~s}, 4 \mathrm{H}), 5.08(\mathrm{~s}, 4 \mathrm{H}), 7.95(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta=14.2,22.7,27.0,29.3,29.8,31.9,45.6,99.7,134.0,141.5,144.4$.


COPBDI-8

According to the general procedure, COPBDI-8 was obtained as a brown solid (57 $\mathrm{mg}, 61 \%$ yield). $\mathrm{Mn}=4100$, $\mathrm{PDI}=1.50 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.82$ (s, 11 H ), 1.24 (m, 36H), 1.98 (m, 7.7H), 5.08 (s, 4H), 7.43 (s, 1H), 7.93 (s, 1.8H), 8.34 (s, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.1,29.3,29.8,31.8,45.5$, 134.8, 140.8, 141.6, 144.5.


PBDI-12

According to the general procedure, PBDI-12 was obtained as a brown solid ( 78 mg , $63 \%$ yield). $\mathrm{Mn}=37200$, $\mathrm{PDI}=1.68 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.84-0.88(\mathrm{~m}$, 6H), 1.22-1.25 (m, 36H), 1.98 (s, 4H), 5.09 (s, 4H), 7.43 (s, 1H), 8.33 (s, 1H) ppm. ${ }^{13} \mathrm{C}_{\mathrm{NMR}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=14.2,22.8,27.1,29.46,29.51,29.79,29.83,32.1$,


## PBDI-18

According to the general procedure, PBDI-18 was obtained as a brown solid ( 140 mg , $85 \%$ yield). $\mathrm{Mn}=44500, \mathrm{PDI}=1.32 .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.86(\mathrm{~s}, 6 \mathrm{H})$, $1.25(\mathrm{~s}, 60 \mathrm{H}), 1.99(\mathrm{~s}, 4 \mathrm{H}), 5.10(\mathrm{~s}, 4 \mathrm{H}), 7.43(\mathrm{~s}, 1 \mathrm{H}), 8.33(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=14.2,22.8,27.1,29.46,29.51,29.79,29.84,32.1,45.6,89.3$, 110.3, 134.9, 140.8, 144.0.

## VI. References

1. A. J. Boydston, P. D. Vu, O. L. Dykhno, V. Chang, A. R. Wyatt, A. S. Stockett, E. T. Ritschdorff, J. B. Shear and C. W. Bielawski, J. Am. Chem. Soc., 2008, 130, 3143.
2. Z.-L. Zhou, S.-T. Lam, L. Zhao, US20130237725.

## VII. Absorption spectra of the polymers



Fig. S1. Absorption spectra of PBDI-4, PBDI-8, PBDI-12, PBDI-18, PBDI-8' and COPBDI-8 in $\mathrm{CHCl}_{3}$.
VIII. Emission spectra of the polymers in PS film


Fig. S2. Emission spectra of PBDI-8, PBDI-12, PBDI-18, PBDI-8' and COPBDI-8 in PS (c = $1 \mathrm{wt} \%$ ) film.
IX. TGA curves of the polymers



Fig. S3. TGA curves of PBDI-4, PBDI-8, PBDI-12 and PBDI-18 (left); TGA curves of PBDI-8' and COPBDI-8 (right).

## X. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY and HMQC $\left({ }^{13} \mathrm{C}^{-1} \mathrm{H}\right.$ COSY $)$ spectra of PBDI-8



Fig. S4. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY and $\operatorname{HMQC}\left({ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}\right.$ COSY $)$ spectra of PBDI-8.

## XI. Copies of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra



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