

Red luminescence and ferromagnetism in europium oxynitridosilicates with β -K₂SO₄ structure

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Supplementary Information

Experimental

Polycrystalline samples of LaM_{1-x}Eu_xSiO₃N ($x=0, 0.01, 0.02, 0.05, 0.1$) and La_{1-x}Ce_xMSiO₃N ($x=0, 0.01, 0.02, 0.05, 0.1$) (M= Sr) were prepared by solid state reaction in N₂/H₂ (95%:5%, Air liquide, 99.999%) of stoichiometric mixtures of La₂O₃ (Aldrich, 99.99 %), SrO or BaO, Si₃N₄ (α -phase, Alfa Aesar, 99.9%), SiO₂ (Aldrich, 99.995 %) and Eu₂O₃ (Aldrich, 99.99 %) or CeO₂ (Aldrich, 99.995 %). LaEuSiO₃N was prepared from stoichiometric mixtures of La₂O₃, Si₃N₄ and Eu₂O₃ adding 1.4 % (w/w) of graphitic C (Aldrich) as oxygen scavenger. La₂O₃, Eu₂O₃, SiO₂ and CeO₂ were previously dried at 950 °C for 4 h for dehydration. SrO and BaO were prepared by overnight decomposition of SrCO₃ (Alfa Aesar, 99.994%) and BaCO₃ (Aldrich 99.99 %) at 1000 °C under a dynamic vacuum of 5x10⁻² mbar. Handling of SrO and BaO was carried out in a Glovebox under recirculating argon atmosphere. The powders were thoroughly mixed in an agate mortar for 30 min, pressed into a pellet, placed in a molybdenum crucible and covered with a zirconium foil as oxygen/water scavenger. The mixtures were fired at 1350 °C (for M=Sr, Eu) or 1500 °C (for M=Ba) during 3 to 9 h with heating and cooling rates of 300°C/h. N contents were determined by combustion analysis using a Thermo Fisher Scientific instrument and by thermogravimetric analysis in pure oxygen using a NETZSCH -STA 449 F1 Jupiter equipment.

Powder X-ray diffraction data were collected on a Rigaku diffractometer using Cu K α radiation ($\lambda= 1.5418 \text{ \AA}$). Powder neutron diffraction data from a 0.5 g LaBaSiO₃N sample were recorded at room temperature on the high-resolution diffractometer HRPD at the ISIS spallation source, Rutherford Appleton Laboratory, U.K. Rietveld refinements were performed using the program Fullprof. Structural models from neutron diffraction data were refined simultaneously against data collected from backscattering and 90° detector banks, which provide d -spacing ranges of 0.6 – 2.6 Å and 0.9 – 3.8 Å, respectively. Electron diffraction micrographs were obtained in a JEOL 1210 transmission electron microscope operating at 120 KV, equipped with a side-entry 60/30° double tilt GATHAN 646 specimen holder. The samples were prepared by dispersing the powders in hexane and depositing of a droplet of this suspension on a carbon coated holey film supported on a copper grid.

Diffuse reflectance spectra were registered at room temperature on a UV-Vis-NIR Varian Cary 5000 spectrophotometer, with operational range of 190-3300 nm. The excitation spectra were measured on a Horiba Jobin Yvon FluoroMax-4 spectrofluorometer equipped with an integrating sphere, a xenon source, a photomultiplier detector (range 200 nm to 850 nm) and a photodiode reference detector. The wavelength range is 200 nm to 950 nm. Photoluminescence (PL) spectra were measured at room temperature using the 405 nm line of a solid-state laser for excitation and collected using a LabRam HR800 spectrometer equipped with a charge-coupled device detector. PL spectra were corrected for the spectral response of the spectrometer by normalizing each spectrum employing the detector and grating characteristics. The incident light power density was about 1 W/cm².

Magnetic measurements were performed between 2 and 300 K in fields of 0.0025 and 0.5 T using a Quantum Design SQUID magnetometer. M(H) curves were measured up to 7 T for temperatures between 2 K and 50 K.

Table S1. Crystallographic and refinement data for LaSrSiO₃N and LaBaSiO₃N

	LaSrSiO ₃ N	LaEuSiO ₃ N	LaBaSiO ₃ N
Radiation	X-Ray Cu k.	X-Ray Cu k.	Neutron
T(K)	298	298	298
Space group	Pmnb	Pmnb	Pmnb
a(Å)	5.64362(14)	5.63246(14)	5.733(4)
b(Å)	7.10719(17)	7.11675(18)	7.316(5)
c(Å)	9.8062(2)	9.7998(2)	9.902(7)
V(Å ³)	393.329(16)	392.825(17)	415.3(5)
N _p , N _{ireff} ^(a)	4751, 264	4751, 240	1911, 281
P _p , P _i , P _g ^(b)	9, 17, 8	9, 15, 8	14, 18, 11
R _{Bragg} , R _f , χ ²	5.37, 4.52, 1.67	6.24, 4.62, 1.70	5.30, 4.4, 1.71
R _p , R _{wp} , R _{exp} ^(c)	9.65, 13.2, 14.9	9.19, 12.2, 9.36	2.43, 2.69, 2.06

(a) N_p, N_{ireff} refer to the number of experimental points and independent reflections.

(b) P_p, P_i, P_g refer to the number of profile, intensity-affecting and global refined parameters, respectively.

(c) Conventional Rietveld R-factors (R_p, R_{wp}, R_{exp}) in %.

Table S2. Atomic coordinates, cation and anion occupancies for LaSrSiO₃N^(a)

Site	Wyckoff position	x	y	z	occupation factor
La1/ Sr1	4c	0.25	0.6552(3)	0.5782(3)	0.242(3)/0.758
La2/ Sr2	4c	0.25	0.0078(3)	0.30391(19)	0.758/ 0.242
Si	4c	0.25	0.2192(11)	0.5828(12)	1
O1/N1	4c	0.25	0.992(2)	0.5663(17)	1/0
O2/N2	4c	0.25	0.316(3)	0.436(2)	0.5/0.5
O3/N3	8d	0.014(2)	0.2874(17)	0.6652(19)	0.75/0.25

(a) Estimated standard deviations in parentheses are shown once for each independent variable. La/Sr occupation factors were refined subject to the ideal stoichiometry. O/N occupation factors were considered fixed to those obtained in LaBaSiO₃N from neutron diffraction. Refined isotropic B-factors were 1.01(7) Å² for La/Sr atoms, 1.32(18) Å² for silicon and 2.1(2) Å² for anions.

Table S3. Atomic coordinates, cation and anion occupancies for LaEuSiO₃N^(a)

Site	Wyckoff position	x	y	z	occupation factor
La1/ Eu1	4c	0.25	0.6547(3)	0.5780(2)	0.188(18)/0.812
La2/ Eu2	4c	0.25	0.0067(3)	0.30338(19)	0.812/ 0.188
Si	4c	0.25	0.2203(13)	0.5826(14)	1
O1/N1	4c	0.25	0.994(3)	0.569(2)	1/0
O2/N2	4c	0.25	0.312(4)	0.430(3)	0.5/0.5
O3/N3	8d	0.015(3)	0.283(2)	0.659(2)	0.75/0.25

(a) Estimated standard deviations in parentheses are shown once for each independent variable. La/Eu occupation factors were refined subject to the ideal stoichiometry. O/N occupation factors were fixed to those obtained in LaBaSiO₃N from neutron diffraction. Refined isotropic B-factors were 1.03(7) Å² for La/Eu atoms, 1.06(7) Å² for silicon and 2.6(3) Å² for anions.

Table S4. Bond distances and cell parameters for LaSrSiO₃N, LaEuSiO₃N, LaBaSiO₃N (this work) and related silicates with β -K₂SO₄ structure.^(a)

	LaSrSiO ₃ N	Sr ₂ SiO ₄	LaEuSiO ₃ N	Eu ₂ SiO ₄	LaBaSiO ₃ N	Ba ₂ SiO ₄
M1-X1	2.397(14)	2.380(9)	2.42(2)	2.416(19)	2.500(9)	2.640(7)
M1-X2	2.79(2)	2.775(7)	2.84(3)	2.83(2)	2.920(12)	3.083(7)
M1-X2 (x2)	2.8326(18)	2.846(6)	2.827(3)	2.838(1)	2.870(2)	2.915(7)
M1-X3 (x2)	3.055(13)	3.019(7)	3.063(15)	3.013(11)	3.166(8)	3.174(7)
M1-X3 (x2)	2.998(17)	2.985(7)	3.038(19)	2.997(12)	3.036(12)	3.060(7)
M1-X3 (x2)	2.843(17)	2.849(6)	2.796(19)	2.821(12)	2.857(12)	2.904(7)
Average	2.868	2.855	2.869(16)	2.858	2.928	2.983
M2-X1	2.575(17)	2.602(9)	2.61(2)	2.607(17)	2.589(10)	2.846(7)
M2-X1 (x2)	3.096(7)	3.111(3)	3.081(8)	3.095	3.141(5)	3.114(7)
M2-X2	2.54(2)	2.623(7)	2.50(3)	2.57(2)	2.610(10)	2.699(7)
M2-X2	2.72(2)	2.622(8)	2.67(3)	2.646(19)	2.657(11)	2.759(7)
M2-X3 (x2)	2.591(12)	2.617(7)	2.572(15)	2.609(11)	2.630(7)	2.755(7)
M2-X3 (x2)	2.488(14)	2.507(6)	2.544(17)	2.503(12)	2.533(7)	2.686(7)
Average	2.687	2.702	2.685(14)	2.693	2.718	2.824
Si-X1	1.623(16)	1.612(9)	1.62(2)	1.578(19)	1.622(11)	1.614(7)
Si-X2	1.60(2)	1.635(8)	1.63(3)	1.60(2)	1.682 (18)	1.652(7)
Si-X3 (x2)	1.632(15)	1.616(6)	1.585(19)	1.615(13)	1.677(10)	1.632(7)
a (Å)	5.6436(14)	5.682(1)	5.63246	5.665(2)	5.733(4)	5.805(1)
b(Å)	7.10719(17)	7.090(1)	7.11675	7.137(3)	7.316(5)	7.499(1)
c(Å)	9.8062(2)	9.773(2)	9.7998	9.767(3)	9.902(7)	10.200(3)
V (Å ³)	393.329(16)	393.7	392.825	396	415.3(5)	444.02

(a) From powder X-ray diffraction data for LaSrSiO₃N and LaEuSiO₃N, powder neutron diffraction data for LaBaSiO₃N and single crystal X-ray diffraction for Sr₂SiO₄ (Catti, M.; Gazzoni, G.; Ivaldi, G.; Zanini, G. *Acta Cryst.* **1983**, B39, 674-679), Eu₂SiO₄ (Marchand, R.; l'Hariloon, P.; Laurent, Y. *J. Solid State Chem.*, **1978**, 24, 71-76) and Ba₂SiO₄ (Grosse, H-P; Tillmanns, E. *Cryst. Struct. Comm.* **1974**, 3, 599-602.).

Table S5. Bond Valence Sums (BVS) for LaMSiO₃N compounds

Site	LaSrSiO ₃ N	LaBaSiO ₃ N	LaEuSiO ₃ N
La1/ M1	2.05(3)/1.69(2)	1.71(1)/2.35(2) ^a	2.02(4)/1.88(3)
La2/ M2	2.98(5)/2.44(3)	2.74(2)/3.78(3) ^a	2.96(5)/1.97(7)
Si	4.36(8)	3.86(5)	4.5(1)
O1	1.98(5)	2.04(3)	1.97(7)
O2/N2	2.13(7)/2.96(9)	2.03(4)/3.02(6)	2.10(9)/3.1(1)
O3/N3	2.03(4)/2.85(6)	1.99(3)/2.95(4)	2.14(6)/3.15(8)

(a) Deviations of BVS with respect to the formal charges of the cations for these sites are larger because of the mixed occupation and the differences between the ionic radii of Ba²⁺ and La³⁺.

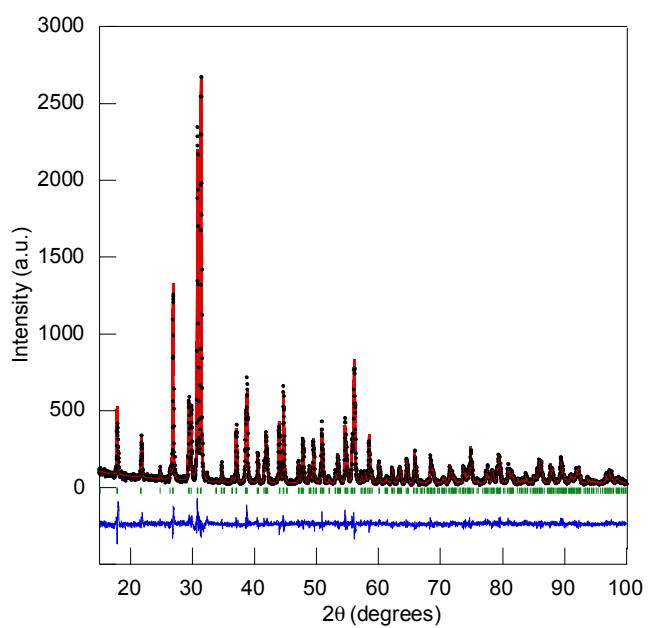


Figure S1. Observed and calculated powder X-ray diffraction patterns for LaSrSiO₃N.

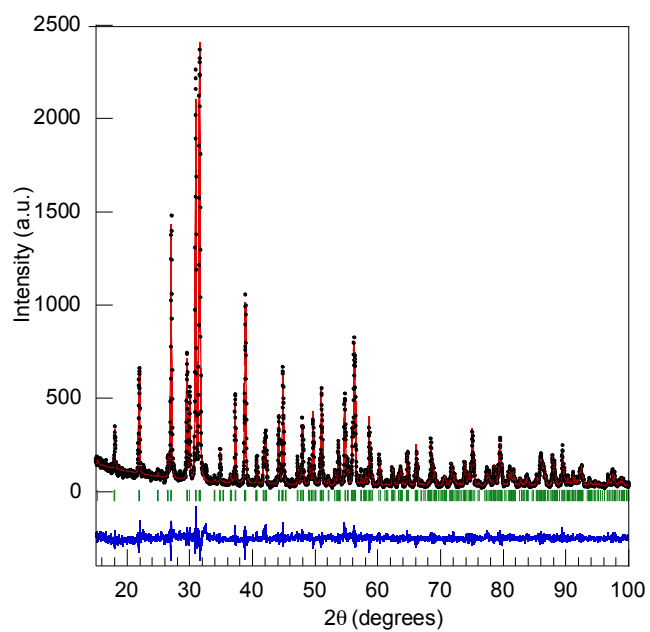


Figure S2. Observed and calculated powder X-ray diffraction patterns for LaEuSiO₃N.

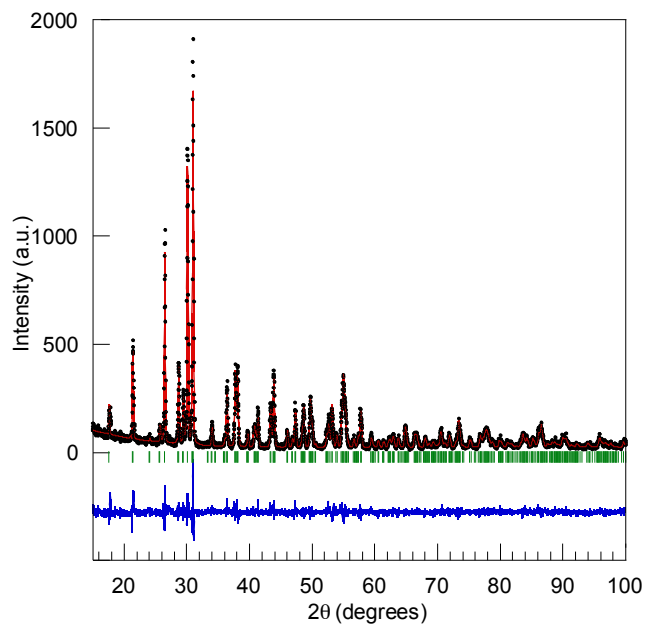


Figure S3. Observed and calculated powder X-ray diffraction patterns for LaBaSiO₃N.

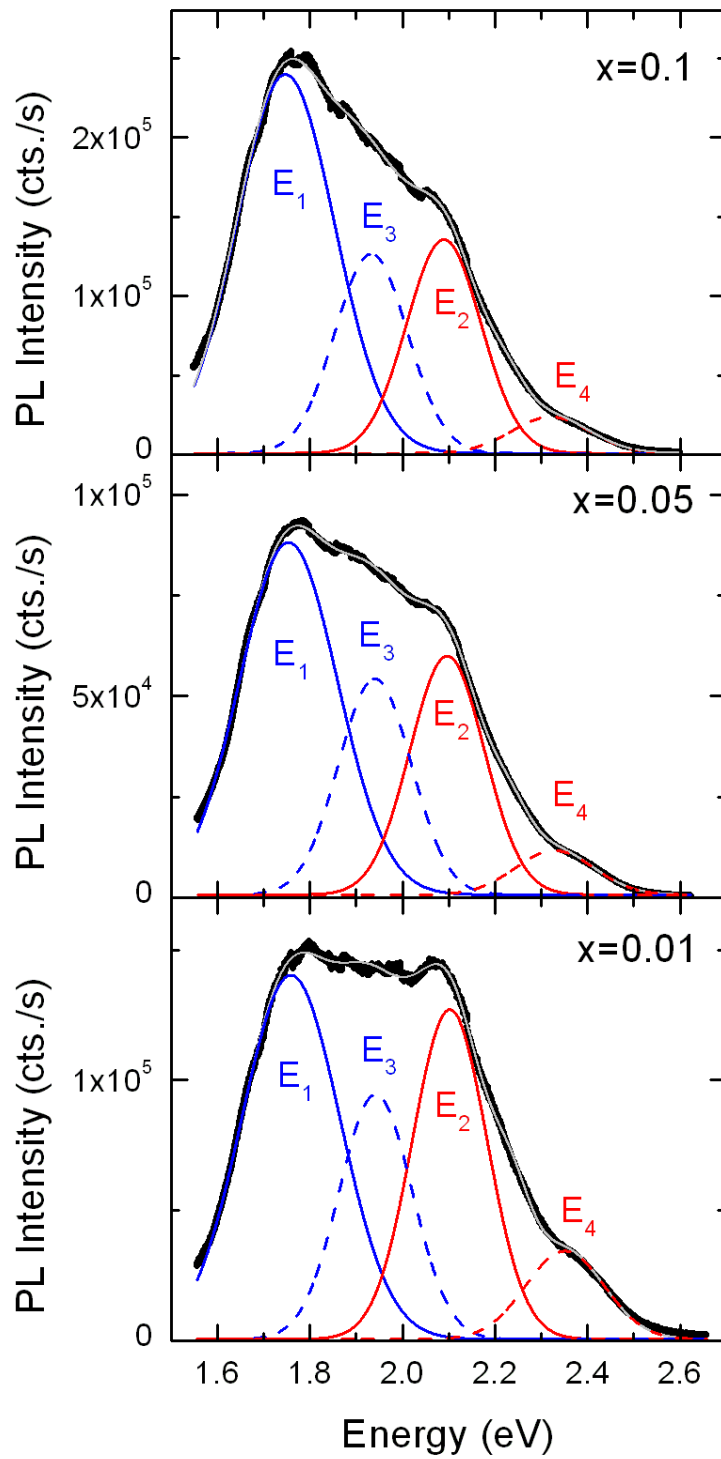


Figure S4. Deconvolution of emission spectra of $\text{LaSr}_{1-x}\text{Eu}_x\text{SiO}_3\text{N}$ under excitation at 405 nm. Bands corresponding to M1 and M2 sites are depicted with red and blue lines respectively.

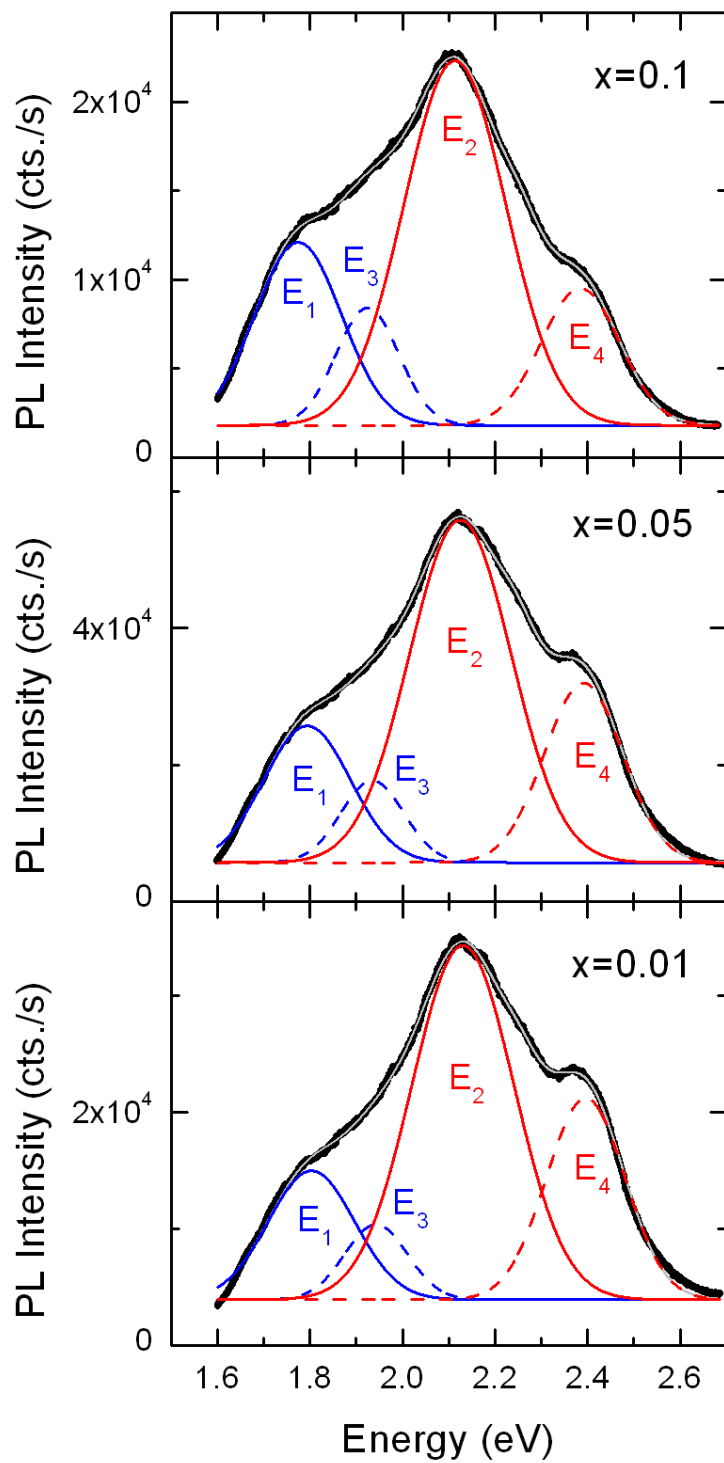


Figure S5. Deconvolution of emission spectra of $\text{La}_{1-x}\text{Ce}_x\text{SrSiO}_3\text{N}$ under excitation at 405 nm. Bands corresponding to M1 and M2 sites are depicted with red and blue lines respectively.

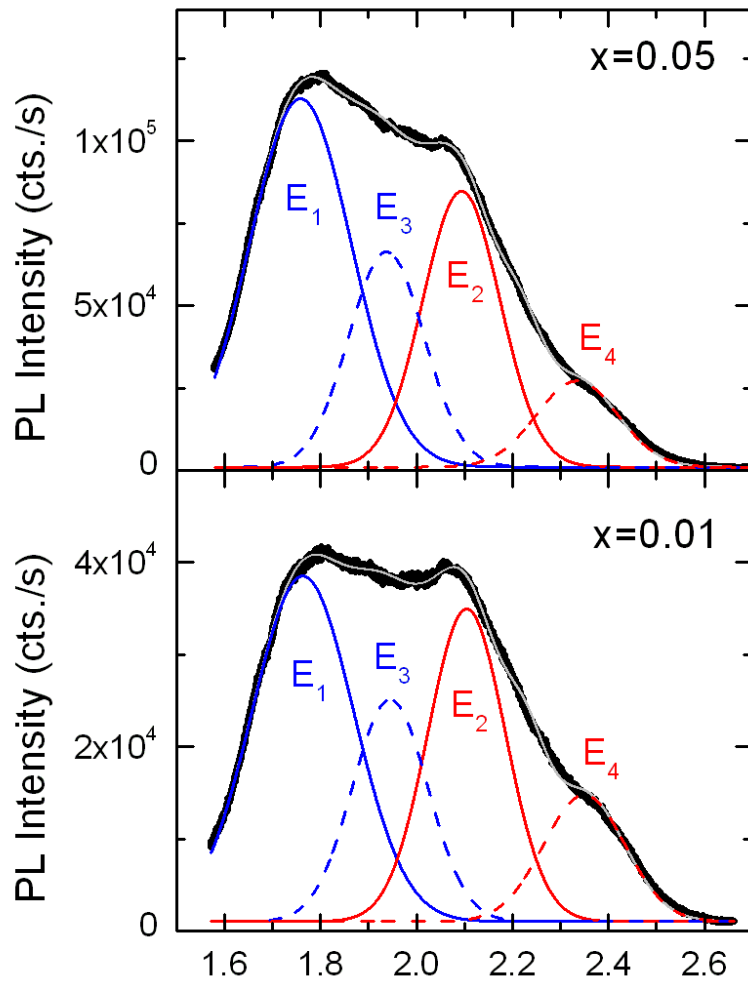


Figure S6. Deconvolution of emission spectra of $\text{LaBa}_{1-x}\text{Eu}_x\text{SiO}_3\text{N}$ under excitation at 405 nm. Bands corresponding to M1 and M2 sites are depicted with red and blue lines respectively.