

Table 1. Absorbance of BPDE adducts formed in MCF-7 cells at 346nm. Relation of absorbance (A), extinction coefficient ( $\epsilon$ ), concentration (c), and path length (l) is written with the following equation:  $A = \epsilon cl$ . ( $\epsilon = 29,000 \text{ M}^{-1} \text{ cm}^{-1}$ ,  $l = 1 \text{ cm}$ ).

	set	BPDE absorbance	Absorbance average	BPDE concentration
<b>Control</b>	<b>1</b>	0	0	0
	<b>2</b>	0		
	<b>3</b>	0		
<b>BP 0.2<math>\mu</math>M</b>	<b>1</b>	0.004	0.0033 $\pm$ 0.001	1.14 $\times$ 10 <sup>-7</sup> M
	<b>2</b>	0.0021		
	<b>3</b>	0.0037		
<b>BP 2<math>\mu</math>M</b>	<b>1</b>	0.0155	0.0125 $\pm$ 0.0034	4.31 $\times$ 10 <sup>-7</sup> M
	<b>2</b>	0.00889		
	<b>3</b>	0.0132		
<b>BP 5<math>\mu</math>M</b>	<b>1</b>	0.0321	0.0314 $\pm$ 0.0052	1.08 $\times$ 10 <sup>-6</sup> M
	<b>2</b>	0.0259		
	<b>3</b>	0.0363		
<b>BP 10<math>\mu</math>M</b>	<b>1</b>	0.0356	0.0359 $\pm$ 0.0023	1.24 $\times$ 10 <sup>-6</sup> M
	<b>2</b>	0.0338		
	<b>3</b>	0.0384		