

Supplementary information

Reversible Crystal Deformation of a Single-Crystal Host of Copper(II)1-Naphthoate—Pyrazine through Crystal Phase Transition Induced by Methanol Vapor Sorption

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Other supplementary material

Video for crystal shape change of **1** observed under a microscope: Takamizawa_mov1.mov

(a) Experimental information

(Preparation of crystal)

Introducing pyrazine vapor (0.26 g, 3.3 mmol) into the methanol solution (60 mL) of copper(II) acetate monohydrate (0.102 g, 0.509 mmol) and 1-naphthoic acid (0.859 g, 4.99 mmol) gave blue green single crystals of **1** with a size of 100-500 μm in 30.9% yield (70.1 mg).

(X-ray single crystal diffraction analysis)

Single-crystal X-ray structural analysis of **1** was performed at 90 K and 293 K on a Bruker Smart APEX CCD area diffractometer (Bruker AXS K.K.) with a nitrogen-flow temperature controller using graphite-monochromated Mo $K\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$). Empirical absorption corrections were applied using the SADABS program. The structure was solved by direct methods (SHELXS-97) and refined by full-matrix least-squares calculations on F^2 (SHELXL-97) using the SHELX-TL program package. Non-hydrogen atoms were refined anisotropically; hydrogen atoms were fixed at calculated positions by riding model approximation.

(Gas adsorption measurement)

CO₂ gas and MeOH vapor adsorption measurements were performed at 195 K and 293 K, respectively, on a Belsorp-max Measurement System (BEL JAPAN, Inc.) after drying the microcrystalline sample of **1** (13.80 mg for CO₂ gas adsorption and 24.70 mg for methanol vapor adsorption) at 373 K for 6 hours.

(b) Crystallographic data and ortep drawings

Table S1. Crystallographic data of **1**•0.93(MeOH) and **1**.

Crystal phase	1 •0.93(MeOH) (β)	1 (α)	1 (α)
Empirical formula	C _{48.93} H _{35.72} Cu ₂ N ₂ O _{8.93}	C ₄₈ H ₃₂ Cu ₂ N ₂ O ₈	C ₄₈ H ₃₂ Cu ₂ N ₂ O ₈
Crystal size /mm ³	0.50 × 0.40 × 0.10	0.50×0.30×0.10	0.78×0.12×0.10
M	921.69	891.84	891.84
Crystal system	Triclinic	Triclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
T /K	90	90	293
a /Å	10.7482(17)	10.4745(11)	10.6065(8)
b /Å	12.230(2)	11.6064(12)	11.6485(9)
c /Å	16.366(3)	17.0370(17)	17.4513(14)
α /deg	95.110(3)	97.583(2)	80.612(2)
β /deg	96.968(3)	104.918(2)	75.2150(10)
γ /deg	102.936(3)	90.081(2)	89.821(2)
V /Å ³	2066.2(3)	1982.5(4)	2055.1(3)
Z	2	2	2
D _{calcd} / Mg m ⁻³	1.482	1.494	1.441
μ (Mo K α) / mm ⁻¹	1.091	1.133	1.093
Reflections collected	10274	9616	10052
Independent reflections (<i>R</i> _{int})	8394(0.0306)	7440(0.0242)	6592(0.0259)
Goodness of fit	1.062	1.036	1.018
<i>R</i> ₁ (<i>I</i> > 2 σ (all data))	0.0580(0.0701)	0.0429(0.0623)	0.0509(0.0882)
<i>wR</i> ₂ (<i>I</i> > 2 σ (all data))	0.1598(0.1699)	0.1139(0.1264)	0.1365(0.1625)
Largest diff. peak (hole) /eÅ ³	1.468(-1.016)	0.952(-0.757)	0.677(-0.596)

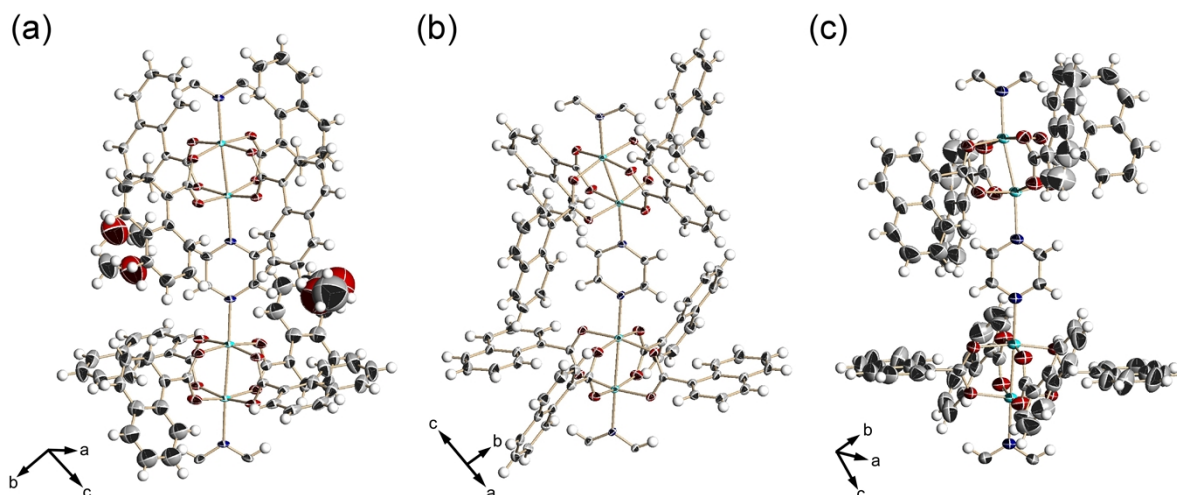


Figure S1. Molecular structures of (a) **1**•0.93(MeOH) at 90 K, (b) **1** at 90 K, and (c) **1** at 293 K as ortep view at 50% probability level for the each ellipsoid. The methanol molecules in **1**•0.93(MeOH) were disordered.

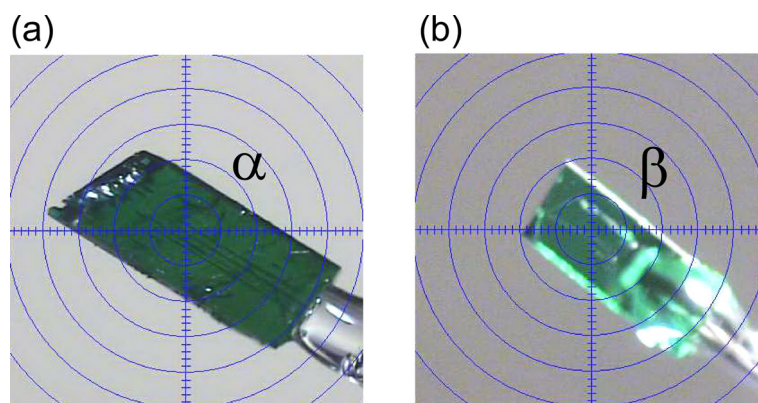


Figure S2. Single crystal of (a) **1** (α phase) and (b) **1**•0.93(MeOH) (β phase) used in single-crystal X-ray diffraction measurements at 90 K.

(c) Connections of the crystal structures at the α/β interface

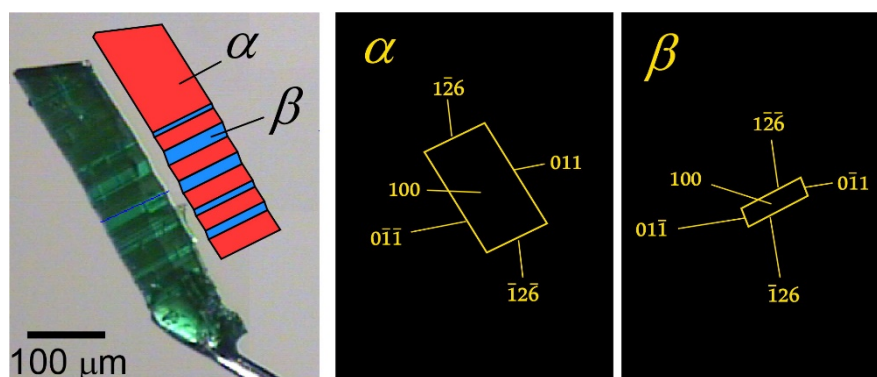


Figure S3. Crystal face indexing of **1** in α - β coexisting state at 293 K.

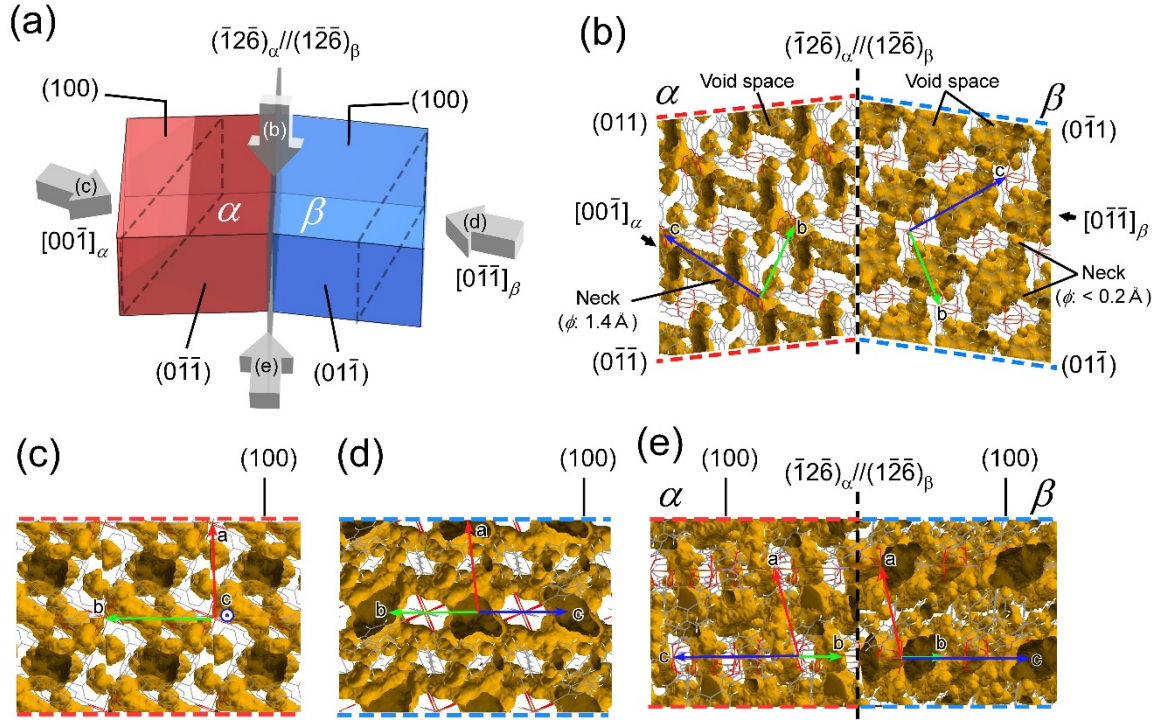


Figure S4. (a) Scheme for connections of α and β phases on the interface. (b-e) Pore structures viewed along the four arrows illustrated in Fig. S4a. (b) Cross section on $(100)_\alpha$ and $(100)_\beta$. (c), (d) Cross sections indicated as dotted squares in Fig. S4a, which are oriented perpendicular to $[00\bar{1}]_\alpha$ and $[0\bar{1}\bar{1}]_\beta$ respectively. (e) Cross section perpendicular to $(100)_\alpha$, $(100)_\beta$ and $(\bar{1}2\bar{6})_\alpha // (\bar{1}2\bar{6})_\beta$. Each void space in the α and β phase was illustrated by Mercury software with a probe radius of $0.7 \text{ \AA}$ and grid spacing of $0.2 \text{ \AA}$.