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Polymorphism and phase transformations in cobaltacarborane molecular crystals: Evidence for very weak intermolecular C–H···H–B interactions

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Contents

- 1) Table for bond lengths of both polymorphs of **1**.
- 2) Complete solid IR for forms I and II
- 3) Phase transformation by IR
- 4) Solution and gas phase IR for **1**.

	Form I	Form II
C1–C2	1.637(6)	1.630(3)
C1-B5	1.698(6)	1.692(3)
C1-B4	1.697(6)	1.719(3)
C1–B6	1.724(6)	1.721(3)
C1–Co1	1.991(4)	2.0092(18)
C2-B11	1.692(6)	1.691(3)
C2-B6	1.717(6)	1.718(3)
C2–B7	1.709(6)	1.723(3)
C2–Co1	2.005(4)	2.0047(18)
C3–C7	1.417(6)	1.402(4)
C3–C4	1.421(6)	1.409(3)
C3–Co1	2.059(4)	2.056(2)
C4–C5	1.415(7)	1.406(3)
C4–Co1	2.030(5)	2.036(2)
C5–C6	1.404(7)	1.422(4)
C5–Co1	2.028(5)	2.035(2)
C6–C7	1.384(6)	1.418(4)
C6–Co1	2.044(5)	2.046(2)
C7–Co1	2.055(4)	2.069(2)
B4–B9	1.792(6)	1.786(3)
B4–B5	1.800(6)	1.789(3)
B4–B8	1.803(6)	1.808(3)
B4–Co1	2.054(5)	2.076(2)
B5–B6	1.758(6)	1.771(3)
B5-B10	1.770(6)	1.779(3)
B5–B9	1.768(6)	1.784(3)
B6-B11	1.768(6)	1.763(3)
B6-B10	1.768(5)	1.769(3)
B7-B12	1.772(7)	1.778(3)
B7–B11	1.778(7)	1.793(3)
B7–B8	1.812(6)	1.817(3)
B7–Co1	2.065(5)	2.069(2)
B8–B9	1.784(6)	1.791(3)
B8-B12	1.783(6)	1.794(3)
B8-Co1	2.106(4)	2.106(2)
B9-B12	1.777(6)	1.786(3)
B9-B10	1.768(6)	1.790(3)
B10-B11	1.766(6)	1.773(3)
B10-B12	1.766(6)	1.786(3)
B11-B12	1.761(7)	1.781(3)

Table S1. Comparison of Bond Lengths (Å) for both polymorphs of 1

Figure S1. IR spectrum of form I.



Figure S2. IR spectrum of form II.



Figure S3. Phase transformation by IR.



Figure S4. Comparison between experimental (Bottom: 0.047M solution in CH_2Cl_2) and calculated in the gas phase (top) IR Spectra of *closo*-[3-Co(η^5 -C₅H₅)-1,2-C₂B₉H₁₁] (1). Only the relevant section with the C-H and B-H stretching bands are shown.

