



Figure S1. Variation of the interaction energy of the water dimer, methane-water and methane-fluoromethane complex and its components with the $\langle\text{O-H}\cdots\text{O}$, $\langle\text{C-H}\cdots\text{O}$ and $\langle\text{C-H}\cdots\text{F}$ angles. The following symbols are used to indicate each IMPT component: \circ for the electrostatic, Δ for the exchange-repulsion, ∇ for the polarization, \square for the dispersion one). The total interaction energy (\bullet , obtained by adding all components) and the sum of the electrostatic and dispersion components (\blacksquare) are also given for comparison.