

Electronic Supplementary Information

CdSnO₃ Micro-Cubes with Porous Architecture: Synthesis and Gas-Sensing Properties

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The thermal behavior of CdSnO₃·3H₂O was characterized by DSC-TG measurements in the air of N₂ during the temperature of 70~800 °C. As detailed in Fig. S1, in the process of thermal treatment, there is one strong exothermic peak at 270 °C, while a weak exothermic peak at about 100 °C, which could be caused by loss of crystal water and the slow release of absorption water, respectively. The strong loss of weight is about 15.20%, which was close to the calculated loss of 16.21% in theory. Furthermore, there is an endothermic peak at about 700 °C, which may be the result of the crystal form transition and can be further testified in the XRD patterns in Fig. S2. Fig. S2(a) was sintered at 500 °C by process temperature control, but with an amorphous nature. Fig. S2(b) was sintered at 700 °C by process temperature control and it can be indexed to rhombohedral CdSnO₃ (JCPDS 34-0758) with good crystallinity.

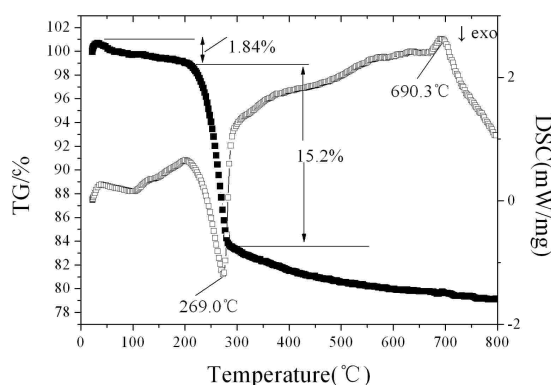


Fig. S1 DSC-TGA curves of CdSnO₃·3H₂O

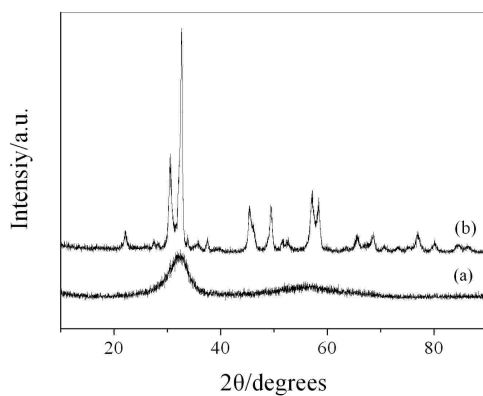


Fig. S2 XRD patterns of CdSnO₃ (a) sintered at 500 °C (b) sintered at 700 °C

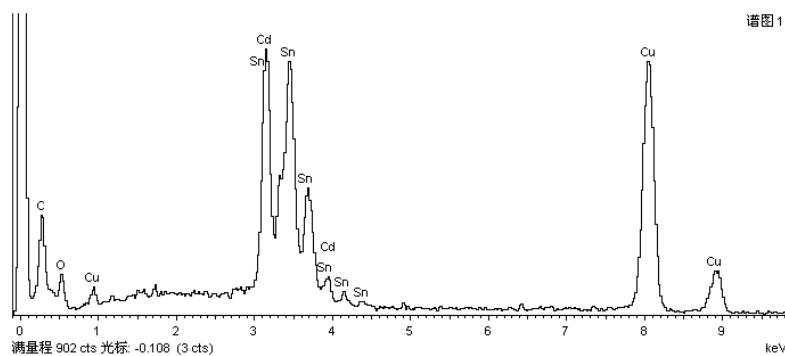


Fig. S3 The EDS of the as-synthesized CdSnO₃

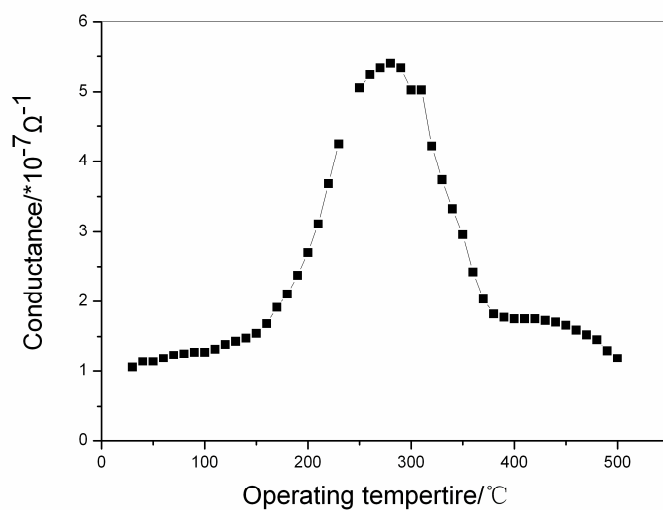


Fig. S4 The conductance-temperature behavior of ilmenite structure-type CdSnO₃