

Supporting Information

Syntheses, structures, photoluminescence and magnetic properties of five compounds with 1,3,5-benzenetricarboxylate acid and imidazole ligands

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Table S1 Selected Bond Distances (Å) and Angles (deg) for complexes

Complex 1			
Zn1—N1	2.041(3)	Zn1—N4_a	1.993(3)
Zn1—O1	1.952(3)	Zn1—O4_a	1.989(3)
O1—Zn1—O4_b	106.01(13)	O1—Zn1—N4_b	113.73(14)
O4_b—Zn1—N4_b	122.56(13)	O1—Zn1—N1	95.47(13)
O4_b—Zn1—N1	109.33(13)	N4_b—Zn1—N1	106.33(15)
C23—N4—Zn1_a	131.3(3)	C9—O1—Zn1	116.9(3)
C7—O4—Zn1_a	105.1(3)	C21—N4—Zn1_a	122.9(3)
Complex 2			
Zn1—N1	2.009(5)	Zn2—N4_a	2.033(4)
Zn2—N6_b	2.026(5)	Zn1—O1_c	1.992(4)
Zn2—O3_d	1.958(4)	Zn1—O6	1.946(4)
Zn1—O7	1.932(4)	Zn2—O7	1.938(4)
O7—Zn1—O6	106.01(17)	O7—Zn1—O1_b	107.54(17)
O6—Zn1—O1_b	107.94(17)	O7—Zn1—N1	109.04(18)
O6—Zn1—N1	122.53(19)	O1_b—Zn1—N1	103.02(17)
O7—Zn2—O3_e	129.36(18)	O7—Zn2—N6_c	114.79(18)
O3_e—Zn2—N6_c	102.15(19)	O7—Zn2—N4_a	104.87(18)
O3_e—Zn2—N4_a	101.04(19)	N6_c—Zn2—N4_a	100.21(18)
C9—N1—Zn1	124.1(4)	C11—N1—Zn1	126.4(4)
C18—N4—Zn2_a	122.1(4)	C19—N4—Zn2_a	131.1(4)
C23—N6—Zn2_b	126.5(4)	C22—N6—Zn2_b	125.3(4)
C7—O1—Zn1_c	119.1(3)	C24—O3—Zn2_d	109.6(4)
C8—O6—Zn1	117.0(4)	Zn1—O7—Zn2	130.2(2)
Complex 3			
Zn1—N1	2.042(5)	Zn2—N3	2.044(5)
Zn2—O1	2.167(4)	Zn1—O2	2.005(3)
Zn1—O3_a	2.315(4)	Zn1—O4_a	2.083(4)
Zn1—O6_b	2.063(4)	Zn2—O7	2.153(3)
Zn1—C27_a	2.526(6)		
O2—Zn1—N1	107.91(16)	O2—Zn1—O6_c	95.12(16)
N1—Zn1—O6_c	93.27(18)	O2—Zn1—O4_d	108.21(16)
N1—Zn1—O4_d	98.00(18)	O6_c—Zn1—O4_d	149.31(16)
O2—Zn1—O3_d	88.20(15)	N1—Zn1—O3_d	155.66(17)
O6_c—Zn1—O3_d	103.57(15)	O4_d—Zn1—O3_d	58.90(14)
O2—Zn1—C27_d	100.60(17)	N1—Zn1—C27_d	126.8(2)
O6_d—Zn1—C27_c	128.14(19)	O4_c—Zn1—C27_c	29.46(17)
O3_c—Zn1—C27_c	29.49(16)	N3_e—Zn2—N3	180.0

N3_e—Zn2—O7_e	88.72(16)	O7_e—Zn2—O7_e	180.0
N3_e—Zn2—O1	93.21(16)	O7_e—Zn2—O1	86.24(14)
O1—Zn2—O1_e	180.00(16)	O4—C27—Zn1_a	55.2(3)
O3—C27—Zn1_a	65.8(3)	C24—C27—Zn1_a	173.1(5)
C1—N1—Zn1	122.4(4)	C2—N1—Zn1	132.8(4)
C10—N3—Zn2	128.4(4)	C11—N3—Zn2	127.5(4)
C19—O1—Zn2	134.1(4)	C19—O2—Zn1	128.6(4)
C27—O3—Zn1_a	84.7(4)	C27—O4—Zn1_a	95.3(4)
C26—O6—Zn1_b	99.6(4)	Zn2—O7—H7B	109.1
Zn2—O7—H7C	109.1		

Complex 4

N1—Ni2	2.016(4)	Ni2—O5_c	2.082(3)
N3—Ni1	2.052(4)	Ni2—O3_b	2.193(3)
Ni1—O7_a	2.075(3)	Ni2—O6_c	2.208(3)
Ni1—O1	2.088(3)	Ni2—C5_a	2.448(5)
Ni2—O2	2.037(3)	Ni2—C8_c	2.459(5)
Ni2—O4_b	2.054(3)		
N3—Ni1—N3_a	180.000	O7—Ni1—O1	86.86(13)
O1—Ni1—O1_a	180.000(14)	O3_b—Ni2—O6_c	87.28(12)
N3—Ni1—O7	92.07(14)	N1—Ni2—O2	101.41(15)
O7_a—Ni1—O7	180.00(13)	N1—Ni2—O4_b	97.99(15)
N3—Ni1—O1	86.73(13)	O2—Ni2—O4_b	102.57(13)
N1—Ni2—O5_c	90.84(14)	O5_c—Ni2—O3_a	106.98(12)
O2—Ni2—O5_c	92.64(12)	N1—Ni2—O6_c	91.78(14)
O4_b—Ni2—O5_c	160.19(13)	O2—Ni2—O6_c	151.00(12)
N1—Ni2—O3_b	158.92(13)	O4_b—Ni2—O6_c	100.87(12)
O2—Ni2—O3_b	89.06(12)	O5_c—Ni2—O6_c	61.10(11)
O4_b—Ni2—O3_b	61.55(12)	N1—Ni2—C5_b	128.27(16)
O4—C5—Ni2_d	57.0(2)	O2—Ni2—C5_b	98.57(14)
O3—C5—Ni2_d	63.3(3)	O5_c—Ni2—C5_b	135.11(15)
C4—C5—Ni2_d	173.5(3)	O6_c—Ni2—C5_b	92.88(14)
O6—C8—Ni2_e	63.6(2)	N1—Ni2—C8_c	88.86(15)
O5—C8—Ni2_e	57.8(2)	O2—Ni2—C8_c	123.23(14)
C7—C8—Ni2_e	169.0(3)	O4_b—Ni2—C8_c	131.10(15)
C10—N1—Ni2	125.6(3)	C21—N3—Ni1	127.6(3)
C11—N1—Ni2	129.0(3)	C19—N3—Ni1	126.4(3)
O3_b—Ni2—C8_c	100.60(14)	C5—O3—Ni2_d	85.9(3)
O6_c—Ni2—C8_c	30.48(13)	C5—O4—Ni2_d	92.3(3)
C5_b—Ni2—C8_c	117.82(16)	C1—O2—Ni2	128.3(3)
C1—O1—Ni1	136.5(3)	Ni1—O7—H7A	119.8
Ni1—O7—H7	109.5		

Complex 5

Cu1—N1	1.953(3)	Cu1—O5	1.972(2)
Cu2—N3	1.951(3)	Cu2—O2_a	1.963(3)
Cu2—O3_b	2.024(3)	Cu2—O4_b	2.063(2)
Cu2—O6	2.208(2)	Cu2—C26_b	2.357(4)
N1—Cu1—N1_c	180.000(1)	N1—Cu1—O5	86.97(11)
O5—Cu1—O5_c	180.0	N1—Cu1—O7	92.02(15)
O5—Cu—O7	104.89(13)	N3—Cu2—O2_b	92.04(12)
N3—Cu2—O3_d	96.25(12)	O2_b—Cu2—O3_d	154.90(10)
N3—Cu2—O4_d	159.72(13)	O2_a—Cu2—O4_d	103.67(10)
O3_d—Cu2—O4_d	64.24(10)	N3—Cu2—O6	104.72(11)
O2_a—Cu2—O6	94.08(10)	O3_d—Cu2—O6	106.55(10)
O4_d—Cu2—O6	87.12(10)	N3—Cu2—C26_d	128.17(14)
O2_a—Cu2—C26_d	131.11(11)	O3_d—Cu2—C26_d	32.45(10)
O4_d—Cu2—C26_d	31.89(11)	O6—Cu2—C26_d	99.73(11)
C24—C26—Cu2_b	171.9(3)	C3—N1—C1	105.6(3)
C3—N1—Cu1	125.2(3)	C1—N1—Cu1	128.7(3)
C12—N3—Cu2	125.0(3)	C10—N3—Cu2	130.2(3)
C19—O2—Cu2_e	106.5(2)	C27—O5—Cu1	124.2(2)
C27—O6—Cu2	134.3(2)		

Symmetry transformations used to generate equivalent atoms:

a = $x - 1/2, -y + 3/2, z - 1/2$; b = $x + 1/2, -y + 3/2, z + 1/2$. (compound 1)

a = $-x + 2, -y + 1, -z + 1$; b = $-x + 2, -y + 1/2, -z + 3/2$; c = $-x + 2, -y - 1/2, -z + 3/2$;
d = $-x + 3, -y - 1/2, -z + 3/2$; e = $-x + 3, -y + 1/2, -z + 3/2$. (compound 2)

a = $x + 1, y, z$; b = $x + 1/2, -y + 1/2, z + 1/2$; c = $x - 1/2, -y + 1/2, z - 1/2$; d = $x - 1, y, z$;
e = $-x, -y, -z + 1$ (compound 3)

a = $-x, -y + 1, -z$; b = $x - 2, y, z$; c = $x - 1/2, -y + 1/2, z - 1/2$; d = $x + 2, y, z$; e = $x + 1/2,$
 $-y + 1/2, z + 1/2$. (compound 4)

a = $x - 1/2, -y + 1/2, z - 1/2$; b = $x + 1, y, z$; c = $-x + 1, -y + 1, -z$; d = $x - 1, -y + 1/2, z +$
 $1/2$; e = $x - 1/2, -y + 1/2, z + 1/2$. (compound 5)

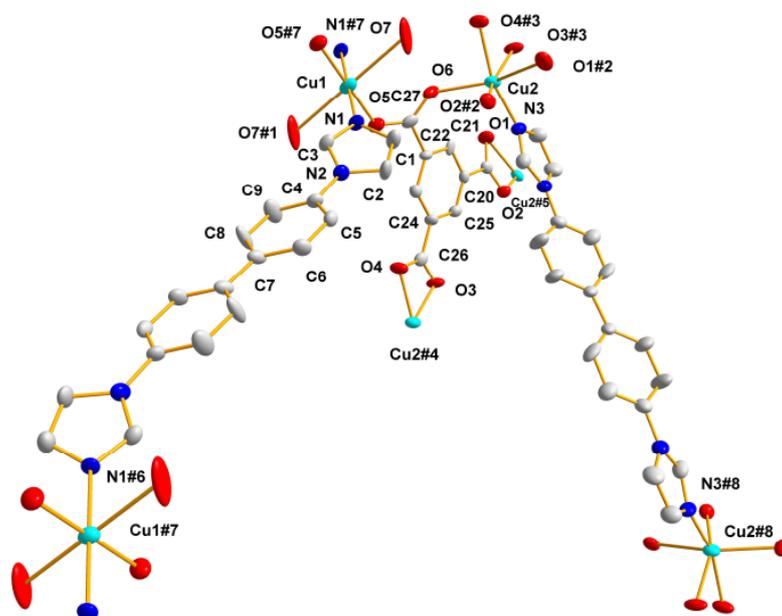


Figure S1. An ORTEP drawing of **5** showing 30% ellipsoid probability (hydrogen atoms are omitted for clarity). Symmetry codes: #1 = 1 - x, 1 - y, - z; #2 = -0.5 + x, 0.5 - y, -0.5 + z; #3 = -1 + x, y, z; #4 = 1 + x, y, z; #5 = 0.5 + x, 0.5 - y, 0.5 + z; #6 = 2 - x, 1 - y, - 1 - z; #7 = 1 + x, y, - 1 + z; #8 = 2 - x, -y, -z.

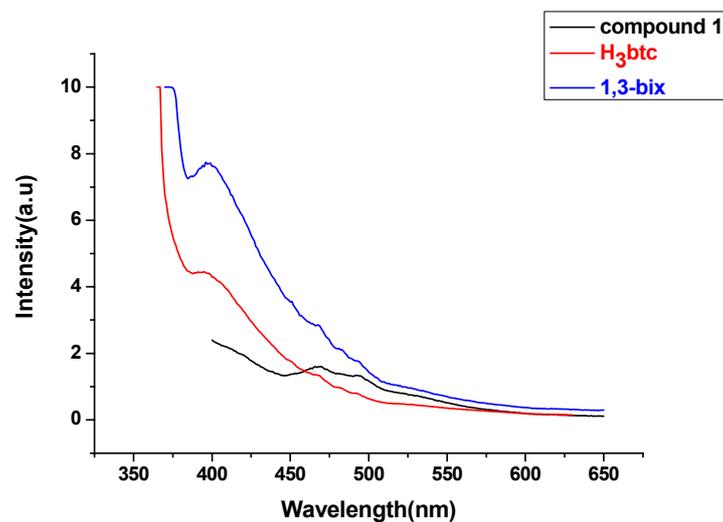


Figure S2. Emission spectra of free H₃btc, 1,3-bix and compound **1** in the solid state at room temperature.

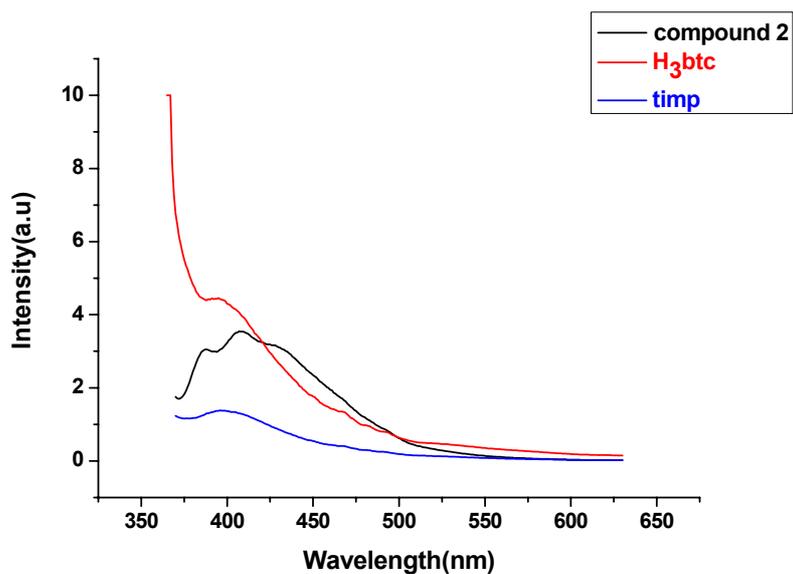


Figure S3. Emission spectra of free H₃btc,timp and compound **2** in the solid state at room temperature.

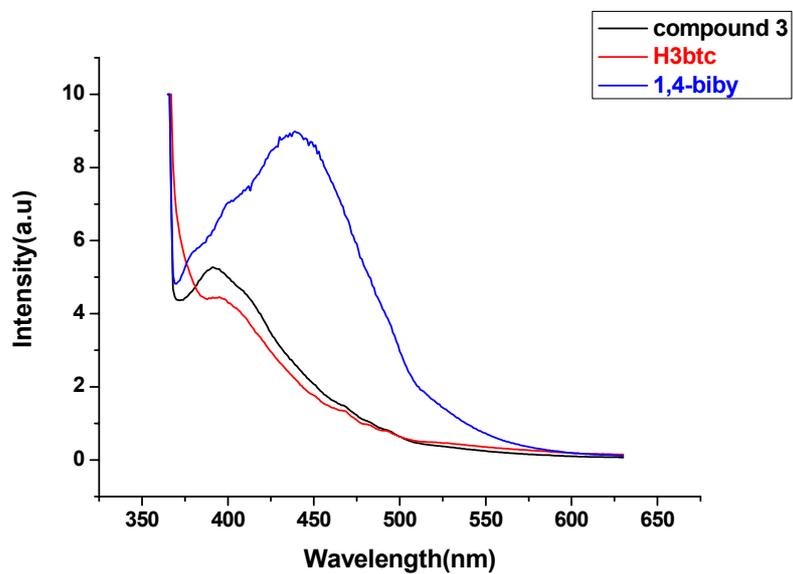


Figure S4. Emission spectra of free H₃btc, 1,4-biby and compound **3** in the solid state at room temperature.

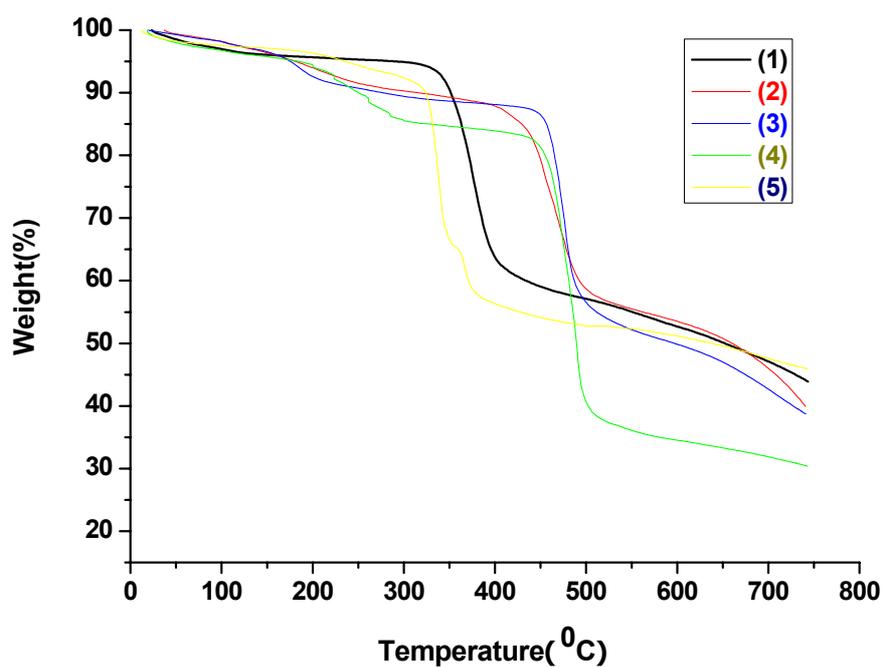


Figure S5. TGA plot of compounds 1 – 5.

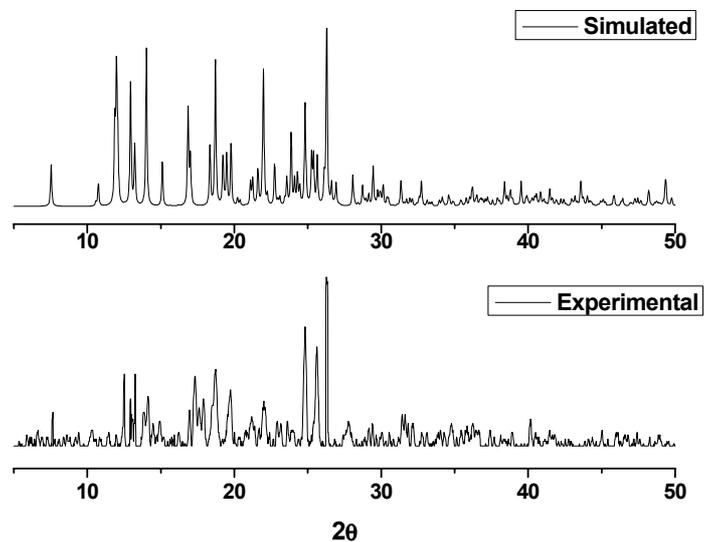


Figure S6. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound 1 at 293K.

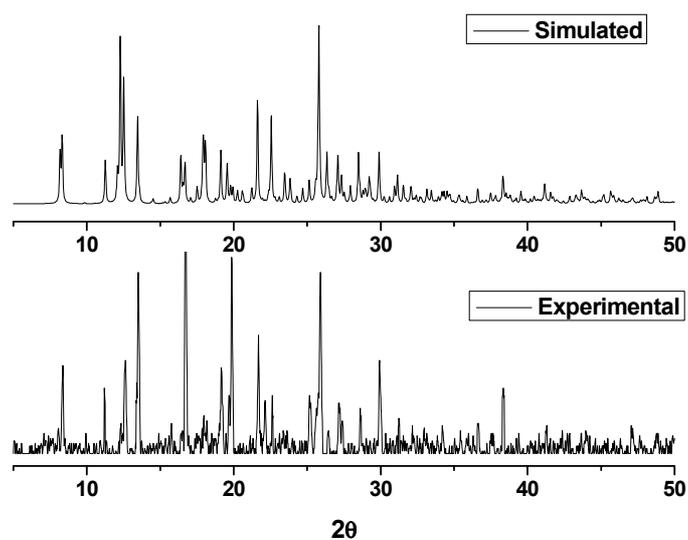


Figure S7. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **2** at 293K.

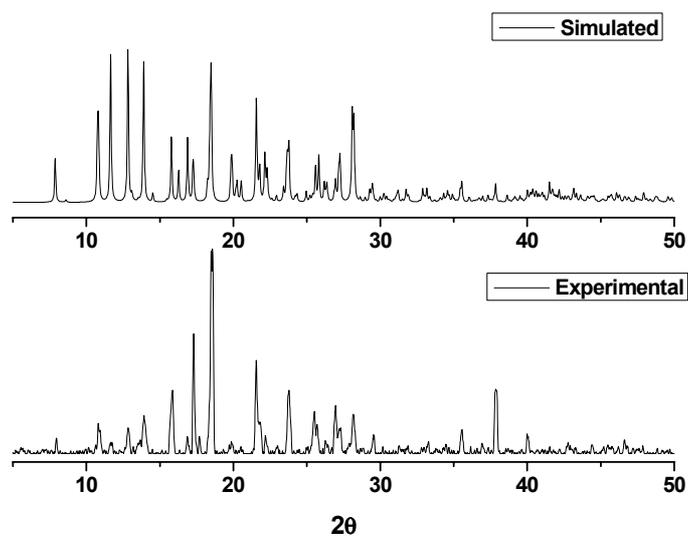


Figure S8. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **3** at 293K.

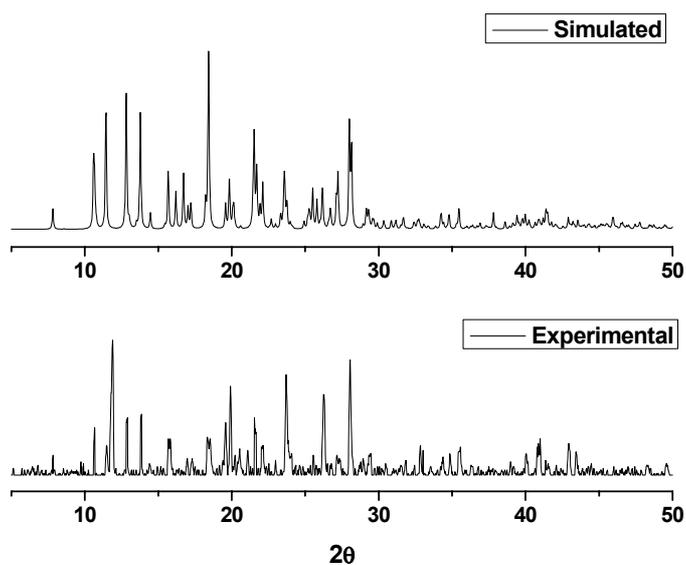


Figure S9. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **4** at 293K.

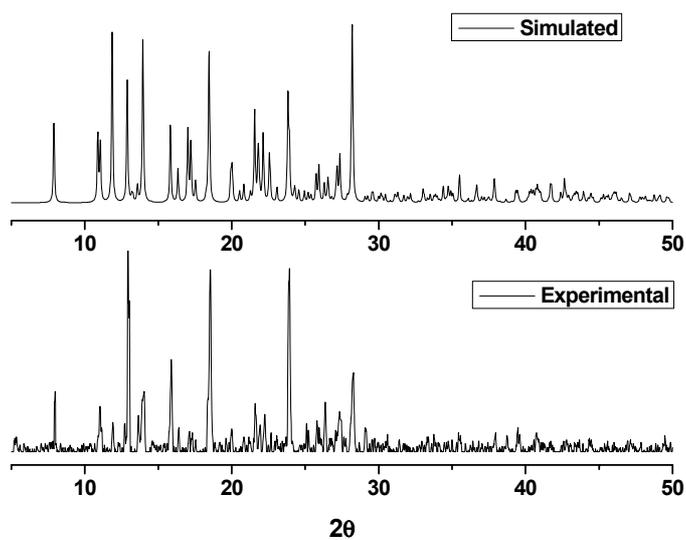


Figure S10. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **5** at 293K.