Supporting Information

Solvothermal synthesis, structures and physical properties of four new complexes constructed from multivariant tricarboxylate ligand and pyridyl-based ligands

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Complex 1					
Cd1-O1	2.252(5)	Cd1-O2	2.564(4)		
Cd1-O8	2.267(5)	Cd1-O3a	2.572(4)		
Cd1-O7	2.295(4)	Cd2-O5	2.219(5)		
Cd1-O4a	2.292(5)	Cd2-O10	2.309(5)		
Cd1-O9	2.314(5)	Cd2-O11	2.246(6)		
O1-Cd1-O8	138.92(17)	O9-Cd1-O2	97.11(16)		
O1-Cd1-O7	88.54(17)	O1-Cd1-O3a	136.16(15)		
O8-Cd1-O7	95.38(19)	O8-Cd1-O3a	84.79(15)		
O1-Cd1-O4a	84.03(16)	O7-Cd1-O3a	82.75(16)		
O8-Cd1-O4a	136.77(18)	O4a-Cd1-O3a	53.10(15)		
O7-Cd1-O4a	89.12(18)	O9-Cd1-O3a	90.46(16)		
O1-Cd1-O9	96.87(17)	O2-Cd1-O3a	167.20(14)		
O8-Cd1-O9	83.30(19)	O5b-Cd2-O5	171.4(3)		
O7-Cd1-O9	173.17(18)	O5-Cd2-O11	85.69(13)		
O4a-Cd1-O9	87.31(17)	O5b-Cd2-O10	107.25(18)		
O1-Cd1-O2	53.24(14)	O5-Cd2-O10	78.81(18)		
O8-Cd1-O2	85.86(16)	O11-Cd2-O10	133.00(15)		
O7-Cd1-O2	89.45(17)	O10-Cd2-O10b	94.0(3)		
O4a-Cd1-O2	137.27(16)				
Complex 2					
Cd1-O1	2.321(3)	Cd1-07	2.366(4)		
Cd1-O2	2.413(3)	Cd1-O8	2.333(4)		
Cd1-O5a	2.228(3)	Cd1-N1	2.214(4)		
N1-Cd1-O5a	134.03(14)	O1-Cd1-O7	86.51(14)		

Table S1 Selected Bond Lengths (Å) and Angles (deg) for Complexs 1-4

N1-Cd1-O1	143.40(14)	O8-Cd1-O7	176.39(12)			
O5a-Cd1-O1	81.97(12)	N1-Cd1-O2	89.97(13)			
N1-Cd1-O8	89.46(14)	O5a-Cd1-O2	135.48(11)			
O5a-Cd1-O8	91.85(12)	O1-Cd1-O2	53.60(11)			
O1-Cd1-O8	96.71(15)	O8-Cd1-O2	95.55(12)			
N1-Cd1-O7	86.97(14)	O7-Cd1-O2	85.02(11)			
O5a-Cd1-O7	90.22(12)					
Complex 3						
Mn1-O5a	2.118(3)	Mn1-N1	2.221(3)			
Mn1-O3	2.149(3)	Mn1-O5b	2.314(4)			
Mn1-O7	2.196(3)	Mn1-O4	2.369(3)			
O5a-Mn1-O3	144.66(11)	O7-Mn1-O5b	175.10(12)			
O5a-Mn1-O7	98.87(12)	N1-Mn1-O5b	88.85(13)			
O3-Mn1-O7	93.97(12)	O5a-Mn1-O4	87.95(11)			
O5a-Mn1-N1	120.58(13)	O3-Mn1-O4	57.91(10)			
O3-Mn1-N1	92.34(12)	O7-Mn1-O4	96.03(12)			
O7-Mn1-N1	88.69(13)	N1-Mn1-O4	150.08(12)			
O5a-Mn1-O5b	78.77(12)	O5b-Mn1-O4	88.21(12)			
O3-Mn1-O5b	90.37(12)					
Complex 4						
Co1-O3	2.012(3)	Co1-07	2.097(4)			
Co1-O5a	2.027(3)	Co1-O5b	2.298(3)			
Co1-N1	2.114(4)					
O3-Co1-O5a	139.33(13)	N1-Co1-O5b	90.95(14)			
O5a-Co1-N1	121.38(15)	O3-Co1-O5b	87.80(13)			
O3-Co1-N1	96.25(15)	O5a-Co1-O5b	77.53(14)			
O3-Co1-O7	94.68(13)	N1-Co1-O7	90.89(15)			

O5a-Co1-O7	99.20(14)	O7-Co1-O5b	176.73(13)
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Symmetry codes: for **1**: (a) *x*+1/2, *y*-1/2, *z*; (b) -*x*, *y*, -*z*; for **2**: (a) *x*, *y*+1, *z*; for **3**: (a) *x*, *y*+1, *z*; (b) -*x*+1, -*y*+1, -*z*+1; for **4**: (a) *x*, *y*+1, *z*; (b) -*x*+1, -*y*+1, -*z*+1.



Fig. S1 Coordination environment of the Co(II) ion in **4** with 30% ellipsoid probability. The hydrogen atoms are omitted for clarity. Symmetry codes: #1 = 1-x, 1-*y*, 1-*z*; #2 = x, 1+*y*, *z*; #3 = 1-x, 2-*y*, 2-*z*; #4 = 1-x, 1-*y*, 2-*z*; #5 = x, 1+*y*, 1+*z*; #6 = x, -1+*y*, *z*. Dashed lines mean weak interactions.



Fig. S2 View of a 1D ribbonlike chain of Co atoms in **4** bridged by HL^{2-} ligands. The hydrogen atoms and bipy ligands are omitted for clarity. Dashed lines mean weak interactions.



Fig. S3 The 2D layer formed with ribbonlike chains pillared by rigid bipy ligands of **4**. The hydrogen atoms and benzoyloxy groups are omitted for clarity.



Fig. S4 Solid-state photoluminescent spectra of $1, 2, H_3L$ ligand and timp ligand at room temperature.

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Fig. S5 The TGA diagrams of complexes 1-4



Fig. S6 Powder X-ray diffraction patterns of complex 1



Fig. S7 Powder X-ray diffraction patterns of complex 2



Fig. S8 Powder X-ray diffraction patterns of complex 3



Fig. S9 Powder X-ray diffraction patterns of complex 4