Electronic Supplementary Information (ESI)

How do substituent groups in the 5-position of 1,3-benzenedicarboxylate affect the construction of supramolecular frameworks? [†]

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[Zn(H	IBT	C)(1,4-bpe	eb)] _n ((2);	(c) {	[Zn(5-]	NO ₂ -1,3-B	DC)(1,4	-bpeb)]∙	$2(H_2O)_{0.5}$;} _n	(3);	(d)
[Zn(5	-Me	e-1,3-BDC)(1,4-bp	eb)] _n	(4).	Black:	simulated	d from	single	crystal	ana	lysis	and
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Fig. S1 Powder XRD patterns of (a) $[Zn(OH_2)(5-HO-1,3-BDC)(1,4-bpeb)]_n$ (1); (b) $[Zn(HBTC)(1,4-bpeb)]_n$ (2); (c) $\{[Zn(5-NO_2-1,3-BDC)(1,4-bpeb)]\cdot 2(H_2O)_{0.5}\}_n$ (3); (d) $[Zn(5-Me-1,3-BDC)(1,4-bpeb)]_n$ (4). Black: simulated from single crystal analysis and experimental; Red: as synthesized.



Fig. S2 (a) The O1–H2W…N2 H-bonding interaction in **1.** (b) The O1–H1W…O5 H-bonding interaction in **1.** (c) The O6–H6A…O3 H-bonding interaction in **1.**

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Fig. S3 The O5–H5A…N2 H-bonding interaction in **2.**



Fig. S4 The O7–H1W…O5 and O8–H2W…O6 H-bonding interactions in 3.



Fig. S5 Emission spectra of 1–4 in the solid state at ambient temperature.



Fig. S6 The TGA curves for 1 (black), 2 (red), 3 (blue) and 4 (purple).

The thermal properties of **1-4** were described as follows. The thermogravimetric analyses revealed that **1-4** were stable up to 202 °C (**1**), 415°C (**2**), 193°C (**3**), and 400 °C (**4**) (Fig. S5†). For **1** and **3**, the first weight loss of 3.56% from 202 to 258 °C (**1**) or 3.51% from 193 to 265 °C (**3**) corresponds roughly to the loss of one coordinated water molecule per formula unit in **1** (calculated 3.29%) or two one-half of the uncoordinated water molecules per formula unit in **3** (calculated 3.12%). The second weight loss of 81.30% from 258 to 820 °C (**1**) or 82.62% from 265 to 830 °C (**3**) approximately amounts to the loss of all 1,4-bpeb and 5-HO-1,3-BDC or 5-NO₂-1,3-BDC ligands (calculated 81.85% for **1** and 82.77% for **3**). For **2** and **4**, only one weight loss of 1,4-bpeb and H₃BTC or 5-Me-1,3-BDC ligands (calculated 85.41% for **2** and 84.58% for **4**). In all cases, the decomposition residue species, according to X-ray fluorescence analysis, was assumed to be ZnO (15.14% *vs* calculated 14.86% (**1**), 14.62% *vs* calculated 14.59% (**2**), 13.87% *vs* calculated 14.11% (**3**), and 15.31% *vs* calculated 15.42% (**4**)).