

## Supporting Information

### Syntheses, structures, photoluminescence, and magnetic properties of four new metal-organic frameworks based on imidazole ligands and aromatic polycarboxylate acids

**Table S1 Selected Bond Distances (Å) and Angles (deg) for complexes**

Complex 1			
Cu1—O2	1.989(2)	Cu1—N1	1.991(3)
Cu1—O1_a	1.999(3)	Cu1—N5_b	1.999(3)
Cu1—O3	2.349(3)	Cu1—O6_c	2.394(3)
Cu2—N10_b	1.957(3)	Cu2—N6	1.979(4)
Cu2—O4	1.985(3)	Cu2—O5	1.988(3)
O2—Cu1—N1	87.64(13)	O2—Cu1—O1_a	179.29(11)
N1—Cu1—O1_a	91.66(13)	O2—Cu1—N5_b	89.69(13)
N1—Cu1—N5_b	176.46(14)	O1_a—Cu1—N5_b	91.02(13)
O2—Cu1—O3	91.10(11)	N1—Cu1—O3	90.19(13)
O1—Cu1—O3_a	88.82(11)	N5—Cu1—O3_b	92.19(13)
O2—Cu1—O6_c	95.55(11)	N1—Cu1—O6_c	89.66(12)
O1_a—Cu1—O6_c	84.53(11)	N5_b—Cu1—O6_c	88.27(12)
O3—Cu1—O6_c	173.34(10)	N10—Cu2—N6_b	171.00(15)
N10—Cu2—O4	93.05(14)	N6—Cu2—O4	90.18(13)
N10—Cu2—O5_b	87.80(14)	N6—Cu2—O5	90.15(14)
O4—Cu2—O5	172.13(12)	N10_b—Cu2—O4_d	94.48(13)
N6—Cu2—O4_d	94.41(12)	O4—Cu2—O4_d	78.13(11)
O5—Cu2—O4_d	94.00(11)	C3—N1—Cu1	124.8(3)
C1—N1—Cu1	128.9(3)	N11—O1—Cu1	124.3(2)
N11—O2—Cu1_c	122.6(2)	N12—O3—Cu1	126.5(2)
N12—O4—Cu2	117.2(2)	Cu2—O4—Cu2_d	101.87(11)
Complex 2			
N1—Cd1	2.222(3)	Cd1—O2	2.205(2)
O2—Cd1—N1	111.88(9)	O2—Cd1—N1	111.88(9)
N1—Cd1—N1_a	94.28(14)	C3—N1—Cd1	122.8(2)
C1—N1—Cd1	130.7(2)	C10—O2—Cd1	102.32(19)
O2—Cd1—O2_a	88.78(11)		
Complex 3			
Co1—O1	2.078(3)	Co1—N3_a	2.150(4)

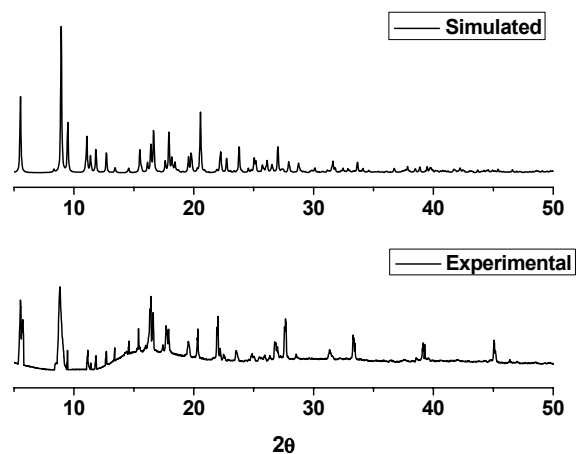
Co1—N1	2.155(4)	Co2—N5	2.091(4)
Co2—O3	2.123(3)	Co2—O5	2.124(4)
O1—Co1—O1_b	180.00	O1—Co1—N3_a	93.02(15)
N3_a—Co1—N3_c	180.00(19)	O1—Co1—N1	92.62(15)
N3—Co1—N1_a	96.85(17)	N1—Co1—N1_b	180.0(19)
N5—Co2—N5_d	180.000	N5—Co2—O3	93.41(14)
O3—Co2—O3_d	180.000(1)	O3—Co2—O5	93.41(14)
N5—Co2—O5_d	90.44(16)	O5—Co2—O5	180.0
C18—N1—Co1	122.3(4)	C19—N1—Co1	129.4(4)
C29—N3—Co1_e	126.5(4)	C27—N3—Co1_e	128.1(4)
C9—N5—Co2	122.7(4)	C11—N5—Co2	132.5(4)
C1—O1—Co1	138.5(4)	C8—O3—Co2	126.4(4)
<b>Complex 4</b>			
Zn1—N1	1.991(2)	Zn1—N3	1.986(3)
Zn1—O3_a	2.0080(2)	Zn1—O6	1.970(2)
O6- Zn1 -N3	109.03(12)	O6- Zn1- N1	110.33(12)
N3 -Zn1- N1	117.12(12)	O6- Zn1 -O3_b	102.79(11)
N3- Zn1- O3_b	105.02(12)	N1 -Zn1 -O3_b	111.48(12)
C3- Zn1-N1	127.9(3)	C1-Zn1-N1	126.4(3)
C15- Zn1-O6	110.1(2)	C5- Zn1-N3	126.0(3)
C23 -Zn1-O3_a	107.8(2)		

$a = -x + 1, y - 1/2, -z + 3/2$ ;  $b = x + 1, y, z$ ;  $c = -x + 1, y + 1/2, -z + 3/2$ ;  $d = -x + 1, -y + 1, -z + 2$  ( compound 1)

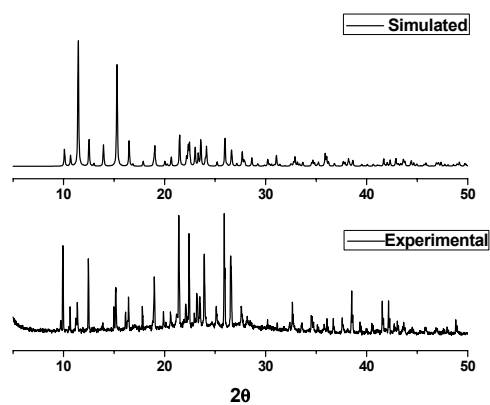
$a = x, -y + 3/2, -z + 1/2$ . (compound 2)

$a = x + 1, y - 2, z$ ;  $b = -x + 1, -y, -z + 1$ ;  $c = -x, -y + 2, z + 1$ ;  $d = -x + 1, -y + 2, -z$ ;  $e = x - 1, y + 2, z$ .( compound 3)

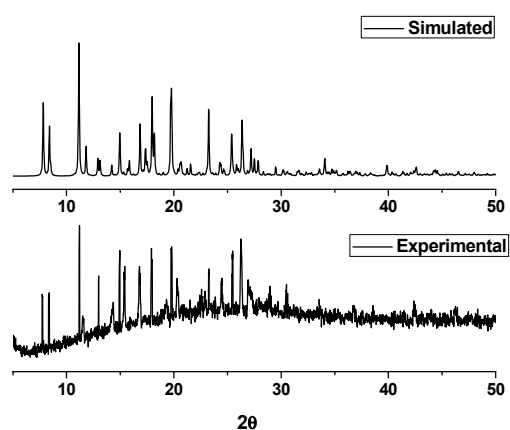
$a = x - 1, y, z$ ;  $b = x + 1, y, z$ .( compound 4)



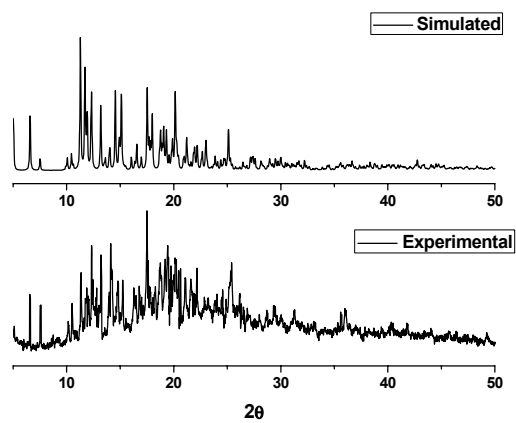
**Figure S1.** Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **1** at 293K.



**Figure S2.** Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **2** at 293K.



**Figure S3.** Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **3** at 293K.



**Figure S4.** Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of compound **4** at 293K.