SUPPORTING INFORMATION

Schiff base derived from 2-hydroxy-1-naphthaldehyde and liquid-assisted mechanochemical synthesis of its isostructural Cu(II) and Co(II) complexes

Dominik Cinčić* and Branko Kaitner

Laboratory of General and Inorganic Chemistry, Department of Chemistry, Faculty of Science, University of Zagreb, Horvatovac 102a, HR-10000 Zagreb, Croatia,

Email: dominik@chem.pmf.hr Fax: +385 1 4606 341 Tel: +385 1 4606 362

Table of Contents

Figure S1.	IR spectrum for compound 1	3
Figure S2.	IR spectrum for compound 2 synthesised by solution-based method	3
Figure S3.	IR spectrum for compound 2 synthesised via grinding	3
Figure S4.	IR spectrum for compound 3 synthesised by solution-based method	4
Figure S5.	IR spectrum for compound 3 synthesised via grinding	4
Figure S6.	IR spectrum for the product of grinding of $Co(II)$ acetate and compound 1, i.e. the acetic acid solvate of compound 3	4
Figure S7.	IR spectra for compound 2 synthesised by solution-based method (black) and via grinding (red)	5
Figure S8.	IR spectra for compound 3 synthesised by solution-based method (black) and via grinding (red)	5
Figure S9.	IR spectra for the product of grinding of $Co(II)$ acetate and compound 1, i.e. the acetic acid solvate of compound 3 (red) and compound 3 after three days standing in acetic acid vapour (black)	5
Figure S10.	DSC curve for compound 1	6
Figure S11.	TG curve for compound 1	6
Figure S12.	DSC curve of nonisothermal crystallisation experiment for compound 1	7
Figure S13.	PXRD patterns for compound 1 before (red) and after (black) experiment of nonisothermal crystallisation	7
Figure S14.	TG (black) and DTA (red) curves for compound 2 synthesised by solution- -based method	8
Figure S15. Figure S16.	TG (black) and DTA (red) curves for compound 2 synthesised via grinding TG (black) and DTA (red) curves for compound 3 synthesised by solution-	8
	-based method	9
Figure S17.	TG (black) and DTA (red) curves for compound 3 synthesised via grinding	9
Figure S18.	TG (black) and DTA (red) curves for the product of grinding of Co(II) acetate and compound 1, i.e. the acetic acid solvate of compound 3	10
Figure S19.	TG curve of isothermal experiment at 160 °C for the product of grinding of Co(II) acetate and compound 1 , i.e. the acetic acid solvate of compound 3 .	10
Figure S20.	PXRD patterns of: a) pure 2-hydroxy-1-naphtaldehyde, b) compand 1 and c)	
	simulated pattern for compound 1	11
Figure S21.	PXRD patterns for mechanochemical and solution-based experiments involving copper(II) acetate monohydrate and the compound 1: a)	

compound 1, b) copper acetate monohydrate, c) compound 2 obtained by grinding of Cu(II) acetate monohydrate and compound 1 in the presence of 30 μ L triethylamine, d) compound 2 obtained by solution-based method, e) compound 2 after three days standing in acetic acid vapour and f) simulated pattern for compound 2

- Figure S22. PXRD patterns for mechanochemical and solution-based experiments involving cobalt(II) acetate tetrahydrate and compound 1: a) compound 1, b) cobalt(II) acetate tetrahydrate, c) LAG product obtained by grinding of Co(II) acetate tetrahydrate and compound 1 in the presence of 40 μL triethylamine (the acetic acid solvate of compound 3 is a major product and compound 3 is in traces), d) LAG product obtained by grinding of Co(II) acetate tetrahydrate and compound 1 in the presence of 50 μL triethylamine (the acetic acid solvate of compound 3 is a major product and compound 3 is in traces), e) compound 3 is a major product and compound 3 is in traces), e) compound 3 obtained by annealing of LAG product, f) compound 3 obtained by solution-based method, g) simulated pattern for compound 3 and h) compound 3 after three days standing in acetic acid vapour
- **Figure S23.** PXRD patterns for mechanochemical and solution-based experiments involving copper(II) acetate monohydrate and the compound **1**: a) compound **1**, b) copper acetate monohydrate, c) product obtained by neat grinding, d) LAG product obtained by grinding in the presence of 30 μL acetonitrile (compound **2**), e) LAG product obtained by grinding in the presence of 30 μL methanol (compound **2**), f) LAG product obtained by grinding in the presence of 30 μL triethylamine (compound **2**) and g) compound **2** obtained by solution-based method.
- **Figure S24.** PXRD patterns for mechanochemical experiments involving cobalt(II) acetate tetrahydrate and 1: a) compound 1, b) cobalt(II) acetate tetrahydrate, c) product obtained by neat grinding, d) LAG product obtained by grinding in the presence of 30 μ L acetonitrile (acetic acid solvate of compound 3 is a major product and cobalt(II) acetate tetrahydrate is in traces), e) LAG product obtained by grinding in the presence of 30 μ L methanol (acetic acid solvate of compound 3), f) LAG product obtained by grinding in the presence of 30 μ L methanol (acetic acid solvate of compound 3), f) LAG product obtained by grinding in the presence of 30 μ L triethylamine (acetic acid solvate of compound 3 is a major product and compound 3 is in traces), g) compound 3 obtained by solution-based method and h) compound 3 after three days standing in acetic acid vapour.

12

13

11

14

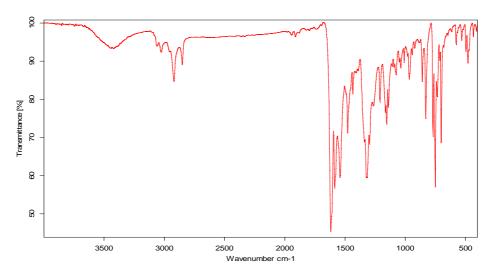


Figure S1. IR spectrum for compound 1.

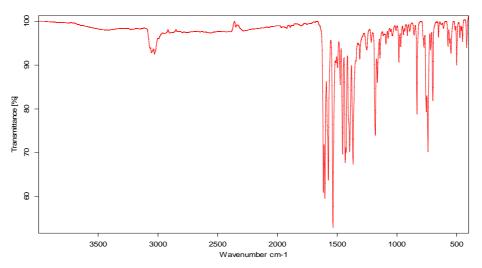


Figure S2. IR spectrum for compound 2 synthesised by solution-based method.

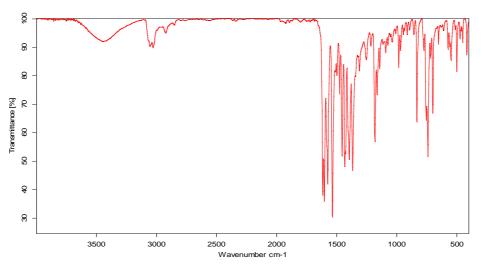


Figure S3. IR spectrum for compound 2 synthesised via grinding.

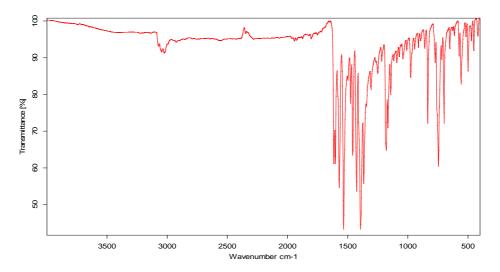


Figure S4. IR spectrum for compound 3 synthesised by solution-based method.

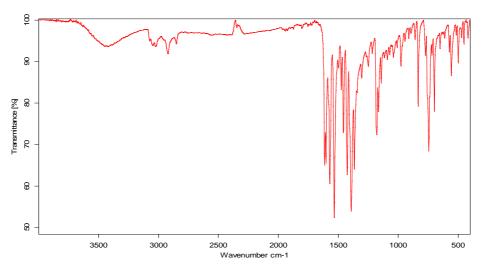


Figure S5. IR spectrum for compound 3 synthesised via grinding.

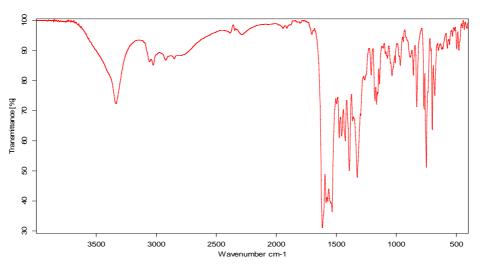


Figure S6. IR spectrum for the product of grinding of Co(II) acetate and compound **1**, i.e. the acetic acid solvate of compound **3**.

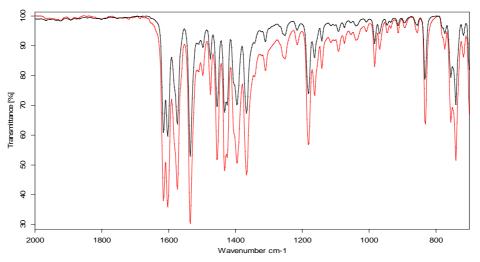


Figure S7. IR spectra for compound 2 synthesised by solution-based method (black) and via grinding (red).

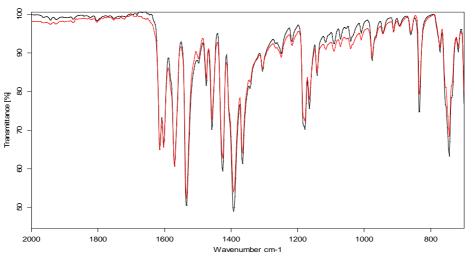


Figure S8. IR spectra for compound 3 synthesised by solution-based method (black) and via grinding (red).

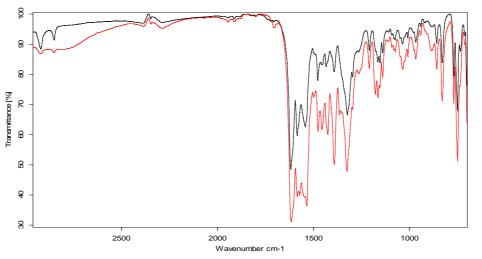


Figure S9. IR spectra for the product of grinding of Co(II) acetate and compound **1**, i.e. the acetic acid solvate of compound **3** (red) and compound **3** after three days standing in acetic acid vapour (black).

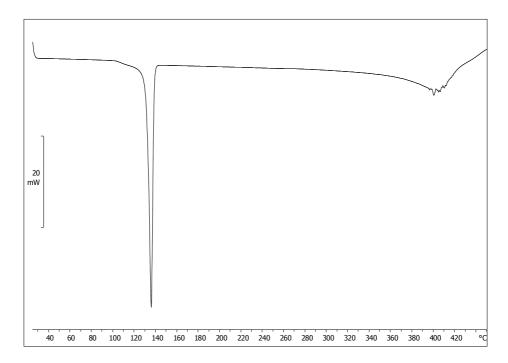


Figure S10. DSC curve for compound 1.

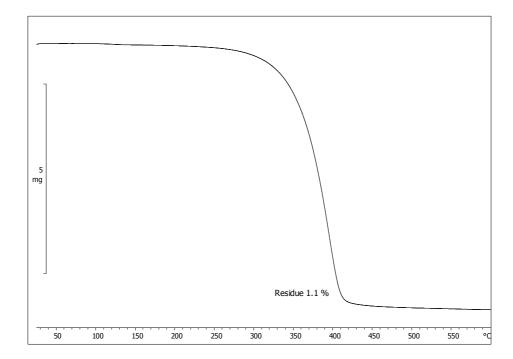


Figure S11. TG curve for compound 1.

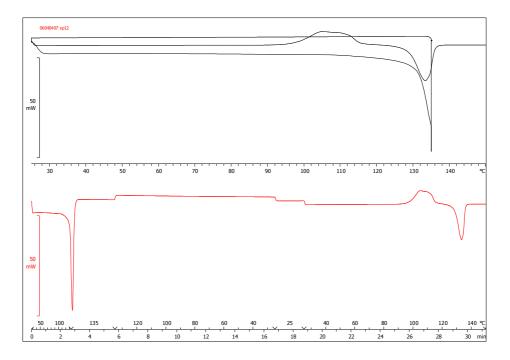


Figure S12. DSC curve of nonisothermal crystallisation experiment for compound 1.

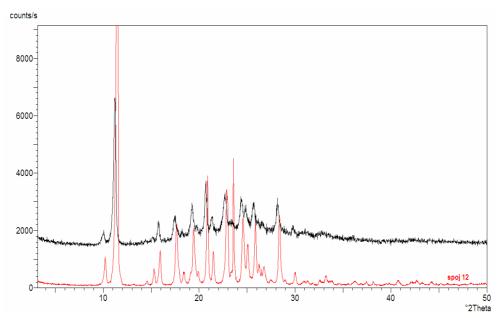


Figure S13. PXRD patterns for compound 1 before (red) and after (black) experiment of nonisothermal crystallisation.

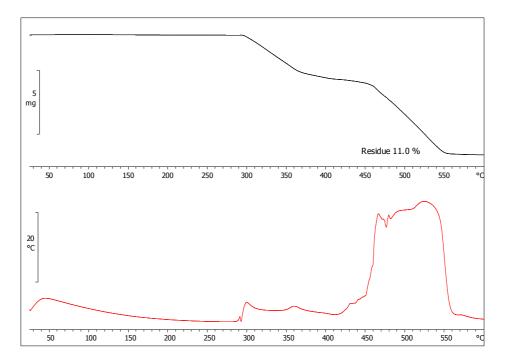


Figure S14. TG (black) and DTA (red) curves for compound 2 synthesised by solution-based method.

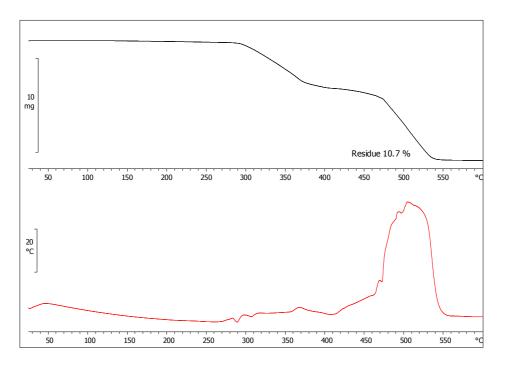


Figure S15. TG (black) and DTA (red) curves for compound 2 synthesised via grinding.

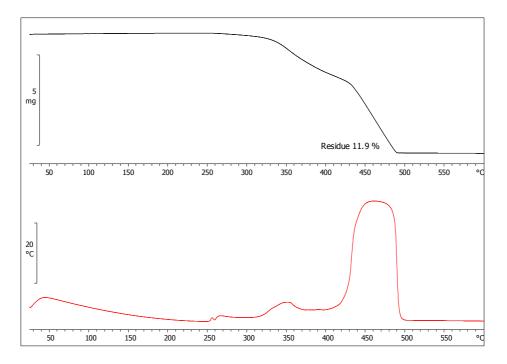


Figure S16. TG (black) and DTA (red) curves for compound 3 synthesised by solution-based method.

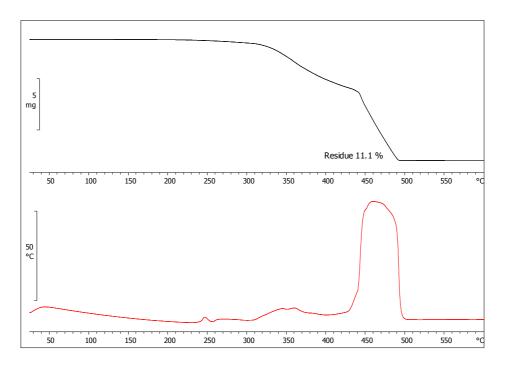


Figure S17. TG (black) and DTA (red) curves for compound 3 synthesised via grinding.

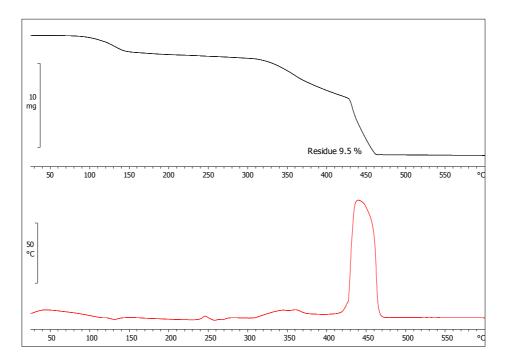


Figure S18. TG (black) and DTA (red) curves for the product of grinding of Co(II) acetate and compound **1**, i.e. the acetic acid solvate of compound **3**.

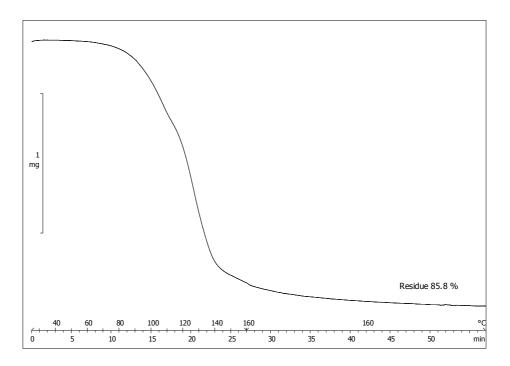


Figure S19. TG curve of isothermal experiment at 160 °C for the product of grinding of Co(II) acetate and compound **1**, i.e. the acetic acid solvate of compound **3**.

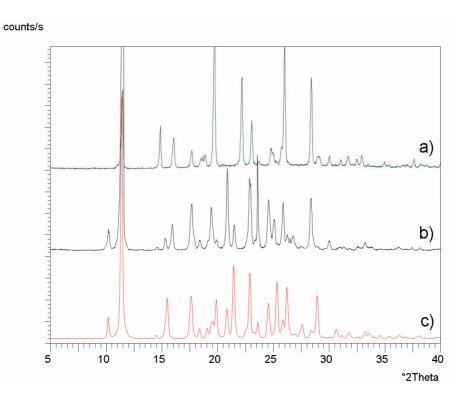


Figure S20. PXRD patterns of: a) pure 2-hydroxy-1-naphtaldehyde, b) compound 1 and c) simulated pattern for compound 1.

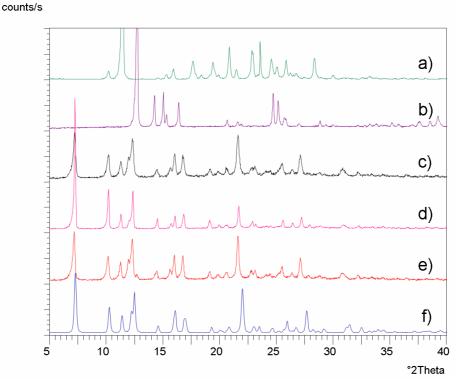


Figure S21. PXRD patterns for mechanochemical and solution-based experiments involving copper(II) acetate monohydrate and the compound 1: a) compound 1, b) copper acetate monohydrate, c) compound 2 obtained by grinding of Cu(II) acetate monohydrate and compound 1 in the presence of 30 μ L triethylamine, d) compound 2 obtained by solution-based method, e) compound 2 after three days standing in acetic acid vapour and f) simulated pattern for compound 2.

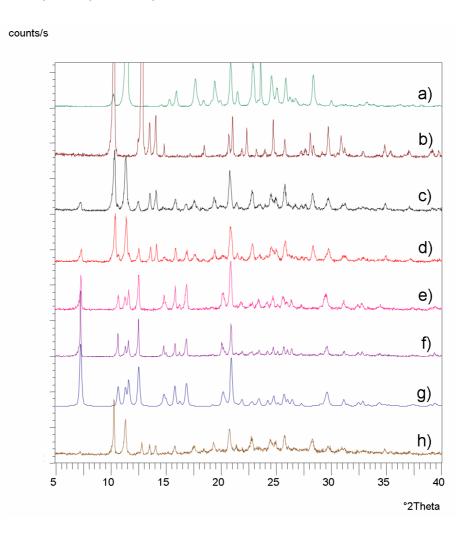


Figure S22. PXRD patterns for mechanochemical and solution-based experiments involving cobalt(II) acetate tetrahydrate and compound **1**: a) compound **1**, b) cobalt(II) acetate tetrahydrate, c) LAG product obtained by grinding of Co(II) acetate tetrahydrate and compound **1** in the presence of 40 μ L triethylamine (the acetic acid solvate of compound **3** is a major product and compound **3** is in traces), d) LAG product obtained by grinding of Co(II) acetate tetrahydrate and compound **3** is in traces), e) compound **3** is a major product and compound **3** is in traces), e) compound **3** obtained by annealing of LAG product, f) compound **3** obtained by solution-based method, g) simulated pattern for compound **3** and h) compound **3** after three days standing in acetic acid vapour.

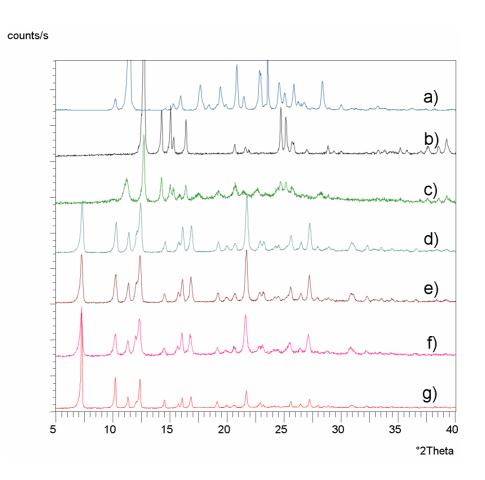


Figure S23. PXRD patterns for mechanochemical and solution-based experiments involving copper(II) acetate monohydrate and the compound 1: a) compound 1, b) copper acetate monohydrate, c) product obtained by neat grinding, d) LAG product obtained by grinding in the presence of 30 μ L acetonitrile (compound 2), e) LAG product obtained by grinding in the presence of 30 μ L methanol (compound 2), f) LAG product obtained by grinding in the presence of 30 μ L triethylamine (compound 2) and g) compound 2 obtained by solution-based method.

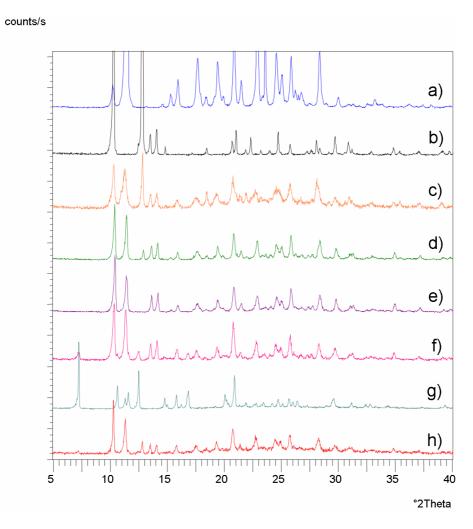


Figure S24. PXRD patterns for mechanochemical experiments involving cobalt(II) acetate tetrahydrate and 1: a) compound 1, b) cobalt(II) acetate tetrahydrate, c) product obtained by neat grinding, d) LAG product obtained by grinding in the presence of 30 μ L acetonitrile (acetic acid solvate of compound 3 is a major product and cobalt(II) acetate tetrahydrate is in traces), e) LAG product obtained by grinding in the presence of 30 μ L methanol (acetic acid solvate of compound 3), f) LAG product obtained by grinding in the presence of 30 μ L methanol (acetic acid solvate of compound 3), f) LAG product obtained by grinding in the presence of compound 3 is a major product and compound 3 is in traces), g) compound 3 obtained by solution-based method and h) compound 3 after three days standing in acetic acid vapour.