

Supporting Information

Syntheses, structures, magnetic and photoluminescence properties of metal-organic frameworks based on aromatic polycarboxylate acids

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Table S1 Selected Bond Distances (Å) and Angles (deg) for Compounds 1-3 (Co, Mn, and Cd).

Co compound			
Co1—O1	2.0644(16)	Co1—O3a	2.0762(15)
Co1—O4b	2.1590(16)		
O1—Co1—O1d	92.30(9)	O1—Co1—O3c	176.83(6)
O1—Co1—O3a	87.87(6)	O3c—Co1—O3a	92.15(8)
O1—Co1—O4e	83.17(6)	O3c—Co1—O4e	93.66(6)
O3a—Co1—O4e	85.52(6)	O1—Co1—O4f	97.65(6)
O4e—Co1—O4f	178.82(9)		
Mn compound			
Mn1—O1	2.210(2)	Mn1—O3	2.144(2)
Mn1—O4	2.181(2)		
O3—Mn1—O1	95.03(8)	O4—Mn1—O1	85.95(8)
O3—Mn1—O4	179.01(13)	O3a—Mn1—O3	94.00(12)
O3a—Mn1—O4	86.77(8)	O4a—Mn1—O4	92.47(11)
O3a—Mn1—O1	85.70(9)	O4a—Mn1—O1	93.36(8)
O1a—Mn1—O1	179.01(13)		
Cd compound			
Cd1—N2	2.331(3)	Cd1a—O4	2.332(2)
Cd1—O2	2.348(2)	Cd1—N1	2.386(3)
Cd1—O3	2.391(2)	Cd1b—O4	2.401(2)
Cd1a—O5	2.494(3)		
N2—Cd1—O4a	157.20(9)	N2—Cd1—O2	89.40(10)
O4a—Cd1—O2	102.18(9)	N2—Cd1—N1	69.98(10)
O4a—Cd1—N1	92.95(9)	O2—Cd1—N1	154.55(9)
N2—Cd1—O3	92.77(9)	O4a—Cd1—O3	109.98(8)
O2—Cd1—O3	55.03(8)	N1—Cd1—O3	137.58(9)
N2—Cd1—O4b	89.84(9)	O4a—Cd1—O4b	72.17(9)
O2—Cd1—O4b	84.12(8)	N1—Cd1—O4b	81.18(9)
O3—Cd1—O4b	139.00(8)	N2—Cd1—O5a	134.29(10)
O4a—Cd1—O5a	53.72(8)	O2—Cd1—O5a	122.57(9)
N1—Cd1—O5a	82.88(9)	O3—Cd1—O5a	83.01(8)
O4b—Cd1—O5a	122.26(8)	Cd1c—O4—Cd1b	107.83(9)

Symmetry transformations used to generate equivalent atoms:

a = -x, -y, -z; b = -x + 1/2, -y + 1/2, z + 1/2. c = -x, -y, z - 1/2; d = -x, y, -z - 1/2;

e = -x + 1/2, -y + 1/2, z - 1/2; f = x - 1/2, -y + 1/2, -z. (Co compound **1**)

a = -x, y, -z + 3/2. (Mn compound **2**)

a = x - 1/2, -y + 1, z - 1/2; b = -x + 3/2, y, -z + 3/2; c = x - 1/2, -y + 2, z - 1/2. (Cd compound **3**)

Table S2 Selected Bond Distances (Å) and Angles (deg) for compounds 4-6 (Eu, Nd, and Pr).

Eu compound			
Eu1 — O3	2.359(7)	Eu1 — O5	2.443(6)
Eu1a — O2	2.446(6)	Eu1b — O6	2.490(6)
Eu1 — O7	2.524(7)	Eu1c — O1	2.548(6)
Eu1c — O2	2.587(6)	O4d — Eu1	2.334(7)
O4d — Eu1 — O3	85.9(2)	O4d — Eu1 — O5	121.9(3)
O3 — Eu1 — O5	74.8(2)	O4d — Eu1 — O2a	81.0(2)
O3 — Eu1 — O2a	132.3(2)	O5 — Eu1 — O2a	74.2(2)
O4d — Eu1 — O6b	81.1(2)	O5 — Eu1 — O6b	135.8(2)
O3 — Eu1 — O6b	148.8(2)	O4d — Eu1 — O7	78.5(3)
O2a — Eu1 — O6b	73.4(2)	O3 — Eu1 — O7	77.6(2)
O5 — Eu1 — O7	143.7(2)	O2a — Eu1 — O7	142.0(2)
O6b — Eu1 — O7	72.1(2)	O4d — Eu1 — O1c	155.3(2)
O3 — Eu1 — O1c	85.0(2)	O5 — Eu1 — O1c	77.4(2)
O2a — Eu1 — O1c	121.6(2)	O6b — Eu1 — O1d	95.2(2)
O7 — Eu1 — O1d	77.1(2)	O4d — Eu1 — O2c	145.8(2)
O3 — Eu1 — O2c	128.0(2)	O5 — Eu1 — O2c	70.7(2)
O2a — Eu1 — O2c	72.1(2)	O6b — Eu1 — O2c	71.3(2)
O7 — Eu1 — O2c	110.5(2)	O1c — Eu1 — O2c	50.5(2)
O4d — Eu1 — Eu1b	115.57(19)	O5 — Eu1 — Eu1b	68.04(17)
O3 — Eu1 — Eu1b	142.74(16)	O6b — Eu1 — Eu1b	67.95(15)
O2a — Eu1 — Eu1b	37.23(14)	O7 — Eu1 — Eu1b	134.1(2)
O1c — Eu1 — Eu1b	84.90(15)	O2c — Eu1 — Eu1b	34.89(13)
Eu1e — O2 — Eu1c	107.9(2)		
Nd compound			
Nd1a—O4	2.333(5)	Nd1b—O3	2.335(5)
Nd1c—O2	2.418(4)	Nd1—O5	2.420(4)
Nd1c—O6	2.472(4)	Nd1—O7	2.496(6)
Nd1—O1	2.534(4)	Nd1—O2	2.589(4)
O4d—Nd1	2.333(5)	O4a—Nd1—O3b	85.47(17)
O3b—Nd1—O2c	131.75(17)	O4a—Nd1—O2c	80.40(16)
O4a—Nd1—O5	121.40(18)	O3b—Nd1—O5	74.49(17)
O2c—Nd1—O5	74.37(16)	O5—Nd1—O6c	136.25(16)
O4a—Nd1—O6c	81.28(17)	O4a—Nd1—O7	78.65(19)
O3b—Nd1—O6c	148.71(16)	O3b—Nd1—O7	77.82(19)
O2c—Nd1—O6c	73.63(15)	O2c—Nd1—O7	141.77(17)
O5—Nd1—O7	143.71(18)	O3b—Nd1—O1	85.13(16)
O6c—Nd1—O7	71.85(18)	O2c—Nd1—O1	122.41(14)
O4a—Nd1—O1	155.16(17)	O5—Nd1—O1	77.79(17)
O6c—Nd1—O1	95.20(17)	O5—Nd1—O2	70.77(16)

O7—Nd1—O1	76.91(18)	O6c—Nd1—O2	71.62(15)
O4a—Nd1—O2	145.83(15)	O7—Nd1—O2	110.86(18)
O3b—Nd1—O2	128.19(15)	O1—Nd1—O2	50.99(14)
O2c—Nd1—O2	72.39(16)	O5—Nd1—O2	70.77(16)
O4a—Nd1—Nd1c	115.48(13)	O3b—Nd1—Nd1c	142.60(13)
O2c—Nd1—Nd1c	37.62(10)	O6c—Nd1—Nd1c	68.25(11)
O5—Nd1—Nd1c	68.16(11)	O7c—Nd1—Nd1	134.27(15)
O1c—Nd1—Nd1	85.27(10)	O2c—Nd1—Nd1	34.76(10)
Nd1c—O2—Nd1	107.61(16)		
Pr compound			
O1—Pr1	2.292(5)	O3—Pr1	2.421(5)
O2—Pr1	2.445(6)	O4—Pr1	2.371(5)
O5a—Pr1	2.493(5)	O8b—Pr1	2.282(5)
O7a—Pr1	2.546(5)	Pr1c—O7	2.375(5)
Pr1d—O7—Pr1a	107.66(18)	O1—Pr1—O4	74.07(19)
O8b—Pr1—O1	85.34(19)	O8b—Pr1—O7	80.52(18)
O8b—Pr1—O4	120.7(2)	O1—Pr1—O7	131.69(19)
O4—Pr1—O7c	74.33(18)	O4—Pr1—O3	136.95(18)
O8b—Pr1—O3	81.62(19)	O7c—Pr1—O3	74.16(17)
O1—Pr1—O3	148.40(18)	O8b—Pr1—O2	79.3(2)
O1—Pr1—O2	77.2(2)	O8b—Pr1—O5a	155.04(19)
O4—Pr1—O2	142.8(2)	O1—Pr1—O5a	84.14(18)
O7c—Pr1—O2	142.75(19)	O4—Pr1—O5a	77.69(19)
O3—Pr1—O2	72.2(2)	O7c—Pr1—O5a	122.89(16)
O3—Pr1—O5a	95.84(18)	O4—Pr1—O7a	71.37(18)
O2—Pr1—O5a	76.4(2)	O7c—Pr1—O7a	72.34(18)
O8b—Pr1—O7a	145.96(17)	O3—Pr1—O7a	71.60(17)
O1—Pr1—O7a	128.18(17)	O2—Pr1—O7a	110.77(19)
O5a—Pr1—O7a	51.71(15)	O4—Pr1—Pr1e	68.52(13)
O8b—Pr1—Pr1e	115.63(14)	O7c—Pr1—Pr1e	37.63(11)
O1—Pr1—Pr1e	142.52(14)	O3—Pr1—Pr1e	68.57(12)
O2—Pr1—Pr1e	134.66(16)	O7a—Pr1—Pr1e	34.72(11)
O5a—Pr1—Pr1e	85.84(11)		

Symmetry transformations used to generate equivalent atoms:

$a = x + 1/2, -y + 1/2, z - 1/2$; $b = -x + 5/2, -y + 1/2, -z + 1$; $c = -x + 2, y, -z + 3/2$;

$d = -x + 2, y, -z + 1/2$; $e = x - 1/2, -y + 1/2, z + 1/2$. (Eu compound **4**)

$a = x, y, z - 1$; $b = -x, y, -z + 3/2$; $c = -x + 1/2, -y + 1/2, -z + 1$; $d = x, y, z + 1$. (Nd compound **5**)

$a = -x, y, -z + 1/2$; $b = -x, y, -z - 1/2$; $c = x + 1/2, -y + 1/2, z - 1/2$; $d = x - 1/2, -y + 1/2, z + 1/2$; $e = -x + 1/2, -y + 1/2, -z$. (Pr compound **6**)

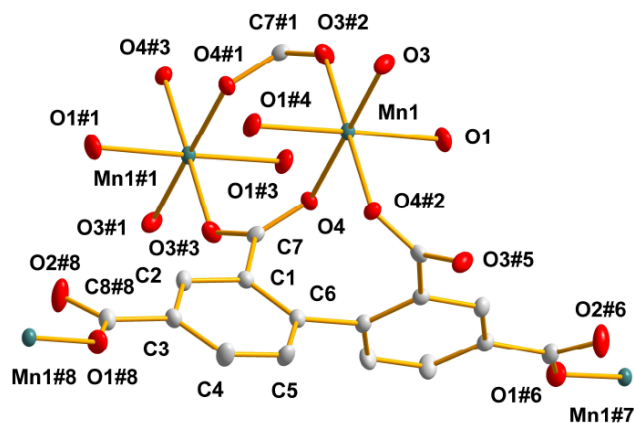


Figure S1. An ORTEP drawing of Mn compound **2** showing 30% ellipsoid probability (hydrogen atoms are omitted for clarity). Symmetry codes: #1= $-x, 2-y, 2-z$; #2= $-x, y, 1.5-z$; #3= $x, 2-y, 0.5+z$; #4= $-x, y, 1.5-z$; #5= $-x, 2-y, 1-z$; #6= $-0.5-x, -0.5+y, z$; #7= $-0.5+x, -0.5+y, 1.5-z$; #8= $0.5+x, -0.5+y, 1.5-z$.

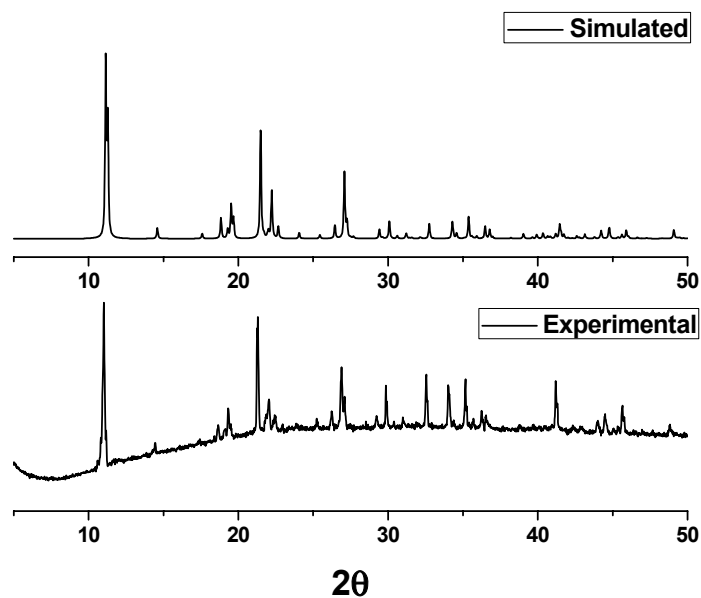


Figure S2. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Co compound **1** at 293K.

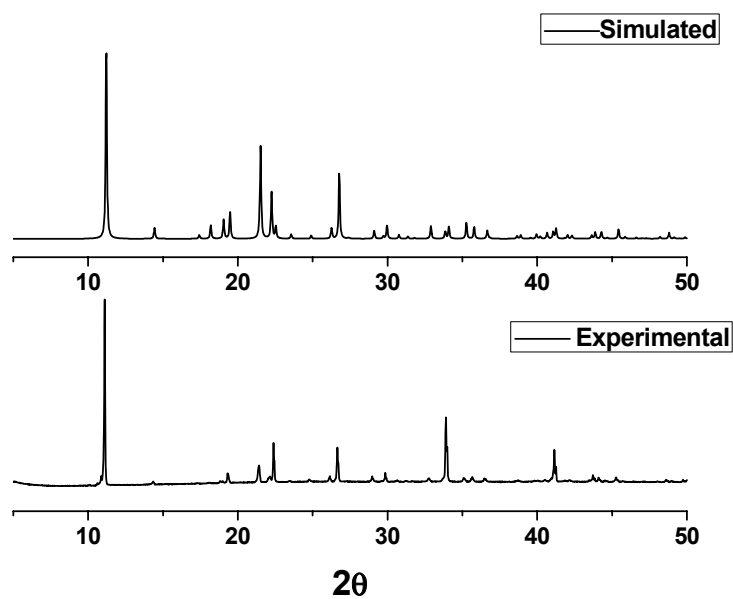


Figure S3. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Mn compound **2** at 293K.

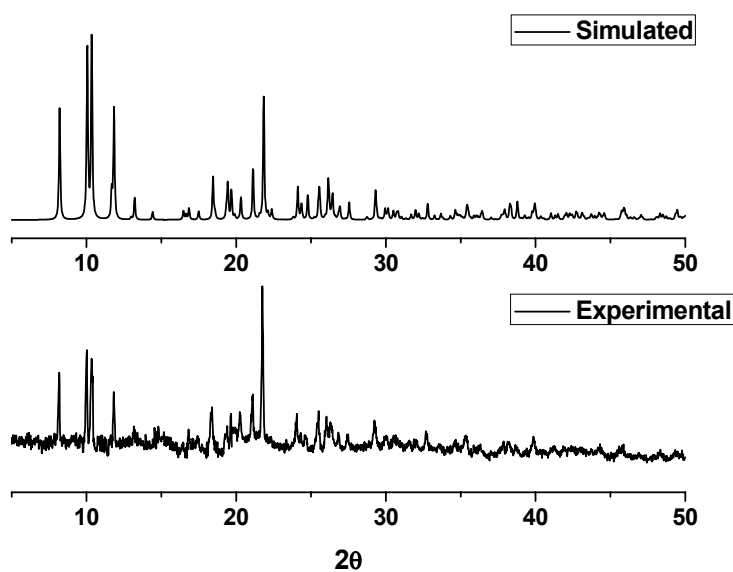


Figure S4. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Cd compound **3** at 293K.

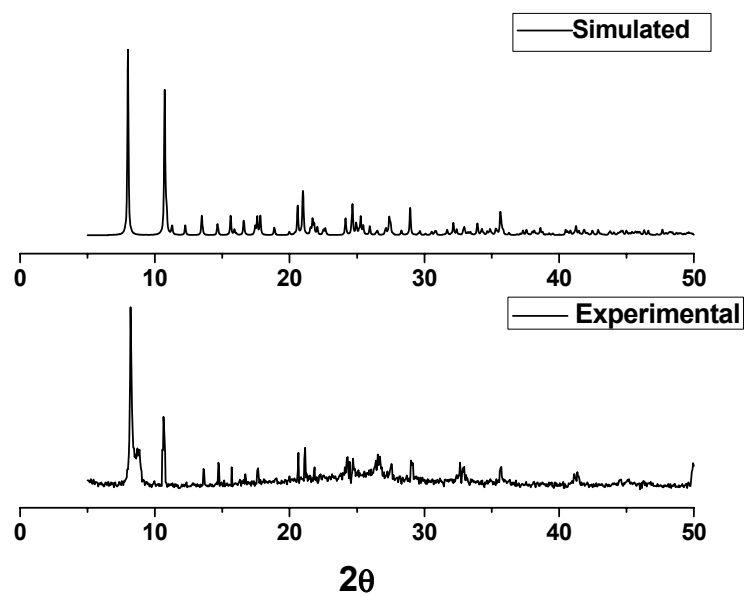


Figure S5. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Eu compound **4** at 293K.

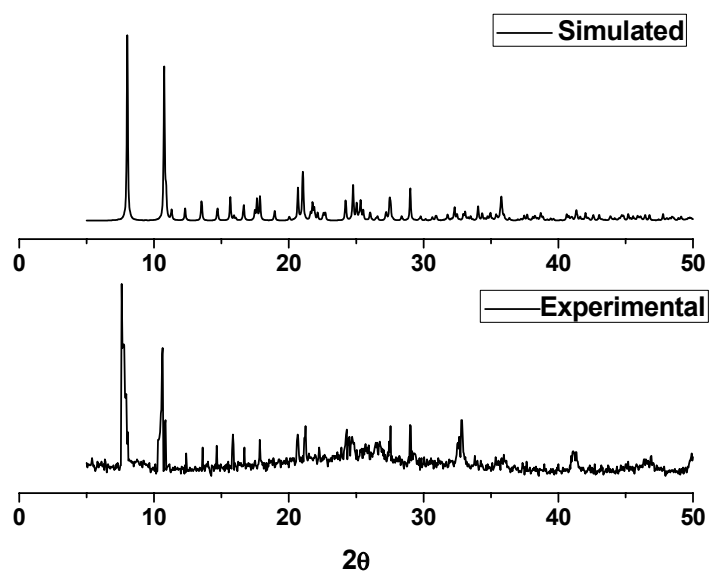


Figure S6. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Nd compound **5** at 293K.

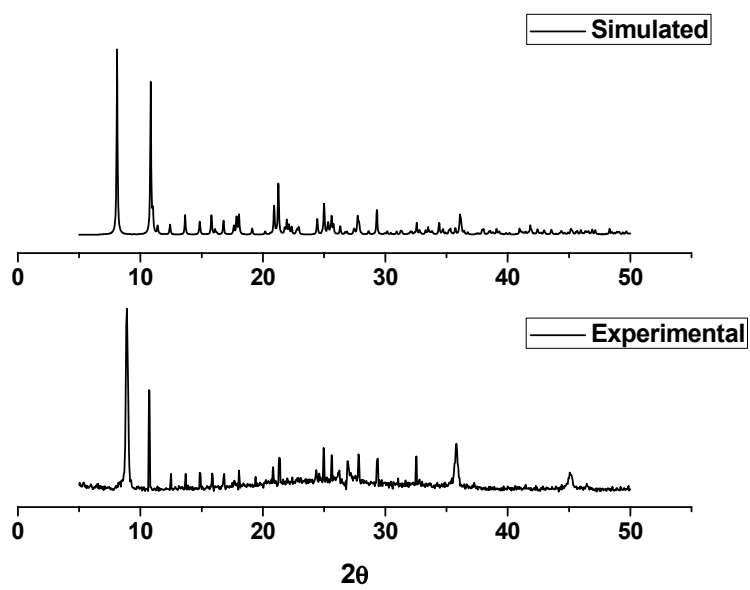


Figure S7. Experimental (bottom) and simulated (top) powder X-ray diffraction patterns of Pr compound **6** at 293K.