

Supporting Information
to the paper entitled

**Supramolecular Assemblies Involving Anion- π and Lone Pair- π
Interactions: Experimental Observation and Theoretical Analysis**

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Table S1. Crystal data and structure refinement for compounds **1** and **2**.

Compound	1	2
Formula	C ₂₆ H ₃₆ N ₈ O ₁₈ Cl ₂ Ni	C ₂₆ H ₃₆ N ₈ O ₁₀ P ₂ F ₁₂ Ni
Mw (g mol ⁻¹)	878.22	969.28
Crystal system	Triclinic	Triclinic
Space group	<i>P</i> $\bar{1}$ (No. 2)	<i>P</i> $\bar{1}$ (No. 2)
<i>a</i> / Å	7.0471(12)	7.0459(10)
<i>b</i> / Å	11.314(2)	11.8570(17)
<i>c</i> / Å	11.747(2)	11.8501(17)
α	86.321(3)°	84.482(2)°
β	83.933(3)°	84.484(2)°
γ	72.684(3)°	72.251(2)°
<i>F</i> (000)	454	494
<i>V</i> / Å ³	888.6(3)	936.2(2)
<i>Z</i>	1	1
<i>T</i> / <i>K</i>	100	100
Theta min-max [°]	1.9 to 26.4	2.4 to 26.4
λ (Mo <i>K</i> α)/ Å	0.71073	0.71073
μ (Mo <i>K</i> α)/ mm ⁻¹	0.787	0.728
Crystal Size [mm ³]	0.40 × 0.45 × 0.50	0.20 × 0.45 × 0.50
<i>R</i> 1, <i>I</i> > 2 σ (<i>I</i>) (all)	0.0340	0.0292
<i>wR</i> 2, <i>I</i> > 2 σ (<i>I</i>) (all)	0.0791	0.0783
<i>S</i> (GOF)	0.97	1.04
Total reflections	7093	7510
Independent reflections (<i>R</i> _{int})	3571 (0.049)	3768 (0.014)
Observed data [<i>I</i> > 2 σ (<i>I</i>)]	2921	3573
Min. and max. resd. dens. [e/ Å ³]	-0.38, 0.56	-0.36, 0.51

Table S2. Selected bond lengths (Å) and angles (deg) for **1**.

Ni(1)–O(1)	2.0137(15)	O(4)–C(3)	1.283(3)
Ni(1)–O(4)	2.0051(15)	O(3)–C(3)	1.232(2)
Ni(1)–O(5)	2.0949(18)	C(1)–C(2)	1.513(3)
O(1)–C(1)	1.275(3)	C(2)–C(3)	1.522(3)
O(2)–C(1)	1.247(3)		
O(1)–Ni(1)–O(4)	90.81(6)	Ni(1)–O(4)–C(3)	128.39(13)
O(1)–Ni(1)–O(5)	94.38(7)	O(4)–C(3)–O(3)	123.10(19)
O(1)–Ni(1)–O(4) [#]	89.19(6)	C(1)–C(2)–C(3)	120.11(18)
O(1)–Ni(1)–O(5) [#]	85.62(7)	O(1)–C(1)–O(2)	122.22(19)
O(4)–Ni(1)–O(5)	91.96(6)	O(4)–C(3)–C(2)	119.38(18)
O(4)–Ni(1)–O(5) [#]	88.05(6)	O(1)–C(1)–C(2)	119.05(19)
Ni(1)–O(1)–C(1)	127.11(13)	O(4)–C(3)–C(2)	119.38(18)

Symmetry code: # = 2–x, 1–y, 2–z

Table S3. Selected bond lengths (Å) and angles (deg) for **2**

Ni(1)–O(1)	2.0218(12)	O(4)–C(3)	1.2841(19)
Ni(1)–O(4)	1.9971(11)	O(3)–C(3)	1.2344(19)
Ni(1)–O(5)	2.0878(14)	C(1)–C(2)	1.520(2)
O(1)–C(1)	1.2704(19)	C(2)–C(3)	1.525(2)
O(2)–C(1)	1.253(2)		
O(1)–Ni(1)–O(4)	91.10(5)	Ni(1)–O(4)–C(3)	129.30(10)
O(1)–Ni(1)–O(5)	94.56(5)	O(4)–C(3)–O(3)	123.00(15)
O(1)–Ni(1)–O(4) [#]	88.90(5)	C(1)–C(2)–C(3)	120.20(14)
O(1)–Ni(1)–O(5) [#]	85.44(5)	O(1)–C(1)–O(2)	122.02(14)
O(4)–Ni(1)–O(5)	92.03(5)	O(4)–C(3)–C(2)	120.08(13)
O(4)–Ni(1)–O(5) [#]	87.97(5)	O(1)–C(1)–C(2)	120.05(13)
Ni(1)–O(1)–C(1)	127.52(10)	O(4)–C(3)–C(2)	120.08(13)

Symmetry code: # = 2–x, 1–y, 2–z

Figure S1. Schematic representation of the reaction pathways used to determine the dimerization energies E_1 and E_2 for monomers A) **M5** and B) **M6**. The two noncovalent forces (H-bonding and lp- π interactions) responsible for the formation of the dimers **D5** and **D6** are indicated.

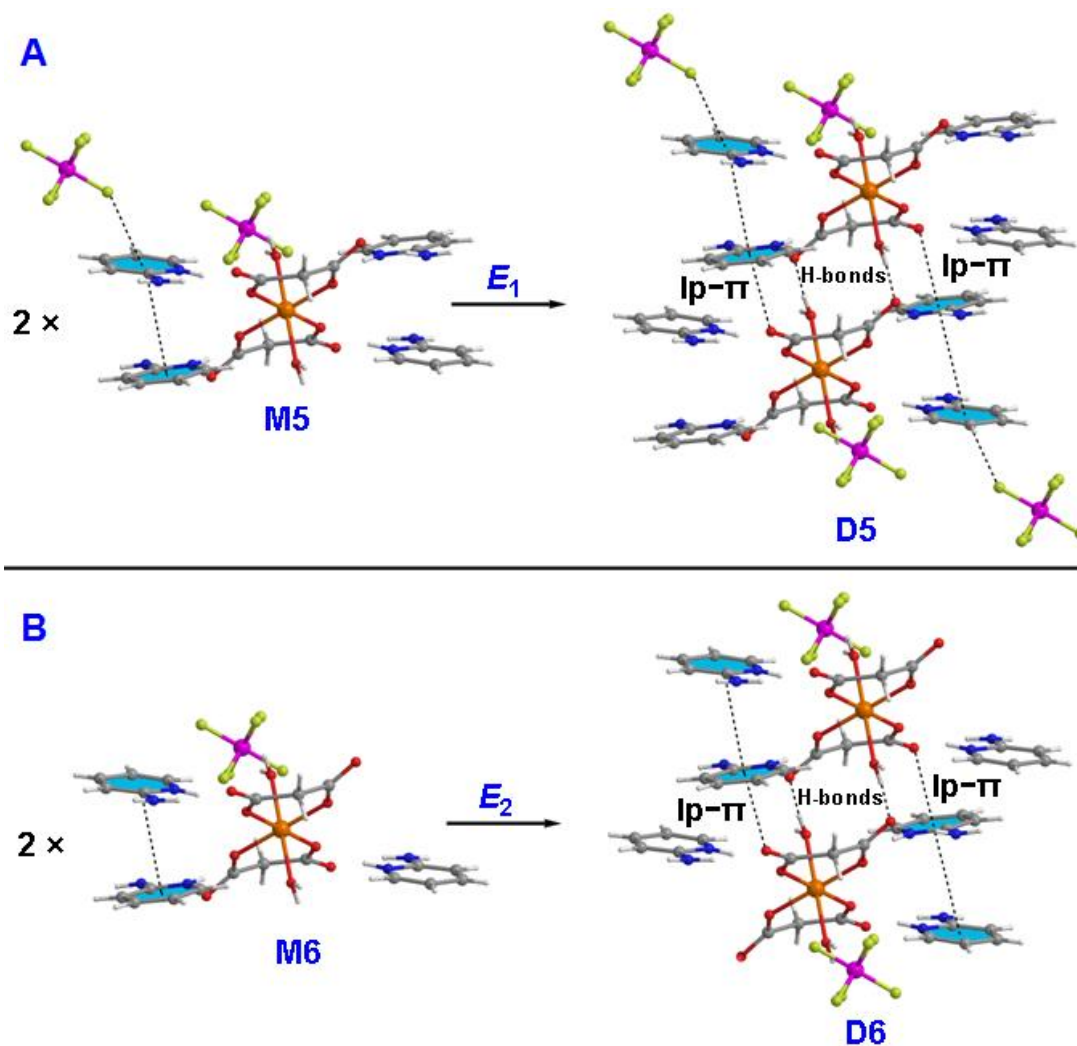


Figure S2. Schematic representation of the reaction used to determine the dimerization energies E_3 and E_4 for monomers A) **M7** and B) **M8**. The two noncovalent forces (H-bonding and lp- π interactions) responsible for the formation of the dimers **D7** and **D8** are indicated.

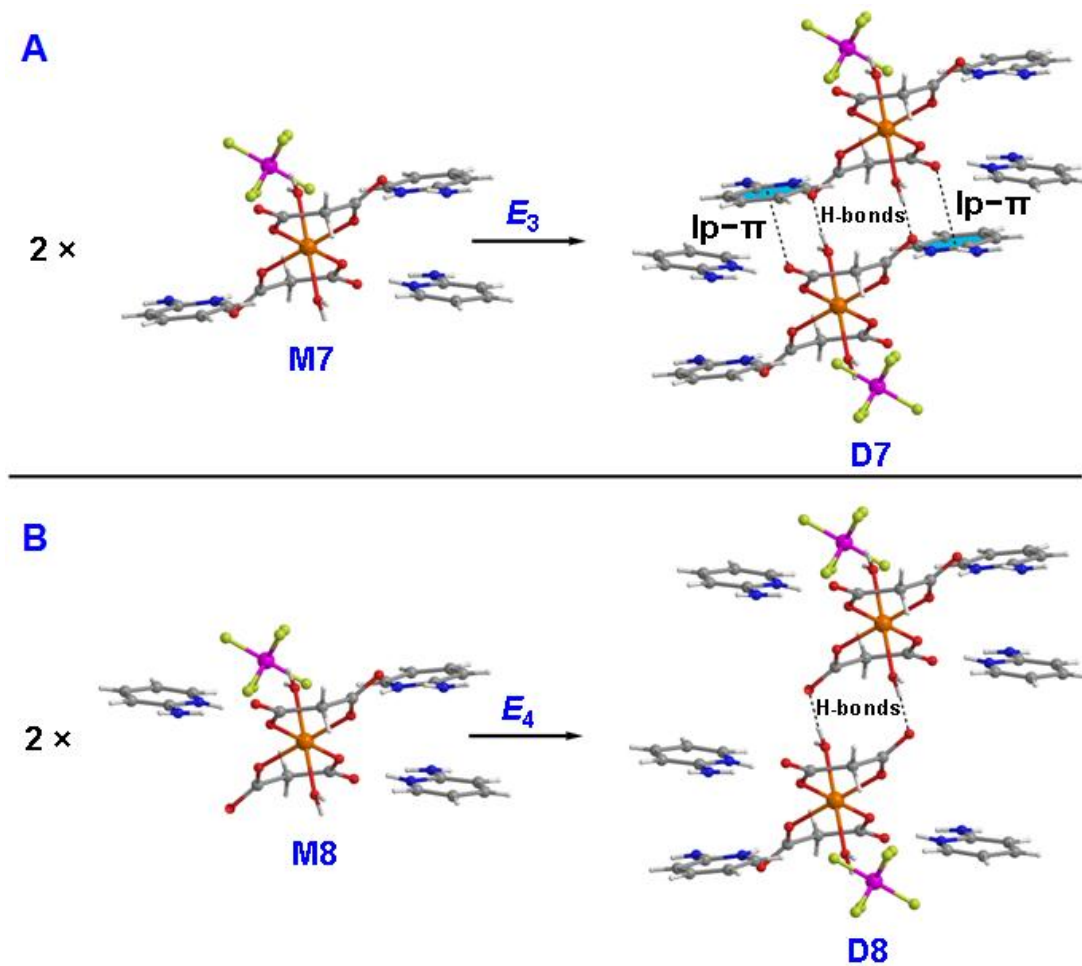


Figure S3. TG plots of compounds 1 and 2.

