

Robust motifs in 2-phenylethylammonium- and related tetrahalometallates

Melanie Rademeyer*, Christos Tsouris, David G. Billing, Andreas Lemmerer and Jonathan Charmant

Supplementary Information

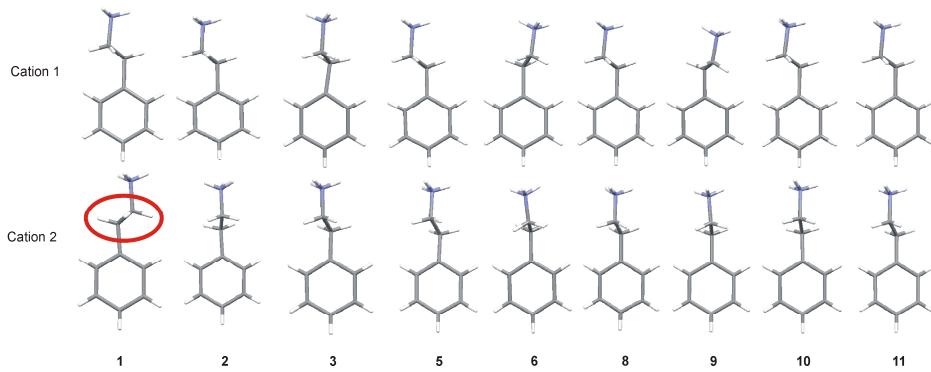


Fig. S1 Comparison of cation geometry in structures **1** to **11**. Cations viewed perpendicular to aromatic plane.

Table S1 Hydrogen bonding parameters involved in formation of the zero-dimensional motifs^{a,b}

Motif 1	D...A (Å)	Symmetry operator	Motif 2	D...A (Å)	Symmetry operator
1 N(1)-H(1C)...Cl(1) N(1)-H(1D)...Cl(4)	3.438(7) 3.314(7)	-x, -y+I, -z+I x, y, z	N(2)-H(2C)...Cl(3) N(2)-H(2D)...Cl(1) N(2)-H(2E)...Cl(1) C(10)-H(10B)...Cl(4)	3.499(9) 3.378(8) 3.310(6) 3.846	x, y, z+I x, y, z+I -x, -y+I, -z+I -x, -y+I, -z+I
2 ¹² N(2)-H(24)...Br(2) N(2)-H(22)...Br(1)	3.527 3.434	x, y, z I-x, -y, -z	N(1)-H(20)...Br(3) N(1)-H(21)...Br(3) N(1)-H(20)...Br(2)	3.596 3.487 3.502	x, ½ - y, -½ + z -x, -½ + y, ½ - z x, ½ - y, -½ + z
3 N(1)-H(1C)...I(2) N(1)-H(1C)...I(1) N(1)-H(1D)...I(4)	3.640(8) 3.612(9) 3.627(10)	x, y, z x, y, z -x+I, -y+I, -z+I	N(2)-H(2C)...I(3) N(2)-H(2C)...I(4) N(2)-H(2E)...I(4)	3.689(11) 3.860(11) 3.838(9)	-x+I, -y+I, -z+I x, y, z+I -x+I, -y+I, -z+I
5 N(1)-H(1C)...Br(1) N(1)-H(1D)...Br(4)	3.510(7) 3.418(7)	-x, -y+I, -z+I x, y, z	N(2)-H(2D)...Br(1) N(2)-H(2E)...Br(1) C(10)-H(10A)...Br(4)	3.493(8) 3.603(6) 3.784	x, y, z+I -x, -y+I, -z+I 2-x, ½ + y, ½ - z
6 N(1)-H(1D)...I(4) N(1)-H(1E)...I(1)	3.716(12) 3.601(17)	x, y, z -x, y-I/2, -z	N(2B)-H(2CB)...I(1) N(2B)-H(2CB)...I(3) N(2B)-H(2DB)...I(3) N(2B)-H(2DB)...I(1) C(9B)-H(9B2)...I(4) C(10B)-H(1B1)...I(4)	3.65(3) 3.84(2) 3.68(3) 3.78(2) 4.041 4.001	x, y, z and x-I, y, z+I -x, y+½, -z+I x, y, z and x-I, y, z+I -x, y-½, -z+I I-x, ½ + y, -z I-x, -½ + y, -z
8 N(1)-H(1E)...Br(3) N(1)-H(1D)...Br(4)	3.474(7) 3.438(7)	-x+I, -y+I, -z+I x, y, x	N(2)-H(2D)...Br(1) N(2)-H(2E)...Br(1)	3.481(8) 3.602(7)	x, y, z+I -x, -y+I, -z+I
9 N(1)-H(1D)...I(4) N(1)-H(1E)...I(1)	3.615(8) 3.714(13)	x, y, z -x, y-I/2, -z	N(2B)-H(2CB)...I(1) N(2B)-H(2CB)...I(3) N(2B)-H(2DB)...I(3) N(2B)-H(2DB)...I(1) C(9B)-H(9B2)...I(4) C(10B)-H(1B1)...I(4)	3.778(13) 3.878(11) 3.737(13) 3.804(11) 4.080 4.096	x, y, z and x-I, y, z+I -x, y+½, -z+I x, y, z and x-I, y, z+I -x, y-½, -z+I I-x, ½ + y, -z I-x, -½ + y, -z
10 N(1)-H(1C)...Cl(1) N(1)-H(1D)...Cl(2) N(1)-H(1E)...I(1)	3.554(11) 3.412(11) 3.598(9)	-x, -y+I, -z+I x, y, z -x+I, -y+I, -z+I	N(2)-H(2D)...Cl(1) N(2)-H(2E)...Cl(1)	3.387(17) 3.430(12)	x, y, z+I -x, -y+I, -z+I
11 ¹³ N(1)-H(1)...Br(2) N(1)-H(2)...Br(1)	3.500 3.434	x, y, z I-x, ½ + y, ½ - z	N(2)-H(14)...Br(3) N(2)-H(15)...Br(3)	3.487 3.606	-x+I, y- ½ , -z + ½ x, -y+½ , z+½

^a Only the cations with the highest occupancy were considered in structures **6** and **9**.

^b Hydrogen bonding definition: H...A < radius(A) + 2 Å and DHA angle larger than 110°

Table S2 Hydrogen bonding and C-H...Cg interactions in the novel structures **1**, **3**, **5**, **6**, **8**, **9** and **10** involved in formation of the one-dimensional column and three-dimensional structure (Note: Cg(1) refers to the aromatic ring centroid on cation 1 consisting of atoms C(3) to C(8), and Cg(2) refers to the aromatic ring centroid on cation 2, consisting of atoms C(11) to C(16).

C-H...π' interactions linking zero-dimensional motifs to form 1-d chains, and one-dimensional motifs to form 2-d sheets			Hydrogen bonding interactions linking two-dimensional sheets		
Interaction	H...Cg (Å)	Symmetry transformation of acceptor ring	Interaction	D...A (Å)	Symmetry transformation of acceptor atom
1					
1-d chain interaction C(12)-H(12)...Cg(1)	2.79	$I+x, y, z$	N(1)-H(1E)...Cl(3)	3.381(7)	$-x+I, -y+I, -z+I$
2-d sheet interaction C(8)-H(8)...Cg(2)	2.86	$x, \frac{1}{2}-y, \frac{1}{2}+z$	N(2)-H(2C)...Cl(3)	3.298(7)	$-x+I, -y+I, -z+I$
C(15)-H(15)...Cg(1)	2.80	$x, \frac{1}{2}-y, \frac{1}{2}+z$			
3					
1-d chain interaction C(12)-H(12)...Cg(1)	3.07	$-x, y+\frac{1}{2}, -z+\frac{1}{2}+I$	N(1)-H(1D)...I(3)	3.690(10)	$-x, -y+I, -z+I$
2-d sheet interaction C(15)-H(15)...Cg(1)	2.98	$x, \frac{1}{2}-y, \frac{1}{2}+z$	N(2)-H(2D)...I(1)	4.166(11)	$-x, -y+I, -z+I$
5					
1-d chain interaction C(12)-H(12)...Cg(1)	2.98	$I+x, y, z$	N(1)-H(1E)...Br(2)	3.864(7)	$-x+I, -y+I, -z+I$
2-d sheet interaction C(15)-H(12)...Cg(1)	2.90	$x, \frac{1}{2}-y, \frac{1}{2}+z$	N(1)-H(1E)...Br(3)	3.494(6)	$-x+I, -y+I, -z+I$
			N(2)-H(2C)...Br(3)	3.520(7)	$-x+I, -y+I, -z+I$
6			N(2)-H(2C)...Br(3)	3.732(8)	$x, y, z+I$
1-d chain interaction (chain 1 ^a) C(8)-H(8B)...Cg(1)	3.09	$-x, y-\frac{1}{2}, -z+I$	N(1)-H(1C)...I(3)	3.62(2)	$-x, y+\frac{1}{2}, -z$
1-d chain interaction (chain 2 ^a) C(16B)-H(16)...Cg(2)	2.99	$I-x, -\frac{1}{2}+y, I-z$	N(2B)-H(2EB)...I(2)	4.02(2)	$x, y, z+I$
2-d sheet interaction C(13B)-H(13B)...Cg(1)	2.89	$x, I+y, z$			
8					
1-d chain interaction C(12)-H(12)...Cg(1)	3.01	$I+x, y, z$	N(1)-H(1C)...Br(1)	3.517(8)	$-x, -y+I, -z+I$
2-d sheet interaction C(15)-H(15)...Cg(1)	2.89	$x, \frac{1}{2}-y, \frac{1}{2}+z$	N(2)-H(2C)...Br(3)	3.505(7)	$-x+I, -y+I, -z+I$
9					
1-d chain interaction (chain 1 ^a) C(8)-H(8)...Cg(1)	3.18	$-x, y-\frac{1}{2}, -z+I$	N(1)-H(1D)...I(2)	3.571(7)	x, y, z
1-d chain interaction (chain 2 ^a) C(16B)-H(16B)...Cg(2)	3.05	$I-x, -\frac{1}{2}+y, I-z$	N(2B)-H(2EB)...I(2)	3.875(8)	$x, y, z+I$
2-d sheet interaction C(13B)-H(13B)...Cg(1)	2.97	$x, I+y, z$			
10					
1-d chain interaction C(12)-H(12)...Cg(1)	2.89	$I+x, y, z$	N(1)-H(1E)...Br(1)	3.545(10)	$-x+I, -y+I, -z+I$
2-d sheet interaction C(15)-H(15)...Cg(1)	2.88	$x, \frac{1}{2}-y, \frac{1}{2}+z$	N(2)-H(2C)...Br(1)	3.533(16)	$-x+I, -y+I, -z+I$

^a In structures **6** and **9** chain 1 consists of motif 1's and chain 2 consists of motif 2's.

^b Definition: H...Cg < 3.0 Å, and the angle between the X-H...Cg vector and normal to aromatic plane smaller than 30°.

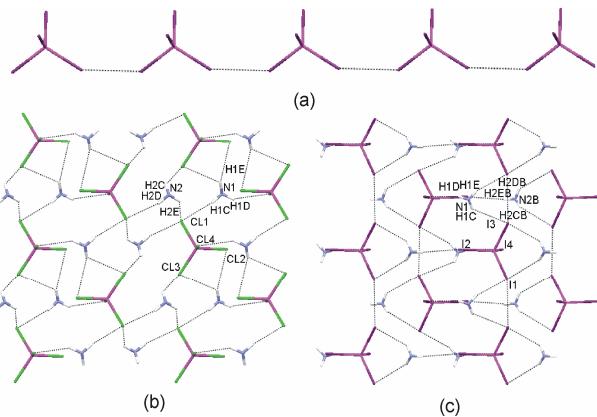


Fig. S2 (a) M-I...I-M close contacts present in structures **6** and **9** (b) hydrogen bonded sheet in structure **1** (c) hydrogen bonded sheet in structure **9**.

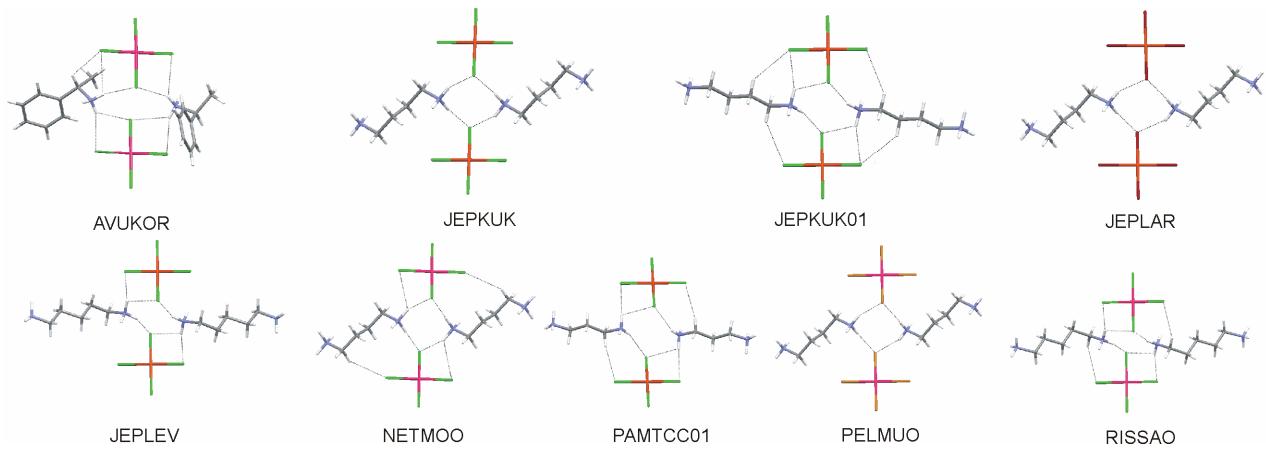


Fig. S3 Zero-dimensional motifs present in structures with square planar anion geometries.

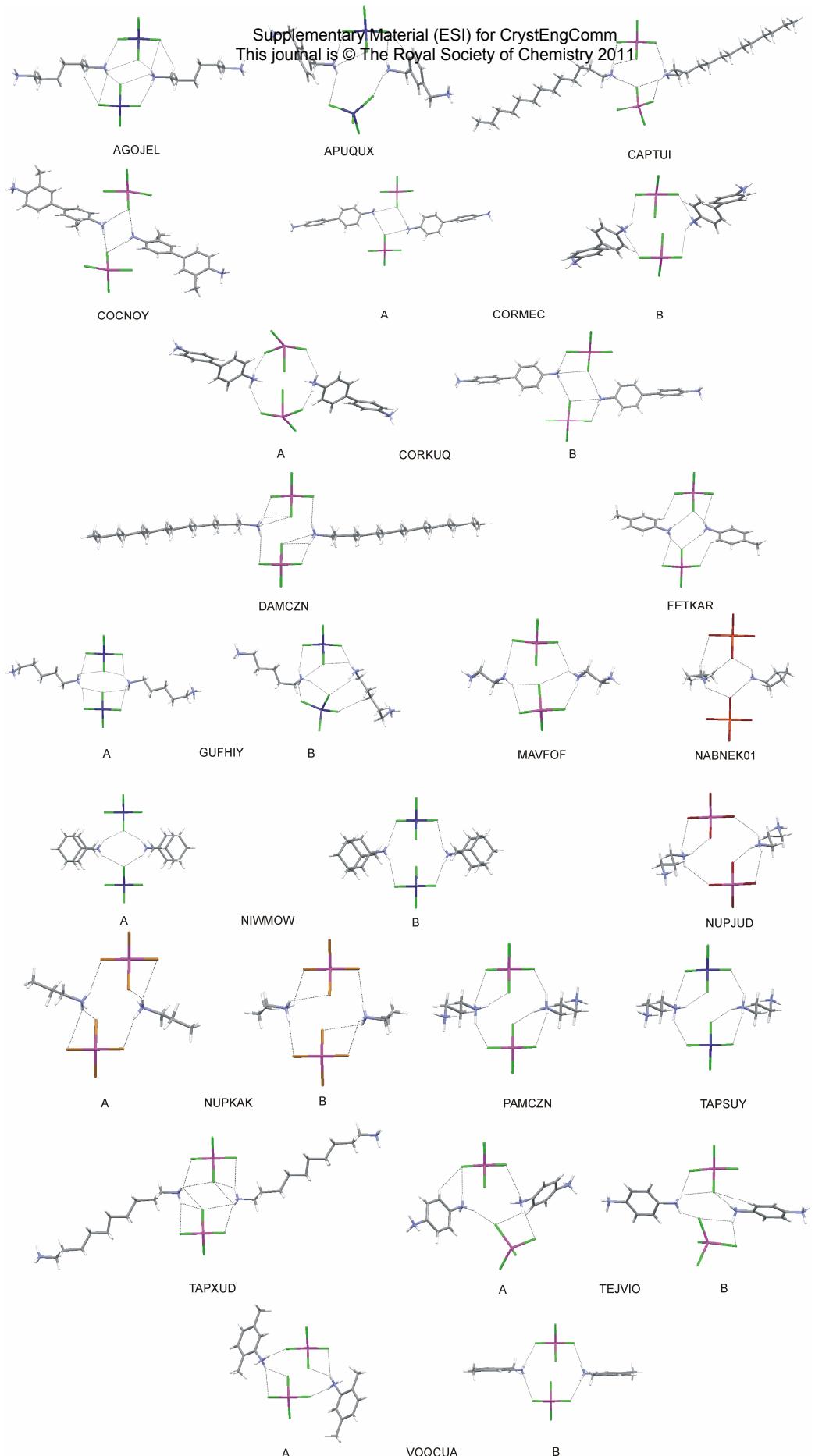


Fig. S4 Zero-dimensional motifs present in structures with tetrahedral geometries.