

Fabrication and growth mechanism of hierarchical porous Fe₃O₄ hollow sub-microspheres and their magnetic properties

Yong Wang,^{ab} Qingshan Zhu^{*b} and Ling Tao^b

^a *Department of Chemistry, Capital Normal University, Beijing 100048, China.*

^b *State Key Laboratory of Multiphase Complex System, Institute of Process Engineering, Chinese Academy of Sciences, Beijing 100080, China.*

S1. XRD patterns of the products with different reaction times

The formation process of hierarchical porous Fe₃O₄ hollow sub-microspheres is indicated in the following series of experiments (2.0 mmol of Fe(acac)₃ was dissolved in 20 mL of diphenyl ether to form a solution, then 10.0 mmol of 1,6-hexanediol was added into the solution. The resulting mixture was then sealed into a 50 mL Teflon-lined autoclave. After that, the autoclave was transferred to an electric oven at 220 °C and kept for 1h, 3h, 4h and 10h respectively). Fig. S1 displays the X-ray diffraction (XRD) patterns of hierarchical porous Fe₃O₄ hollow sub-microspheres prepared at 220°C with reaction time of 1h, 3h, 4h and 10h respectively. The XRD patterns clearly show the phase transformation from Fe₂O₃ to Fe₃O₄ (Fig. S1, ESI). All four systems give the same transformation trend, which means that the pure Fe₂O₃ is the only template in forming Fe₃O₄ hollow sub-microspheres. In Fig. S1a, the sample prepared at 220°C with reaction time of 1h is identified as single phase Fe₂O₃ with rhombohedral structure (a =5.038 Å, c=13.772 Å, JCPDS file No. 24–0072). No peaks from other phases are found, indicating that the as-obtained solid sub-microspheres are single phase α-Fe₂O₃. In Fig. S1b, XRD analyses indicate that the as-obtained sample is a mixture with two phases, Fe₂O₃ with rhombohedral structure (a =5.038 Å, c=13.772 Å, JCPDS file No. 24–0072, signal +) and Fe₃O₄ with cubic structure (a=8.393 Å, JCPDS file No. 85-1436, signal #). The result confirms that the reductive reaction of hematite to magnetite occurs and an incompact layer of Fe₃O₄ is formed on the surface of the initial α-Fe₂O₃ particles. As shown in Fig. S1b-d, moreover, the crystallinity of the Fe₃O₄ in the samples gradually increases

with the reaction time, which indicates that Ostwald ripening (crystallites grow at the expense of the smaller ones) is one of the underlying mechanisms in this hollowing process.

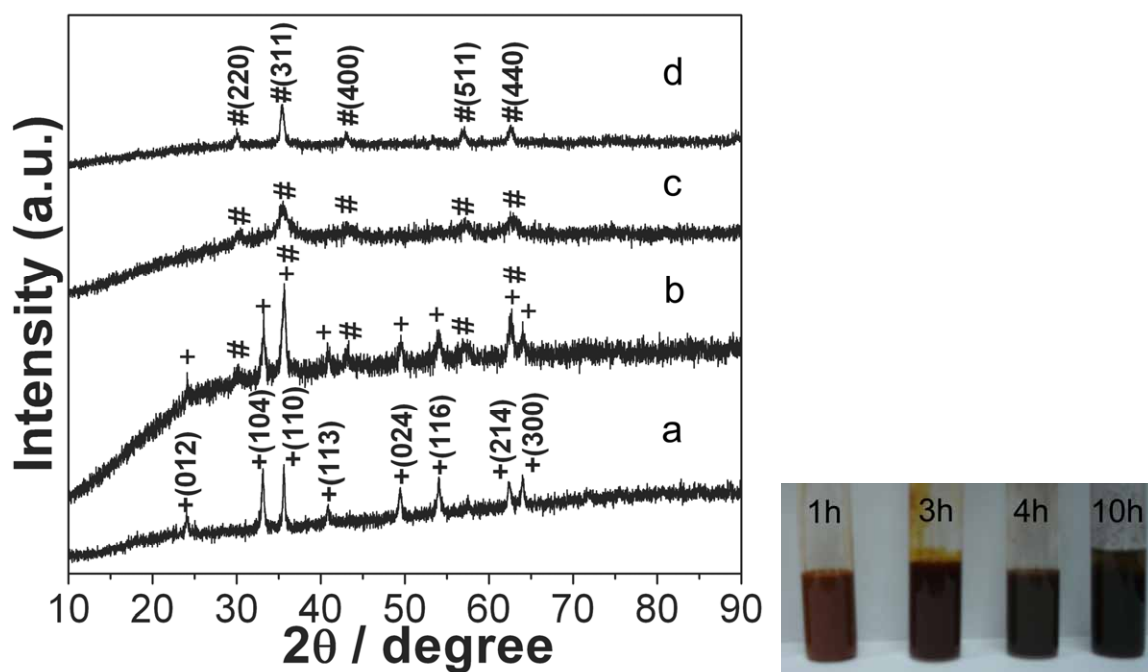


Fig. S1 X-ray diffraction patterns of Fe₃O₄ hierarchical hollow sub-microspheres prepared at 220°C with different experimental time: (a)1h; (b)3 h; (c) 4 h; (d) 10 h. Fe₂O₃ (+) and Fe₃O₄ (#).

S2. Good dispersion of hierarchical Fe₃O₄ hollow sub-microspheres

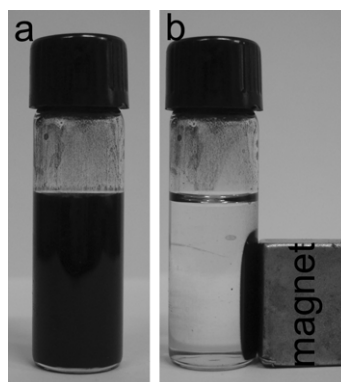


Fig. S2 Photographs of sample (a) before and (b) after using an applied magnetic field for 30 s.