

Synthesis of diisopropyl pyridylphosphonates 2''(B) and 3''(B) (Scheme 1) **by general procedure.**²⁶ To a solution of diisopropyl phosphite (6.0 mmol, 1.0 mL) in dried MeCN (8 mL) was added corresponding bromo-derivative of pyridine (5.0 mmol methyl 5-bromonicotinate²⁵ **2'** or 2.5 mmol 3,5-dibromopyridine **3'**), *N,N*-diisopropylethylamine (6.0 mmol, 1.05 mL), palladium acetate (0.05 mmol, 14 mg) and 1,1'-bis(diphenylphosphino)ferrocene (dppf, 0.055 mmol, 31 mg) at room temperature. The reaction mixture was heated at 85 °C while continuously stirred under argon atmosphere for 24 h. Then the solvent was evaporated under reduced pressure. The residue was extracted with ethyl acetate (5, 2 and 2 mL), insoluble *N,N*-diisopropylethylamine hydrobromide was filtered off, dried in air and weight (gravimetric measurement of reaction yield). The filtrate was evaporate under reduced pressure to obtain a residue. The residue was subjected to column chromatography on silica gel using chloroform-ethyl acetate or ethanol-ethyl acetate as an eluent to obtain pure esters as oils. The eluates were monitored by TLC (R_f data are given in the experimental part below) The structures of final products **2''(B)** and **3''(B)** were confirmed by ^1H , $^{31}\text{P}\{^1\text{H}\}$ and $^{13}\text{C}\{^1\text{H}\}$ NMR in CDCl_3 and data are given in the experimental section below.

Methyl 5-(diisopropoxyphosphoryl)nicotinate 2''(B): 1.0394 g, 69%, $R_f=0.09$ (silica gel on PET foils with fluorescent indicator 254 nm, CHCl_3 -ethyl acetate=5:1 v/v, UV lamp for visualization). ^1H NMR (300 MHz, CDCl_3): δ 1.26 (d, $^3J_{\text{HH}}=5.9\text{Hz}$, 6H, POCCH_3), 1.40 (d, $^3J_{\text{HH}}=5.9\text{Hz}$, 6H, POCCH_3), 3.98 (s, 3H, COOCH_3), 4.69-4.82 (m, 4H, POCH), 8.71 (d, $^2J_{\text{PH}}=13.7\text{Hz}$, 1H, pyridine ring), 9.11 (bs, 1H, pyridine ring), 9.34 (bs, 1H, pyridine ring). $^{31}\text{P}\{^1\text{H}\}$ NMR (121 MHz, CDCl_3): δ 12.29 (s). ^{13}C NMR (75 MHz, CDCl_3): δ 23.90 (d, $^3J_{\text{CP}}=16.2\text{Hz}$, POCC), 24.03 (d, $^3J_{\text{CP}}=16.2\text{Hz}$, POCC), 52.72 (s, COOC), 71.86 (d, $^2J_{\text{CP}}=5.8\text{Hz}$, POCC), 125.69 (d, $J_{\text{CP}}=24.7\text{Hz}$, pyridine ring), 126.87 (d, $J_{\text{CP}}=178\text{Hz}$, pyridine ring), 140.38 (d, $J_{\text{CP}}=9.0\text{Hz}$, pyridine ring), 153.38 (s, pyridine ring), 155.41 (d, $J_{\text{CP}}=12.3\text{Hz}$, pyridine ring), 164.89 (s, COOCH_3).

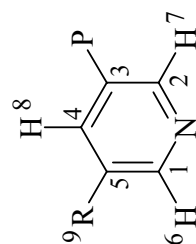
Tetraisopropyl 3,5-pyridinediylldiphosphonate 3''(B): 0.295 g, 29%, oil which solidified after standing, mp 57-58°C (hexane), $R_f=0.06$ (silica gel on PET foils with fluorescent indicator 254 nm, ethyl acetate, UV lamp for visualization). ^1H NMR (300 MHz, CDCl_3): δ 1.24 (d, $^3J_{\text{HH}}=6.2\text{Hz}$, 12H, POCCH_3), 1.38 (d, $^3J_{\text{HH}}=6.2\text{Hz}$, 12H, POCCH_3), 4.75 (d x septet, $^3J_{\text{HP}}=6.3\text{Hz}$, 4H, POCH), 8.44 (t, $^3J_{\text{HP}}=13.2\text{Hz}$, 1H, C4H), 9.08 (bs, 2H, C2H and C6H). $^{31}\text{P}\{^1\text{H}\}$ NMR (121 MHz, CDCl_3): δ 12.77 (s). ^{13}C NMR (75 MHz, CDCl_3): δ 23.96 (d, $^3J_{\text{CP}}=12.9\text{Hz}$, POCC), 71.86 (d, $^2J_{\text{CP}}=2.6\text{Hz}$, POCC), 126.50 (d x d, $^1J_{\text{CP}}=189\text{Hz}$, $^4J_{\text{CP}}=10.0\text{Hz}$, C3 and C5), 142.41 (t, $^2J_{\text{CP}}=8.8\text{Hz}$, C4), 154.81 (d, $^2J_{\text{CP}}=11.2\text{Hz}$, C2 and C6).

Table S1 Observed vibrational bands and their assignment.

1	2	3	Assignment
3425 w, b	3420 w, b	3421 w, b	v N–H
3250 w	3265 w	3260 m	v C–H (ring)
3168 w	3200 w	3190 w-m	
3120 w		3138 w-m	
3087 m, n	3075 m-s	3106 sh	
3056 w	3025 w	3084 m	
		3040 m	
2650 m, b		~2660 m, vb	v (P)O–H
2550 window	~2400 m-s, b	2500 window	
2450 m	2273 window		
	2145 m-s		
2400 min	2185 m-s		and/or
	2050 window	~2280 vb	v (C)O–H
2250 b	2007 m-s	2000 window	
	1950 m-s		
1735 b	1885 vb		
		~ 1000 s, b	v (P)O–H–O(P)
	1708 s		v C=O
1630 vs	1624 m-s	1620 m-s	v C=N (pyridyl ring)
1582 s	1565 w-m	1605 s	
1552 s	1540 w	1550 m-s	
			v C=C (pyridyl ring)
1463 s	1457 w		δ N–H
	1430 sh		
	1417 s, n	1405 s	
	1316 m, n		v C–OH and/or δ O–H
	1266 vs, b		v C–C
1370 s	1375 vw		v P=O
1355 m	1360 vw	1325 w	
1320 w, sh		1260 sh	
1275 m-s	1211 sh	1230 sh	and/or
1192 vs	1176 vs	1205 s	
1148 s, sh			v PO ₂
1126 vs	1125 s	1133 vs	
1114 vs	1100 m	1070 vs	
1060 s, sh	1041 vs	1026 s	v P–OH
1037 vs	1010 vs	1000 vs	
986 vs	978 m, sh	958 vs	
940 vs	955 vs	945 vs	v P–C
	937 vs		
921 sh	925 sh		
900 sh	915 vs		γ (C)O–H
	860 m	865 s, b	
807 vs	776 window	806 vs	
745 window	769 s, asym	758 sh	ring deformation and

730 s-vs, n	750 sh	701 window	ring torsion
685 vs, n	722 w	690	
	680 s, n	685	
	655 s, asym	676 vs	
628 s-vs,	587 s	570 s	
550 vs	556 m, sh	552 vs	
537 vs	536 m		
520 vs	502 m		PO ₃ deformations
447 s	479 m	475 vs	
	440 m	441 s	
	415 m	413 s	

Table S2 Calculated and measured chemical shifts.

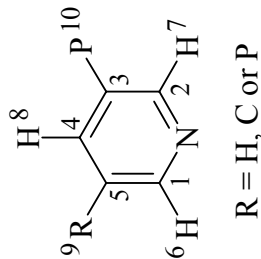


R = H, C or P

	1"			2" (A)			3" (A)						1	2	3
	Calculated		Exp.	Calculated			Exp.	Calculated			Exp.	Exp.	Exp.	Exp.	
	↑ ^a	↓		↓↑	↑↑	↑↓		↓↓↓	↓↑↑	↑↑↑					
1	156.88	157.09	152.89	157.16	157.72	158.07	157.88	152.31	161.3	159.89	158.83	155.05	148.31	148.93	142.43
2	156.20	155.02	152.11	159.98	159.87	161.15	159.29	154.05	161.29	160.64	158.29	155.05	142.06	151.71	142.43
3	129.74	130.90	124.94	132.01	131.19	130.10	132.35	124.25	129.78	130.00	131.35	125.32	136.57	136.26	134.99
4	144.50	142.00	139.45	143.84	144.95	143.19	142.99	139.84	146.55	147.74	145.61	142.73	142.18	139.00	148.37
5	127.10	127.08	123.38	133.25	134.83	133.57	134.91	123.28	129.78	130.44	131.64	125.32	127.33	132.45	134.99
6	8.41	8.40	8.69	9.02	8.82	9.08	8.83	9.34	8.71	8.60	8.55	9.10	8.62	8.70	8.73
7	8.50	8.50	8.89	8.65	8.65	8.76	8.66	9.10	8.71	8.68	8.54	9.10	8.71	8.63	8.73
8	7.73	7.56	8.02	7.91	8.36	7.74	8.21	9.71	7.77	7.85	7.84	8.39	8.58	8.23	8.66
9	7.04	6.99	7.32	-	-	-	-	163.04	-	-	-	-	7.91	162.41	-

^a The arrow indicates the direction of the P=O (↓ or ↑) or C=O (↑ or ↓) vectors with respect to the N_{py} atom: toward (when pointing down) or out (when pointing up).

Table S3 Calculated and measured coupling constants.



	1''		2'' (A)				3'' (A)				1	2	3	
	Calculated		Exp.	Calculated				Exp.	Calculated			Exp.	Exp.	Exp.
	$\uparrow^a$	$\downarrow$		$\downarrow\uparrow$	$\uparrow\downarrow$	$\uparrow\uparrow$	$\uparrow\downarrow$		$\downarrow\downarrow$	$\downarrow\uparrow$	$\uparrow\uparrow$			
10-1	-1.88	-1.79	~1.0	-2.03	-1.86	-1.98	-1.89	0	-1.51	-1.60	-1.64	5.47	0	0
10-2	9.21	13.48	12.15	8.08	13.42	9.31	13.74	12.60	13.46	11.16	7.64	11.07	11.55	14.92
10-3	131.94	150.48	187.95	139.57	132.12	132.92	145.37	124.55	131.79	133.13	145.73	188.90	162.75	174.22
10-4	8.50	2.63	8.25	8.04	3.23	9.02	3.12	9.10	4.50	6.92	9.46	8.92	~1	6.97
10-5	9.04	7.79	11.40	8.26	6.47	7.83	6.53	11.80	6.68	7.46	9.00	10.20	9.37	9.45
10-6	$2.46_{5J_{HP}}$	$2.59_{5J_{HP}}$	$2.39_{5J_{HP}}$	$2.16_{5J_{HP}}$	$2.21_{5J_{HP}}$	$2.31_{5J_{HP}}$	$2.45_{5J_{HP}}$	$1.89_{5J_{HP}}$	2.35	2.48	2.48	$2.32_{5J_{HP}}$	bs	$>1_{5J_{HP}}$
10-7	$4.98_{3J_{HP}}$	$6.09_{3J_{HP}}$	$6.40_{3J_{HP}}$	$4.84_{3J_{HP}}$	$5.45_{3J_{HP}}$	$4.58_{3J_{HP}}$	$5.55_{3J_{HP}}$	$6.39_{3J_{HP}}$	$5.48_{3J_{HP}}$	$5.71_{3J_{HP}}$	$4.72_{3J_{HP}}$	$6.39_{3J_{HP}}$	$7.92_{3J_{HP}}$	$7.73_{3J_{HP}}$
10-8	$12.47_{3J_{HP}}$	$11.26_{3J_{HP}}$	$13.10_{3J_{HP}}$	$13.23_{3J_{HP}}$	$11.41_{3J_{HP}}$	$13.16_{3J_{HP}}$	$12.15_{3J_{HP}}$	$13.58_{3J_{HP}}$	$11.29_{3J_{HP}}$	$11.21_{3J_{HP}}$	$13.41_{3J_{HP}}$	$13.18_{3J_{HP}}$	$10.44_{3J_{HP}}$	$11.92_{3J_{HP}}$
10-9	$3.07_{4J_{HP}}$	$2.79_{4J_{HP}}$	$3.44_{4J_{HP}}$	-	-	-	-	-	-	$3.99_{4J_{PP}}$	$6.00_{4J_{PP}}$	-	$2.63_{4J_{HP}}$	-

6-7	-1.23 $^4J_{HH}$	-1.20 $^4J_{HH}$	1.20 $^4J_{HH}$	-1.27	-1.31	-1.23	-1.29	1.00 $^4J_{HH}$	-1.13	-1.17	-1.21	~ 1.0 $^4J_{HH}$	~ 1.0 $^4J_{HH}$	bs $^4J_{HH}$	>1 $^4J_{HH}$
6-8	0.93 $^4J_{HH}$	0.87 $^4J_{HH}$	1.25 $^4J_{HH}$	1.37 $^4J_{HH}$	1.28 $^4J_{HH}$	1.32 $^4J_{HH}$	1.22 $^4J_{HH}$	1.91 $^4J_{HH}$	1.13 $^4J_{HH}$	1.11 $^4J_{HH}$	1.14 $^4J_{HH}$	1.97 $^4J_{HH}$	bs $^4J_{HH}$	bs $^4J_{HH}$	1.59 $^4J_{HH}$
6-9	4.89 $^3J_{HH}$	4.81 $^3J_{HH}$	4.90 $^3J_{HH}$	-	-	-	-	-	2.35 $^3J_{HH}$	4.82 $^3J_{HH}$	5.09 $^3J_{HH}$	6.39 $^3J_{HP}$	5.69 $^4J_{HH}$	-	7.73 $^3J_{HP}$
7-8	1.10 $^4J_{HH}$	1.12 $^4J_{HH}$	1.50 $^4J_{HH}$	0.92 $^4J_{HH}$	1.10 $^4J_{HH}$	0.95 $^4J_{HH}$	1.12 $^4J_{HH}$	1.92 $^4J_{HH}$	1.13 $^4J_{HH}$	1.17 $^4J_{HH}$	1.13 $^4J_{HH}$	1.97 $^4J_{HH}$	~ 1.0 $^4J_{HH}$	bs $^4J_{HH}$	1.59 $^4J_{HH}$
7-9	0.60 $^5J_{HH}$	0.54 $^5J_{HH}$	1.00 $^5J_{HH}$	-	-	-	-	-	2.35 $^5J_{HH}$	2.28 $^5J_{HH}$	2.40 $^5J_{HH}$	2.32 $^5J_{HP}$	>1 $^5J_{HH}$	-	>1 $^5J_{HP}$
8-9	7.64 $^3J_{HH}$	7.76 $^3J_{HH}$	7.80 $^3J_{HH}$	-	-	-	-	-	11.30 $^3J_{HH}$	12.40 $^3J_{HH}$	12.75 $^3J_{HH}$	13.18 $^3J_{HP}$	7.92 $^3J_{HH}$	-	11.92 $^3J_{HP}$

^a The arrow indicates the direction of the P=O ($\downarrow$ or $\uparrow$) or C=O ($\uparrow$ or $\downarrow$) vectors with respect to the N_{py} atom: toward (when pointing down) or out (when pointing up).