

Article title: Construction of helical networks by using multiple V-shaped mixed ligand systems

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Thermal analysis:

In order to characterize the compounds in terms of thermal stability, their thermal behaviors were studied by TGA. The experiment was performed on crystalline samples of **1, 2, 3, 4** and **6** under nitrogen atmosphere with a heating rate of 10°C/min. Both of the

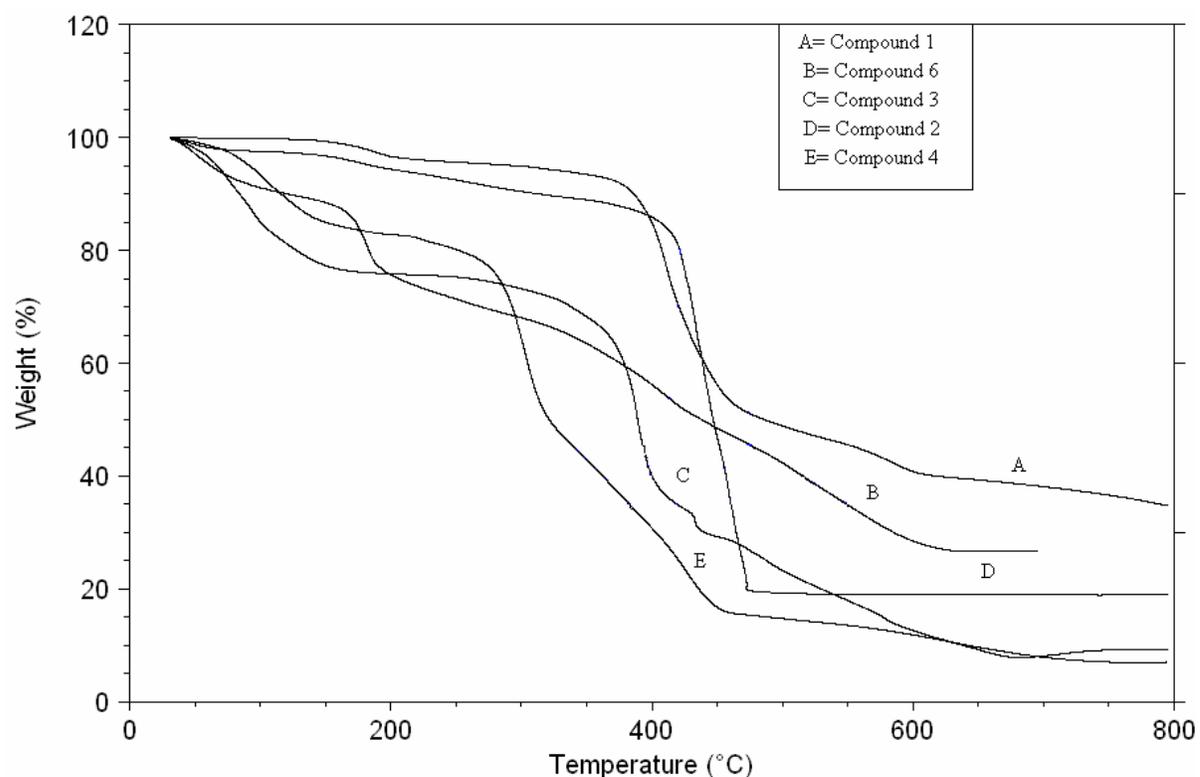


Figure ESI 1: TGA plot for compound **1,2,3,4** and **6**

compounds **1** and **2** do not show any significant weight loss upto 400° C and 450° C, respectively. After that both the compounds start degrading rapidly. This is in good agreement with the crystal structure supporting the absence of any trapped solvent molecule inside the framework. Compound **3**, however, shows almost 25% weight loss around 150° C indicating the removal of the solvent molecules. No further weight loss is observed upto 350° C and thereafter rapid degradation takes place. Compound **4** also shows an approximate 20% weight loss between room temperature and 275° C corresponding to the removal of the trapped solvent molecules and after that it starts degrading quickly. Compound **6** shows a 25% weight loss *ca.* 200° C corresponding to the removal of water molecules present in the framework and then shows a rapid degradation.

TOPOS data of Compound 1.

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1:C25 H24 N4 O5 Zn

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Structure consists of molecules (ZD1). The composition of molecule is Zn

Topology for ZD1

Atom ZD1 links by bridge ligands and has

Common vertex with	R(A-A)			f	Total SA
ZD 1 1.3012 1.6988 0.9167 (1 1 0)	9.649A	1	23.72		
ZD 1 -0.6988 -0.3012 0.9167 (-1-1 0)	9.649A	1	23.72		
ZD 1 1.6988 1.3976 0.7500 (1 1-1)	14.181A	1	26.28		
ZD 1 -0.3976 -0.6988 1.0833 (0-1 0)	14.181A	1	26.28		

Structural group analysis

Structural group No 1

Structure consists of 3D framework with ZD

Coordination sequences

ZD1: 1 2 3 4 5 6 7 8 9 10

Num 4 12 36 72 122 188 264 354 456 570

Cum 5 17 53 125 247 435 699 1053 1509 2079

TD10=2079

Vertex symbols for selected sublattice

ZD1 Point (Schlafli) symbol: {7^5;9}

Extended point symbol:[7(2).9(2).7(3).7(3).7(3).7(3)]

Point (Schlafli) symbol for net: {7^5;9}

4-c net; uninodal net

Topological type: qzd quartz-dual, "dense" net (topos&RCSR.ttd) {7^5;9} - VS [7(2).*.7(3).7(3).7(3).7(3)]
(67371 types in 9 databases)

Non-equivalent circuits

Circuit No 1; Type=7a; Centroid: (0.787,0.213,0.917)

Atom x y z

ZD1 0.3012 0.6988 0.9167

ZD1 -0.3976 -0.6988 1.0833

ZD1 -0.3976 -1.6988 1.0833

ZD1 0.3012 -0.3012 0.9167

ZD1 1.3012 0.6988 0.9167

ZD1 2.6988 1.3976 0.7500

ZD1 1.6988 1.3976 0.7500

Circuit No 2; Type=7b; Centroid: (0.215,0.785,0.917)

Atom x y z

ZD1 0.3012 0.6988 0.9167

ZD1 -0.3976 -0.6988 1.0833

ZD1 -0.3976 0.3012 1.0833

ZD1 0.3012 1.6988 0.9167

ZD1 -0.6988 0.6988 0.9167

ZD1 0.6988 1.3976 0.7500

ZD1 1.6988 1.3976 0.7500

Circuit No 3; Type=9a; Centroid: (0.679,0.321,0.917)

Atom x y z

ZD1 0.3012 0.6988 0.9167

ZD1 1.3012 1.6988 0.9167

ZD1 0.6024 0.3012 1.0833

ZD1 0.6024 -0.6988 1.0833

ZD1 1.3012 0.6988 0.9167

ZD1 0.3012 -0.3012 0.9167

ZD1 1.6988 0.3976 0.7500

ZD1 0.6988 0.3976 0.7500

ZD1 -0.6988 -0.3012 0.9167

Circuit No 4; Type=9b; Centroid: (0.234,0.766,0.917)

Atom x y z

ZD1 0.3012 0.6988 0.9167

ZD1 -0.6988 -0.3012 0.9167

ZD1 -1.3976 -1.6988 1.0833

ZD1 -1.3976 -0.6988 1.0833

ZD1 -0.6988 0.6988 0.9167

ZD1 0.3012 1.6988 0.9167

ZD1 1.6988 2.3976 0.7500

ZD1 2.6988 2.3976 0.7500

Elapsed time: 5.83 sec.

Table S-1: Selected Bond Lengths (Å) and Angles (°) for Compounds 1-9

Compound 1				Compound 2			
Zn ₁ -O ₁	1.939(5)	O ₁ -Zn ₁ -O ₁ ⁱ	106.2(3)	Co ₁ -O ₁	1.930(2)	O ₁ -Co ₁ -O ₁ ⁱ	108.90(15)
Zn ₁ -O ₁ ⁱ	1.939(5)	O ₁ -Zn ₁ -N ₁	114.4(2)	Co ₁ -O ₁ ⁱ	1.930(2)	O ₁ -Co ₁ -N ₁	114.73(11)
Zn ₁ -N ₁	1.974(5)	O ₁ ⁱ -Zn ₁ -N ₁	100.2(2)	Co ₁ -N ₁	1.995(3)	O ₁ ⁱ -Co ₁ -N ₁	100.23(1)
Zn ₁ -N ₁ ⁱ	1.974(5)	O ₁ -Zn ₁ -N ₁ ⁱ	100.2(2)	Co ₁ -N ₁ ⁱ	1.995(3)	O ₁ -Co ₁ -N ₁ ⁱ	100.23(11)
		O ₁ ⁱ -Zn ₁ -N ₁ ⁱ	114.4(2)			O ₁ ⁱ -Co ₁ -N ₁ ⁱ	114.73(11)
		N ₁ -Zn ₁ -N ₁ ⁱ	121.0(4)			N ₁ -Co ₁ -N ₁ ⁱ	118.35(18)
Compound 3				Compound 4			

Ni ₁ -O ₃ ⁱⁱ	2.021(3)	N ₃ ⁱⁱⁱ -Ni ₁ -O ₂			O ₄ -Cu ₁ -N ₁	94.21(14)
Ni ₁ -N ₃ ⁱⁱⁱ	2.056(4)	89.10(15)	Cu ₁ -O ₃		O _{3A} -Cu ₁ -N ₁	94.31(14)
Ni ₁ -N ₁	2.059 (4)	N ₁ -Ni ₁ -O ₂	1.969(3)		O _{4A} -Cu ₁ -N ₁	98.50(14)
Ni ₁ -O _{1W}	2.098(4)	163.82(14)	Cu ₁ -O ₄		O ₃ -Cu ₁ -Cu ₁ ^{iv}	82.27(10)
Ni ₁ -O ₂	2.117(3)	O _{1W} -Ni ₁ -O ₂	1.970(3)		O ₄ -Cu ₁ -Cu ₁ ^{iv}	84.75(10)
Ni ₁ -O ₁	2.148(3)	87.06(14)	Cu ₁ -O _{3A}		O _{3A} -Cu ₁ -Cu ₁ ^{iv}	80.77(10)
Ni ₁ -C ₂₄	2.476(5)	O ₃ ⁱⁱ -Ni ₁ -O ₁	1.972(3)		O _{4A} -Cu ₁ -Cu ₁ ^{iv}	86.41(10)
O ₃ ⁱⁱ -Ni ₁ -N ₃ ⁱⁱⁱ	88.03(14)	155.47(13)	Cu ₁ -O _{4A}		N ₁ -Cu ₁ -Cu ₁ ^{iv}	175.00(11)
O ₃ ⁱⁱ -Ni ₁ -N ₁	102.63(15)	N ₃ ⁱⁱⁱ -Ni ₁ -O ₁	2.158(4)		O ₁ -Cu ₂ -O _{2A}	88.90(14)
N ₃ ⁱⁱⁱ -Ni ₁ -N ₁	93.07(16)	91.36(14)	Cu ₁ -Cu ₁ ^{iv}		O ₁ -Cu ₂ -O ₂	166.75(13)
O ₃ ⁱⁱ -Ni ₁ -O _{1W}	91.24(14)	N ₁ -Ni ₁ -O ₁	2.6935(11)		O _{2A} -Cu ₂ -O ₂	88.22(14)
N ₃ ⁱⁱⁱ -Ni ₁ -O _{1W}	176.04(16)	O _{1W} -Ni ₁ -O ₁	1.960(3)		O ₁ -Cu ₂ -O _{1A}	89.00(14)
N ₁ -Ni ₁ -O _{1W}	90.88(16)	87.69(13)	Cu ₂ -O ₁		O _{2A} -Cu ₂ -O _{1A}	167.26(14)
O ₃ ⁱⁱ -Ni ₁ -O ₂	93.47(13)	O ₂ -Ni ₁ -O ₁	Cu ₂ -O _{2A}	1.971(3)	O ₂ -Cu ₂ -O _{1A}	90.97(14)
		62.00(12)	Cu ₂ -O ₂	1.972(3)	O ₁ -Cu ₂ -N ₃	97.23(14)
		O ₃ ⁱⁱ -Ni ₁ -C ₂₄	Cu ₂ -O _{1A}		O _{2A} -Cu ₂ -N ₃	99.88(15)
		124.28(15)	1.981(3)		O ₂ -Cu ₂ -N ₃	96.00(14)
		N ₃ ⁱⁱⁱ -Ni ₁ -C ₂₄	Cu ₂ -N ₃		O _{1A} -Cu ₂ -N ₃	92.85(15)
		89.99(16)	2.142(4)		O ₁ -Cu ₂ -Cu ₂ ^v	85.76(10)
		N ₁ -Ni ₁ -C ₂₄	Cu ₂ -Cu ₂ ^v	2.6824(11)	O _{2A} -Cu ₂ -Cu ₂ ^v	89.11(10)
		133.07(16)	O ₃ -Cu ₁ -O ₄			
		O _{1W} -N ₁ -C ₂₄	167.00(13)			
		87.20(15)	O ₃ -Cu ₁ -O _{3A}			
		O ₂ -Ni ₁ -C ₂₄	90.62(14)			
		30.82(14)	O ₄ -Cu ₁ -O _{3A}			
		O ₁ -Ni ₁ -C ₂₄	88.02(13)			
		31.19(14)	O ₃ -Cu ₁ -O _{4A}			
			87.78(14)			
			O ₄ -Cu ₁ -O _{4A}			
			90.69(13)			
			O _{3A} -Cu ₁ -			
			O _{4A}	167.18(13)		
			O ₃ -Cu ₁ -N ₁			
			98.78(14)			

Compound 5			Compound 6			
Zn ₁ -O ₁ ^{vi}	1.898(3)	O ₁ ^{vi} -Zn ₁ -O ₁ 110.6(2)	Co ₁ -O ₁	1.919(3)	O ₁ -Co ₁ -O ₁ ^{vii} 110.8(2)	
Zn ₁ -O ₁	1.898(3)	O ₁ ^{vi} -Zn ₁ - N ₁ ^{vi} 109.41(12)	Co ₁ -O ₁ ^{vii}	1.919(3)	O ₁ -Co ₁ - N ₁ ^{vii} 107.85(12)	
Zn ₁ -N ₁ ^{vi}	1.982(3)	O ₁ -Zn ₁ -N ₁ ^{vi} 108.23(11)	Co ₁ -N ₁ ^{vii}	1.984(3)	O ₁ ^{vii} -Co ₁ - N ₁ ^{vii} 110.32(13)	
Zn ₁ -N ₁	1.982(3)	O ₁ ^{vi} -Zn ₁ -N ₁ 108.23(11)	Co ₁ -N ₁	1.984(3)	O ₁ -Co ₁ - N ₁ 110.32(13)	
		O ₁ -Zn ₁ -N ₁ 109.41(12)			O ₁ ^{vii} -Co ₁ - N ₁ 107.85(12)	
		N ₁ ^{vi} -Zn ₁ -N ₁ 110.99(16)			N ₁ ^{vii} -Co ₁ -N ₁ 109.69(18)	
Compound 7			Compound 8			
Cd ₁ -O ₂	2.282(3)	O ₄ -Cd ₁ -O ₃ 54.16(10)	Ni ₁ -N ₁	2.100(3)	O ₁ -Ni ₁ -O ₁ 172.87(16)	
Cd ₁ -N ₁	2.301(3)	N ₃ -Cd ₁ -O ₃ 78.22(9)	Ni ₁ -O ₁	2.1034(15)	N ₁ -Ni ₁ -N ₃ 176.67(14)	
Cd ₁ -O ₄	2.351(3)	O ₂ -Cd ₁ -O ₁ 96.04(9)	Ni ₁ -N ₃	2.110(3)	N ₁ -Ni ₁ -N ₃ 90.40(8)	
Cd ₁ -N ₃	2.370(3)	N ₁ -Cd ₁ -O ₁ 82.35(9)	N ₁ -Ni ₁ -N ₁ 92.92(16)		O ₁ -Ni ₁ -N ₃ 86.19(9)	
Cd ₁ -O ₃	2.474(3)	O ₄ -Cd ₁ -O ₁ 74.97(10)	N ₁ -Ni ₁ -O ₁	93.73(9)	O ₁ -Ni ₁ -N ₃ 88.60(9)	
Cd ₁ -O ₁	2.545(3)	N ₃ -Cd ₁ -O ₁ 176.88(9)	N ₁ -Ni ₁ -O ₁	91.18(9)	N ₃ -Ni ₁ -N ₃ 86.27(15)	
Cd ₁ -C ₂₅	2.744(4)	O ₃ -Cd ₁ -O ₁ 104.57(8)				
O ₂ -Cd ₁ -N ₁ 126.86(10)		O ₂ -Cd ₁ -C ₂₅ 111.31(12)			Compound 9	
O ₂ -Cd ₁ -O ₄ 90.29(10)		N ₁ -Cd ₁ -C ₂₅ 121.77(11)	Cu ₁ -O ₁	1.9730(15)	N ₁ -Cu ₁ -O ₄ 162.74(7)	
N ₁ -Cd ₁ -O ₄ 138.43(10)		O ₄ -Cd ₁ -C ₂₅ 26.99(11)	Cu ₁ -N ₁	1.9795(17)	O ₁ -Cu ₁ -O ₃ 145.57(7)	
O ₂ -Cd ₁ -N ₃ 83.00(10)		N ₃ -Cd ₁ -C ₂₅ 93.21(11)	Cu ₁ -O ₄	2.0108(17)	N ₁ -Cu ₁ -O ₃ 102.17(7)	
N ₁ -Cd ₁ -N ₃ 95.84(10)		O ₃ -Cd ₁ -C ₂₅ 27.18(11)	Cu ₁ -O ₃	2.0287(17)	O ₄ -Cu ₁ -O ₃ 64.57(7)	
O ₄ -Cd ₁ -N ₃ 107.96(10)		O ₁ -Cd ₁ -C ₂₅ 89.90(11)	Cu ₁ -N ₃	2.1911(18)	O ₁ -Cu ₁ -N ₃ 104.40(7)	
O ₂ -Cd ₁ -O ₃ 130.45(10)			O ₁ -Cu ₁ -N ₁ 90.47(7)		N ₁ -Cu ₁ -N ₃ 97.29(7)	
N ₁ -Cd ₁ -O ₃ 100.55(9)			O ₁ -Cu ₁ -O ₄ 95.25(7)		O ₄ -Cu ₁ -N ₃ 97.03(8)	
					O ₃ -Cu ₁ -N ₃ 105.53(7)	

Symmetry operations: (i) $-y+1, -x+1, -z+1/6$; (ii) $x+1/2, y-1/2, z$; (iii) $-x-1, y, -z+1/2$; (iv) $-x, -y, -z$; (v) $-x-1, -y+1, -z+1$; (vi) $-x+5/4, y, -z+1/4$; (vii) $-x+1/4, y, -z+1/4$.