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Electronic Supplementary Information (ESI)

**Inorganic anion induced supramolecular architectures and
luminescent properties of flexible bis(pyridyl) based ionic salts**

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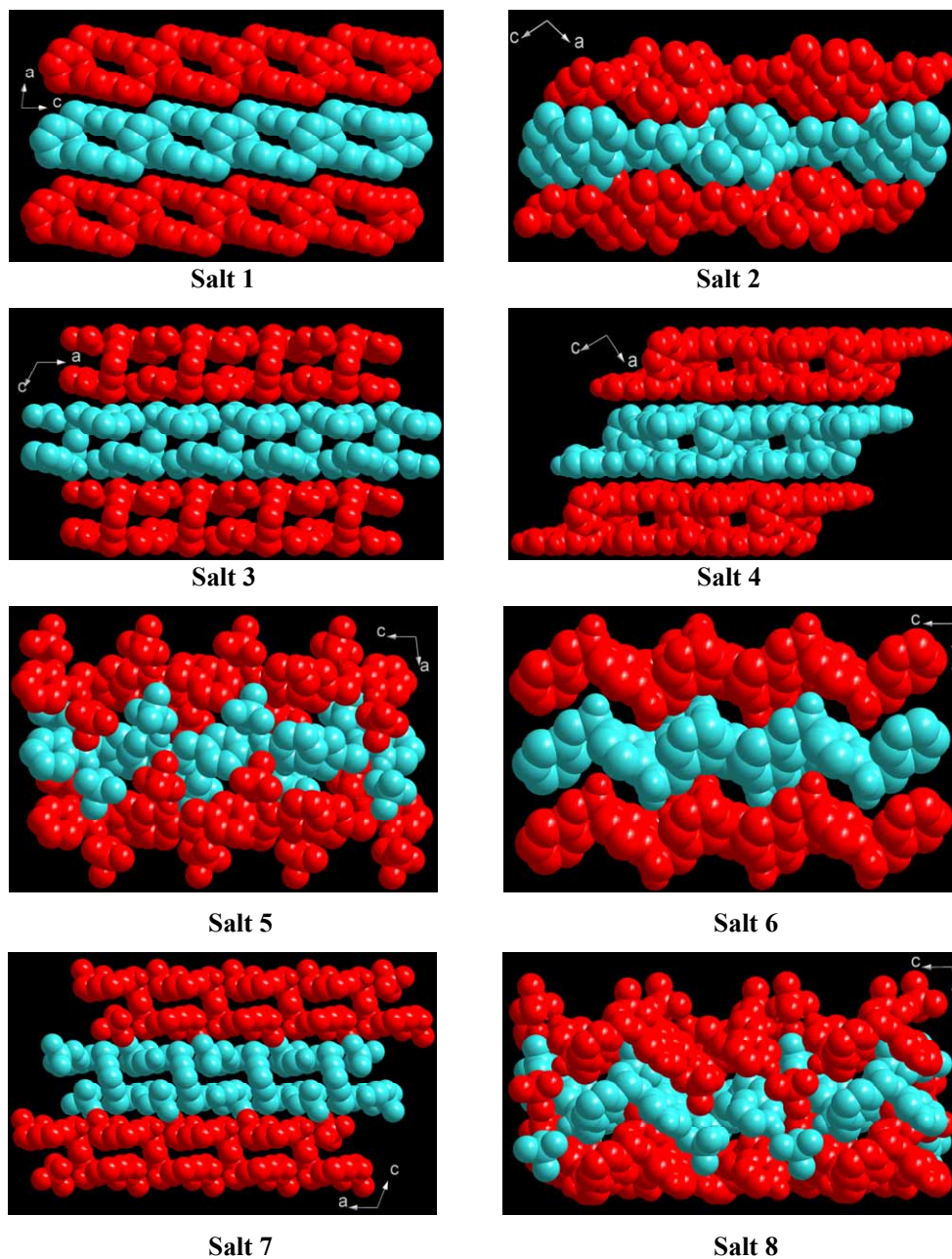


Fig. S1 Space-filling modes of the packing diagram in salts **1-8** along the [010] direction.

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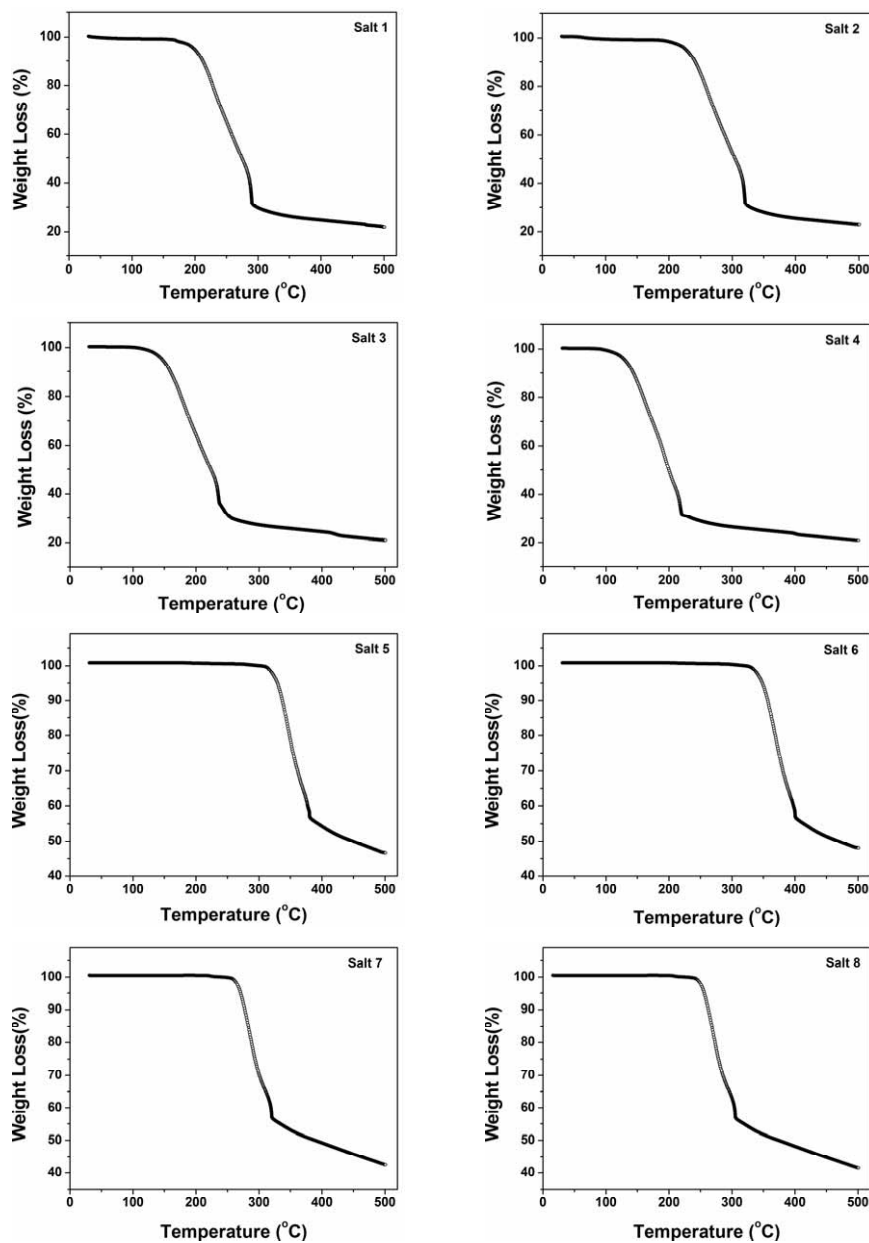


Fig. S2 TG curves of salts **1-8** at N₂ atmosphere.

The thermal stability of the eight salts was analyzed on crystalline samples by thermogravimetric analyses (TGA) from 30 to 500 °C at a rate of 10 °C min⁻¹ under N₂ atmosphere. As shown in Fig. S2, the thermal stability of the perchlorates is higher than the corresponding nitrates. The sharp weight loss of the eight salts is falling in the range of 162-290 °C, 190-320 °C, 107-258 °C, 90-220 °C, 305-380 °C, 320-402 °C, 247-320 °C, and 235-305 °C, respectively.

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Table S1 Selected bond angles ($^{\circ}$) data of salts **1-8**

		1	
O(2)-N(3)-O(1)	121.1(3)	O(1)-N(3)-O(3)	119.1(3)
O(2)-N(3)-O(3)	119.8(3)		
		2	
O(2)-N(3)-O(1)	118.9(2)	O(3)-N(3)-O(1)	118.8(3)
O(2)-N(3)-O(3)	122.3(3)		
		3	
O(2)-N(3)-O(1)	120.5(2)	O(3)-N(3)-O(1)	120.1(2)
O(3)-N(3)-O(2)	119.4(2)		
		4	
O(1)-N(5)-O(2)	121.2(5)	O(4)-N(6)-O(5)	122.5(5)
O(2)-N(5)-O(3)	118.6(5)	O(5)-N(6)-O(6)	118.0(5)
O(1)-N(5)-O(3)	120.2(5)	O(4)-N(6)-O(6)	119.5(5)
		5	
O(1)-Cl(1)-O(2)	107.3(6)	O(4)-Cl(1)-O(3)	111.0(4)
O(3)-Cl(1)-O(2)	107.6(4)	O(1)-Cl(1)-O(4)	112.1(5)
		6	
O(2)-Cl(1)-O(1)	110.2(2)	O(3)-Cl(1)-O(4)	109.4(2)
O(2)-Cl(1)-O(3)	109.6(2)	O(4)-Cl(1)-O(1)	108.7(2)
		7	
O(2)-Cl(1)-O(1)	112.6(4)	O(3)-Cl(1)-O(4)	110.1(4)
O(2)-Cl(1)-O(3)	110.2(3)	O(4)-Cl(1)-O(1)	106.2(3)
		8	
O(1)-Cl(1)-O(2)	110.4(3)	O(5)-Cl(2)-O(6)	109.4(3)
O(2)-Cl(1)-O(3)	108.1(3)	O(6)-Cl(2)-O(7)	108.1(3)
O(3)-Cl(1)-O(4)	108.3(3)	O(8)-Cl(2)-O(7)	108.7(3)
O(1)-Cl(1)-O(4)	110.6(3)	O(5)-Cl(2)-O(8)	109.1(3)