

Supplementary Information

Room-Temperature Ferromagnetism in Hierarchically Branched MoO₃ Nanostructures

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TABLE S1: Raman Spectra Comparison of Our Branched MoO₃ Nanostructures^a and Bulk MoO₃ Single Crystal.^b

branched MoO ₃ nanostructures /cm ⁻¹	bulk MoO ₃ single crystal /cm ⁻¹
994	998
817	822
664	668
468	473
375	381
335	338
282	285
244	247
214	219
196	200
157	160
126	131

^a Branched MoO₃ nanostructures in this experiment. ^b See ref 1.

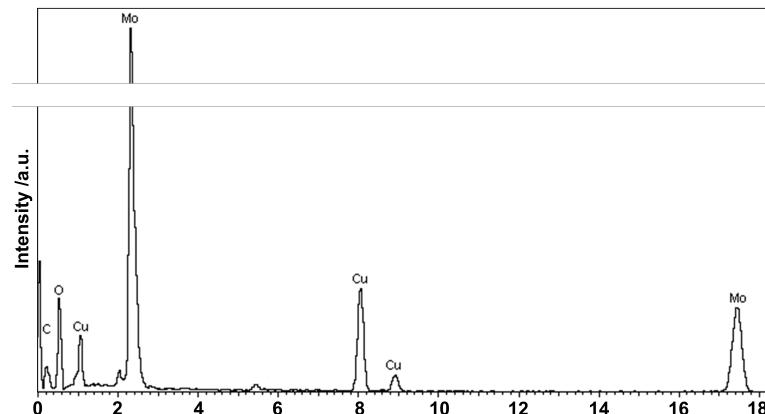


Fig. S1 EDS spectrum of the as-deposited samples before annealing.

The composition of the as-deposited samples before annealing was also analyzed by EDS, as shown in Fig. S1. Except the C and Cu peaks originating from the TEM grid, only the elements Mo and O are detected in the as-deposited samples before annealing. The stoichiometry Mo/O in the samples is about 3/2. This result is consistent with the XRD data, which indicates that the as-deposited sample consists of amorphous molybdenum oxide.

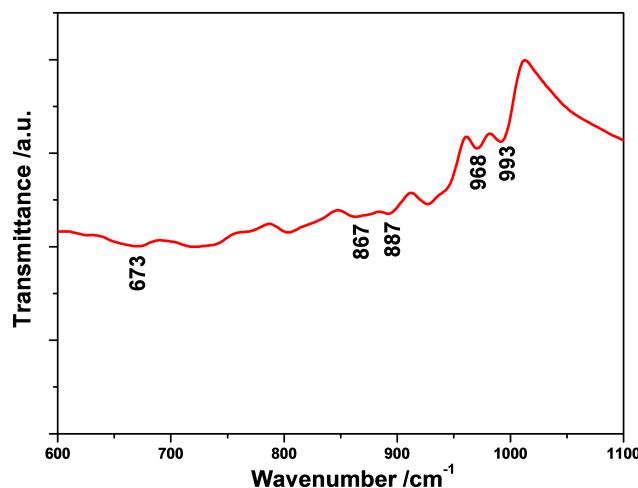


Fig. S2. FTIR spectrum of branched MoO₃ nanostructures.

Fig. S2 demonstrates the Fourier transform infrared (FTIR) spectrum of the branched MoO₃ nanostructures. The peak at 993 cm⁻¹ is assigned to a characteristic mode of the orthorhombic MoO₃.² The stretching mode of the Mo=O bond is located at 968 cm⁻¹, and the peak at 887 cm⁻¹ corresponds to

the stretching mode of the Mo atom with doubly connected O.³ The signal at 867 cm⁻¹ is a result of the stretching vibration of the O₍₃₎ atoms linked to two Mo atoms.⁴ The absorption band between 700 and 635 cm⁻¹ is associated with the bending mode of the O–Mo–O skeleton.³

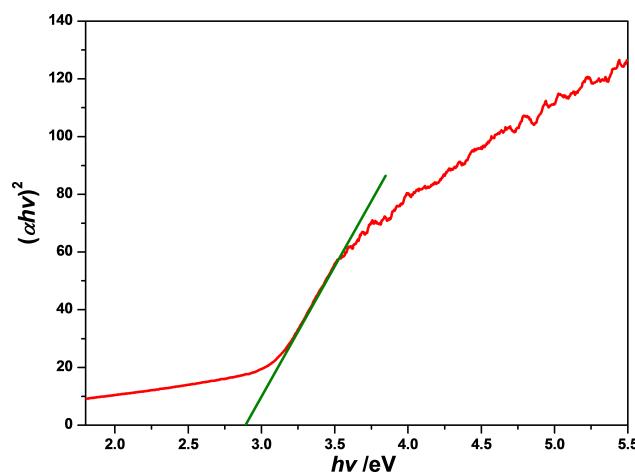


Fig. S3. $(\alpha h\nu)^2$ vs $h\nu$ curve for branched MoO₃ nanostructures.

References

- 1 M. A. Py, P. E. Achmid and J. T. Vallin, *Nuovo Cim.*, 1977, **38**, 271.
- 2 S. B.Umbarkar, T. V.Kotbagi, A. V. Biradar, R. Pasricha, J. Chanale, M. K. Dongare, A.-S. Mamede, C. Lancelot and E. Payen, *J. Mol. Catal. A: Chem.*, 2009, **310**, 150.
- 3 S. Hu, X. Ling, T. Lan and X. Wang, *Chem. Eur. J.* 2010, **16**, 1889.
- 4 L. Mai, B. Hu, W. Chen, Y. Qi, C. Lao, R. Yang, Y. Dai and Z. L. Wang, *Adv. Mater.*, 2007, **19**, 3712.