

Electronic Supplementary Information

Fabrication of nanoporous MTiO₃ (M=Pb, Ba, Sr) perovskite array films with unprecedented high structural regularity

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Experimental Procedures

Electropolishing and two step anodization to fabricate the nanoporous titanate

First, bare Ti foil (1x1.5cm, 0.25mm thickness) was cleaned by ultrasonication in acetone, isopropyl alcohol, and ethanol. The electropolishing of cleaned Ti foil was carried in a solution with perchloric acid, butanol, and ethanol at 20V for 5min below -20°C. The two-step anodization process was performed in a two-electrode electrochemical cell equipped with Ti foil (anode) and platinum mesh (cathode) under constant voltage. The first potentiostatic anodization process was performed at 60V for 3h at 20°C in a solution of 0.3wt% NH₄F and 2 vol% deionized water dissolved in ethylene glycol. As-prepared first anodic TiO₂ was removed mechanically. The second anodization was carried on the textured Ti surface at 60V for 30min at 20°C in the same solution.

Hydrothermal treatment to obtain nanoporous MTiO₃ (M=Pb, Ba, Sr) perovskite arrays

The hydrothermal treatment was carried out in a Teflon-vessel fitted stainless steel reactor containing 0.002M Pb acetate trihydrate, 0.05M Ba hydroxide octahydrate, or 0.05M Sr hydroxide dissolved in 80ml CO₂-free water under N₂ atmosphere. The reactor was placed in a convection oven at 280°C ($PbTiO_3$), 200°C ($BaTiO_3$), and 200°C ($SrTiO_3$), for 6h without disruption. After the reaction, as-prepared thin film was washed with deionized water and dried in a vacuum oven.

Sample Characterization

The Top surface and cross sectional morphology images were obtained by FE-SEM (JEOL JSM-7401F). The crystal structure of the sample was determined by X-ray diffraction (XRD, PANalytical X'Pert diffractometer with an X'Celerator detector) with Cu K α radiation.

Perovskite titanate films reported in the literature¹⁻³

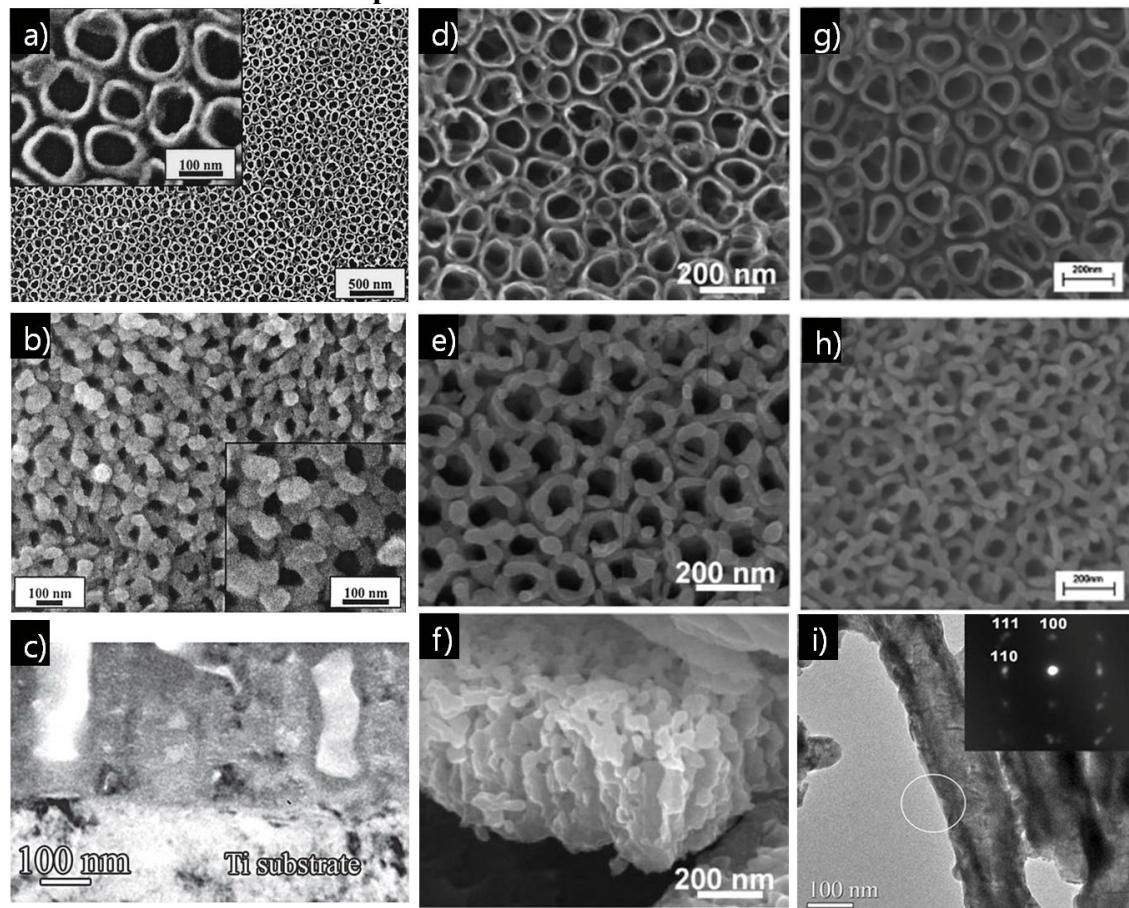


Figure S1. SEM images: (a) Anodic TiO₂ nanotubes¹, and (b,c) BaTiO₃ nanotubes by hydrothermal synthesis of anodic TiO₂ nanotubes shown in (a); (d) Anodic TiO₂ nanotubes², and (e,f) PbTiO₃ nanocellular structure by electrodeposition and thermal conversion of (d); (g) Anodic TiO₂ nanotubes³, and (h,i) PbTiO₃ nanotubes by hydrothermal method of anodic TiO₂ nanotubes shown in (g). c, f and i are cross sectional SEM images.

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- [3] Yang, Y.; Wang, X.; Zhong, C.; Sun, C.; Yao, G.; Li, L. *J. Am. Ceram. Soc.* 2008 91 3388-3390.