

Supporting Information

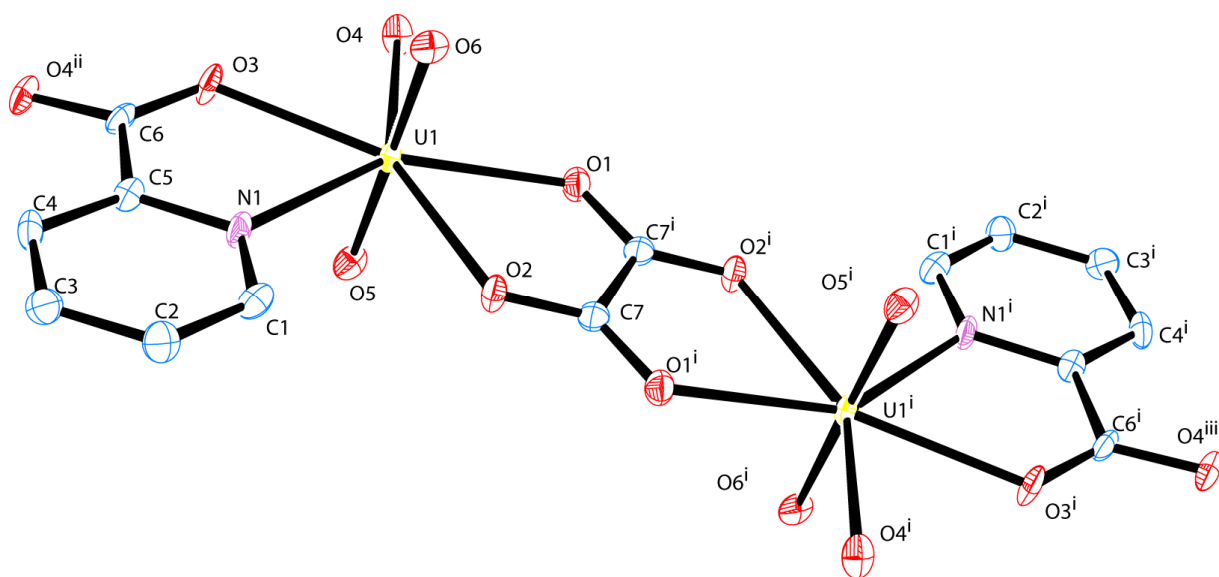


Figure S1. ORTEP illustration of **1**. Ellipsoids shown at 50% probability level. Superscript donates symmetry transformations $i = -x+1, -y+1, -z$; $ii = -x+2, y+1/2, -z+1/2$; $iii = x-1, -y+1/2, z-1/2$.

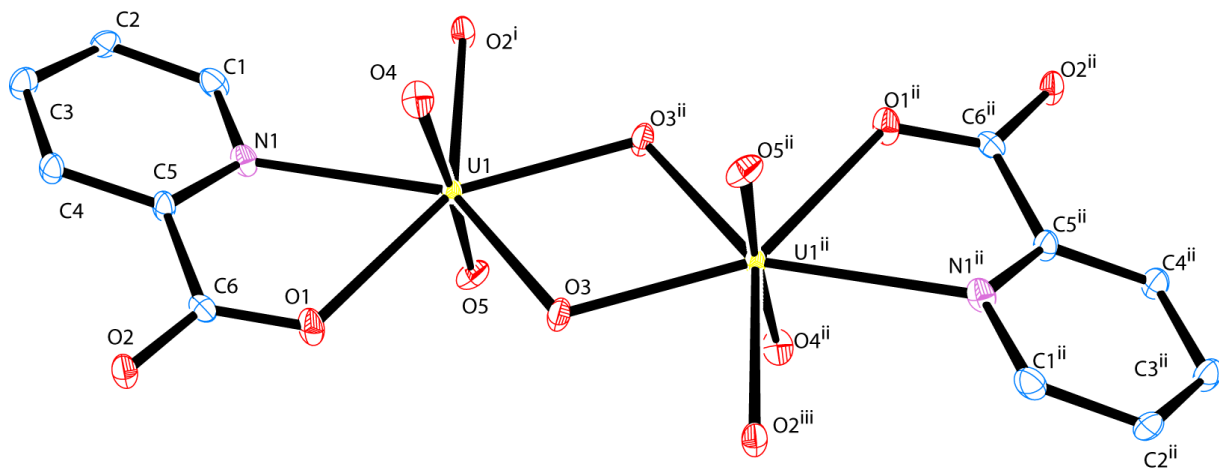


Figure S2. ORTEP illustration of **2**. Ellipsoids shown at 50% probability level. Superscript donates symmetry transformations $i = x-1, y, z$; $ii = -x+1, y-2, z+1$; $iii = -x+2, -y+2, -z-1$.

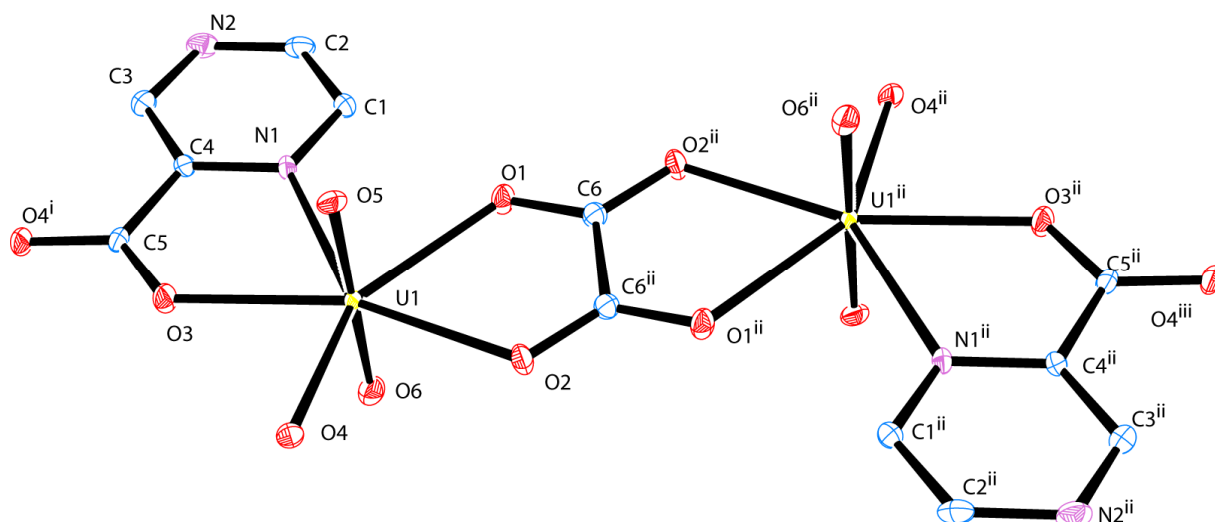


Figure S3. ORTEP illustration of **3**. Ellipsoids shown at 50% probability level. Superscript donates symmetry transformations $i = -x+1, -y-1/2, -z+3/2$; $ii = -x+2, -y+1, -z+2$; $iiix+1, -y+3/2, z+1/2$.

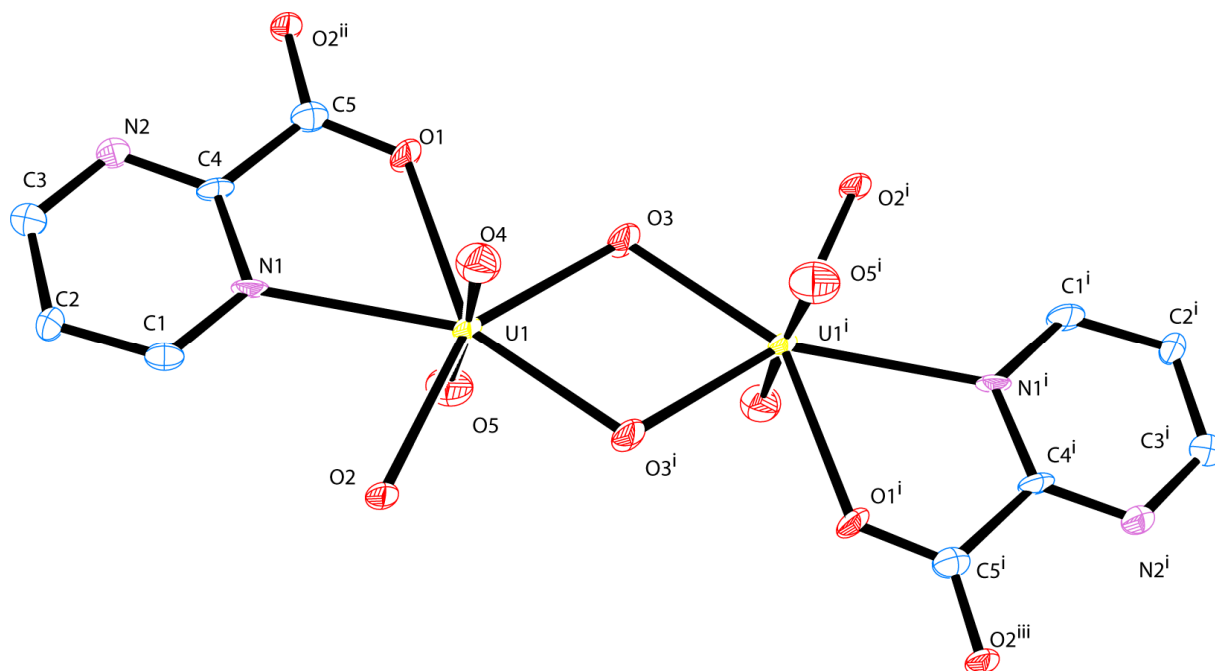


Figure S4. ORTEP illustration of **4**. Ellipsoids shown at 50% probability level. Superscript donates symmetry transformations $i = -x+2, -y+1, -z+1$; $ii = x-1, y, z$; $iii = x+3, -y+1, -z+1$.

Powder X-Ray Diffraction

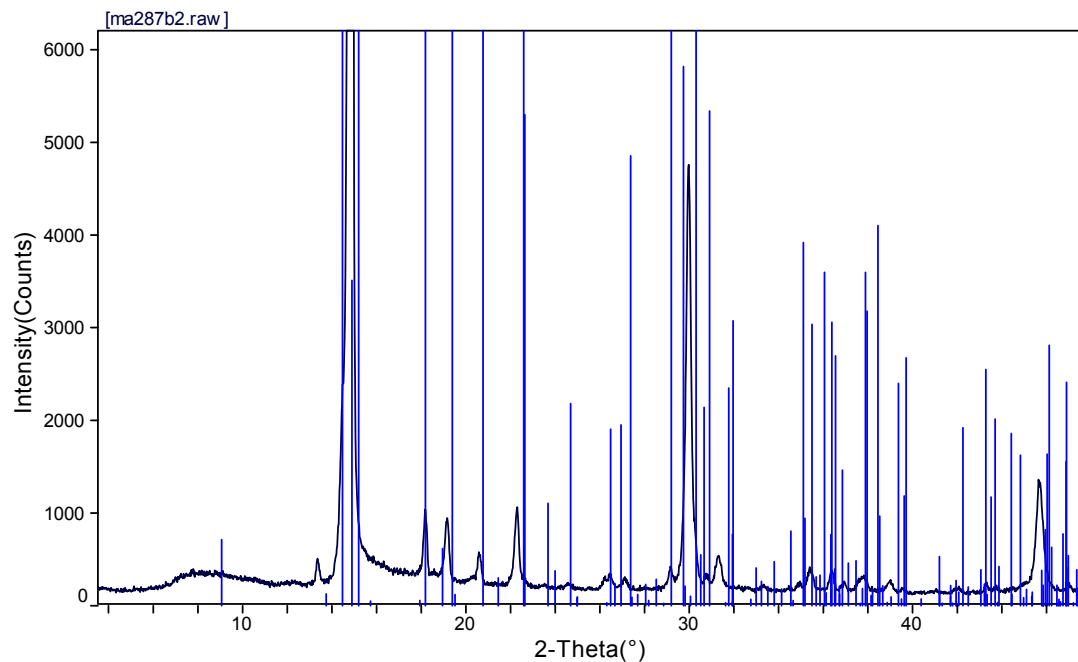


Figure S5. Observed PXRD pattern (black) for Compound **1** compared to calculated from CIF (blue)

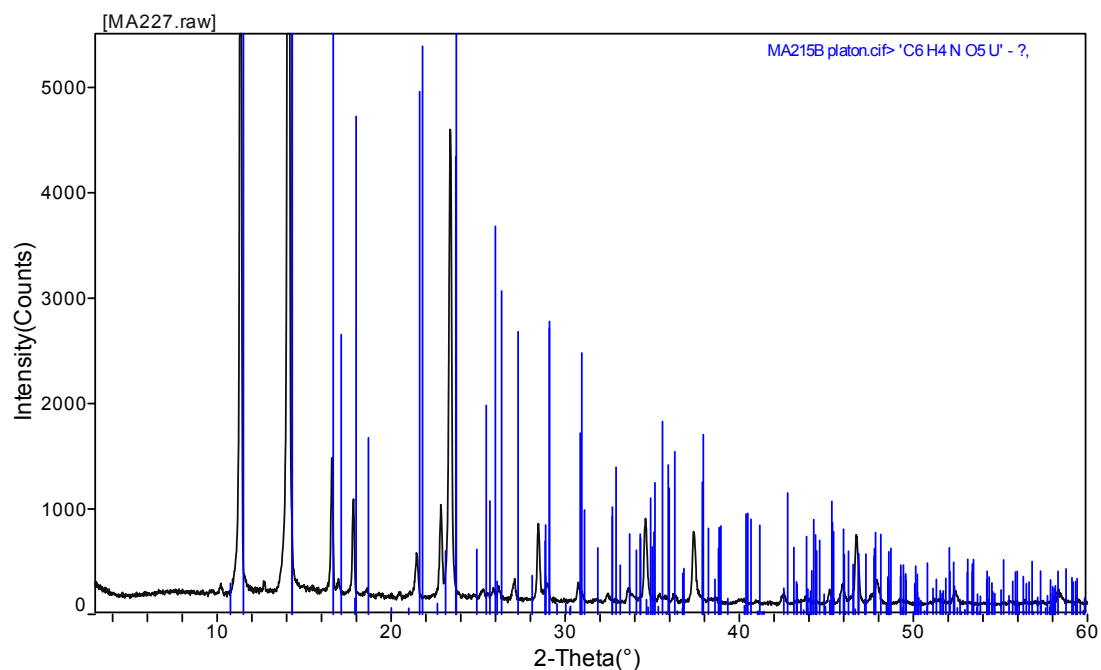


Figure S6. Observed PXRD pattern (black) for Compound **2** compared to calculated from CIF (blue)

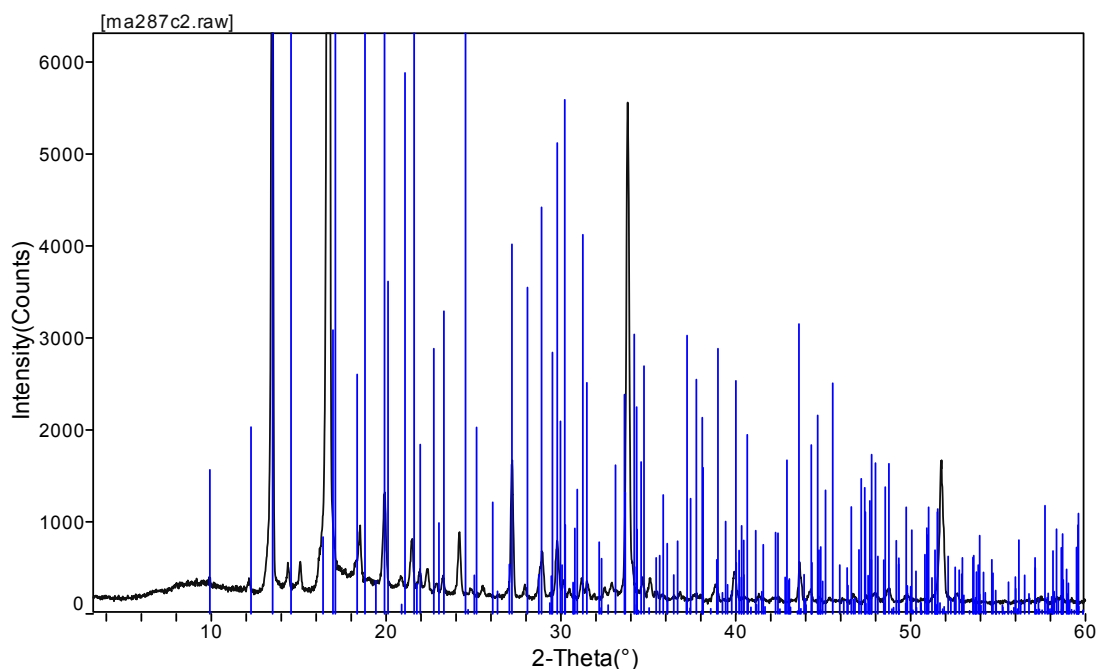


Figure S7. Observed PXR D pattern (black) for Compound **3** compared to calculated from CIF (blue)

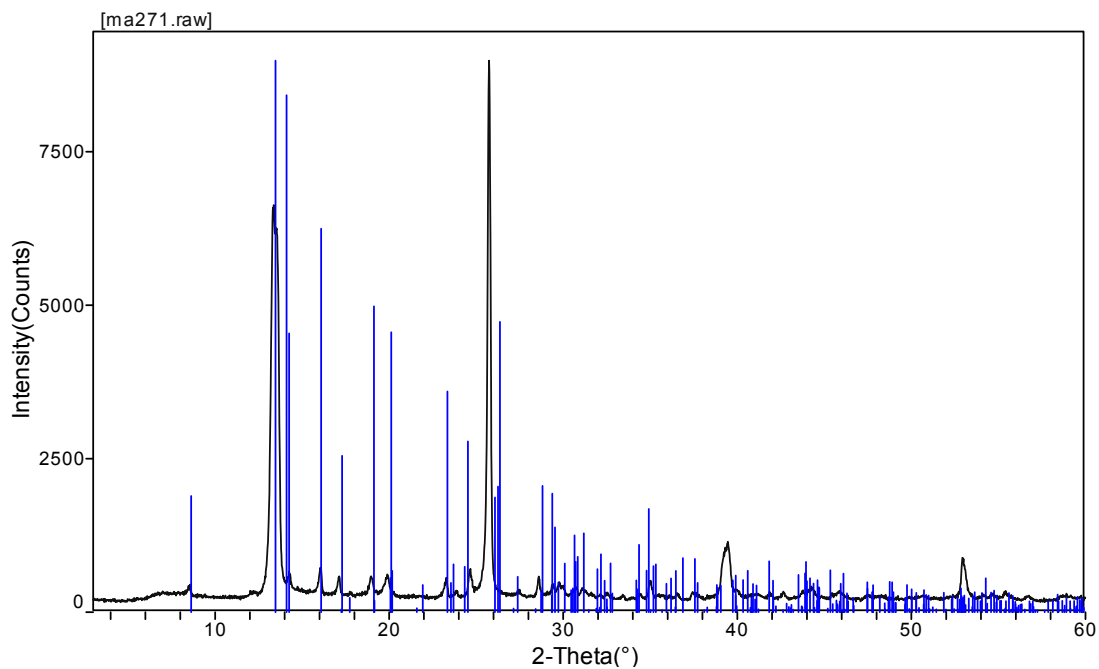


Figure S8. Observed PXR D pattern (black) for Compound **4** compared to calculated from CIF (blue)

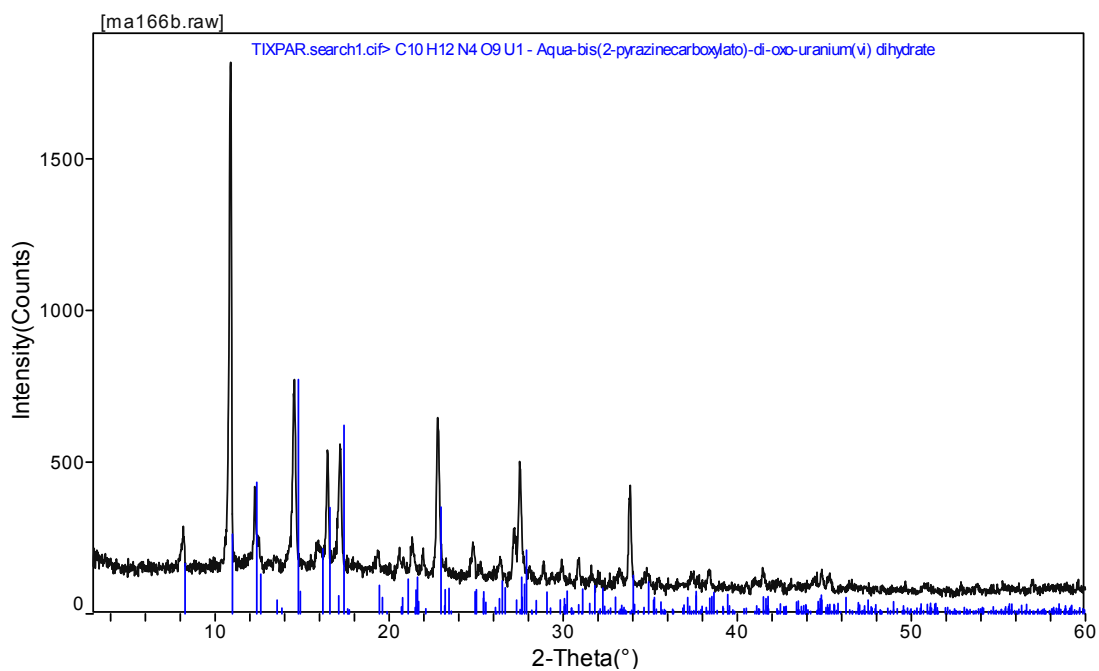


Figure S9. Observed PXR D pattern (black) for known uranyl pyrazine compound $[\text{UO}_2(\text{C}_5\text{H}_3\text{N}_2\text{O}_2)_2(\text{H}_2\text{O})] \cdot 2(\text{H}_2\text{O})$ (CSD Refcode: TIXPAR) compared to calculated from CIF (blue).

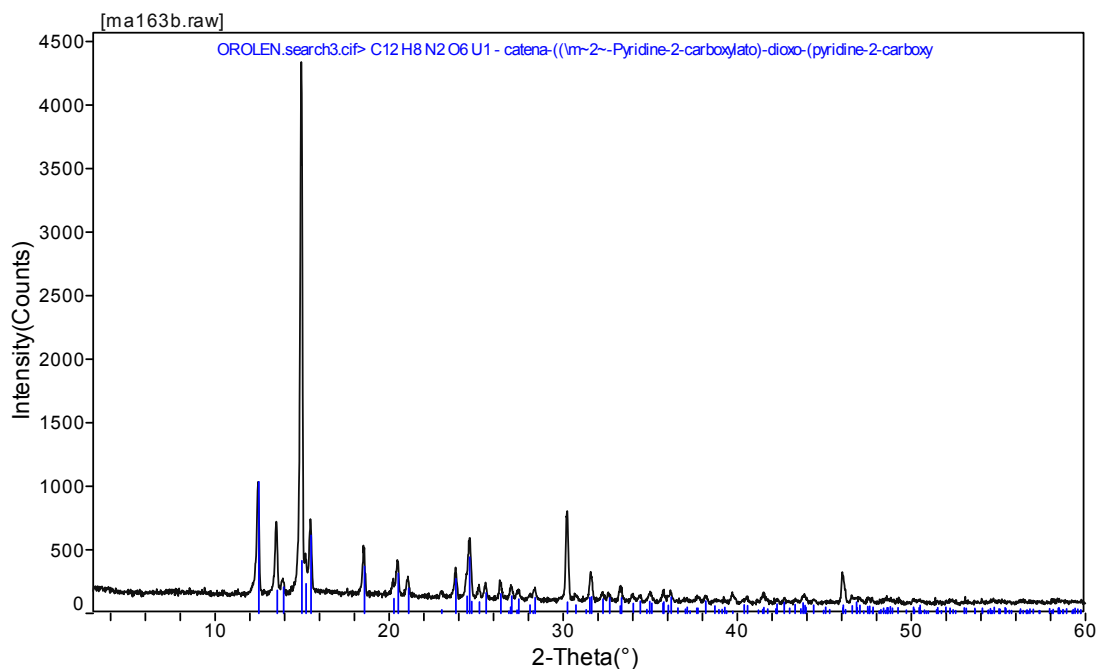


Figure S10. Observed PXR D pattern (black) for known uranyl pyridine material $[\text{UO}_2(\text{C}_6\text{H}_4\text{N}_1\text{O}_2)_2]^2$ compared to calculated from CIF (blue)

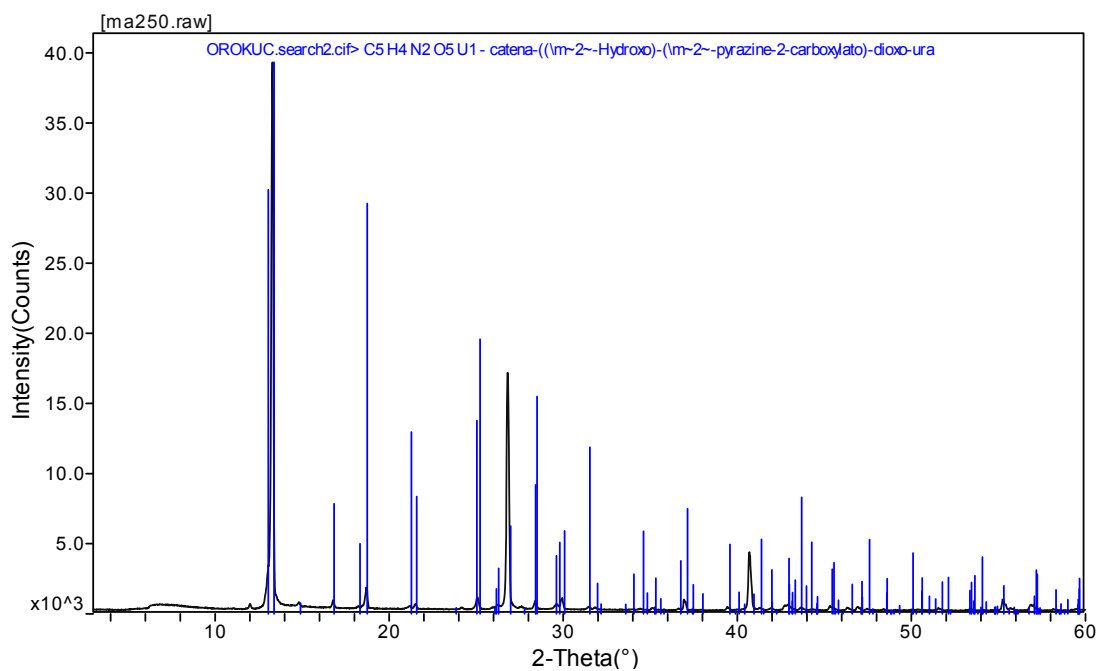


Figure S11. Observed PXRD pattern (black) for known uranyl pyrazine compound $[\text{UO}_2(\text{C}_5\text{H}_3\text{N}_2\text{O}_2)(\text{OH})]^{2-}$ compared to calculated from CIF (blue)

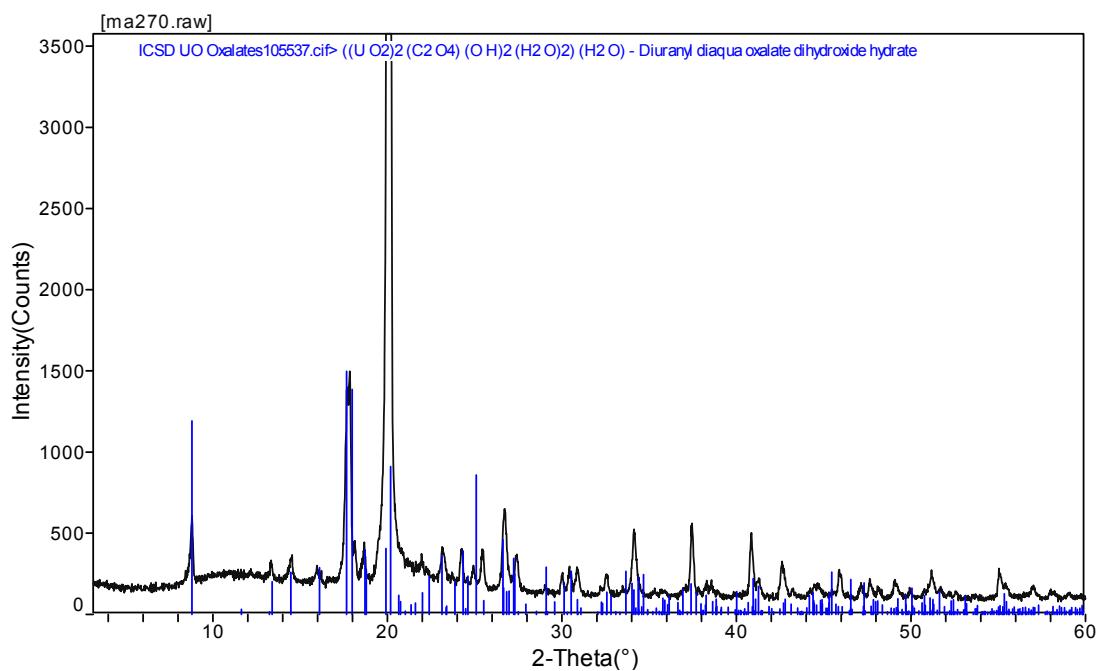


Figure S12. Observed PXRD pattern (black) for known uranyl oxalate compound $[(\text{UO}_2)_2(\text{C}_2\text{O}_4)(\text{OH})_2(\text{H}_2\text{O})] \cdot 2(\text{H}_2\text{O})$ (ICSD Refcode 105537).³ compared to calculated from CIF (blue).

Thermogravimetric Analysis

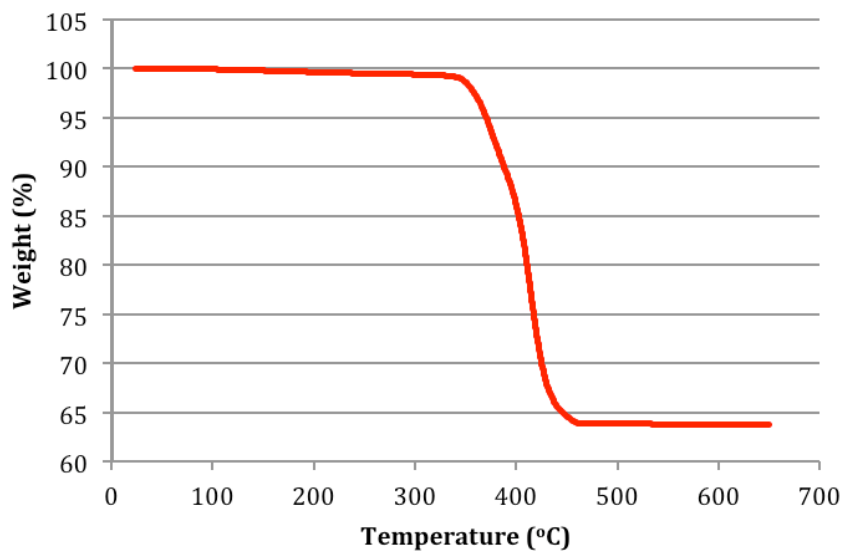
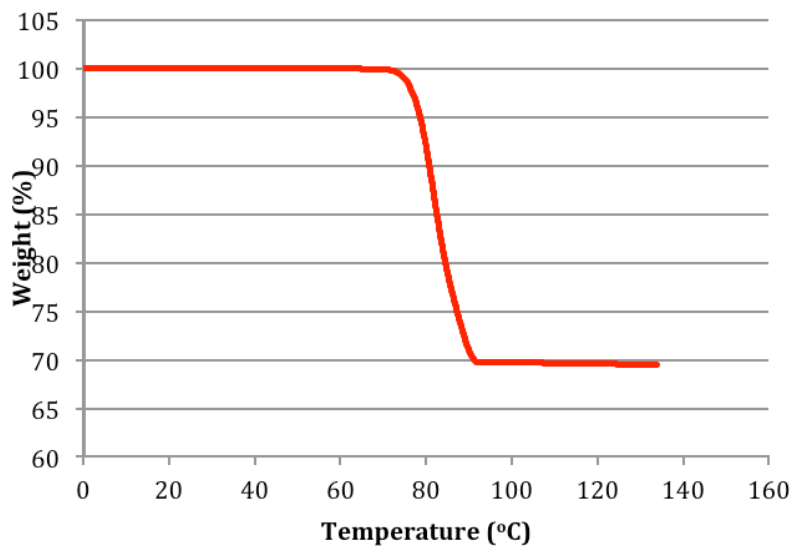
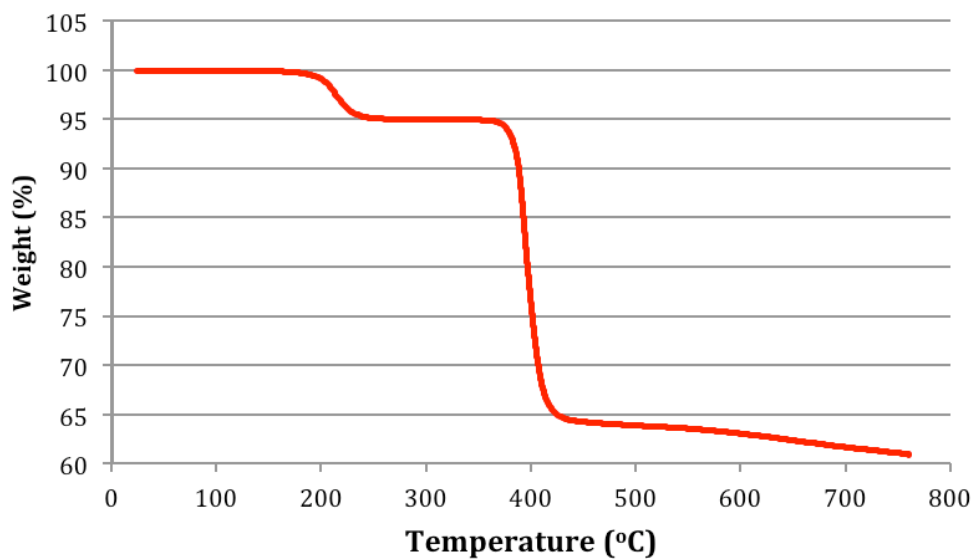


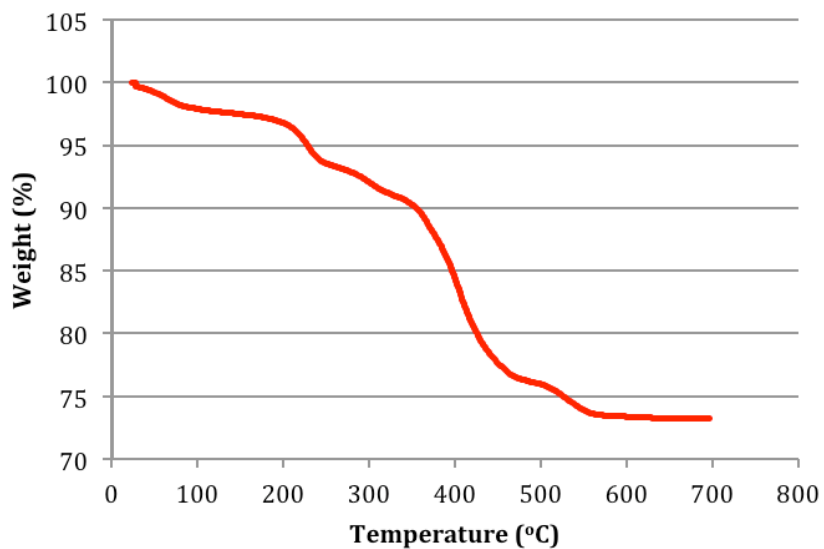
Figure S13. TGA of compound **1** showing 36% decrease in mass. Calculated mass of organic = 34%



SI Figure S14. TGA of compound **2** showing 30% decrease in mass. Calculated mass of organic = 30%



SI Figure S15. TGA of compound **3** showing 38% decrease in mass. Calculated mass of organic = 35%



SI Figure S16. TGA of compound **4** showing 28% decrease in mass. Calculated mass of organic = 31%

Elemental Analysis

Compound 1

ELEMENT	OBS	CALC
C	19.44 %	19.26 %
N	3.08 %	3.21 %
H	1.01 %	0.92 %

Compound 2

ELEMENT	OBS	CALC
C	17.80 %	17.60 %
N	3.31 %	3.42 %
H	1.31 %	1.22 %

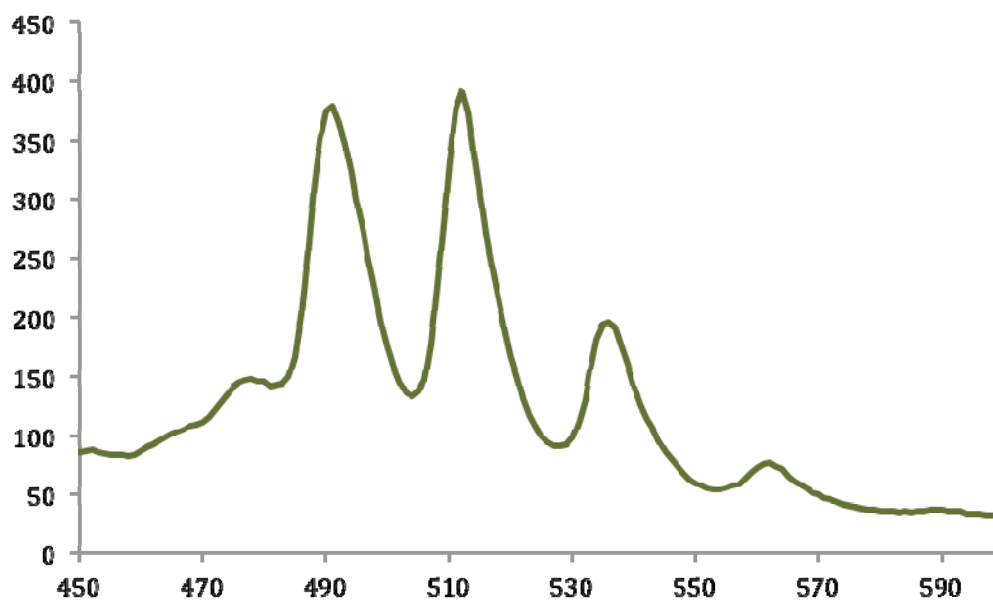
Compound 3

ELEMENT	OBS	CALC
C	15.99 %	15.89 %
N	6.08 %	6.18 %
H	1.18 %	0.66 %

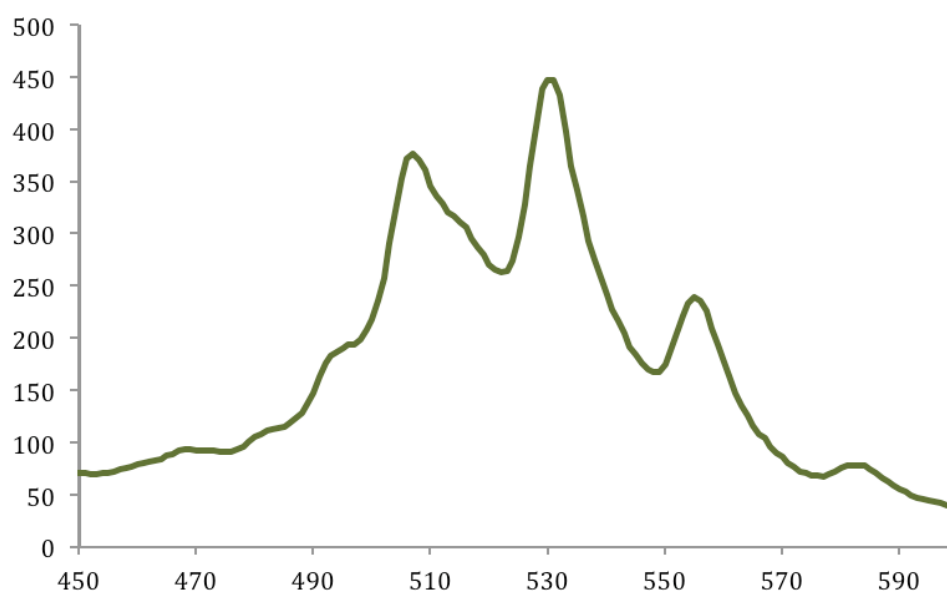
Compound 4

ELEMENT	OBS	CALC
C	15.12 %	14.08 %
N	6.11 %	6.61 %
H	1.42 %	0.93 %

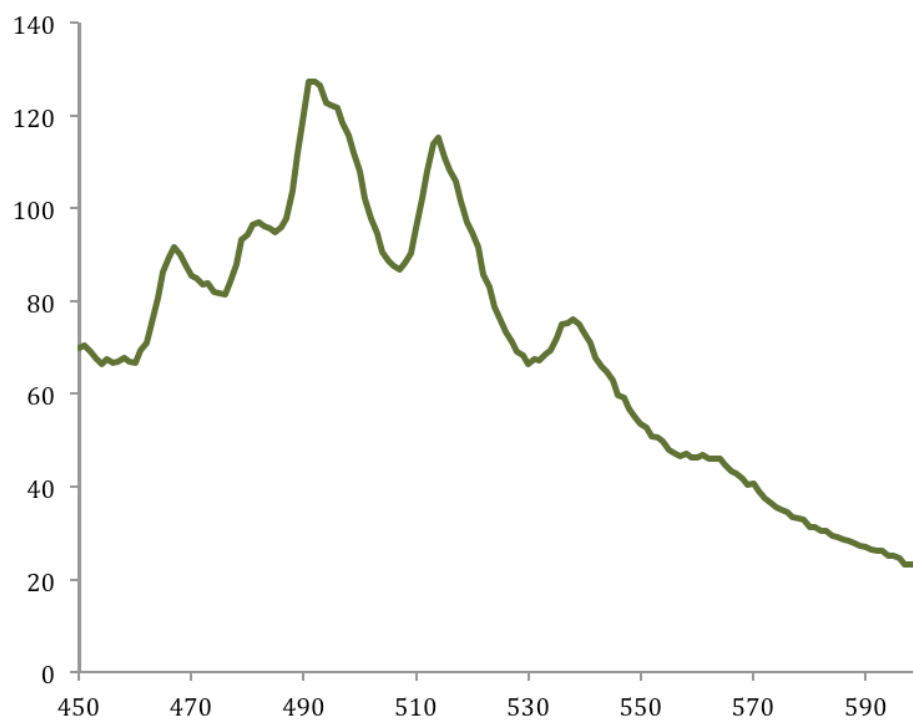
Fluorescence



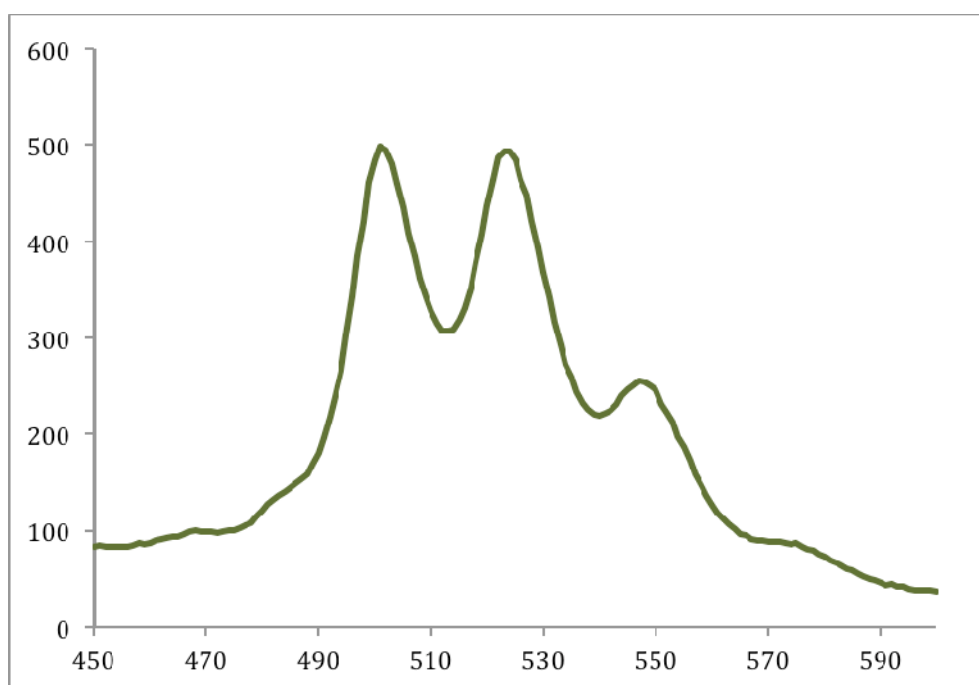
SI Figure 17. Fluorescence of compound 1



SI Figure 18. Fluorescence of compound 2



SI Figure 19. Fluorescence of compound 3



SI Figure 20. Fluorescence of compound 4

Bond valence calculation for bridging oxygen O3 in Compound 2^{4,5}

Atom	Bond Length	Valence Contribution
U1	2.306	0.604780212
<u>U2</u>	2.306	0.604780212
Total Valence		1.209560423

Bond valence calculation for bridging oxygen O3 in Compound 4^{4,5}

Atom	Bond Length	Valence Contribution
U1	2.333	0.574122038
<u>U2</u>	2.317	0.592097016
Total Valence		1.166219055

1. N. W. Alcock, T. J. Kemp, S. M. Roe and J. Leciejewicz, *Inorganica Chimica Acta*, 1996, 248, 241-246.
2. R. C. Severance, S. A. Vaughn, M. D. Smith and H.-C. zur Loye, *Solid State Sciences*, 2011, 13, 1344-1353.
3. L. Duvieubourg, G. Nowogrocki, F. Abraham and S. Grandjean, *Journal of Solid State Chemistry*, 2005, 178, 3437-3444.
4. N. E. Brese and M. O'keeffe, *Acta Crystallographica Section B-Structural Science*, 1991, 47, 192-197.
5. P. C. Burns, R. C. Ewing and F. C. Hawthorne, *Canadian Mineralogist*, 1997, 35, 1551-1570.

