Electronic Supplementary Information

Probing the O...Br-Br halogen bonding in the x-ray crystal structures with *ab initio* calculations

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Cartesian coordinates for 1-6 at MP2/6-311+G* level of theory.

1			
С	3.65471100	0.73330900	0.02185200
0	2.94497600	-0.50397700	-0.05783800
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Η	3.54571200	1.19637500	1.00670500
Η	3.30882900	-1.12521000	0.58044700
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Br	-2.01068600	0.14028300	0.01498200
2			

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Н	-4.00915600	1.19470800	-1.13604600
Н	-4.22829700	2.02221500	0.43104000
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Н	-1.80392000	2.02547200	-0.37336100
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$\mathbf{3}_{\mathrm{ax}}$

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4eq

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4ax

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Н	-3.71167900	2.01254100	-0.69369700	Н	3.10258300	1.71247500	0.00000000
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Cartesian coordinates for 1-6 at CBS-QB3 level of theory.

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0	2.89656300	-0.52235000	-0.03275200	Br	2.32012600	0.00010600	0.08618000
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Н	3.46634500	1.21904400	-0.91143900	0	4.36745500	0.00032200	0.45503600
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C H	-3.19760300 -4.27293800	-1.18575100 -1.22657000	0.20817300 -0.00391900	H H H	3.970764004.255464002.07051700	-2.02476100 -1.22911200	0.38543300 1.23196400
C H H	-3.19760300 -4.27293800 -2.70167300	-1.18575100 -1.22657000 -2.02751000	0.20817300 -0.00391900 -0.27269700	H H H	3.970764004.255464002.070517001.79424200	-2.02476100 -1.22911200 -2.03104300	0.38543300 1.23196400 -0.33814400
C H H H	-3.19760300 -4.27293800 -2.70167300 -3.04110700	-1.18575100 -1.22657000 -2.02751000 -1.24091500	0.20817300 -0.00391900 -0.27269700 1.29258400	H H H H	3.970764004.255464002.070517001.794242002.07003000	-2.02476100 -1.22911200 -2.03104300 1.22883500	0.38543300 1.23196400 -0.33814400 1.23189000
С Н Н С	-3.19760300 -4.27293800 -2.70167300 -3.04110700 -3.19668900	-1.18575100 -1.22657000 -2.02751000 -1.24091500 1.18639100	0.20817300 -0.00391900 -0.27269700 1.29258400 0.20793100	н Н Н Н Н	3.970764004.255464002.070517001.794242002.070030001.79333400	-2.02476100 -1.22911200 -2.03104300 1.22883500 2.03052000	0.38543300 1.23196400 -0.33814400 1.23189000 -0.33827400
С Н Н С Н	-3.19760300 -4.27293800 -2.70167300 -3.04110700 -3.19668900 -4.27189100	-1.18575100 -1.22657000 -2.02751000 -1.24091500 1.18639100 1.22819300	0.20817300 -0.00391900 -0.27269700 1.29258400 0.20793100 -0.00465200	н Н Н Н Н Н	 3.97076400 4.25546400 2.07051700 1.79424200 2.07003000 1.79333400 3.97026500 	-2.02476100 -1.22911200 -2.03104300 1.22883500 2.03052000 1.21849900	0.38543300 1.23196400 -0.33814400 1.23189000 -0.33827400 -1.17891900

Br

Br

3_{ax}

O C C

O C

С

Η

H H

H H

Η

H H

Br Br

4eq

С

O C

С

Η

Η

Н

Η

Η

H Br

Br

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			Ο	3.66078200	-1.22945400	-0.16356600
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-3.38955100	-1.23971300	-1.36986900	Н	-3.39153900	0.88458900	-1.61582400
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-3.38875400	1.23683000	-1.37288900	Н	-1.94697300	1.59689700	0.99718700
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0.73907800	-0.00065000	-0.31411800	С	-3.60107300	0.21946300	1.10328300
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			Н	-3.74553700	0.38643700	2.17033400
2.29299400	-1.15849500	0.10515100	Ο	-2.84027100	-0.98707200	0.98701200
1.70428100	0.02155600	-0.43582900	5			
2.28969600	1.20580500	0.13316700	С	-3.14364400	0.03276200	-0.00380500
3.80165200	1.18963300	-0.08785900	Ο	-2.22918000	-0.77072500	-0.00087600
2.10893700	-1.18260700	1.19261200	С	-4.57797700	-0.44725200	0.00734400
1.81370300	-1.99970100	-0.38971300	С	-2.91775000	1.52636400	-0.00627800
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1.80709100	2.04909700	-0.35941900	Н	-5.15646700	0.07805300	0.77262300
4.01179800	1.28125000	-1.15667200	Н	-5.04179700	-0.21862800	-0.95808900
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-0.89474400	-0.00023300	-0.16928600	Н	-3.08673200	1.90520600	1.00802800
-3.25351600	0.00228200	0.13327600	Н	-1.89685600	1.76310200	-0.30073200

S5

Br	0.32874100	-0.27221900	-0.00200300	0	1.07874300	2.88238900	0.00000000
Br	2.67396900	0.14241400	0.00154700	Br	0.00000000	0.45221700	0.00000000
				Br	-0.80154700	-1.77961200	0.00000000
6				Н	2.89972000	2.00241100	0.00000000
С	2.28352800	2.91901400	0.00000000	Н	2.82331500	3.88320900	0.00000000



Fig. S1 MP2/6-311+G* optimized geometries of σ -type interaction for methanol, dimethyl ether, acetone and formaldehyde with bromine molecule. [Distances are in Å] [BSSE corrected binding energies given in kcal/mol] [Gray = carbon; white = hydrogen; red = oxygen; dark red = bromine]



Fig. S2 Different geometries of methanol, 1,3-dioxane, acetone and formaldehyde with bromine molecule taken from x-ray crystal structure. [Distances are in Å] [Gray = carbon; white = hydrogen; red = oxygen; dark red = bromine]



Fig. S3 The maximum electrostatic potentials ($V_{S,max}$,kcal/mol) of bromine atom for 1,4dioxane:Br₂ and acetone:Br₂ complexes. The distances are given in A. [Gray = carbon; white = hydrogen; red = oxygen; dark red = bromine]



Fig. S4 MP2/6-311+G* optimized bridging geometry of acetone with bromine molecule. [Distances are in Å] [Binding energies (BSSE corrected) are in kcal/mol] [Gray = carbon; white = hydrogen; red = oxygen; dark red = bromine]



Fig. S5 CBS-QB3 optimized geometries of σ -type interaction for methanol, dimethyl ether and formaldehyde with bromine molecule. [Distances are in Å] [Binding energies are in kcal/mol] [Gray = carbon; white = hydrogen; red = oxygen; dark red = bromine]