Supplementary Information

Two-dimensional frameworks built from Single-Molecule Magnets

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Experimental details

Complex 1: $Mn(ClO_4)_2 \cdot 6H_2O$ (0.330g, 0.900mmol) and saoH₂(0.122g, 0.900mmol) were dissolved in MeOH (20mL). MeONa (1.000mmol) was then added under stirring (solution A) and the solution stirred for another 40 min. A methanolic solution (10mL) containing tmaH₃ (0.025g, 0.100mmol) was then added to solution A and the solution stirred for about 1 min. The resulting solution left undisturbed. X-ray quality, dark green, crystals of 1 were obtained in 40% yield. Elemental analysis for 1: Calculated (%) for C₅₅₂₅ H₅₁₂₅O₂₄₅ N₆ Mn₆ (vacuum dried complex 1): C: 43.63; H: 3.60; N: 5.53; Found: C: 43.58; H: 3.55; N: 5.59.

Complex **2**: To a methanolic solution of $Mn(ClO_4)_2 GH_2O_6(0.330g, 0.900mmol)$, saoH₂(0.122g, 0.900mmol) and tmaH₃(0.025g, 0.100mmol), MeONa (1.000mmol) was added in small portions under stirring to form a dark green slurry solution. The solution turns to clear dark green after a stirring period of 14 mins. The resulting solution left undisturbed to form dark green crystals of **2** in 40% yield. Elemental analysis for **2**: Calculated (%) for C_{51.33} H₄₈O_{22.67} MMn₆ (vacuum dried complex **2**): C: 42.78; H: 3.36; N: 5.83; Found: C: 42.72; H: 3.28; N: 5.90.



Fig. S1 Plot of χ_{M}^{T} versus T for complexes 1 and 2. The data could not be successfully fitted because each complex has more than one crystallographically independent [Mn] in the crystal. For comparison the data for [Mn] with similar torsion angles, reference 3c, are also plotted as 3 and 4.

Fig. S2 Plot of magnetisation $(M/N\mu_{\rm B})$ versus field (*H*) at 2 K for **1** and **2**. The data for each could not be successfully fitted using a simple model that assumes only population of the ground state. The reason, as with most members of this family, is the presence of multiple low-lying excited



states and the existence of more than one crystallographically independent [Mn] in the crystal.



Fig. S3 Plot of χ_{M}^{T} versus T for complex **1** at the indicated temperature and frequency ranges.

Fig. S4 Plot of $\chi_M T$ versus T for complex **2** at the indicated temperature and frequency





Fig. S5 Plot of χ_{M} " versus T for complex 2 at the indicated temperature and frequency ranges. Inset: Arrhenius plots constructed from the χ " data to afford $\tau_{0} = 5.6 \times 10^{-8}$ s and $U_{eff} \approx 24$ K.



Fig. S6 The IR spectra (KBr disks) of complex 1 (left) and 2 (right) in the 4000 - 400 cm⁻¹ region.

Table S1 [.] Dimensi	ions of the uniqu	e hydrogen	bonds (distances	in Å and angles in °) for 1
rable 51. Dimensi	ions of the uniqu	c nyurogen	bonus (uistances	In A and angles in) IOI I .

$D^{a}-H\cdots A^{b}$	$D^{a} \cdots A^{b}$	$H \cdots A^{\flat}$	$< D^{a}HA^{b}$	${}^{a}_{b}$ = donor atom.
O32K-H32K…	2.698(7)	1.88(1)	164(1)	A = acceptor
O1N				atom.
[3-x,3-y,-z]	2	1.07(1)	17(1)	
$O1W-H1W\cdots$	2.694(7)	1.86(1)	1/6(1)	
$\begin{bmatrix} 3 & 3 & 4 & 7 \end{bmatrix}$				
053D-H53D	2 660(7)	1 84(1)	163(1)	
01B	2.000(7)	1.0 ((1)	105(1)	
[1-x,2-y,1-z]				
O1Q-H1Q…O	2.773(7)	1.94(1)	169(1)	