

Supporting Information

Facile Synthesis of GdBO₃ Spindle Assemblies and Microdisks as Versatile Host Matrixes for Lanthanides Doping

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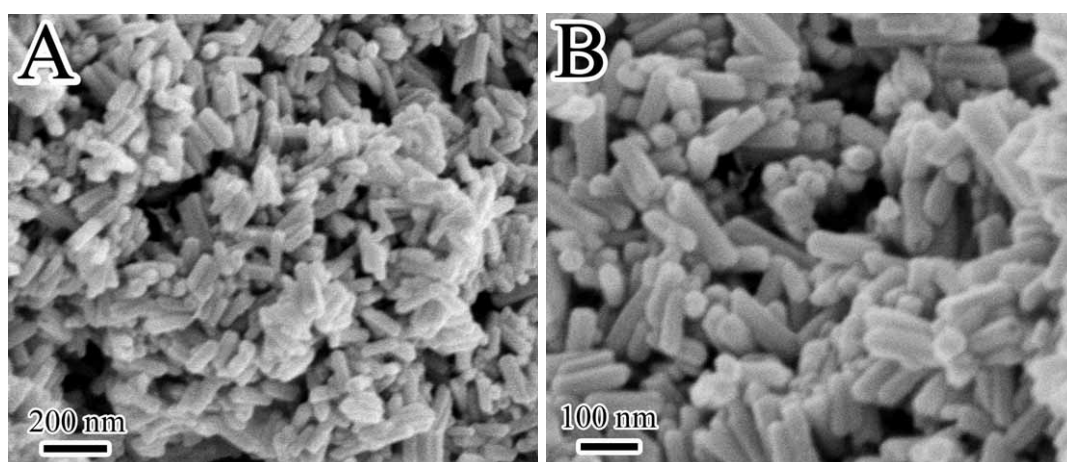


Figure S1. SEM images of Gd(OH)₃ nanorods synthesized under the same condition as that in Figure 2 except the absence of H₃BO₃. It indicates that the intermediate Gd(OH)₃ colloids may possess a rod-like shape in the synthesis of GdBO₃ assemblies.

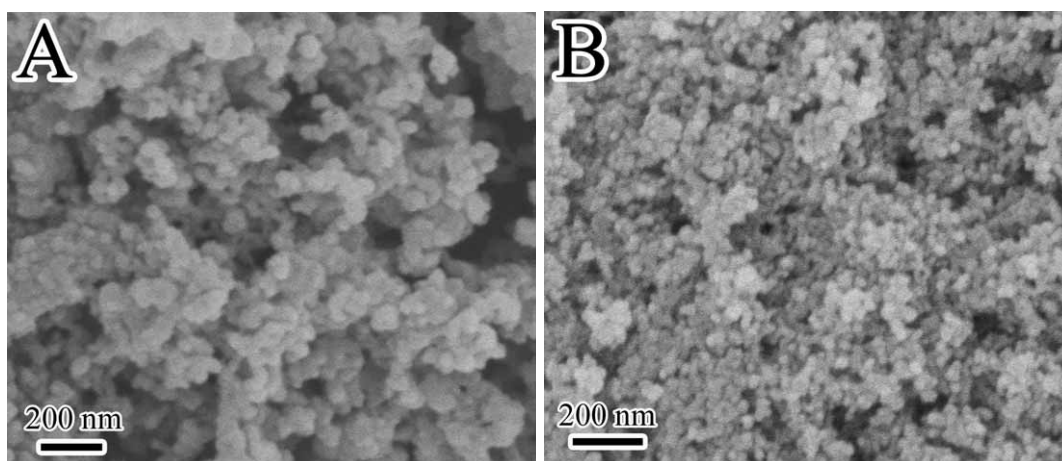


Figure S2. SEM images of GdBO_3 nanoparticles prepared under the same condition as that in Figure 2 except (A) the use of 2-mL ammonia (28. wt.%) to control pH; (B) the use of soluble $\text{Na}_2\text{B}_4\text{O}_7$ instead of H_3BO_3 powder. In both cases, the solution maintains alkaline during the entire process. As a result, GdBO_3 nanoparticles are of irregular shapes rather than rod-like shape, suggesting that the pH change in the synthesis holds the key to the formation of GdBO_3 spindle assemblies.

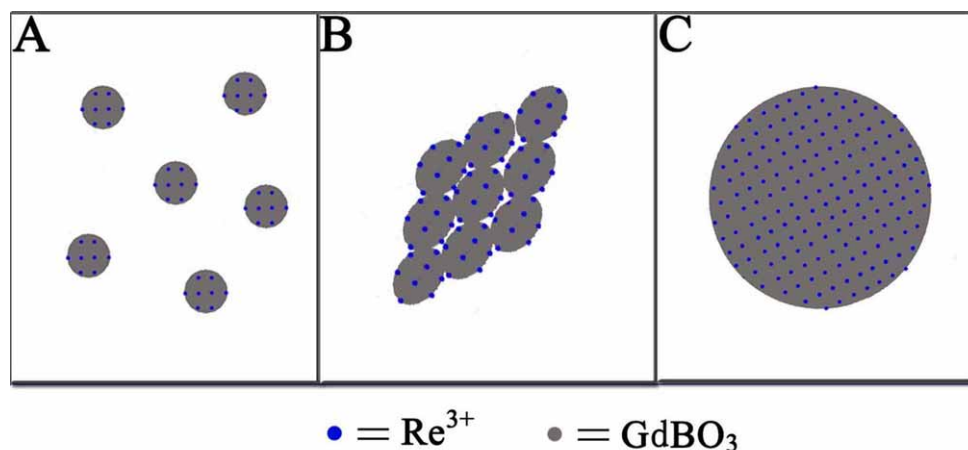


Figure S3. Schematic illustration for various GdBO_3 nanostructures containing luminescent lanthanide ions: (A) discrete nanocrystals; (B) a spindle assembly and (C) a single microdisk. The nanocrystals are assembled together to form a spindle assembly (from Panel A to Panel B). In the consideration that each assembly represents a single spot in bioimaging applications, local fluorescence from each spot in Panel B should be stronger than that from each discrete nanocrystal in Panel A. Furthermore, when multiple nanocrystals are assembled and then crystallized into a single-crystal microdisk at a bigger size (from Panel A to Panel C), surface defects will be greatly eliminated, enabling stronger fluorescent intensities in both the down-conversion or up-conversion spectra.

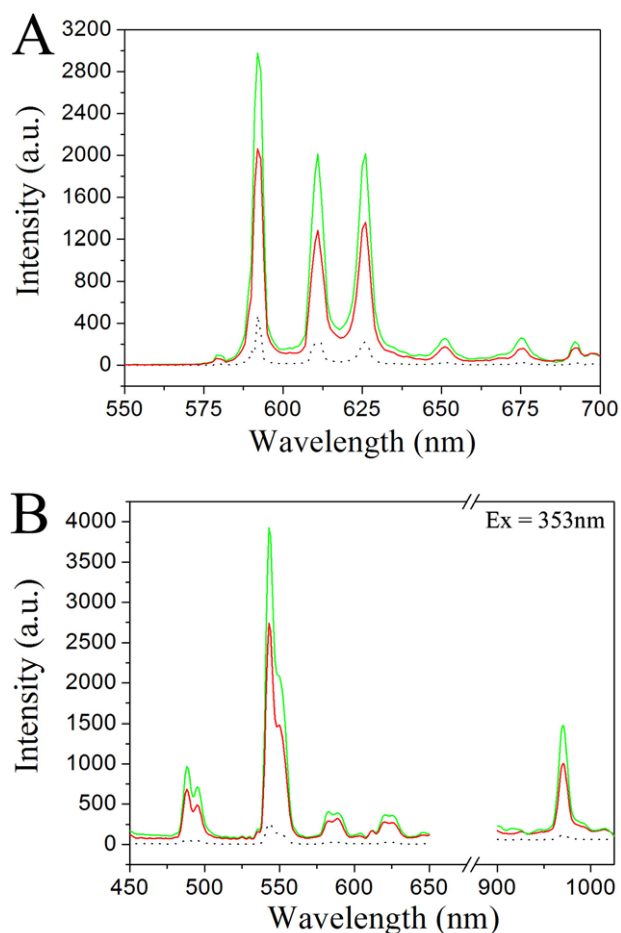


Figure S4. PL spectra of (A) GdBO₃:5%Eu and (B) GdBO₃:1%Ce,5%Tb,5%Yb samples with different structures: green line: microdisks; red line: spindle assemblies; dash line: small nanocrystals prepared with ammonia solution (see their morphologies in Fig. S2A). Fluorescence intensity of the spindle assemblies is nearly 5 times and 13 times stronger than the small nanocrystals in the Eu-doped and Ce/Tb/Yb co-doped samples, respectively. No obvious up-conversion fluorescence was observed from the small GdBO₃:10%Yb,1%Er nanocrystals under the same measuring condition.

Table S1. EDX analysis for the compositions of intermediate samples collected from different reaction stages during the hydrothermal process. The listed data of each sample are obtained by averaging those from three spots in the sample. It is found that the atomic percentage of element B in samples gradually increased in the first 24 h, suggesting that $\text{Gd}(\text{OH})_3$ colloids were increasingly transformed into GdBO_3 as the reaction proceeded. Note that element H could not be detected by the EDX technique. Signals from the supporting substrates have been excluded during the data analysis.

Atomic% Samples	Element Gd	Element O	Element B
0h	27%	73%	0%
3h	23%	74%	3%
6h	22%	67%	11%
12h	21%	64%	15%
24h	19%	62%	19%
36h	17%	65%	18%
48h	19%	63%	18%