

## Supplementary Materials

# Evolutionary Metadynamics: a Novel Method to Predict Crystal Structures

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### Potential used for Al<sub>2</sub>SiO<sub>5</sub> and MgSiO<sub>3</sub>

We studied Al<sub>2</sub>SiO<sub>5</sub> and MgSiO<sub>3</sub> with simple classical simulations based on the ionic shell model. The ionic shell model included pairwise interactions described by the following expression:

$$U_{ij}(R_{ij}) = \frac{z_i z_j}{R_{ij}} + b_{ij} \exp\left(-\frac{R_{ij}}{\rho_{ij}}\right) - \frac{c_{ij}}{R_{ij}^6} \quad (1)$$

a core-shell term for polarizable oxide ions (only for Al<sub>2</sub>SiO<sub>5</sub>):

$$U_s = k_s (\Delta r)^2 \quad (2)$$

and three-body angle-bending potentials (only for Al<sub>2</sub>SiO<sub>5</sub>):

$$U_{ijk} = k_b (\varphi - \varphi_0)^2 \quad (3)$$

The values of the parameters are given in Tables S1 and S2.

**Table S1. Parameters of the ionic shell model for Al<sub>2</sub>SiO<sub>5</sub>**

(Z<sub>Si</sub>=+4.0; Z<sub>Al</sub>=+3.0; Z<sub>O</sub>=-2.0)

Pair Potentials			
Interaction	B <sub>ij</sub> , eV	ρ <sub>ij</sub> ,	C <sub>ij</sub> , eV* Å <sup>6</sup>
O-O 2	023.8	0.2674	12.83
Al-O 1	474.4	0.3006	0
Si-O 1	283.9	0.3205	10.66
Shell parameters			

$k_s(O) = 74.9204 \text{ eV}/\text{\AA}^2$	$q(\text{shell}) = -2.84819$	$Q(\text{core}) = +0.84819$
Three-body potentials		
$k_b(\text{O-Al-O}) = k_b(\text{O-Si-O}) = 2.09724 \text{ eV}/\text{rad}^2$		

**Table S2. Parameters of the ionic shell model for MgSiO<sub>3</sub>**

(Z<sub>Si</sub>=+2.9043; Z<sub>Al</sub>=+1.9104; Z<sub>O</sub>=-1.6049)

Pair Potentials			
Interaction	$B_{ij}$ , eV	$\rho_{ij}$ ,	$C_{ij}$ , eV* Å <sup>6</sup>
O-O 2023	.800	0.2674	12.83
Mg-O 1007	.526	0.2866	0
Si-O 1108	.983	0.2827	0