

*Supporting Information for the Manuscript:*

**A series of five divalent zinc and cadmium coordination polymers based on a new bifunctional ligand: syntheses, crystal structures, and properties**

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**This PDF file includes Table S1– S2 and Fig. S1– S7.**

**Table S1.** Hydrogen Bonds parameters in **1–3**.

**Table S2.** Emission and excitation maxima wavelengths of **1–5**.

**Fig. S1.** Top view of the open-ended, hollow nanotube of **4**.

**Fig. S2.** TG curves for compounds **1–5**.

**Fig. S3** X-ray powder diffraction of complex **1** (a) simulated from single crystal data,  
(b) observed for complex **1**.

**Fig. S4** X-ray powder diffraction of complex **2** (a) simulated from single crystal data,  
(b) observed for complex **2**.

**Fig. S5** X-ray powder diffraction of complex **3** (a) simulated from single crystal data,  
(b) observed for complex **3**.

**Fig. S6** X-ray powder diffraction of complex **4** (a) simulated from single crystal data,  
(b) observed for complex **4**.

**Fig. S7** X-ray powder diffraction of complex **5** (a) simulated from single crystal data,  
(b) observed for complex **5**.

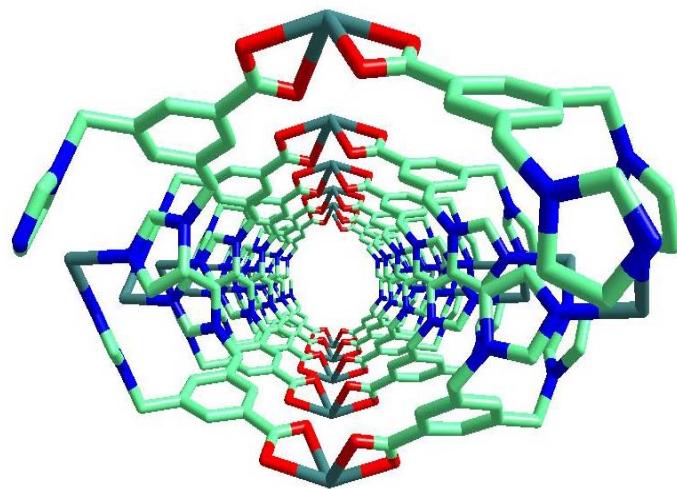
**Table1.** Selected Hyrdogen-Bond Geometry ( $\text{\AA}$ ) for **1-3** (in  $\text{\AA}$  and  $^{\circ}$ )

D-H $\cdots$ A	$d(\text{D}\cdots\text{H})$	$d(\text{H}\cdots\text{A})$	$d(\text{D}\cdots\text{A})$	$\angle\text{DHA}$
Compound <b>1<sup>a</sup></b>				
O(3W)-H(3WA)...O(1)	0.85	1.87	2.718(5)	170.6
O(2W)-H(2WA)...O(2)#4	0.84	1.87	2.687(4)	164.9
O(1W)-H(1WB)...O(2W)#5	0.84	1.92	2.713(5)	157.2
O(1W)-H(1WA)...O(3W)#6	0.84	1.88	2.691(5)	162.2
Compound <b>2<sup>b</sup></b>				
O(1W)-H(1WA)...O(2W)#4	0.77	1.97	2.738(2)	172.1
O(1W)-H(1WB)...O(3W)#5	0.79	1.91	2.685(4)	167.5
O(2W)-H(2WA)...O(2)#6	0.84	1.87	2.691(3)	163.2
O(3W)-H(3WB)...O(1)	0.85	1.87	2.719(3)	179.6
O(3W)-H(3WA)...O(1)#7	0.85	1.96	2.738(3)	151.8
Compound <b>3<sup>c</sup></b>				
O(1W)-H(1WA)...Cl(1)#6	0.85	2.54	3.352(5)	159.4
O(1W)-H(1WB)...O(2)#7	0.85	2.04	2.834(5)	154.7
O(2W)-H(2WB)...O(1W)#8	0.85	1.96	2.802(17)	170.2
O(2W)-H(2WA)...O(1W)#9	0.85	2.31	2.757(17)	113.0

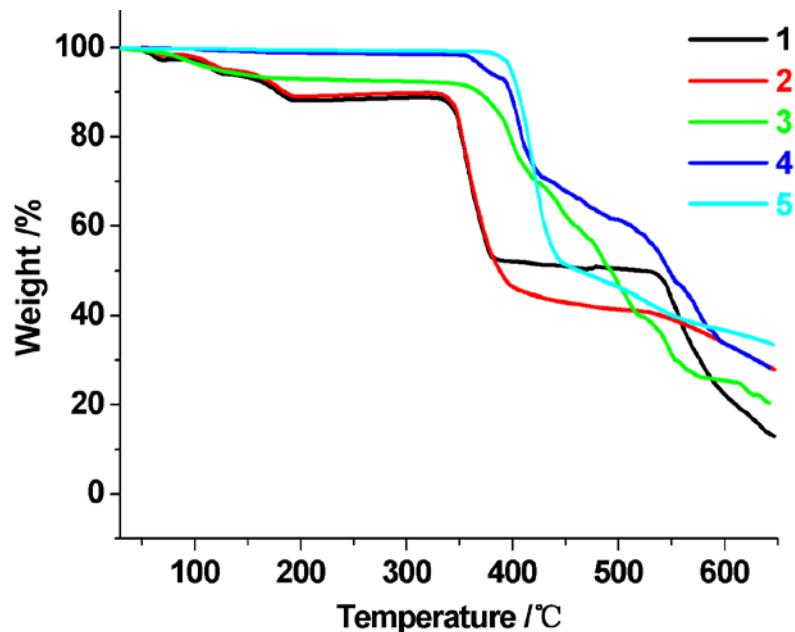
Symmetry codes: <sup>a</sup> #4 - $x+1/2$ , - $y+1/2$ , - $z+1$ ; #5 - $x$ , - $y+1$ , - $z$ ; #6 - $x+1/2$ , - $y+1/2$ , - $z$ . <sup>b</sup> #4  $x$ ,  $y-1$ ,  $z$ ; #5  $x-1/2$ ,  $y-1/2$ ,  $z$ ; #6  $x$ , - $y+1$ ,  $z+1/2$ ; #7 - $x+1/2$ , - $y+1/2$ , - $z$ . <sup>c</sup> #6 - $x$ , - $y+1$ , - $z+1$ ; #7 - $x+1$ , - $y+1$ , - $z+1$ ; #8 - $x+1$ ,  $y-1/2$ , - $z+1/2$ ; #9  $x+1$ , - $y+3/2$ ,  $z+1/2$ .

**Table S2.** Emission and excitation maxima wavelengths (nm)

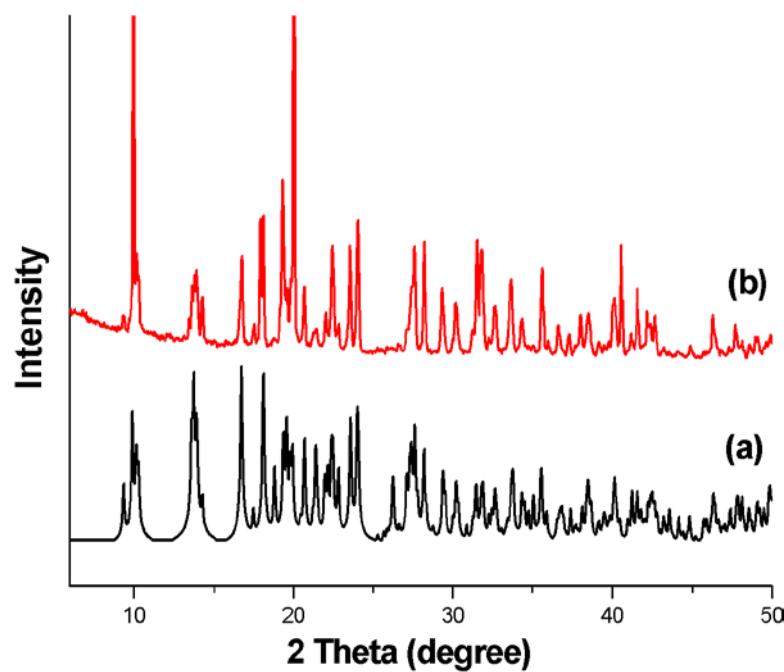
polymer	<b>HL</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>
$\lambda_{\text{em}}$ (nm)	451	436	413	481	449	468
$\lambda_{\text{ex}}$ (nm)	370	375	314	382	365	369



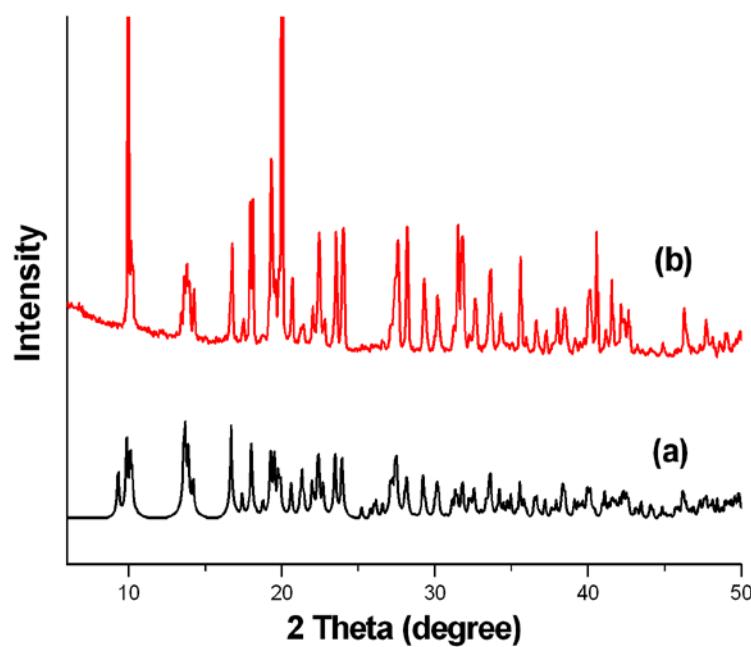
**Fig. S1** Top view of the open-ended, hollow nanotube of **4**.



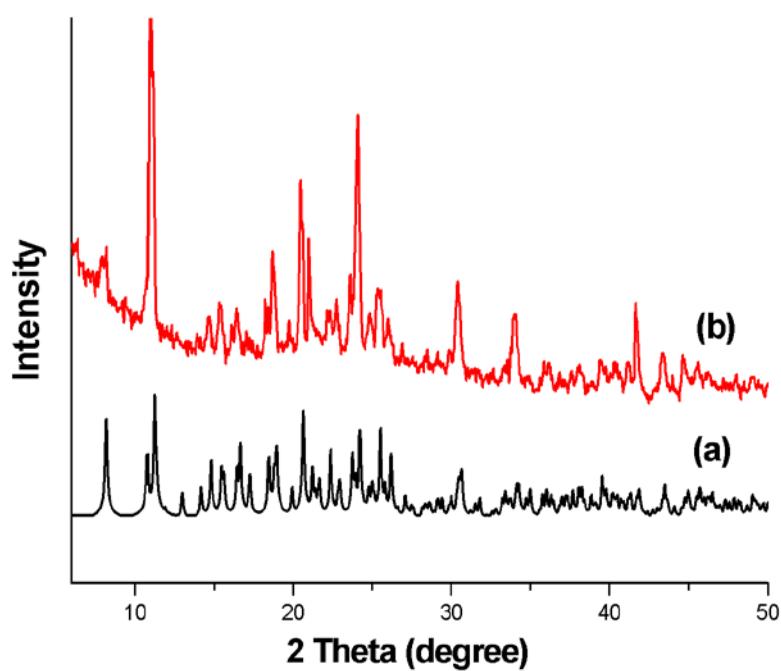
**Fig. S2.** The TG curves of complexes **1**–**5**.



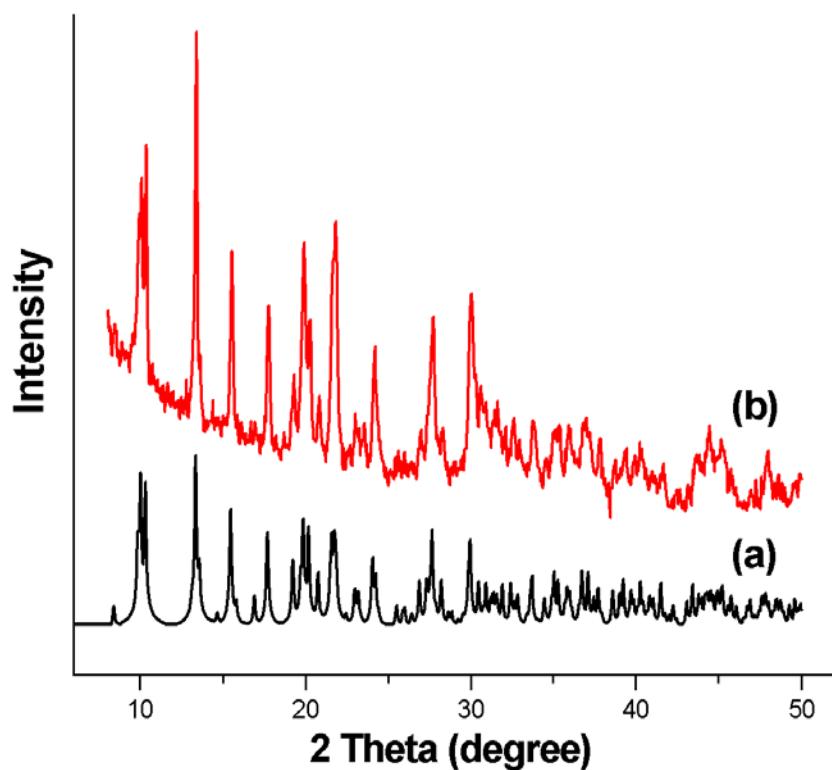
**Fig. S3** X-ray powder diffraction of complex **1** (a) simulated from single crystal data, (b) observed for complex **1**.



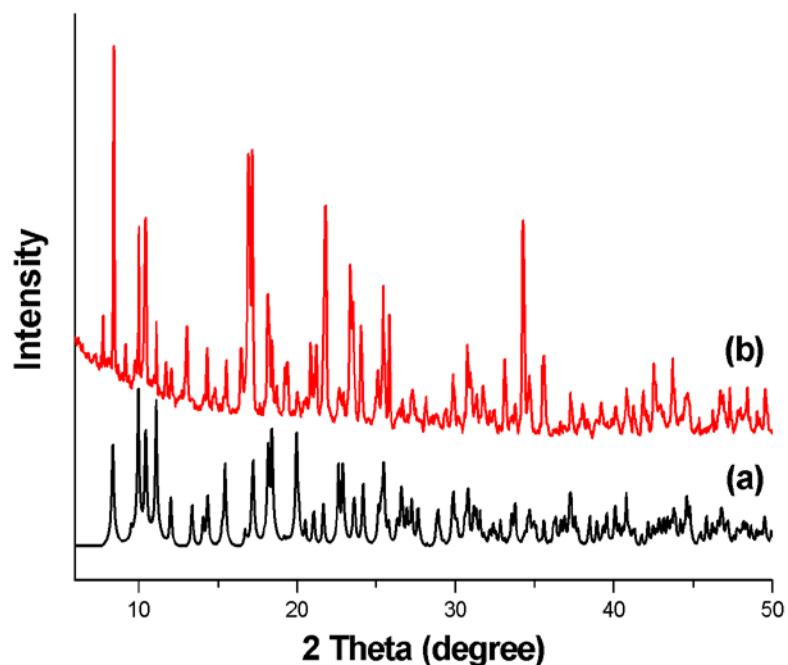
**Fig. S4** X-ray powder diffraction of complex **2** (a) simulated from single crystal data, (b) observed for complex **2**.



**Fig. S5** X-ray powder diffraction of complex 3 (a) simulated from single crystal data, (b) observed for complex 3.



**Fig. S6** X-ray powder diffraction of complex 4 (a) simulated from single crystal data, (b) observed for complex 4.



**Fig. S7** X-ray powder diffraction of complex **5** (a) simulated from single crystal data, (b) observed for complex **5**.