

Electronic Supplementary Information

Hetero-epitaxial growth of vertically-aligned TiO₂ nanorods on m-cut sapphire substrate with (001) SnO₂ buffer layer

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1. Experimental Details

Preparation of SnO₂ buffer layer. M-cut (100) sapphire was used as a substrate. Before deposition, the substrate was ultrasonically cleaned in acetone, ethanol, and de-ionized water for 10 min. Highly (001) oriented epitaxial SnO₂ film was deposited by plasma enhanced atomic layer deposition (PE-ALD). The Sn precursor was dibutyl tin diacetate (DBTDA, ((CH₃CO₂)₂Sn[(CH₂)₃-CH₃]₂). The time sequence for source pulse, first purge, plasma pulse, and second purge was 2, 8, 10, and 8 s, respectively. The deposition was conducted with an rf power of 100W at 240 mTorr for 500 cycles. The SnO₂ buffered m-cut sapphire substrate was annealed at 600 °C to enhance their crystallinity before the TiO₂ nanorod growth.

Synthesis of TiO₂ nanorods. For TiO₂ nanorod growth, 25 mL of de-ionized water was mixed with 25 mL of concentrated hydrochloric acid (36.5% ~ 38% by weight) and stirred at ambient conditions for 5 min. and then 0.8 mL of titanium butoxide (Ti(OBu)₄, 97% Aldrich) was added to the solution and stirred until the solution became transparent. As-prepared solution was transferred to the Teflon-lined stainless steel autoclave with 120 mL capacity. A piece of SnO₂ buffered M-cut sapphire was placed against the wall of the Teflon-liner just below the solution surface. The hydrothermal synthesis was conducted at 150 °C for 0.5~3.0 h in an electric oven. After synthesis, the autoclave was naturally cooled to room temperature. The as-synthesised TiO₂ nanorods were taken out, rinsed with de-ionized water, and dried in ambient air.

Materials Characterization. The morphology of as-synthesized product was observed by field-emission scanning electron microscopy (FE-SEM, JSM-7401F, JEOL). The phase and in- and out of-plane orientation relationships between nanorods and substrate were examined by X-ray diffraction (XRD) and X-ray pole figure. The out-of plane orientation was examined by θ -2 θ X-ray diffraction (Model D8-Advance, BRUKER MILLER Co.) using Cu K α radiation ($\lambda=1.5406\text{\AA}$), and the in-plane orientation was investigated by X-ray pole figure (Model X'Pert Pro, PANalytical, the Netherlands), which was performed in Schulz reflection geometry by

scanning the tilt angle of goniometer, χ (Chi), in the range of 0–85° and the azimuthal angle, ϕ (phi), in the range of 0–360° with a step size of 5°. High-resolution transmission electron microscopy (HR-TEM, JEM-3000F, JEOL) analysis was further performed to investigate the crystal structure of nanorods and interfaces between nanorod, buffer layer, and substrate.

2. Results and Discussion

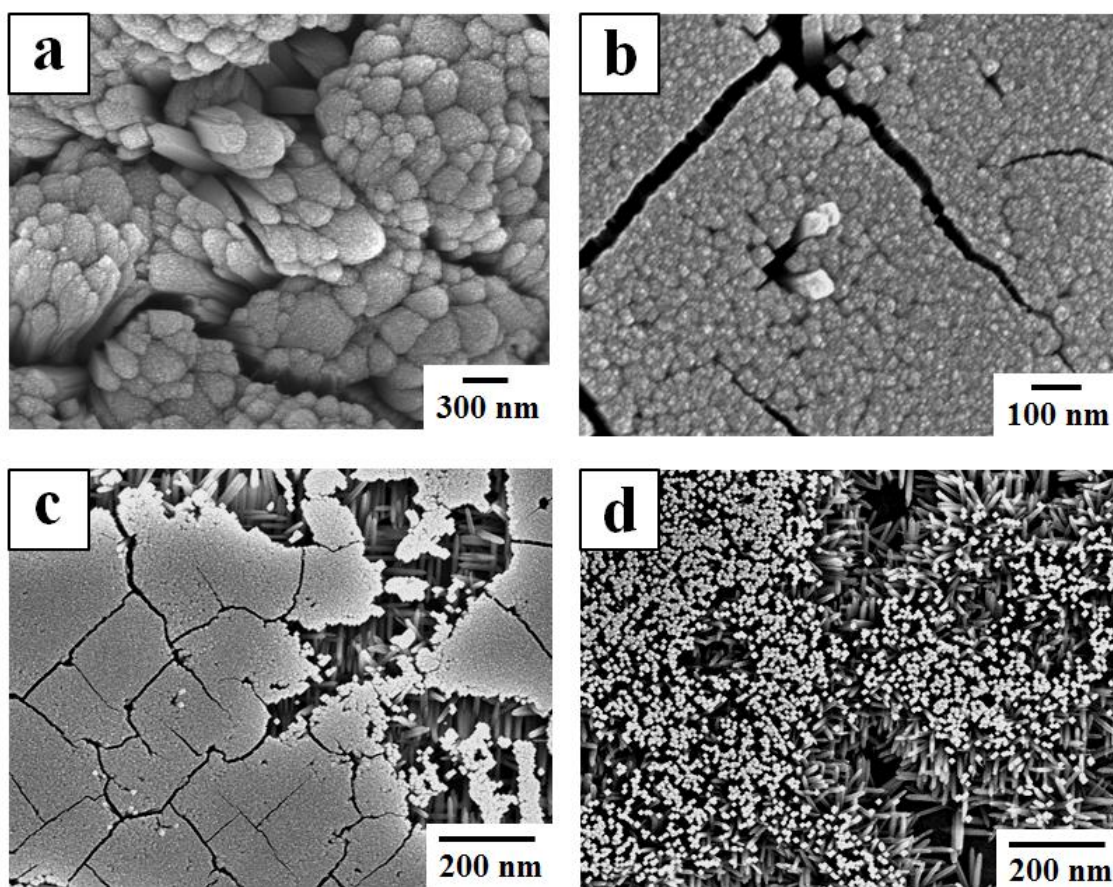


Figure S1. SEM images of TiO₂ rods grown on (a) bare m-cut sapphire and (b-d) SnO₂ buffered m-cut sapphire.

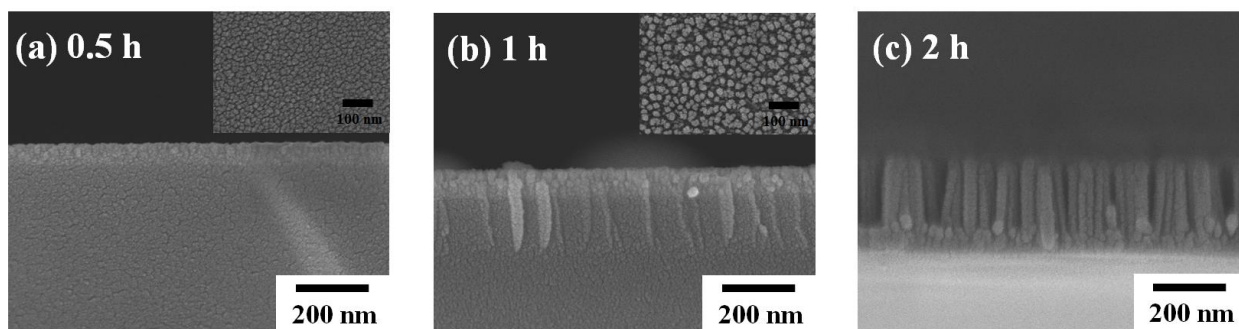


Figure S2. SEM images of TiO₂ rods grown for (a) 0.5 h, (b) 1.0 h, and (c) 2.0 h..

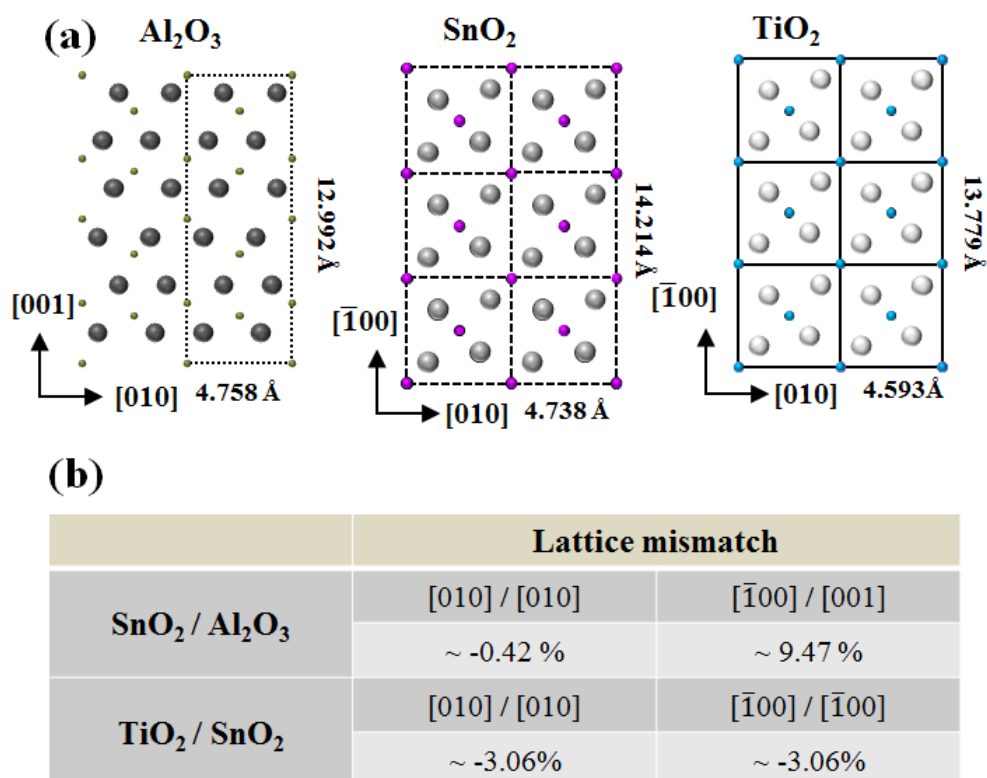


Figure S3. (a) Atomic configurations of m-cut sapphire, SnO_2 buffer layer, and TiO_2 nanorod based on the determined in-plane orientation relationships and (b) estimated lattice mismatches.

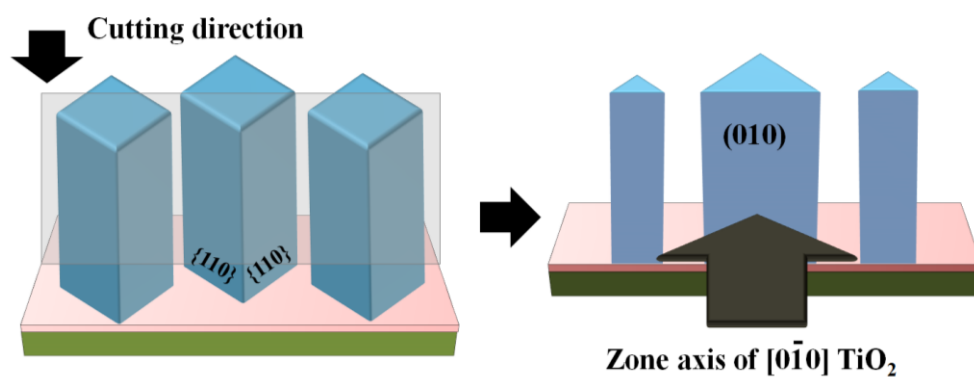


Figure S4. The preparation of cross-sectional TEM sample by FIB.

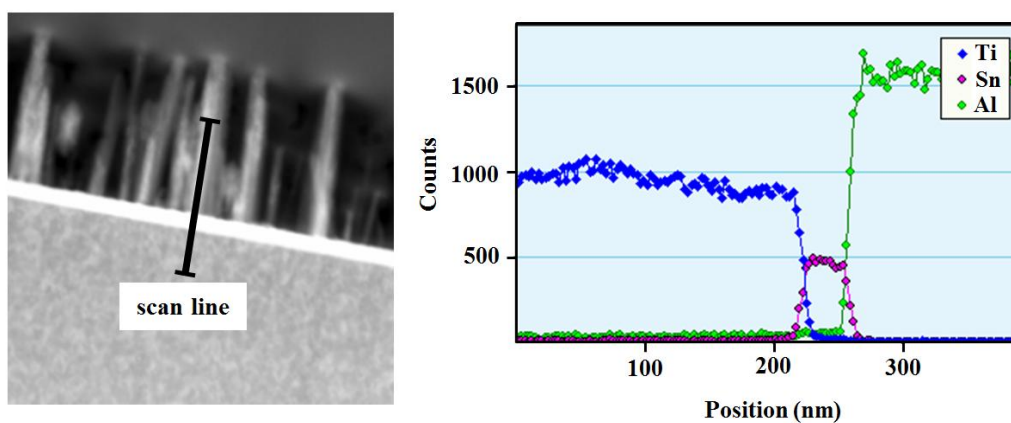


Figure S5. The energy dispersive X-ray spectroscopy (EDS) analysis of TiO₂ nanorods/SnO₂ buffer layer/m-cut sapphire substrate

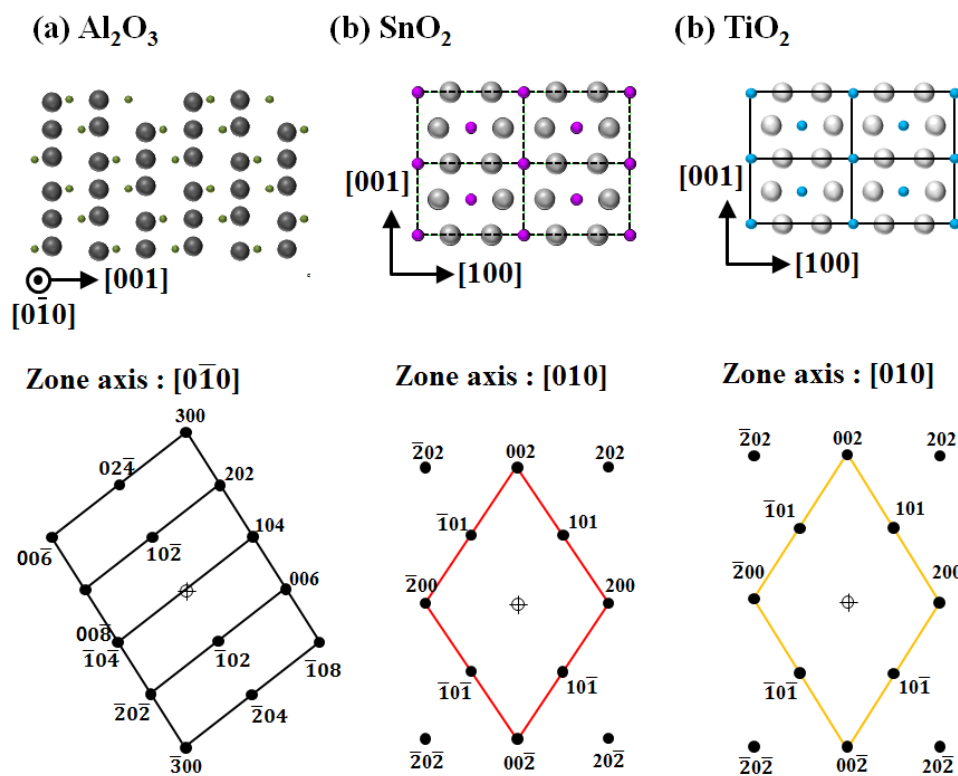


Figure S6. Atomic configurations and simulated diffraction patterns; (a) m-cut sapphire with a view direction of $[\text{0}\bar{1}\text{0}]$, (b) SnO_2 buffer layer with a view direction of $[\text{010}]$, and (c) TiO_2 nanorod with a view direction of $[\text{010}]$.

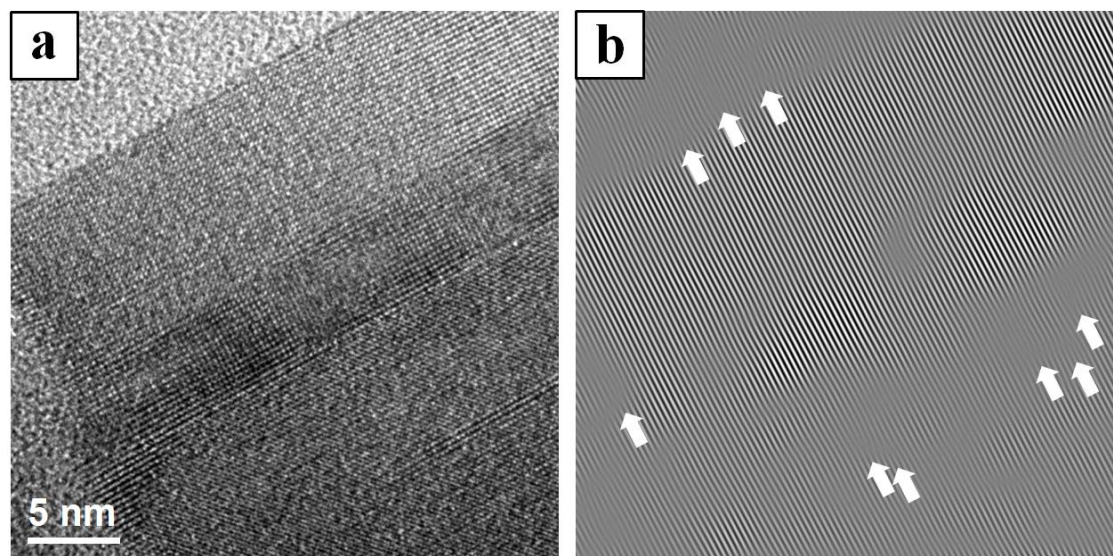


Figure S7. (a) HRTEM image of TiO_2 nanorod and (b) filtered FFT image of (a) (arrows indicate the line defects in the TiO_2 nanorods).

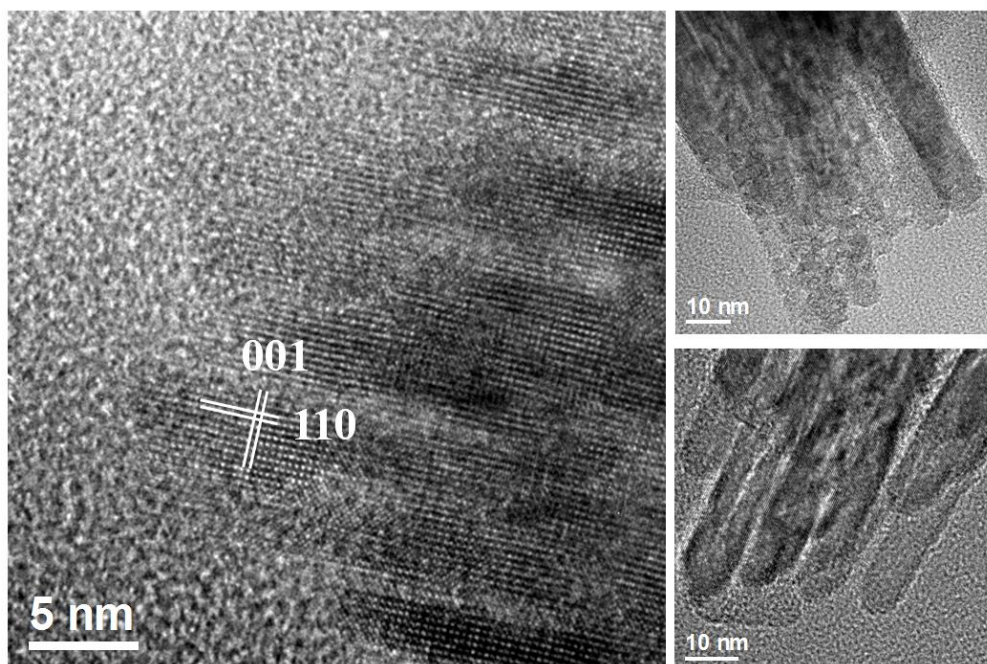


Figure S8. HRTEM images of TiO₂ nanorod tips.