Electronic Supplementary Information

Controlled Synthesis of SnO₂ Decorated SnS₂ Nanosheets with Enhanced Visible Light Photocatalytic Activity

Xianlong Zhou, Tengfei Zhou, Juncheng Hu* and Jinlin Li

Key Laboratory of Catalysis and Materials Science of the State Ethnic Affairs Commission & Ministry of Education, Hubei Province, South-Central University for Nationalities, Wuhan, 430074, People's Republic of China

*E-mail: junchenghuhu@hotmail.com

	Sample	V_{ani} (<i>mL</i>)	V_{eth} (<i>mL</i>)	V_{ani} / V_{eth}	SnO ₂ molar fraction
•	а	20	40	33.3%	29.3%
	b	30	30	50.0%	44.5%
	c	40	20	66.6%	62.1%

Table S1 Com	position of the SnSa	$@SnO_{2} NCs sy$	vnthesized in [hinary solvent
Table DI Com	position of the bilb?	ebil0210035	ynthesized m	omary sorvent



Fig. S1 Degradation percentage of the MO in the presence of $SnS_2@SnS_2-2$ powder (50 mg) under visible light ($\lambda > 420$ nm) irradiation after 45min



Fig. S2 XRD patterns of (a) fresh $SnS_2@SnO_2-2$ catalyst and (b) reused $SnS_2@SnO_2-2$ catalyst for four times.

Samples	Crystallite size of	Crystallite size of	BET surface area
	SnO ₂ (nm)	SnS_2 (nm)	(g/m ²)
SnS ₂		13.2	41.83
SnO ₂	24.9		15.45
SnS ₂ @SnO ₂			
(Veth / Vani = 1:2)	24.9	23.8	17.69

Table S2 Crystallite size and surface area of SnS2, SnO2 and SnS2@SnO2.