

Supporting Information

Co₂GeO₄Nanoplates and Nano-octahedrons from Low-Temperature Controlled Synthesis and Their Magnetic Properties

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To explore the effect of the experimental parameters on the products, a series of experiments are performed. The respective detailed conditions are given in Table S1. Sample A (SA) and Sample B (SB) are the final products of Co₂GeO₄nano particles among these experiments.

Table S1. The products obtained from different conditions.

| Sample | CoCl ₂ ·6H ₂ O(g) | GeO ₂ (g) | CTAB(mmol) | NaOH(mmol) | Hydrothermal Temprature (°C) |
|--------|---|----------------------|------------|------------|------------------------------|
| SA | 0.238 | 0.052 | 1.1 | 0.1 | 120, then annealing at 400 |
| SB | 0.238 | 0.052 | 1.1 | 0.4 | 160 |
| SC | 0.238 | 0.052 | 1.1 | 0.8 | 160 |
| SD | 0.238 | 0.052 | None | 0.1 | 160 |
| SE | 0.238 | 0.052 | 0.275 | 0.1 | 160 |
| SF | 0.238 | 0.052 | 0.55 | 0.1 | 160 |
| SG | 0.238 | 0.052 | None | 0.4 | 160 |

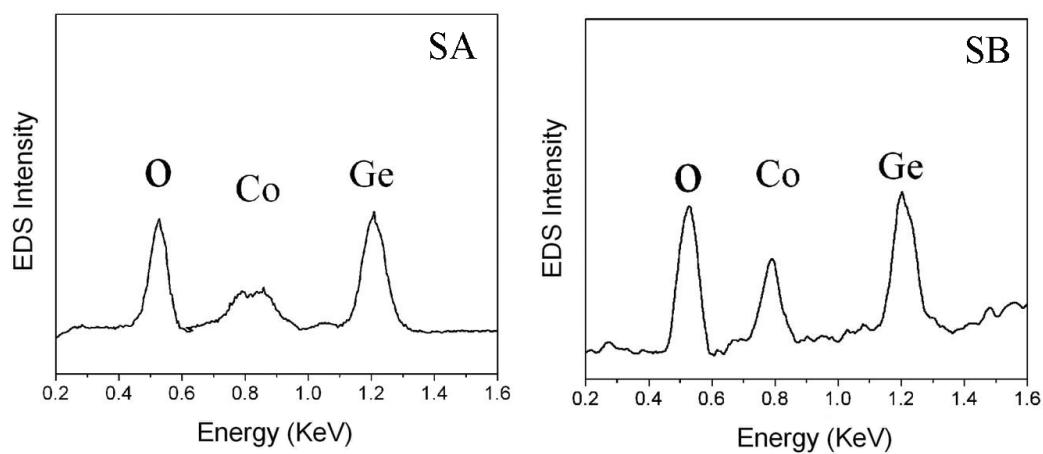


Fig. S1. The EDS patterns of SA and SB (the ratio of the above elements is closed to the theoretical ratio 2:1:4).

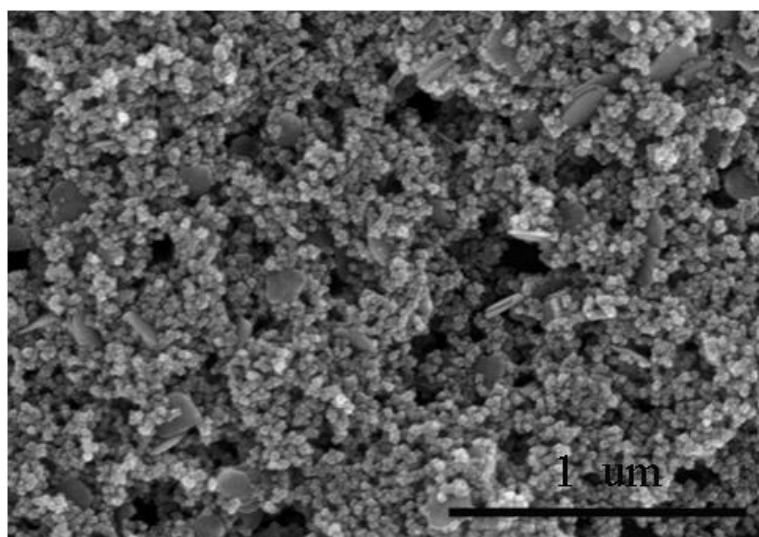


Fig. S2. Typical SEM image of the as-prepared Co_2GeO_4 products when the amount of NaOH is 0.8 mmol.

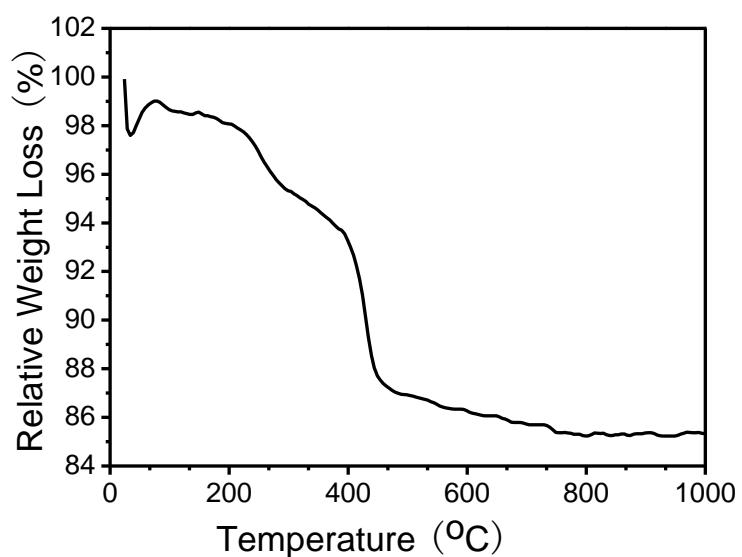


Fig. S3. Thermogravimetric analysis of precursor.

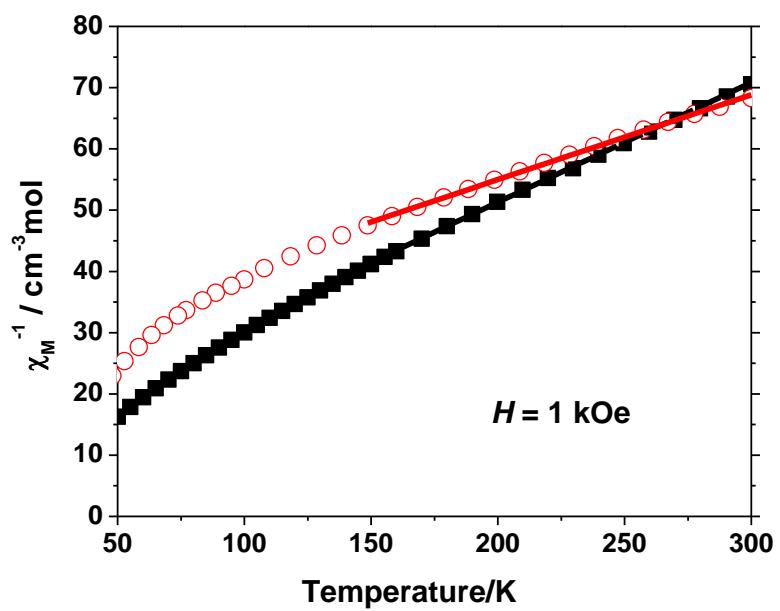


Fig. S4. Plot of χ_M^{-1} vs T for SA (open circle) and SB (filled cubic). The line is linear fitted.