Supporting Information

Conformational Control of Ligands to Create a Finite Metal–Organic Cluster and an Extended Metal–Organic Framework

Lalit Rajput,^a Dongwook Kim,^a and Myoung Soo Lah*^a

^a Interdisciplinary School of Green Energy, Ulsan National Institute of Science & Technology, Ulsan, Korea.

 Table S1. Crystal data and structure refinement for 1.

Empirical formula	$C_{60}H_{42}N_6O_{21}Cu_3$	$C_{60}H_{42}N_6O_{21}Cu_3$	
Formula weight	1373.66		
Temperature	173(2) K		
Wavelength	0.71073 Å		
Crystal system	Cubic		
Space group	Pn-3n		
Unit cell dimensions	a = 32.365(4) Å	$\alpha = 90^{\circ}$	
	b = 32.365(4) Å	$\beta = 90^{\circ}$	
	c = 32.365(4) Å	$\gamma = 90^{\circ}$	
Volume	33902(7) Å ³		
Z	8		
Density (calculated)	0.538 Mg/m^3		
Absorption coefficient	0.402 mm^{-1}		
F(000)	5592		
Crystal size	$0.50 \ge 0.49 \ge 0.43 \text{ mm}^3$		
Theta range for data collection	2.88 to 23.33°.		
Index ranges	$-35 \le h \le 34, -35 \le k \le 30, -32 \le l \le 32$		
Reflections collected	108634	108634	
Independent reflections	4057 [R(int) = 0.2337]		
Completeness to theta = 23.33°	98.3 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.8460 and 0.8241		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	4057 / 82 / 119		
Goodness-of-fit on F ²	1.341		
Final R indices [I>2sigma(I)]	R1 = 0.1515, wR2 = 0.4379		
R indices (all data)	R1 = 0.2465, wR2 = 0.4844		
Largest diff. peak and hole	$0.619 \text{ and } -0.335 \text{ e} \cdot \text{\AA}^{-3}$		

Table S2. Crystal data and structure refinement for 2.

Empirical formula	$C_{278}H_{340}N_{44}O_{94}Ni_{14}$	
Formula weight	6623.88	
Temperature	96(2) K	
Wavelength	0.80000 Å	
Crystal system	Monoclinic	
Space group	C2/c	
Unit cell dimensions	a = 41.121(8) Å	$\alpha = 90^{\circ}$
	b = 20.457(4) Å	$\beta = 106.71(3)^{\circ}$
	c = 39.436(8) Å	$\gamma = 90^{\circ}$
Volume	31773(11) Å ³	
Z	4	
Density (calculated)	1.385 Mg/m ³	
Absorption coefficient	1.179 mm^{-1}	
F(000)	13840	
Crystal size	$0.50 \ge 0.50 \ge 0.33 \text{ mm}^3$	
Theta range for data collection	1.63 to 28.00°	
Index ranges	-48<=h<=48, -23<=k<=23, -46<=l<=46	
Reflections collected	93568	
Independent reflections	26217 [R(int) = 0.0407]	
Completeness to theta = 28.00°	97.5 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.6969 and 0.5901	
Refinement method	Full-matrix-block least-squares on F ²	
Data / restraints / parameters	<u> 26217 / 177 / 2303</u>	
Goodness-of-fit on F ²	1.085	
Final R indices [I>2sigma(I)]	R1 = 0.0643, wR2 = 0.1933	
R indices (all data)	R1 = 0.0803, wR2 = 0.2031	
Largest diff. peak and hole	$0.668 \text{ and } -0.425 \text{ e} \cdot \text{\AA}^{-3}$	



Fig. S1 PXRD patterns of 1. (a) A simulated PXRD pattern from the single crystal structure of 1,(b) from the single crystal structure of 1 with [1, 1, 1] preferred orientation, and (c) a PXRD pattern of the as-synthesized sample.



Fig. S2 The 14 Ni(II) ions of 2 in a crystallographic inversion center are bridged *via* eight μ^3 -hydroxo oxygen atoms (shown in pink).



Fig. S3 PXRD patterns of 2. (a) A simulated PXRD pattern from the single crystal structure of 2 and (b) a PXRD of the as-synthesized sample.