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The novel upconversion properties of LiYbF₄:Er microcrystals compared to the Na counterpart[†]

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Experimental section

Materials

All the Rare-earth oxides in this work including Yb₂O₃ (99.99 %) and Er_2O_3 (99.9 %) were purchased from Sinopharm Chemical Reagent Company of China. Lithium fluoride (LiF), sodium fluoride (NaF), ammonium fluoride (NH₄F), nitric acid (HNO₃) and ethylenediamine tetraacetic acid (EDTA) were of analytical grade and used directly without further purification. Deionized water was used throughout the experiments.

Microcrystals synthesis

Erbium-doped NaYbF₄ (or LiYbF₄) microcrystals have been fabricated via a facile hydrothermal route assisted with EDTA. Briefly, 0.75 mmol of EDTA (0.5 mol/L) was added to a mixture containing 0.3 mmol of rare-earth nitrate (1.5 mL of 0.2 mol/L RE(NO₃)₃, RE=Yb and Er) and 15 mL deionized water. The solution was then thoroughly stirred for 30 min to form a chelated RE-EDTA complex. Then, a solution 8 mL (1.0 mol/L) of NH₄F and 4 mL (1.0 mol/L) of NaF (or LiF) aqueous solutions were dropped into chelated RE-EDTA complex under thoroughly stirring. The pH value of the mixture was tuned around 4.0 with nitric acid. Subsequently, the milky colloidal solution was transferred to a 40.0 mL Teflon-lined autoclave, and heated at 200 $^{\circ}$ C for 24 h. The final product was collected by centrifuging and washed with water and ethanol. The

collected microcrystals were dried under 80 °C for 12 h.

Characterizations

The phase compositions of the as-prepared products were examined by XRD with a D/Max2400 X-ray diffraction meter with Cu *Ka* (40 kV, 100 mA) irradiation (λ =0.15406 nm). SEM micrographs were obtained using a Hitachi S-4800 FE-SEM. The photoluminescence were recorded with a Ti sapphire continuum laser (MBR-110) and a YAG:Nd³⁺ (Quanta Ray Lab-170) laser as the excitation source. All the measurements were performed at room temperature.



Fig. S1 XRD patterns of different fluoride microcrystals.

XRD patterns in Fig. S1 show that the crystal phases of the NaYbF₄ (or LiYbF₄) microparticles closely resemble to those of the NaYF₄ (or LiYF₄) microparticle counterparts except for the shift of the diffraction peaks of NaYbF₄ (or LiYbF₄) toward higher 2 θ angle.

The blue shift of the spectra

The similar phenomenon of the blue shift appear in LiYF4:Er relative to NaYF4:Er, which



indicate that the shift may be ascribed to dissimilar crystal phase construction.

Fig. S2 UC emission spectra of 2 mol% Er doped fluoride microcrystals. The samples were excited with 980 nm photons.

Thermal population

The thermal effect caused by the exposure of the 980 nm laser has to be considered. For Er^{3+} ions, the energy separation between the two nearest excited states ${}^{2}H_{11/2}$ and ${}^{4}S_{3/2}$ is only several hundred wave numbers. This will lead the intensity ratio of ${}^{2}H_{11/2}$ - ${}^{4}I_{15/2}$ to ${}^{4}S_{3/2}$ - ${}^{4}I_{15/2}$ to be sensitive to the Yb³⁺ concentration and excitation power owing to Boltzmann's thermal population. Figure S3 shows the ratio of ${}^{2}H_{11/2}$ - ${}^{4}I_{15/2}$ to ${}^{4}S_{3/2}$ - ${}^{4}I_{15/2}$ as a function of temperature in different samples. The larger value of ratio was found in LiYbF₄, which indicates the higher Yb³⁺ concentration and stronger thermal effect in LiYbF₄.



Fig. S3 Intensity ratios of ${}^{2}H_{11/2}$ - ${}^{4}I_{15/2}$ to ${}^{4}S_{3/2}$ - ${}^{4}I_{15/2}$ as a function of excitation power in different samples.