

# N-Oxide-4,4'-Bipyridine, a Forgotten Ligand in Coordination Chemistry: Structure-Photoluminescence Property Relationships in 2D and 1D Lead-Coordination Polymer

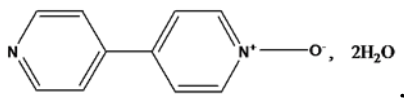
Oksana Toma,<sup>a</sup> Nicolas Mercier,<sup>\*a</sup> Mathilde Bouilland<sup>a</sup> and Magali Allain

## Supporting Information

### A- Synthesis

#### A1- Procedure for the preparation of viologens:

##### Synthesis of the hydrated N-oxide-4,4'-bipyridine (bp4mo, 2H<sub>2</sub>O)



M = 208,22 g/mol

According to the literature [1,2], 3,4 g 4,4'-bipyridine ( $2,1 \cdot 10^{-2}$  mol) are dissolved in 25 ml of acid acetic glacial under heating at 70°C. After, 2,18 g of hydrogen peroxide ( $2,1 \cdot 10^{-2}$  mol) is added drop by drop. Solution is left for agitation under heating at 70°C during 24 hours and after that, cooled down to the ambient temperature. Then, 37 g of NaHCO<sub>3</sub> (0,44 mol) is added to the solution leading to a white solid. Later, all products, which come from 4,4'-bipyridine are extracted in chloroform (4×200 ml). Afterwards, the resulting pink solution which was obtained, is concentrated and is put into the chromatographic column (SiO<sub>2</sub>, 20\*5 cm). The first eluent is acetone, used to extract the residual 4,4'-bipyridine. The second eluent is mixture acetone/methanol (in proportion 4:1 to 3:1), used to extract the intermediate product, N-oxide-4,4'-bipyridine. After concentration, a crystalline white powder (2,27 g, 63% based on 4,4'-bipyridine) of N-oxide-4,4'-bipyridine, 2H<sub>2</sub>O is obtained.

RMN <sup>1</sup>H (300 MHz, D<sub>2</sub>O): δ=8,48 (d, 2H, J=6,3 Hz, ortho-N), 8,27 (d, 2H, J=7,5 Hz, ortho-N<sup>+</sup>-O), 7,76 (d, 2H, J=7,5 Hz, meta-N), 7,56 (d, 2H, J=6,3 Hz, meta-N<sup>+</sup>-O).

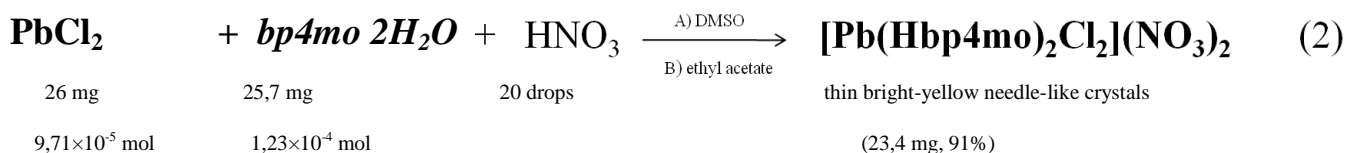
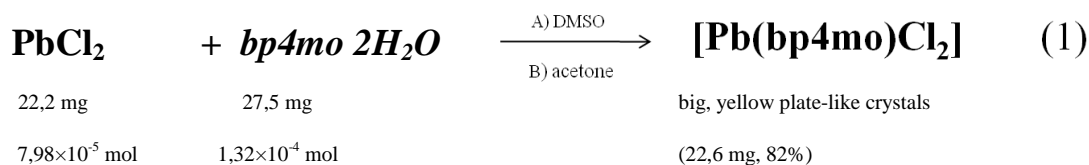
Anal. Elem.: Calc. C, 57,68; H, 5,81; N, 13,45; O, 23,05 – Measured. C, 57,57; H, 5,71; N, 13,50; O, 22,30.

[1] L. A. S. R. Fielden, J Heterocyclic Chem, 1974, 11, 299.

[2] H. Brunner, R. Störiko, F. Rominger, Eur J Inorg Chem, 1998, 771

## A2 - Procedure for the preparation of compounds:

The compounds **1**, **2**, **A** were obtained, with method of slow liquid – gaz diffusion with special apparatus (see picture). In this method, bp4mo,2H<sub>2</sub>O and lead chloride salt are dissolved in the minimum of first solvent, DMSO (for compounds **1**, **2**) in a pillbox (A). The pillbox is then covered with a holed aluminium paper and inserted in a jar of jam filled with contra-solvent: acetone (for compounds **1**), or ethyl acetate (for compound **2**) (B). The jar of jam is then covered with a lid and sealed with parafilm. A few days later, crystals appeared. Sample is then washed with contra-solvent and dried in the oven at 50 °C.



## **B- Single crystal and powder X-ray diffraction analysis for (1), (2)**

### **B-I- [(Pb(bp4mo)Cl<sub>2</sub>)] (1)**

#### **B-I-A- Summary of crystallographic data**

Empirical formula	C10 H8 Cl2 N2 O Pb
Formula weight	450.29
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, C2
Unit cell dimensions	a = 16.459(1) Å    alpha = 90 deg. b = 4.1097(5) Å    beta = 93.72(1) deg. c = 18.072(1) Å    gamma = 90 deg.
Volume	1219.84(18) Å <sup>3</sup>
Z, Calculated density	4, 2.452 Mg/m <sup>3</sup>
Absorption coefficient	14.247 mm <sup>-1</sup>
F(000)	824
Crystal size	0.24 x 0.08 x 0.06 mm
Theta range for data collection	2.66 to 29.53 deg.
Limiting indices	-22<=h<=22, -5<=k<=5, -25<=l<=24
Reflections collected / unique	6612 / 3161 [R(int) = 0.0467]
Completeness to theta = 29.53	98.7 %
Absorption correction	Multiscan
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3 161 / 1 / 147
Goodness-of-fit on F <sup>2</sup>	1.054
Final R indices [I>2sigma(I)]	R1 = 0.0381, wR2 = 0.0686
R indices (all data)	R1 = 0.0585, wR2 = 0.0745
Absolute structure parameter	0.052(15)
Largest diff. peak and hole	1.067 and -1.002 e.Å <sup>-3</sup>

#### **CHECKCIF**

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Bond precision:	C-C = 0.0138 Å	Wavelength=0.71073
Cell:	a=16.459(1)    b=4.1097(5)    c=18.072(1)	alpha=90    beta=93.72(1)    gamma=90
Temperature:	293 K	
	Calculated	Reported
Volume	1219.84(18)	1219.84(18)
Space group	C 2	C 1 2 1
Hall group	C 2y	?
Moiety formula	C10 H8 Cl2 N2 O Pb	?
Sum formula	C10 H8 Cl2 N2 O Pb	C10 H8 Cl2 N2 O Pb
Mr	450.28	450.29
Dx, g cm <sup>-3</sup>	2.452	2.452
Z	4	4
Mu (mm <sup>-1</sup> )	14.247	14.247
F000	824.0	824.0
F000'	812.63	
h, k, lmax	22, 5, 25	22, 5, 25
Nref	1932 [ 3415]	3161
Tmin, Tmax	0.265, 0.425	0.305, 0.475
Tmin'	0.031	

Correction method= MULTI-SCAN  
Data completeness= 1.64/0.93    Theta(max)= 29.530  
R(reflections)= 0.0381( 2537)    wR2(reflections)= 0.0745( 3161)  
S = 1.054    Npar= 147

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The following ALERTS were generated. Each ALERT has the format

**test-name\_ALERT\_alert-type\_alert-level.**

Click on the hyperlinks for more details of the test.

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### Alert level B

[PLAT111\\_ALERT\\_2\\_B](#) ADDSYM Detects (Pseudo) Centre of Symmetry .....  
88 PerFi

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### Alert level C

[PLAT241\\_ALERT\\_2\\_C](#) Check High    Ueq as Compared to Neighbors for    O1  
[PLAT242\\_ALERT\\_2\\_C](#) Check Low    Ueq as Compared to Neighbors for    Pb1  
[PLAT242\\_ALERT\\_2\\_C](#) Check Low    Ueq as Compared to Neighbors for    Pb2  
[PLAT342\\_ALERT\\_3\\_C](#) Low Bond Precision on C-C Bonds .....    0.0138  
Ang  
[PLAT410\\_ALERT\\_2\\_C](#) Short Intra H...H Contact H2 .. H7 ..    1.98 Ang.

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### Alert level G

[REFLT03\\_ALERT\\_4\\_G](#) Please check that the estimate of the number of Friedel pairs is

correct. If it is not, please give the correct count in the  
\_publ\_section\_exptl\_refinement section of the submitted CIF.

From the CIF: \_diffn\_reflns\_theta\_max    29.53

From the CIF: \_reflns\_number\_total    3161

Count of symmetry unique reflns    1932

Completeness ( \_total/calc)    163.61%

TEST3: Check Friedels for noncentro structure

Estimate of Friedel pairs measured    1229

Fraction of Friedel pairs measured    0.636

Are heavy atom types Z>Si present    yes

[PLAT004\\_ALERT\\_5\\_G](#) Info: Polymeric Structure Found with Dimension .    2

[PLAT005\\_ALERT\\_5\\_G](#) No \_iucr\_refine\_instructions\_details in CIF ....    ?

[PLAT113\\_ALERT\\_2\\_G](#) ADDSYM Suggests Possible Pseudo/New Space-group.  
C2/c

[PLAT194\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_reflns\_used datum ....    ?

[PLAT195\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_theta\_max datum ....  
?

[PLAT196\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_theta\_min datum ....    ?

[PLAT199\\_ALERT\\_1\\_G](#) Check the Reported \_cell\_measurement\_temperature  
293 K

[PLAT200\\_ALERT\\_1\\_G](#) Check the Reported \_diffn\_ambient\_temperature  
293 K

[PLAT232\\_ALERT\\_2\\_G](#) Hirshfeld Test Diff (M-X) Pb2 -- Cl1 ..    6.0 su

[PLAT794\\_ALERT\\_5\\_G](#) Note: Tentative Bond Valency for Pb2 (II)    2.20

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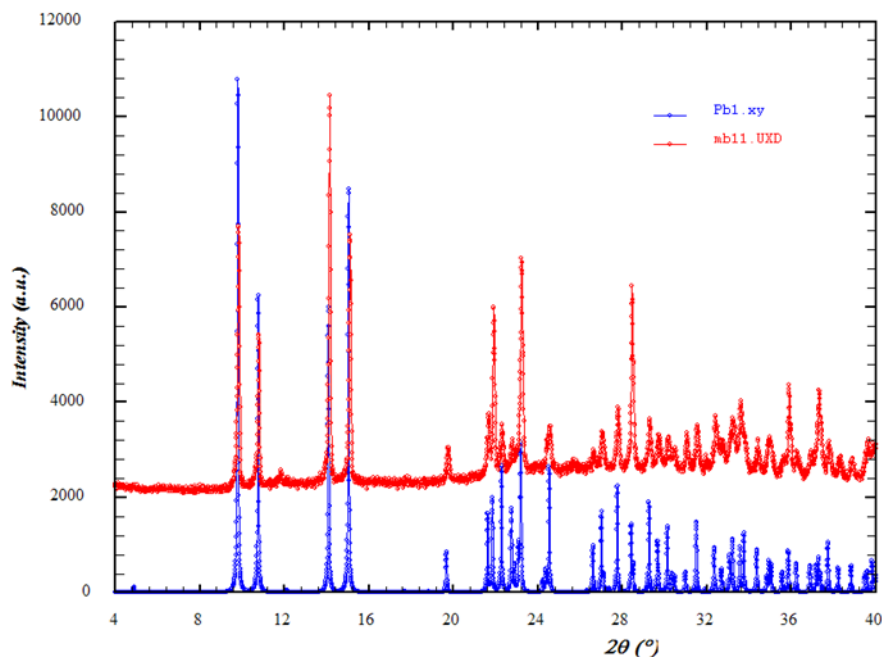
0 **ALERT level A** = Most likely a serious problem - resolve or explain

1 **ALERT level B** = A potentially serious problem, consider carefully

5 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight

11 **ALERT level G** = General information/check it is not something unexpected

### **B-I-B- XRPD of (1) : theoretical (blue) and experimental (red)**



### **B-II- $[Pb(Hbp4mo)_2Cl_2](NO_3)_2$ (2)**

#### **B-II-A- Summary of crystallographic data**

Empirical formula	C <sub>20</sub> H <sub>18</sub> Cl <sub>2</sub> N <sub>6</sub> O <sub>8</sub> Pb
Formula weight	748.50
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, C2/c
Unit cell dimensions	a = 16.847(1) Å    alpha = 90 deg. b = 4.0866(5) Å    beta = 97.55(1) deg. c = 35.249(3) Å    gamma = 90 deg.
Volume	2406.0(4) Å <sup>3</sup>
Z, Calculated density	4, 2.066 Mg/m <sup>3</sup>
Absorption coefficient	7.294 mm <sup>-1</sup>
F(000)	1440
Crystal size	0.16 x 0.08 x 0.07 mm
Theta range for data collection	3.99 to 30.06 deg.
Limiting indices	-23 ≤ h ≤ 23, -5 ≤ k ≤ 5, -48 ≤ l ≤ 49
Reflections collected / unique	13502 / 3392 [R(int) = 0.0637]
Completeness to theta = 30.06	95.8 %
Absorption correction	Multiscan
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3392 / 0 / 168
Goodness-of-fit on F <sup>2</sup>	1.039
Final R indices [I > 2σ(I)]	R1 = 0.0399, wR2 = 0.0617
R indices (all data)	R1 = 0.0856, wR2 = 0.0685
Largest diff. peak and hole	1.283 and -1.883 e.Å <sup>-3</sup>

## CHECKCIF

Bond precision: C-C = 0.0071 Å Wavelength=0.71073  
Cell: a=16.847 (1) b=4.0866 (5) c=35.249 (3)  
alpha=90 beta=97.55 (1) gamma=90  
Temperature: 293 K

	Calculated	Reported
Volume	2405.8 (4)	2406.0 (4)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	?
Moiety formula	C20 H18 Cl2 N4 O2 Pb, 2(N O3)	?
Sum formula	C20 H18 Cl2 N6 O8 Pb	C20 H18 Cl2 N6 O8 Pb
Mr	748.50	748.50
Dx, g cm <sup>-3</sup>	2.067	2.066
Z	4	4
Mu (mm <sup>-1</sup> )	7.295	7.294
F000	1440.0	1440.0
F000'	1428.56	
h, k, lmax	23, 5, 49	23, 5, 49
Nref	3538	3392
Tmin, Tmax	0.501, 0.600	0.405, 0.528
Tmin'	0.308	

Correction method= MULTI-SCAN  
Data completeness= 0.959 Theta(max)= 30.060  
R(reflections)= 0.0399( 2294) wR2(reflections)= 0.0685( 3392)  
S = 1.039 Npar= 168

The following ALERTS were generated. Each ALERT has the format  
**test-name\_ALERT\_alert-type\_alert-level**.

Click on the hyperlinks for more details of the test.

### Alert level B

[PLAT029\\_ALERT\\_3\\_B\\_diffn\\_measured\\_fraction\\_theta\\_full](#) Low ..... 0.958

### Alert level C

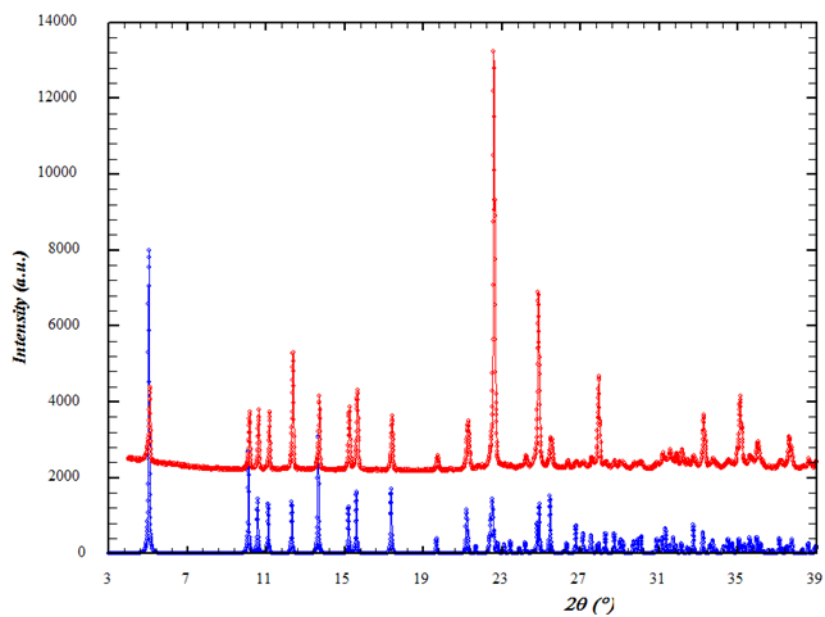
[PLAT241\\_ALERT\\_2\\_C](#) Check High Ueq as Compared to Neighbors for O1  
[PLAT242\\_ALERT\\_2\\_C](#) Check Low Ueq as Compared to Neighbors for Pb1  
[PLAT244\\_ALERT\\_4\\_C](#) Low 'Solvent' Ueq as Compared to Neighbors of N3  
[PLAT790\\_ALERT\\_4\\_C](#) Centre of Gravity not Within Unit Cell: Resd. # 1  
C20 H18 Cl2 N4 O2 Pb

### Alert level G

[PLAT004\\_ALERT\\_5\\_G](#) Info: Polymeric Structure Found with Dimension . 1  
[PLAT005\\_ALERT\\_5\\_G](#) No \_jucc\_refine\_instructions\_details in CIF .... ?  
[PLAT007\\_ALERT\\_5\\_G](#) Note: Number of Unrefined D-H Atoms ..... 1  
[PLAT194\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_reflns\_used datum .... ?  
[PLAT195\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_theta\_max datum .... ?  
[PLAT196\\_ALERT\\_1\\_G](#) Missing \_cell\_measurement\_theta\_min datum .... ?  
[PLAT199\\_ALERT\\_1\\_G](#) Check the Reported \_cell\_measurement\_temperature 293 K  
[PLAT200\\_ALERT\\_1\\_G](#) Check the Reported \_diffn\_ambient\_temperature 293 K  
[PLAT232\\_ALERT\\_2\\_G](#) Hirshfeld Test Diff (M-X) Pb1 -- Cl1 .. 16.6 su

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain  
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9 **ALERT level G** = General information/check it is not something unexpected

**B-II-B- XRPD of (1) : theoretical (blue) and experimental (red)**

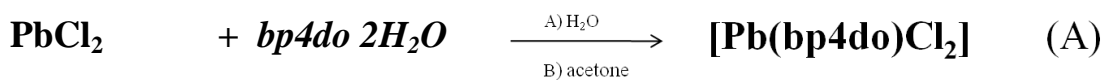


## C- Synthesis and XRPD characterization of (A), (B<sub>β</sub>), (B<sub>α</sub>)

### C-I- [Pb(bp4do)Cl<sub>2</sub>] (A)

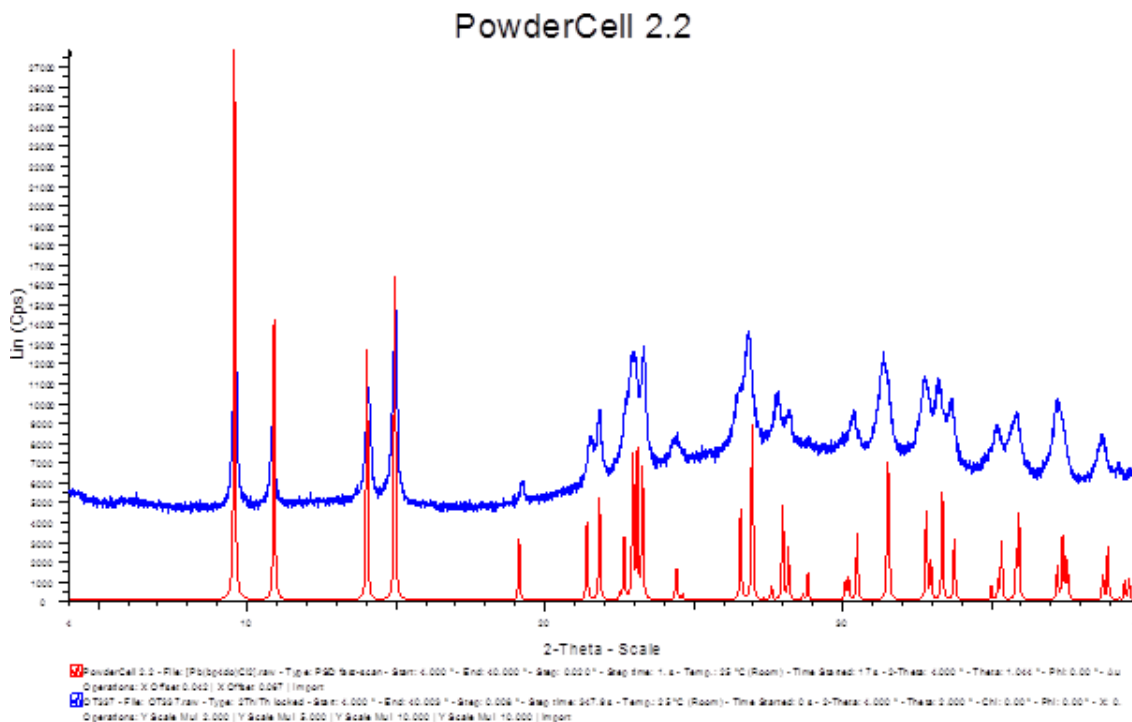
#### C-I-A- Synthesis

A has been synthesized according to the literature (ref 4 in main text) (*bp4do* 2H<sub>2</sub>O commercially available):



32,9 mg	26,5 mg	big and nice orange needle-like crystals
1,18×10 <sup>-4</sup> mol	1,05×10 <sup>-4</sup> mol	(23 mg, 87%)

#### C-I-B- XRPD of (A) : theoretical (red) and experimental (blue)

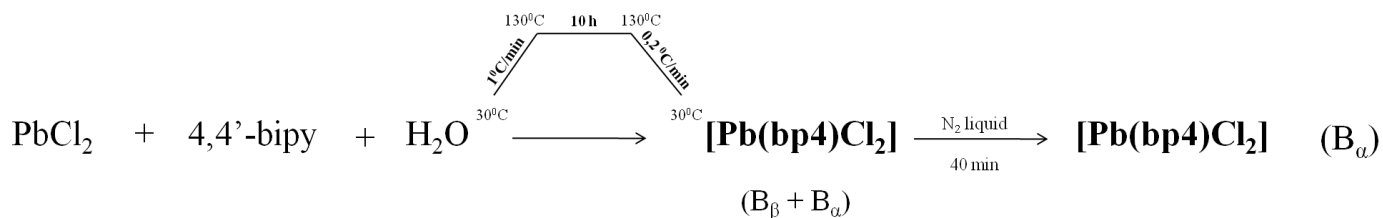




## C-II- [Pb(bp4)Cl<sub>2</sub>] (B<sub>α</sub>)

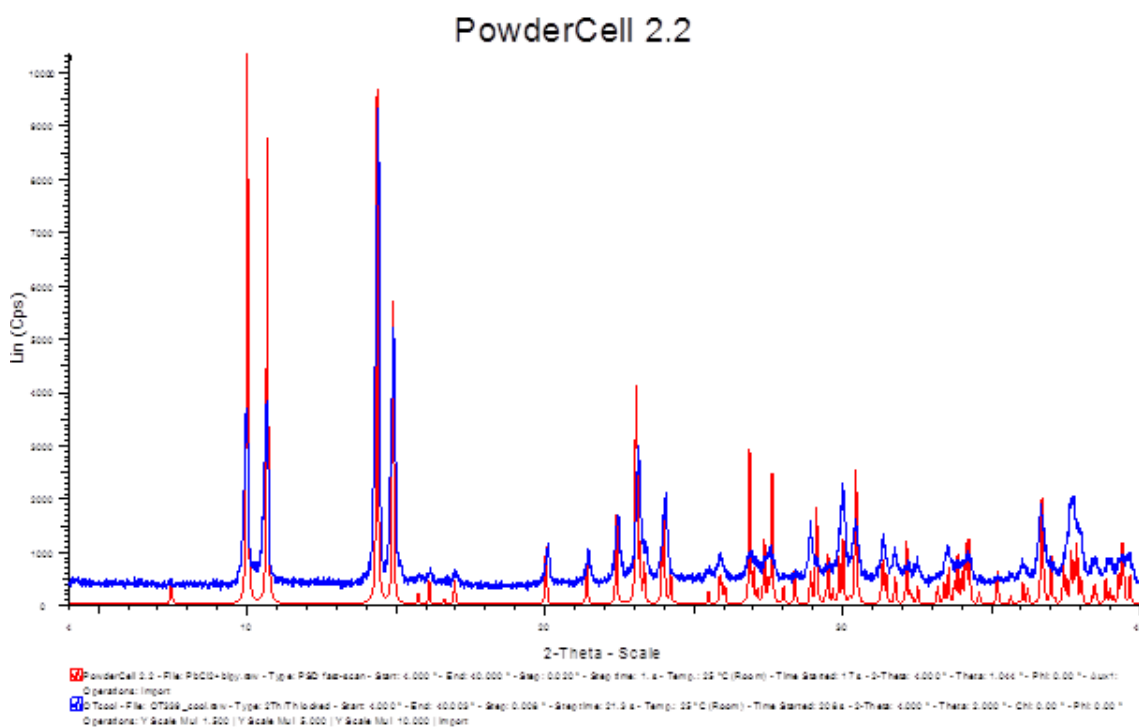
### C-II-A- Synthesis

B<sub>α</sub> has been synthesized in two steps. The synthesis described in the literature (ref 7 in main text) has here lead to a mixture of phase (B<sub>α</sub> and B<sub>β</sub>). B<sub>α</sub> has been obtained as a pure phase after cooling down the mixture at low temperature during one hour:



135,5 mg	62,8 mg	4 ml	white powder
$4,87 \times 10^{-4}$ mol	$4,02 \times 10^{-4}$ mol		(49 mg, 78%)

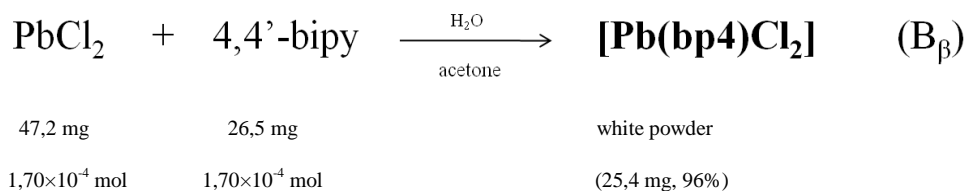
### C-II-B- XRPD of (B<sub>α</sub>) : theoretical (red) and experimental (blue)



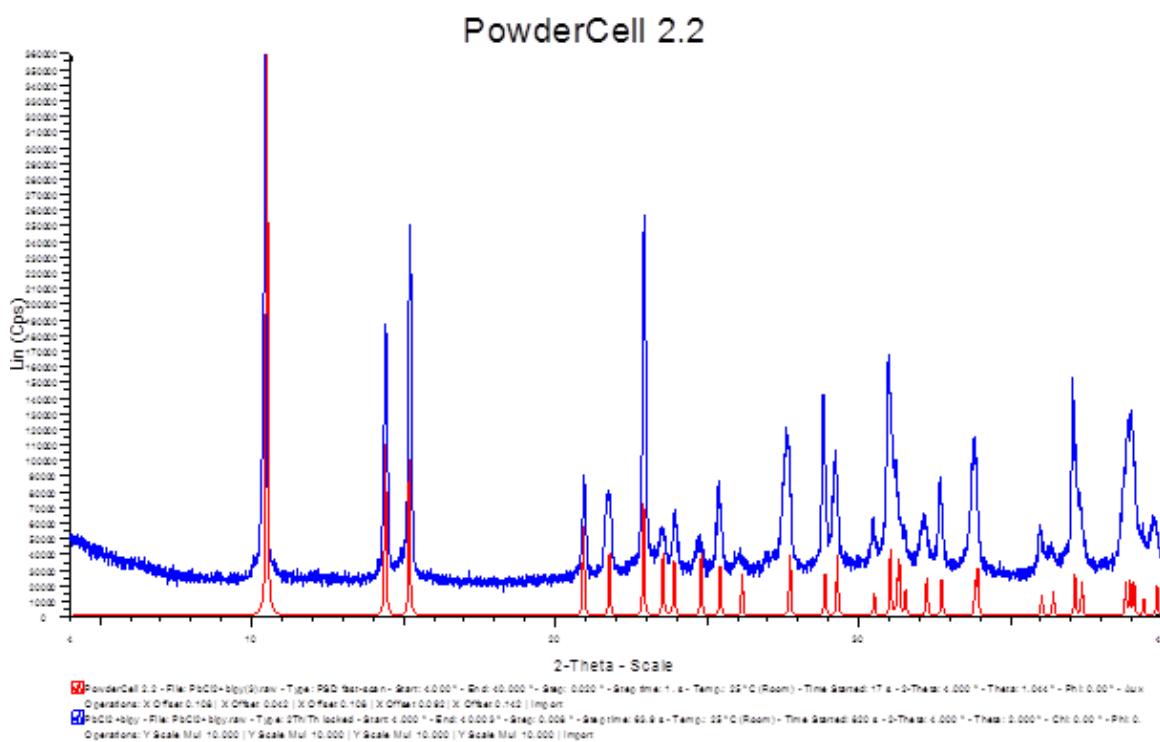
## C-III- [Pb(bp4)Cl<sub>2</sub>] (B<sub>β</sub>)

### C-III-A- Synthesis

B<sub>β</sub> has been synthesized according to the literature (ref 8 in main text):

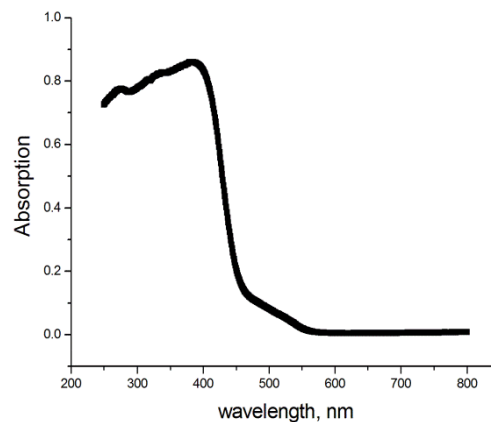
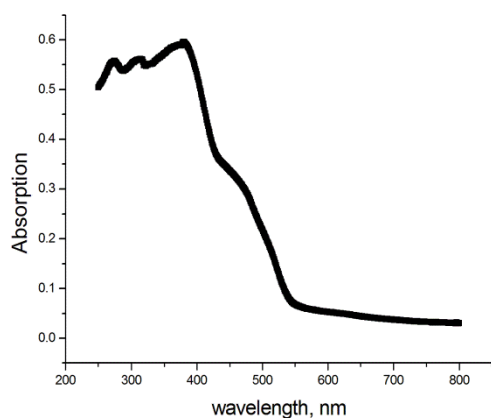


### C-III-B- XRPD of (B<sub>β</sub>) : theoretical (red) and experimental (blue)

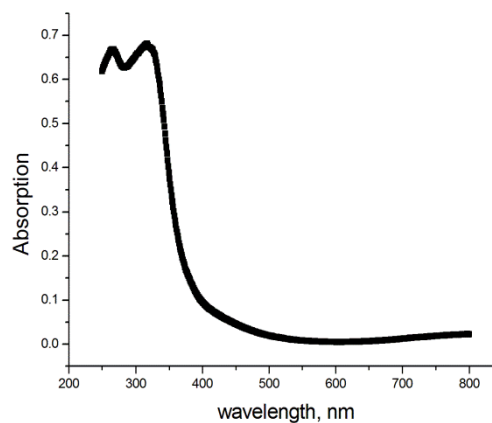
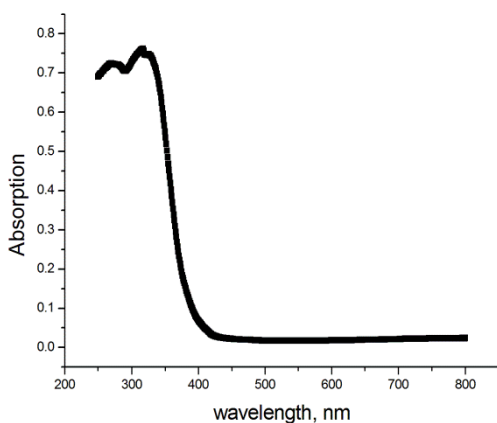


## D- Characterizations of (1), (2), (A), (B<sub>α</sub>), (B<sub>β</sub>): UV-Vis, photoluminescence, TGA-DSC

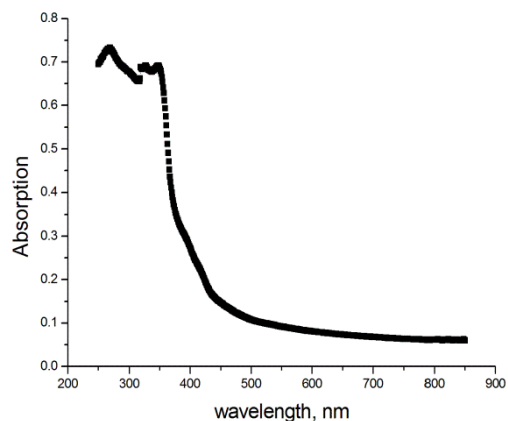
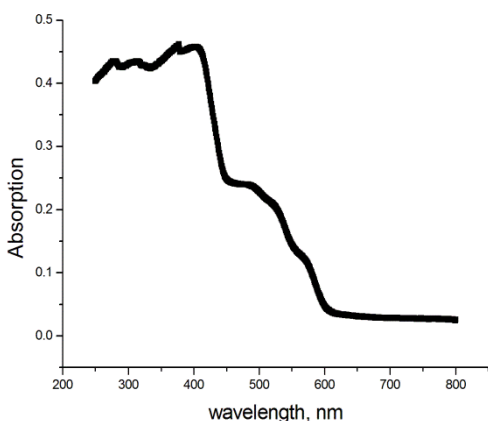
### D-I- UV-VIS spectra of (1), (2), (A), (B<sub>α</sub>), (B<sub>β</sub>) and of starting material (bp4mo, 2H<sub>2</sub>O)



UV-VIS spectra of (1) - [Pb(bp4mo)Cl<sub>2</sub>] (left) and (2) - [Pb(Hbp4mo)<sub>2</sub>Cl<sub>2</sub>](NO<sub>3</sub>)<sub>2</sub> (right)



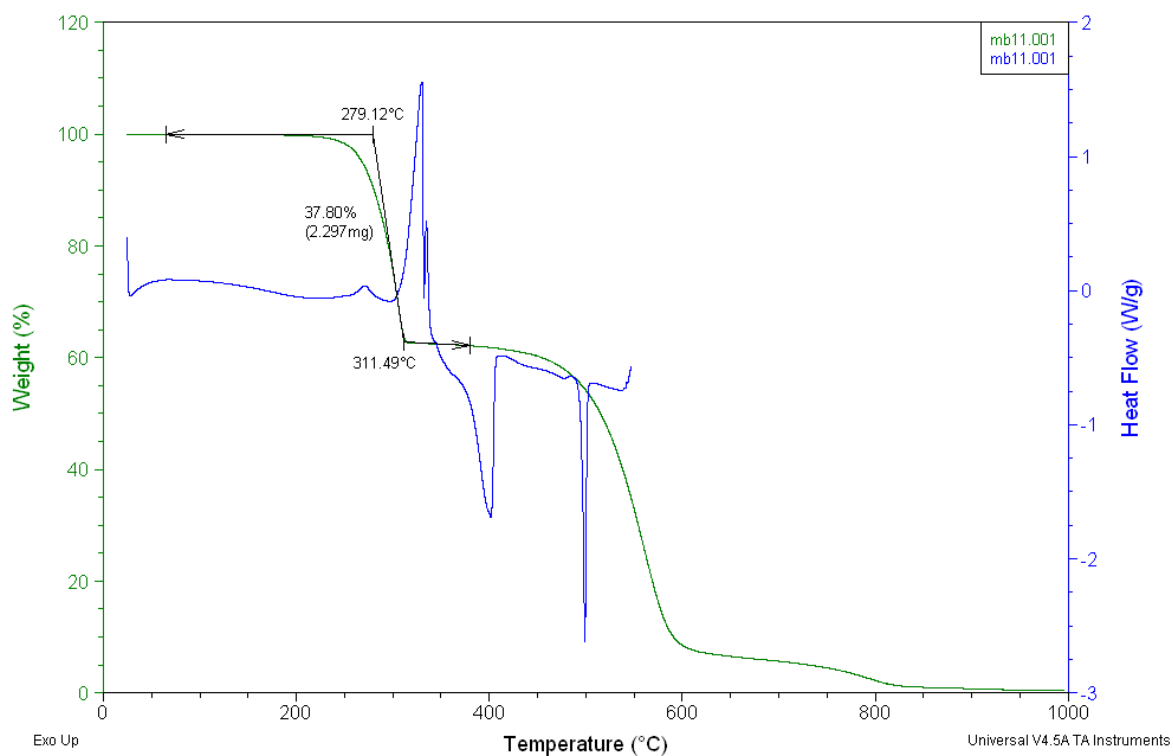
UV-VIS spectra of (B<sub>β</sub>) - [Pb(bp4)Cl<sub>2</sub>] (left) and (B<sub>α</sub>) - [Pb(bp4)Cl<sub>2</sub>] (right)



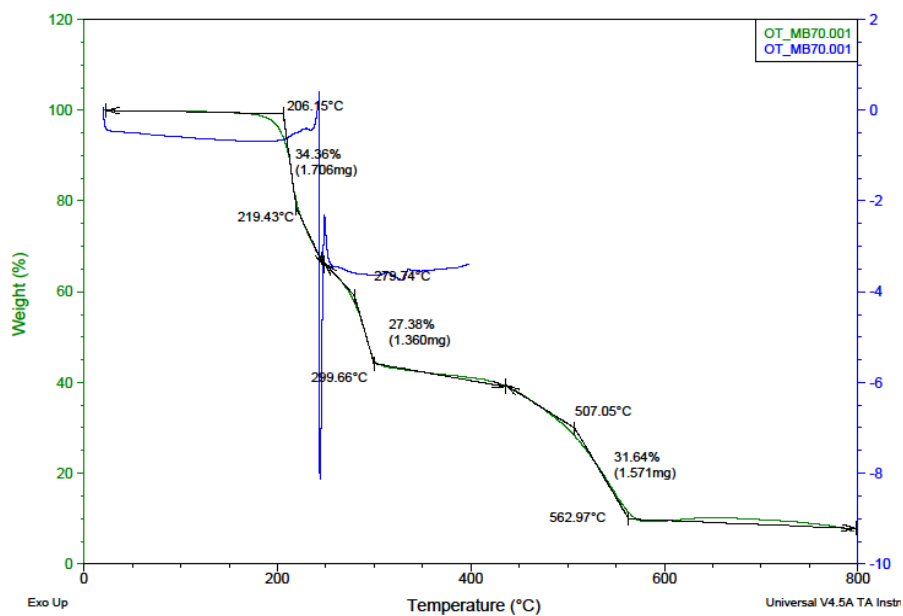
UV-VIS spectra of (A) - [Pb(bp4do)Cl<sub>2</sub>] (left) and starting material - bp4mo, 2H<sub>2</sub>O (right)

## D-II- TGA-DSC analysis of (1) and (2)

### D-II-1- TGA-DSC analysis of compound (2) – $[(Pb(bp4mo)Cl_2)]$

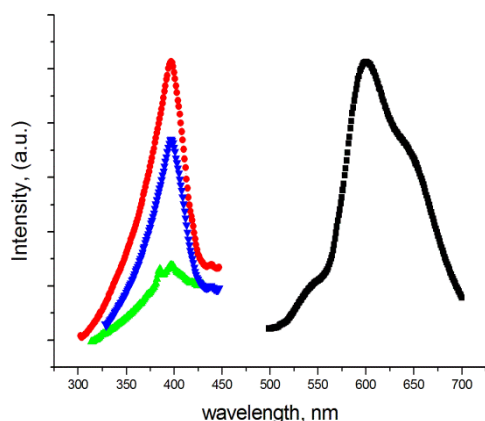


### D-II-2- TGA-DSC analysis of compound (2) – $[(Pb(Hbp4mo)_2Cl_2)](NO_3)_2$



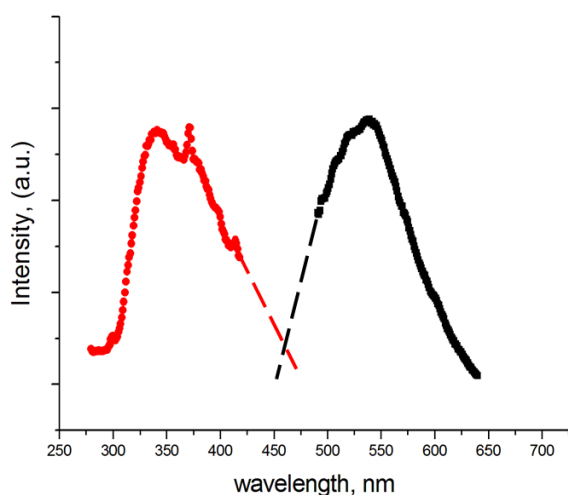
### D-III- Photoluminescence

#### D-III-1- Photoluminescence of (1) - [Pb(bp4mo)Cl<sub>2</sub>]



Solid-state emission ( $\lambda_{\text{ex}}=400\text{ nm}$ ) and excitation (green:  $\lambda_{\text{em}}=550\text{ nm}$ ; blue:  $\lambda_{\text{em}}=640\text{ nm}$ ; red:  $\lambda_{\text{em}}=600\text{ nm}$ ) spectra of (1) - [Pb(bp4mo)Cl<sub>2</sub>] at room temperature.

#### D-III-2- Photoluminescence of (B <sub>$\alpha$</sub> ) - [Pb(bp4)Cl<sub>2</sub>]



Solid-state emission (black line,  $\lambda_{\text{ex}}=350\text{ nm}$ ) and excitation (red line,  $\lambda_{\text{em}}=540\text{ nm}$ ) spectra of (B <sub>$\alpha$</sub> ) - [Pb(bp4)Cl<sub>2</sub>] at room temperature.

#### D-III-2- Photos of samples of 1, 2, B <sub>$\alpha$</sub> and B <sub>$\beta$</sub> taken under white light (top) and UV-light (bottom)

