

## Supporting Information

### Facile Synthesis of Tetrapod-Shaped Ag<sub>3</sub>PO<sub>4</sub> Microcrystals with an Increased Percentage of Exposed {110} Facets and Highly Efficient Photocatalytic Properties

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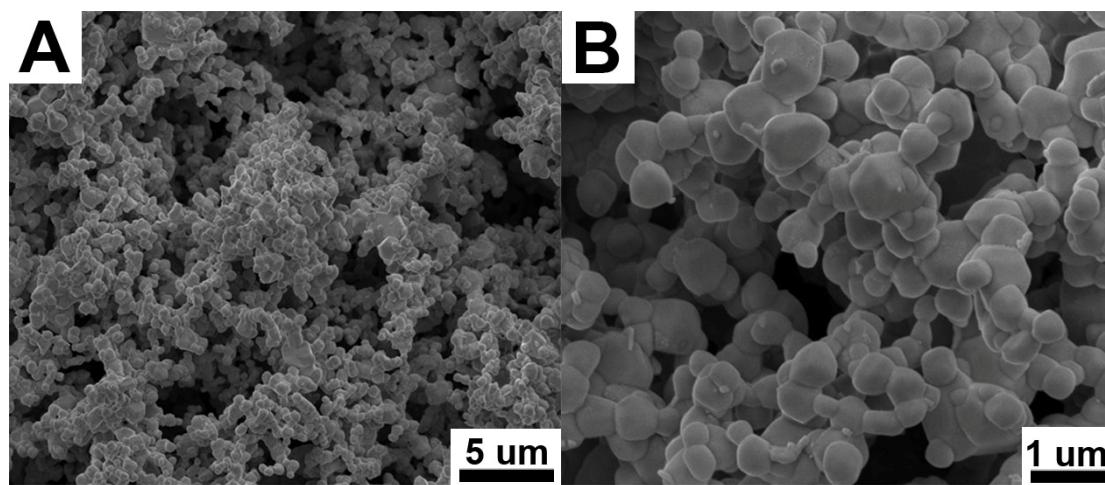
## EXPERIMENT

**Preparation of Ag<sub>3</sub>PO<sub>4</sub> Tetrapods.** The silver-ammino ( $[\text{Ag}(\text{NH}_3)_2]^+$ ) complex were used as the Ag<sup>+</sup> source, and typically, AgNO<sub>3</sub> (100 mg) was dissolved in 6 mL deionized water, then 168  $\mu\text{L}$  concentrated ammonia aqueous solution (NH<sub>3</sub>·H<sub>2</sub>O, mass fraction 25%-28%) was added to form a transparent solution. The Ag<sub>3</sub>PO<sub>4</sub> tetrapods were prepared using a facile precipitation reaction by the direct mixture of the above ( $[\text{Ag}(\text{NH}_3)_2]^+$ ) complex with Na<sub>2</sub>HPO<sub>4</sub> (5 g) aqueous solution at 60 °C. After continuous stirring for 1 h, the products were collected by centrifugation and washed several times

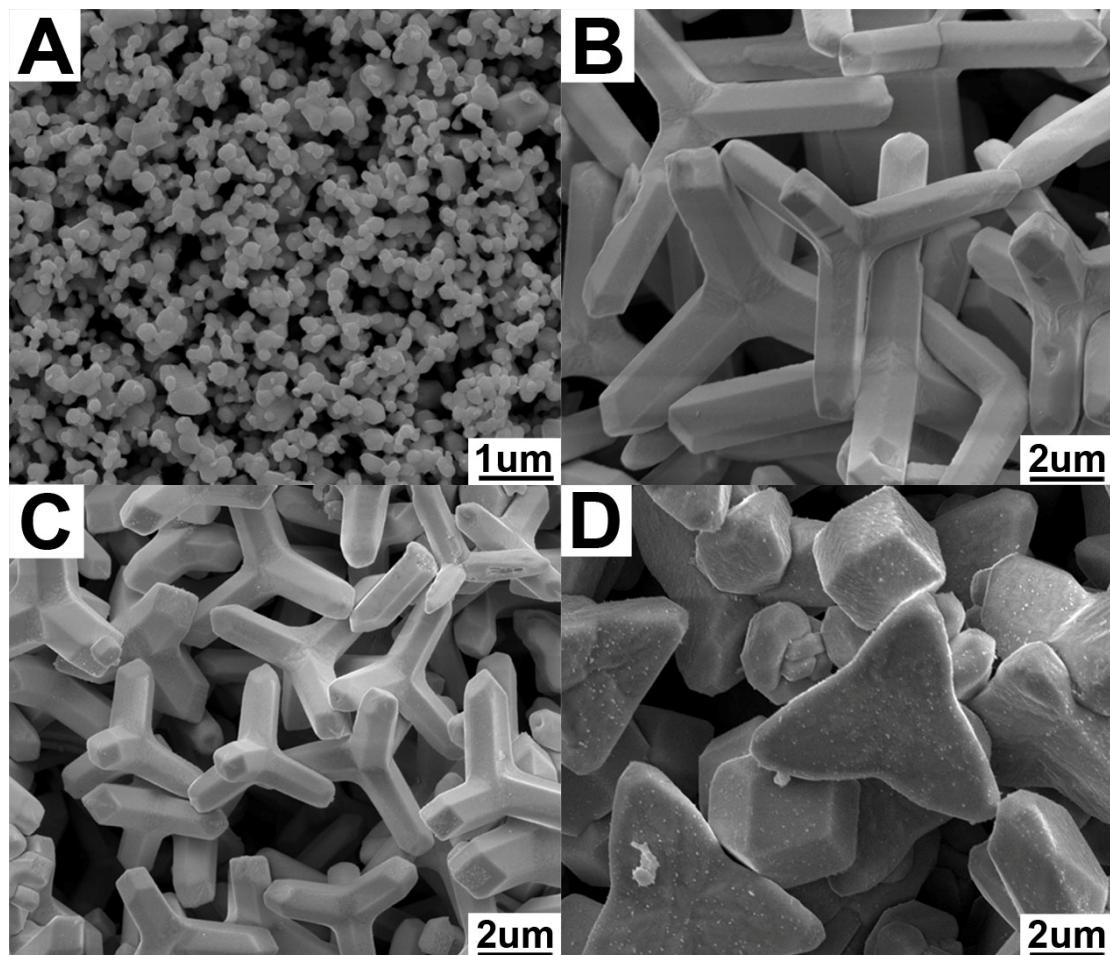
with distilled water, and then dried at room temperature in vacuum overnight. Some control experiments were carried out using the same process by simply adjusting the amount of added  $\text{NH}_3 \cdot \text{H}_2\text{O}$  with the concrete amount of 5 g  $\text{Na}_2\text{HPO}_4$ , and adjusting the amount of added  $\text{Na}_2\text{HPO}_4$  with the concrete amount of 168  $\mu\text{L}$   $\text{NH}_3 \cdot \text{H}_2\text{O}$ . The growth process was also investigated by replacing  $\text{Na}_2\text{HPO}_4$  with other sources of  $\text{PO}_4^{3-}$ , such as  $\text{NaH}_2\text{PO}_4$ ,  $\text{H}_3\text{PO}_4$  and  $\text{Na}_3\text{PO}_4$ . Moreover, the irregular  $\text{Ag}_3\text{PO}_4$  particles were prepared by a direct precipitation reaction with  $\text{AgNO}_3$  and  $\text{Na}_2\text{HPO}_4$  aqueous solutions without  $\text{NH}_3 \cdot \text{H}_2\text{O}$  as coordination agent. The N-doped  $\text{TiO}_2$  photocatalyst was prepared by nitridation of commercially available P25 powder under  $\text{NH}_3$  flow (flow rate of 350  $\text{mL min}^{-1}$ ) at 500 °C for 10 h.

**Characterization.** The morphology test of the samples was carried out on a FEI Quanta 400 scanning electron microscope (FEI Company, Oregon, USA). The X-ray diffraction (XRD) spectra of the samples were recorded by a Rigaku Dmax 2200 X-ray diffractometer with  $\text{Cu K}\alpha$  radiation ( $\lambda=1.5416 \text{ \AA}$ ). Diffuse reflectance absorption spectra of  $\text{Ag}_3\text{PO}_4$ -based photocatalyst were recorded in the range from 200 to 800 nm using a Hitachi U-3010 spectroscopy with  $\text{BaSO}_4$  as reference. Brunauer–Emmett–Teller (BET) nitrogen adsorption–desorption (NOVA 2200e, Quanhachrome, USA).

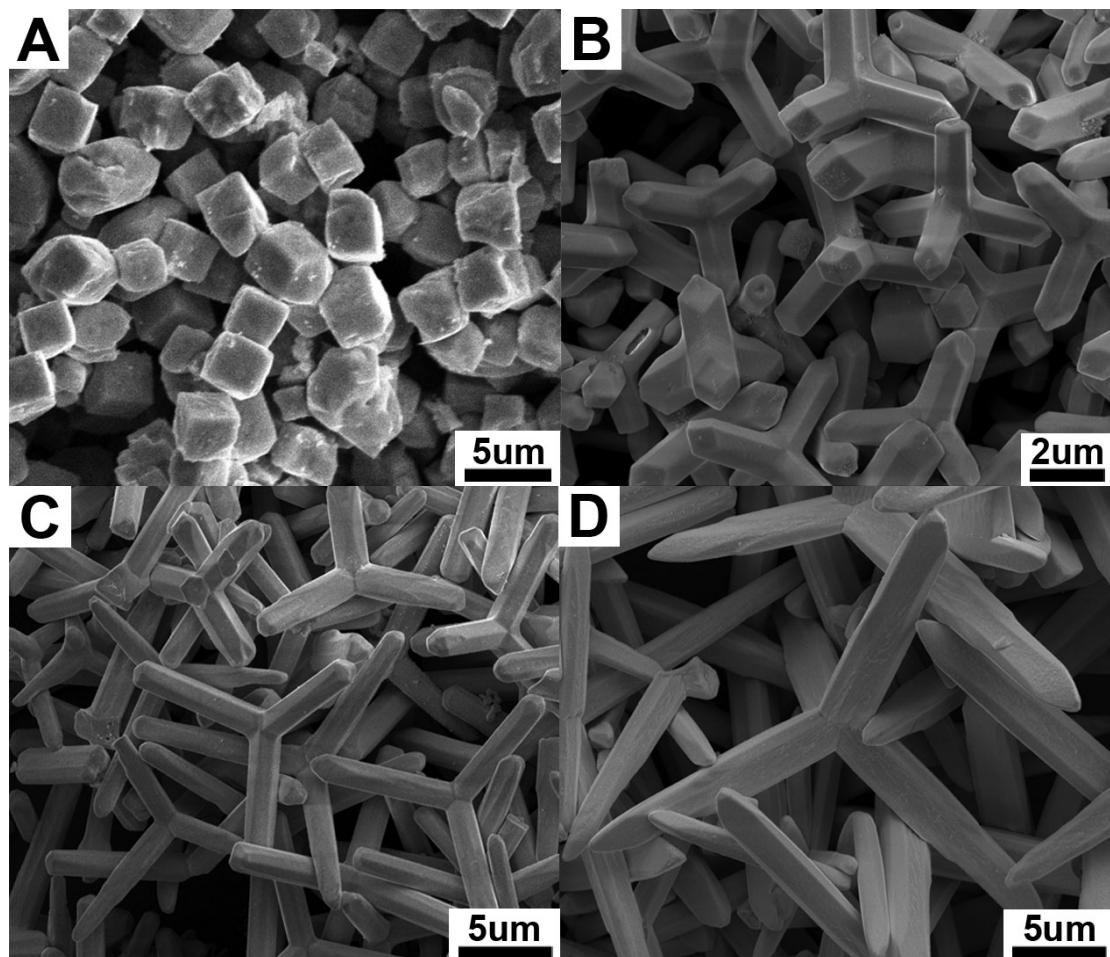
**Photocatalytic Experiment.** Photocatalytic activities of the  $\text{Ag}_3\text{PO}_4$ -based photocatalyst were evaluated by degradation of MO dyes under a 300 W Xe lamp with UV cutoff filter (providing visible light with  $\lambda \geq 400 \text{ nm}$ ). MO solution (100 mL,  $2 \times 10^{-5} \text{ mol} \cdot \text{L}^{-1}$ ) containing 0.1 g  $\text{Ag}_3\text{PO}_4$ -based photocatalysts or N-doped  $\text{TiO}_2$  was placed in a 200 mL cylindrical quartz vessel. Before the light was turned on, the solution was stirred in the dark for 30 min to ensure an adsorption/desorption equilibrium between the catalysts and organic dyes.



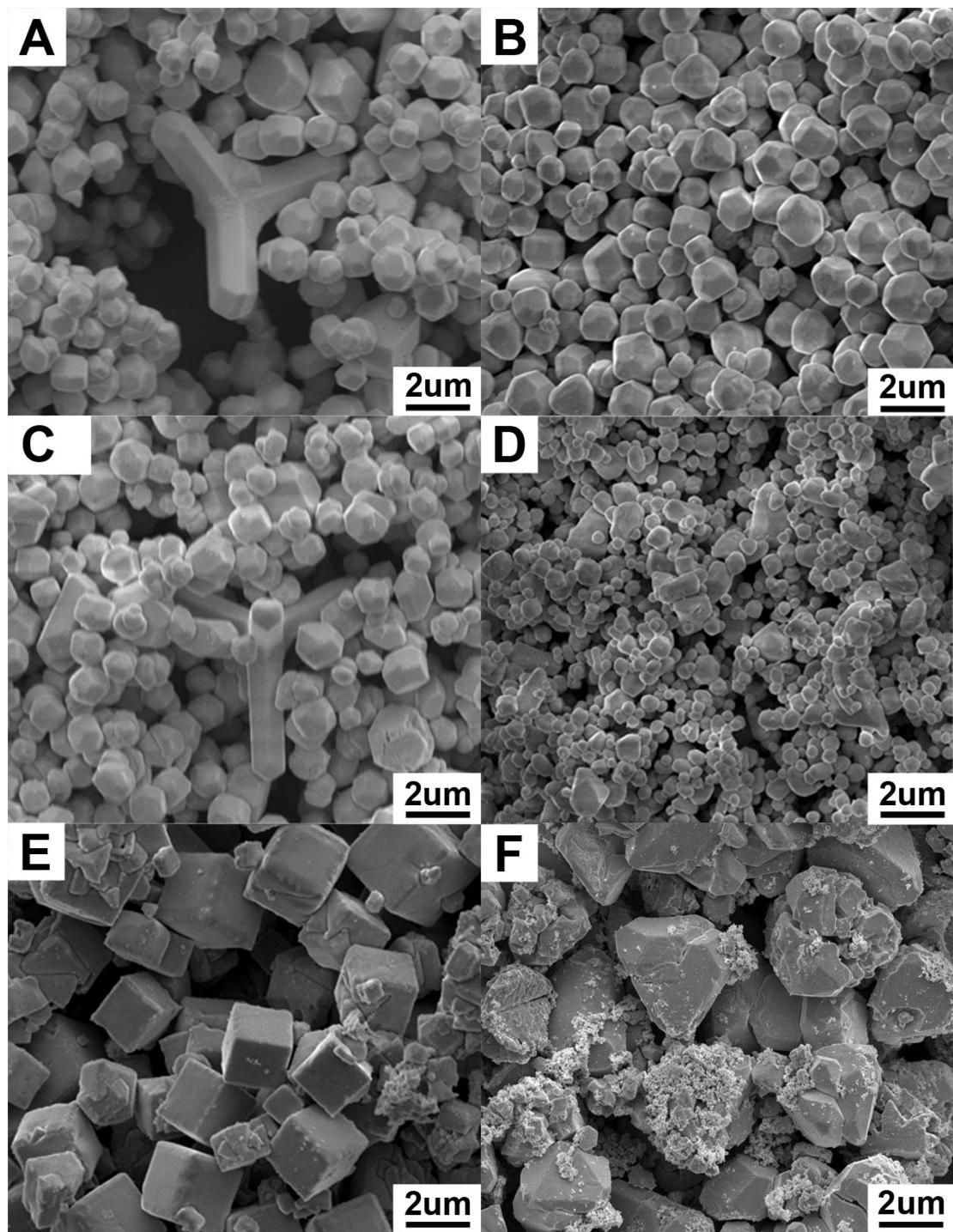
**Fig. S1.** SEM images of irregular  $\text{Ag}_3\text{PO}_4$  particles at low (A) and high (B) magnifications.



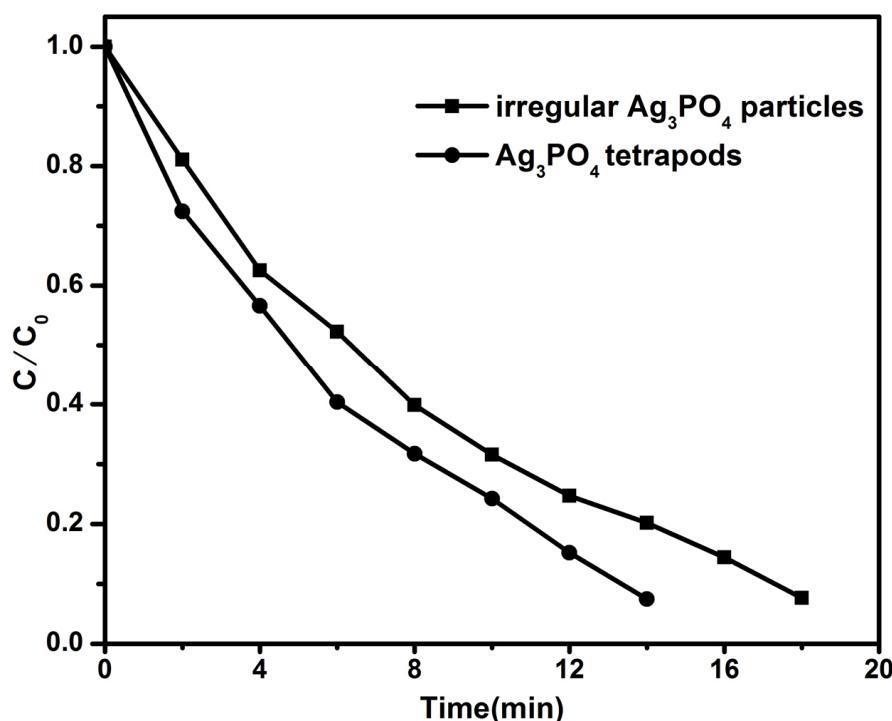
**Fig. S2.** SEM images of as-prepared  $\text{Ag}_3\text{PO}_4$  nanoparticles produced with the different amount of added  $\text{NH}_3 \cdot \text{H}_2\text{O}$ . (A) 64  $\mu\text{L}$  (B) 138  $\mu\text{L}$ , (C) 168  $\mu\text{L}$ , (D) 198  $\mu\text{L}$ .



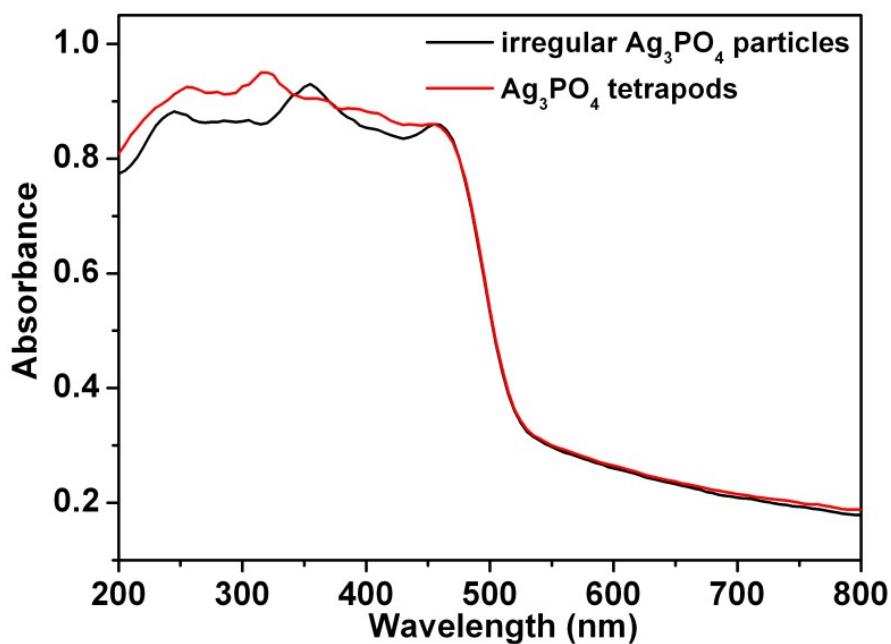
**Fig. S3.** SEM images of as-prepared  $\text{Ag}_3\text{PO}_4$  nanoparticles produced with the different amount of added  $\text{Na}_2\text{HPO}_4$ . (A) 3 g, (B) 5 g, (C) 6 g, (D) 7.5 g.



**Fig. S4.** SEM images of as-prepared  $\text{Ag}_3\text{PO}_4$  nanoparticles produced by replacing  $\text{Na}_2\text{HPO}_4$  with other sources of  $\text{PO}_4^{3-}$ , such as  $\text{NaH}_2\text{PO}_4$  (A,B),  $\text{H}_3\text{PO}_4$  (C,D) and  $\text{Na}_3\text{PO}_4$  (E,F).



**Fig. S5.** Photodegradation of higher concentration MO dyes solution over as-prepared  $\text{Ag}_3\text{PO}_4$  tetrapods and irregular  $\text{Ag}_3\text{PO}_4$  particles.



**Fig. S6.** Diffuse reflectance absorption spectra (DRS) of as-prepared  $\text{Ag}_3\text{PO}_4$  tetrapods (a) and irregular particles (b).