## **Supporting materials**

Tuning	zinc(II)	coordin	ation	polyn	ners	based	on
bis(1,2,4-	-triazol-1-y	l)ethane		and		5-substi	tuted
1,3-benze	enedicarbo	xylates:	Syntl	heses,	stru	ctures	and
propertie	es						

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		1	
Zn(1)-N(3)	2.020(5)	Zn(1)-N(6B)	2.026(5)
Zn(1)-O(1)	2.003(4)	Zn(1)-O(3A)	1.960(4)
N(3)-Zn(1)-N(6B)	113.1(2)	O(1)-Zn(1)-N(3)	107.1(2)
O(3A)-Zn(1)-N(3)	103.6(2)	O(1)-Zn(1)-N(6B)	115.6(2)
O(3A)-Zn(1)-N(6B)	115.0(2)	$O(3^{a})-Zn(1)-O(1)$	101.20(18)
		2	
Zn(1)-N(3)	2.049(4)	Zn(1)-N(6B)	2.009(4)
Zn(1)-O(1)	1.975(3)	Zn(1)-O(3A)	1.975(3)
N(6B)-Zn(1)-N(3)	108.82(14)	O(1)-Zn(1)-N(3)	114.01(13)
O(3A)-Zn(1)-N(3)	108.76(13)	O(1)-Zn(1)-N(6B)	109.87(13)
O(3A)-Zn(1)-N(6B)	112.93(13)	$O(3^{a})-Zn(1)-O(1)$	102.43(11)
		3	
Zn(1)-N(3)	2.123(4)	Zn(1)-N(6B)	2.142(4)
Zn(1)-O(1)	2.104(3)	Zn(1)-O(4A)	2.059(3)
Zn(1)-O(5)	2.104(3)		
N(3)-Zn(1)-N(6B)	93.07(15)	O(1)-Zn(1)-N(3)	134.71(13)
O(4A)-Zn(1)-N(3)	128.05(12)	O(5)-Zn(1)-N(3)	90.50(15)
O(1)-Zn(1)-N(6B)	86.81(14)	O(4A)-Zn(1)-N(6B)	91.78(14)
O(5)-Zn(1)-N(6B)	175.85(15)	O(4A)-Zn(1)-O(1)	97.20(13)
O(5)-Zn(1)-O(1)	89.17(13)	O(4A)-Zn(1)-O(5)	87.67(13)
		4	
Zn(1)-N(3)	2.194(4)	Zn(1)-N(6A)	2.160(4)

Table S1 Selected bond lengths [Å] and angles [°] for 1-6

2.171(4)	Zn(1)-N(12A)	2.193(4)
2.089(4)	Zn(1)-O(2)	2.156(4)
90.54(14)	N(9)-Zn(1)-N(3)	87.68(15)
178.49(14)	N(6A)-Zn(1)-N(9)	176.84(15)
90.11(14)	N(9)-Zn(1)-N(12A)	91.60(15)
88.25(15)	O(2)-Zn(1)-N(3)	90.73(14)
88.50(15)	N(6A)-Zn(1)-O(2)	86.53(15)
88.84(15)	O(2)-Zn(1)-N(9)	96.10(16)
90.41(15)	O(2)-Zn(1)-N(12A)	90.66(14)
174.91(15)		
5	;	
2.004(3)	Zn(1)-N(6B)	2.029(3)
1.974(3)	Zn(1)-O(3A)	1.963(3)
110.99(13)	O(1)-Zn(1)-N(3)	117.09(13)
111.61(13)	O(1)-Zn(1)-N(6B)	110.01(13)
97.47(13)	O(3A)-Zn(1)-O(1)	107.88(12)
6	î	
2.058(5)	Zn(1)-N(3C)	2.058(5)
1.990(4)	Zn(1)-O(6B)	1.990(4)
1.998(4)	Zn(2)-O(1)	1.973(4)
1.946(4)	Zn(2)-O(7)	2.028(4)
93.1(3)	O(6B)-Zn(1)-N(3)	95.42(18)
117.71(18)	O(6A)-Zn(1)-N(3C)	95.42(18)
117.71(18)	O(6A)-Zn(1)-O(6B)	132.0(2)
119.69(17)	O(4D)-Zn(2)-N(6)	119.04(18)
106.06(17)	O(4D)-Zn(2)-O(1)	100.32(16)
106.69(16)	O(4D)-Zn(2)-O(7)	103.47(17)
	$\begin{array}{c} 2.171(4)\\ 2.089(4)\\ 90.54(14)\\ 178.49(14)\\ 90.11(14)\\ 88.25(15)\\ 88.50(15)\\ 88.84(15)\\ 90.41(15)\\ 174.91(15)\\ 174.91(15)\\ 10.99(13)\\ 110.99(13)\\ 111.61(13)\\ 97.47(13)\\ & & & & & & & & & & & & & & & & & & &$	2.171(4) $Zn(1)-N(12A)$ $2.089(4)$ $Zn(1)-O(2)$ $90.54(14)$ $N(9)-Zn(1)-N(3)$ $178.49(14)$ $N(6A)-Zn(1)-N(9)$ $90.11(14)$ $N(9)-Zn(1)-N(12A)$ $88.25(15)$ $O(2)-Zn(1)-N(3)$ $88.50(15)$ $N(6A)-Zn(1)-O(2)$ $88.84(15)$ $O(2)-Zn(1)-N(9)$ $90.41(15)$ $O(2)-Zn(1)-N(9)$ $114.91(15)$ $O(2)-Zn(1)-N(9)$ $90.41(15)$ $O(1)-Zn(1)-N(6B)$ $1.974(3)$ $Zn(1)-O(3A)$ $110.99(13)$ $O(1)-Zn(1)-N(3)$ $111.61(13)$ $O(1)-Zn(1)-N(3)$ $97.47(13)$ $O(3A)-Zn(1)-O(1)$ $6$ $Zn(2)-O(1)$ $2.058(5)$ $Zn(1)-N(3C)$ $1.990(4)$ $Zn(2)-O(1)$ $1.998(4)$ $Zn(2)-O(1)$ $1.998(4)$ $Zn(2)-O(7)$ $93.1(3)$ $O(6B)-Zn(1)-N(3C)$ $117.71(18)$ $O(6A)-Zn(1)-N(3C)$ $117.71(18)$ $O(6A)-Zn(1)-O(6B)$ $119.69(17)$ $O(4D)-Zn(2)-O(1)$ $106.06(17)$ $O(4D)-Zn(2)-O(1)$

Symmetry transformations used to generate equivalent atoms: A x, y, z-1; B -x+1, -y, -z+1 for 1; A x+1, y-1, z; B -x+1, -y, -z for 2; A x, y+1, z; B -x, -y+1, -z+1 for 3; A x, -y+1/2, z+1/2 for 4; A x+1/2, -y+1/2, z+1/2; B -x+1/2, y-1/2, -z+3/2 for 5; A x-1/2, -y+3/2, z+1/2; B -x+3/2, -y+3/2, -z+1; C -x+1, y, -z+3/2; D x-1/2, y+1/2, z for 6.

## Table S2 Hydrogen bondings for 3 (Å and $^{\rm o})$

D-H <sup></sup> A	d(D-H)	d(H <sup></sup> A)	$D(D^{}A)$	<(DHA)
O(5)-H(1W) <sup></sup> O(6) <sup>i</sup>	0.886(19)	1.89(2)	2.751(5)	163(5)
O(5)-H(2W) <sup></sup> O(2) <sup>i</sup>	0.889(19)	1.783(19)	2.681(5)	178(4)
O(6)-H(3W) <sup></sup> O(3)	0.87(2)	1.92(3)	2.746(5)	159(5)
O(6)-H(4W) <sup></sup> N(5) <sup>ii</sup>	0.90(2)	2.46(5)	3.223(6)	143(6)

Symmetry transformations used to generate equivalent atoms: i -x+1/2, -y+1/2, -z+1; ii x, y-1, z.

D-H <sup></sup> A	d(D-H)	d(H <sup></sup> A)	D(D <sup></sup> A)	<(DHA)
O(1)-H(1W) <sup></sup> O(10)	0.80(5)	1.96(5)	2.742(6)	166(6)
O(1)-H(2W) <sup></sup> O(12)	0.76(6)	1.92(6)	2.681(6)	171(6)
O(2)-H(3W) <sup></sup> O(10) <sup>i</sup>	0.84(6)	2.08(6)	2.913(6)	174(6)
O(2)-H(4W) <sup></sup> O(8)	0.88(6)	1.85(6)	2.721(6)	167(5)
O(5)-HWA <sup></sup> O(4) <sup>ii</sup>	0.88(2)	1.67(2)	2.534(5)	166(5)
O(10)-H(5W) <sup></sup> O(7) <sup>iii</sup>	0.74(6)	2.25(6)	2.916(6)	152(7)
O(10)-H(6W) <sup></sup> O(11)	0.78(6)	2.03(6)	2.734(6)	149(6)
O(11)-H(7W) <sup></sup> O(3) <sup>iv</sup>	0.92(7)	1.96(7)	2.802(5)	152(6)
O(11)-H(8W) <sup></sup> O(13)	0.91(6)	1.99(7)	2.893(6)	174(6)
O(12)-H(9W) <sup></sup> O(8) <sup>iii</sup>	0.95(6)	1.93(7)	2.835(7)	159(6)
O(12)-H(10W) <sup></sup> O(7) <sup>v</sup>	0.82(4)	2.38(5)	3.142(7)	154(6)
O(13)-H(11W) <sup></sup> N(11) <sup>vi</sup>	0.87(2)	2.16(2)	3.030(6)	174(5)
O(13)-H(12W) <sup></sup> O(4) <sup>ii</sup>	0.90(2)	2.14(2)	3.021(5)	169(6)
$O(13)^{-11}(1200) O(4)$	0.90(2)	2.1 T(2)	5.021(5)	107(0)

Table S3 Hydrogen bondings for 4 (Å and °)

Symmetry transformations used to generate equivalent atoms: i -x+1, y-1/2, -z+3/2; ii -x, y+1/2, -z+3/2; iii x, -y+1/2, z+1/2; iv x, y+1, z; v -x+1, y+1/2, -z+3/2; vi -x, -y+1, -z+1.

## Table S4 Hydrogen bondings for 5 (Å and °)

D-H <sup></sup> A	d(D-H)	d(H <sup></sup> A)	D(D <sup></sup> A)	<(DHA)
O(6)-H(1W) <sup></sup> O(5) <sup>i</sup>	0.894(19)	1.73(2)	2.624(4)	175(4)
O(7)-H(2W) <sup></sup> O(8) <sup>ii</sup>	0.908(19)	2.06(4)	2.818(8)	140(4)
O(7)-H(3W) <sup></sup> O(9) <sup>ii</sup>	0.900(19)	1.83(3)	2.749(8)	162(5)

Symmetry transformations used to generate equivalent atoms: i -x+3, -y, -z+1; ii x+1, y, z.

<b>Table S5</b> Hydrogen bondings for 6 (Å and $^{\circ}$ )	

D-H <sup></sup> A	d(D-H)	d(H <sup></sup> A)	D(D <sup></sup> A)	<(DHA)
O(7)-H(1W) <sup></sup> O(2) <sup>i</sup>	0.84(2)	1.88(2)	2.704(6)	168(6)
O(7)-H(2W) <sup></sup> O(8) <sup>ii</sup>	0.83(2)	1.86(2)	2.684(6)	169(6)
O(8)-H(3W) <sup></sup> O(8) <sup>iii</sup>	0.85(2)	1.99(3)	2.772(6)	152(5)
O(8)-H(4W) <sup></sup> O(9) <sup>iv</sup>	0.847(19)	1.91(3)	2.736(6)	166(5)

Symmetry transformations used to generate equivalent atoms: i -x+3/2, -y+1/2, -z+1; ii -x+1, -y+1, -z+1; iii -x+3/2, -y+3/2, -z+1; iv x-1/2, y+1/2, z.

Ntriazole-(C-C)etahne-Ntriazole and Zii Zii distances bridged by bie in 1-0					
compound	dihedral angle (°)	torsion angle (°)	$Zn \cdot \cdot \cdot Zn$ distance (Å)		
1	53.0(2)	47.4(8)	6.540(2)		
2	51.8(2)	-62.7(5)	7.711(3)		
3	67.2(2)	-49.6(6)	7.9020(10)		
4	53.28(4), 59.62(2)	65.5(5), 64.6(5)	8.327(2)		
5	40.71(2)	179.0(4)	9.392(2)		
6	41.20(2)	71.5(7)	8.1715(11)		

**Table S6** The dihedral angle between the two triazol rings, and the torsion angle  $N_{triazole}$ -(C-C)<sub>etahne</sub>-N<sub>triazole</sub> and Zn<sup>...</sup>Zn distances bridged by bte in **1-6** 



Fig. S1 The coordination environment of The Zn(II) atom in 1.



Fig. S2 The coordination environment of The Zn(II) atom in 2.



Fig. S3 The coordination environment of The Zn(II) atom in 3.



Fig. S4 The coordination environment of The Zn(II) atom in 4.



Fig. S5 The hydrogen bond interactions in 4.



Fig. S6 The coordination environment of The Zn(II) atom in 5.



Fig. S7 The coordination environment of the Zn(II) atoms in 6.



Fig. S8 Each Zn1 atom connects two  $[Zn(btc)]_n$  chain based on Zn2 atoms and btc ligands in 6.



Fig. S9 Solid state emissions of compounds 1, 2, 3, 4, 5 and 6.



Fig. S10 The TG curves of compounds 1, 2, 3, 4, 5 and 6.