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Calcium sulfate hemihydrate-mediated crystallization of gypsum on Ca²⁺-activated cellulose thin films

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Supplementary Information

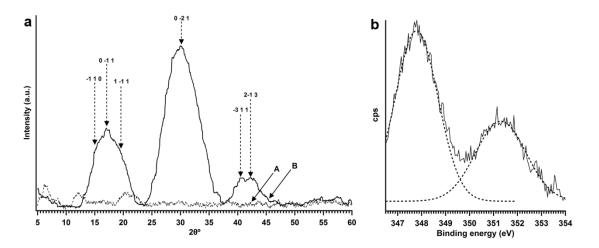


Fig. S1. (a) X-ray diffraction profiles of (A) untreated and (B) Ca-treated cellulose thin films. Peaks at 17, 30.5, 40.5 and 42° in profile **B** are associated with the evaporation-induced *in situ* crystallization of CaCl₂(H₂O)₄ during sample preparation. (b) XPS spectrum showing Ca $2p_{3/2}$ binding energies (Demri, B. *J. Mater. Processing Techn.* **55**, 311-314 (1995)).The peak at 351.3 eV corresponds to Ca $2p_{1/2}$ -Ca $2p_{3/2}$ spin-orbit splitting, while the peak at 347.8 eV is attributed to the binding energy in the CaCl₂ evaporite.

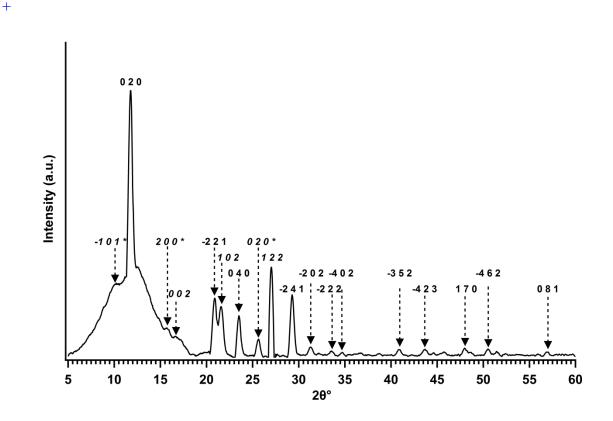


Fig. S2. XRD powder patterns for sample prepared from calcium sulfate hemihydrate (bassanite)-mediated crystallization on vertically oriented spin-coated thin films of Camodified cellulose. Reflections observed at 20 values of 11.8, 21.0, 23.5, 29.3, 31.3, 33.5, 34.5, 41.0, 43.5, 48.0, 51.0 and 56.7° indicated that gypsum (CaSO₄. 2H₂O) was produced on the films at room temperature under stirred reaction conditions. Peaks at 10.0, 15.5 and 25.7° correspond to residual undissolved bassanite. Peaks at 16.5, 21.7 and 27.0° are related to Na₂SO₄(H₂O)₂ contamination .

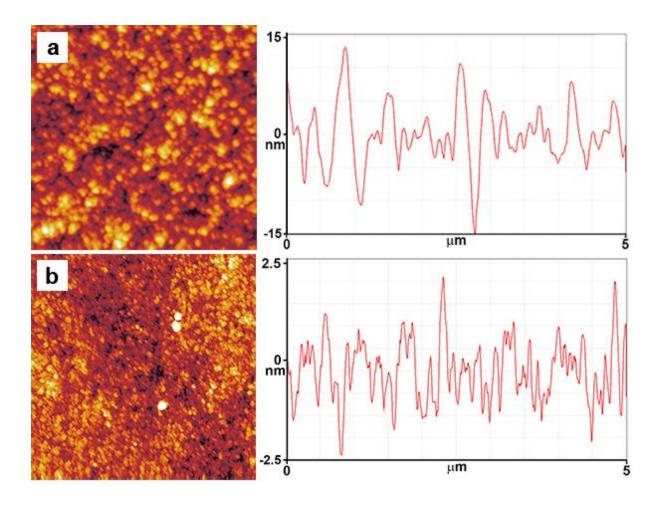


Fig. S3. AFM z-height images of the surface roughness of (a) Ca-activated and (b) untreated cellulose films. Corresponding roughness profiles are shown in the right hand side panels.

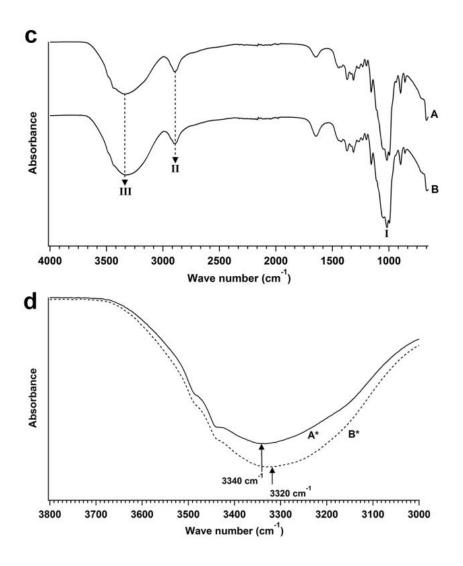


Fig. S4. ATR-FTIR spectra across a range of (a) 4000-600 cm⁻¹ and (b) 3800-3000 cm⁻¹ for untreated cellulose film (A/A*) and Ca-activated cellulose films (B/B*). Arrows in (b) show peak shift in the presence of calcium.