<Electronic Supplementary Information>

## Insight into the crystallization process: relationships between crystal structures and properties of copper(II) coordination polymers containing dimethylbis(4-pyridyl)silane

So Yun Moon, Tae Hwan Noh and Ok-Sang Jung\*

Department of Chemistry, Pusan National University, Pusan 609-735, Korea



**Fig. S1.** Powder XRD data of  $[Cu(L)_2Br_2]$  (a),  $[Cu(L)_2Br]Br \cdot 2Me_2SO$  (b), and  $[Cu(L)Br_2(O-Me_2SO)] \cdot Me_2SO$  (c).



	С	Ν	В	0	F	Si	Cl	Br	Cu
[Cu(L) <sub>2</sub> Br]Br ·2Me <sub>2</sub> SO	62.292	5.884	_	-	_	11.418	_	13.939	6.467
$[Cu(L)_2Br]BF_4$	62.262	8.490	2.662	-	10.884	7.3981	-	4.003	3.718
[Cu(L) <sub>2</sub> Br]ClO <sub>4</sub>	58.844	6.740	_	14.212	_	7.838	3.710	4.675	3.980
$[Cu(L)_2Br]NO_3$	59.217	11.552	_	12.578	_	8.296	_	4.876	3.481

Fig. S2. SEM images and EDX data for  $[Cu(L)_2Br]Br$  (a),  $[Cu(L)_2Br]BF_4$  (b),

 $[Cu(L)_2Br]ClO_4\ (c),\ and\ [Cu(L)_2Br]NO_3\ (d).$ 



**Fig. S3.** TGA (red) and DSC (blue) curves of  $[Cu(L)_2Br][Br \cdot 2H_2O] \cdot 4CH_3OH$ .

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