

Supporting Materials

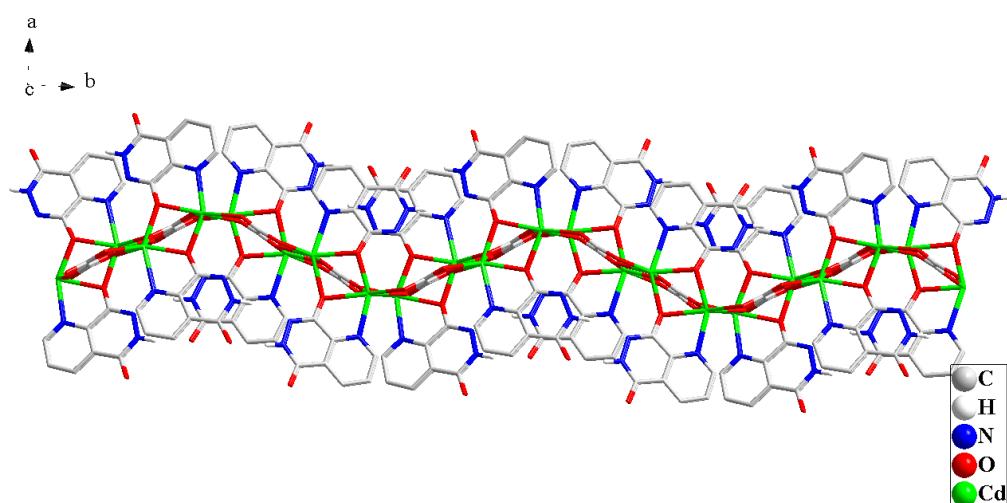
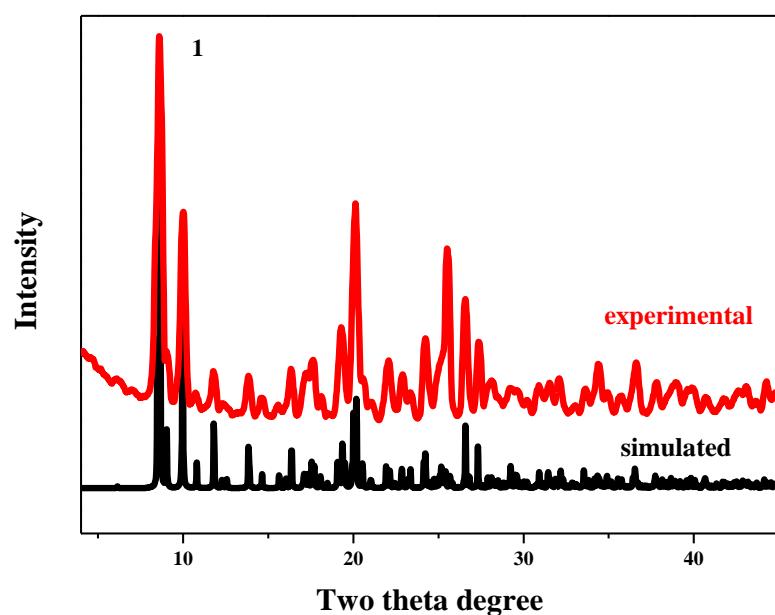


Fig. S1 The projection plot of 2-D layer for compound **2** in *ab* plane.



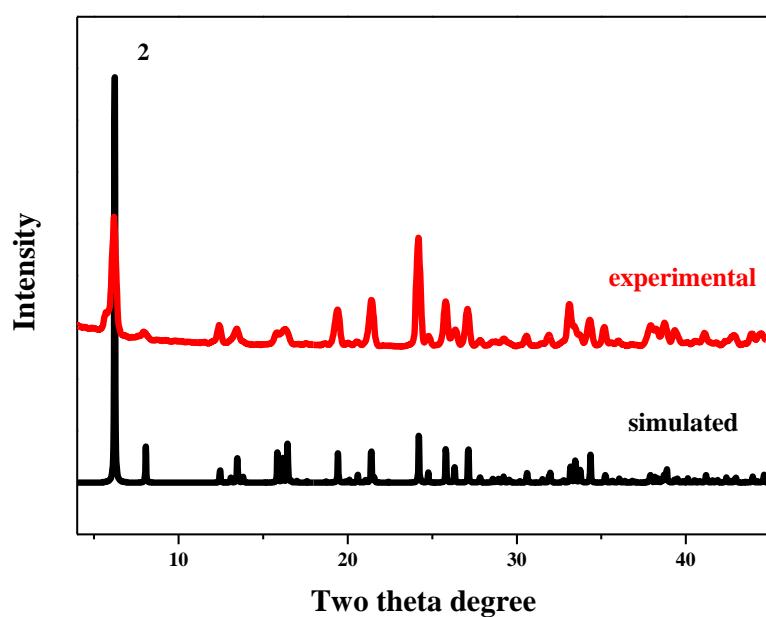
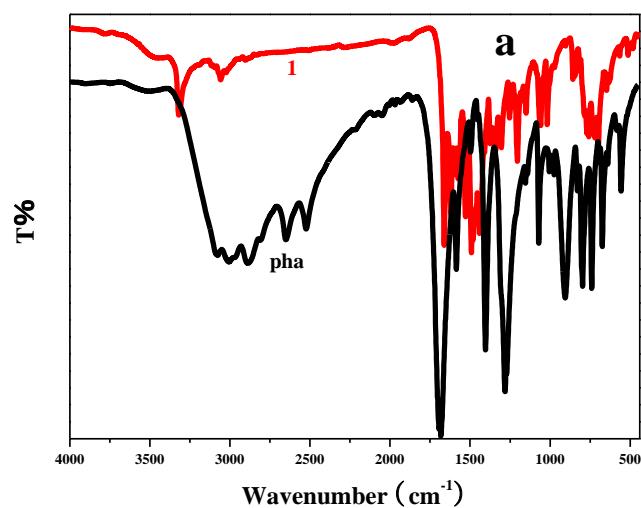


Fig. S2 The simulated (black) and experimental (red) powder XRD patterns for compounds **1** and **2**.



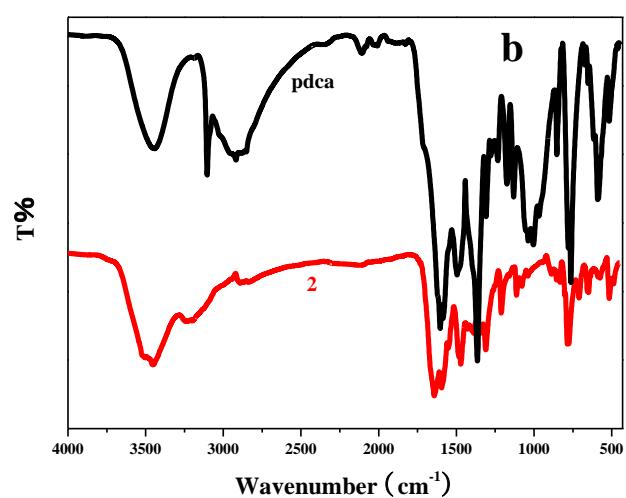


Fig. S3 The IR spectrum of compounds **1** and **2**.

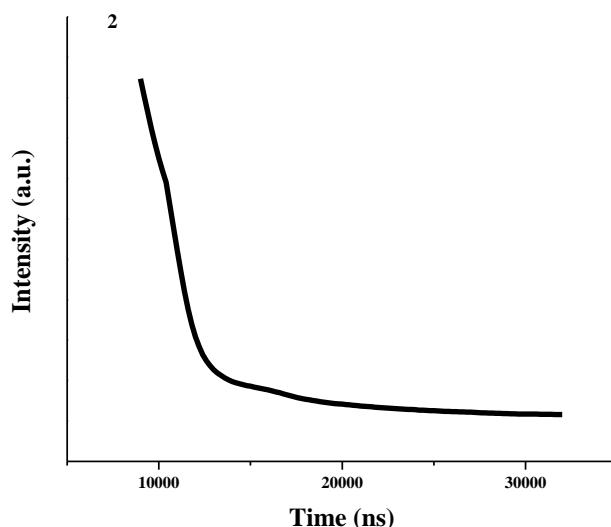


Fig. S4 The experimental decay curve for the solid state compound **2**.

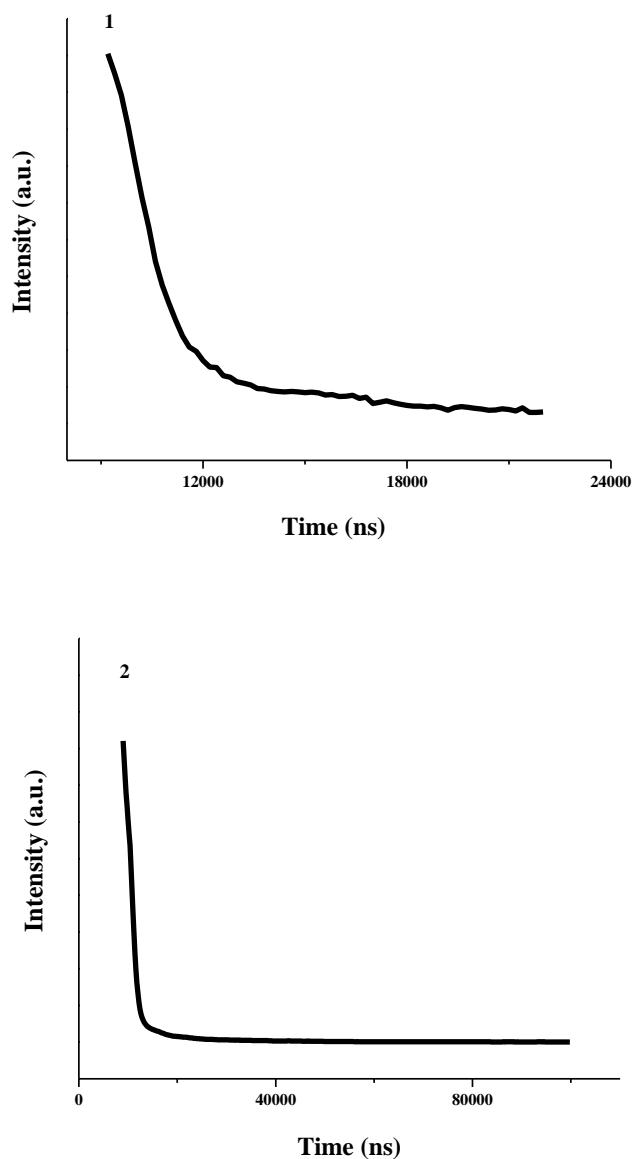
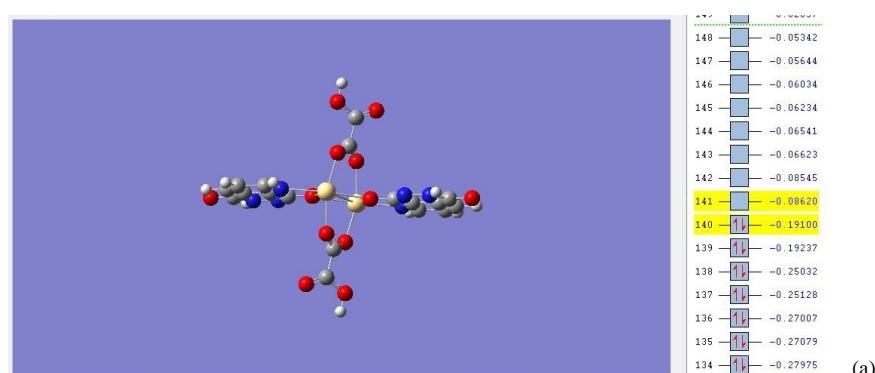


Fig. S5 The experimental decay curves for compounds **1** (above) and **2** (bottom) in an aqueous solution.



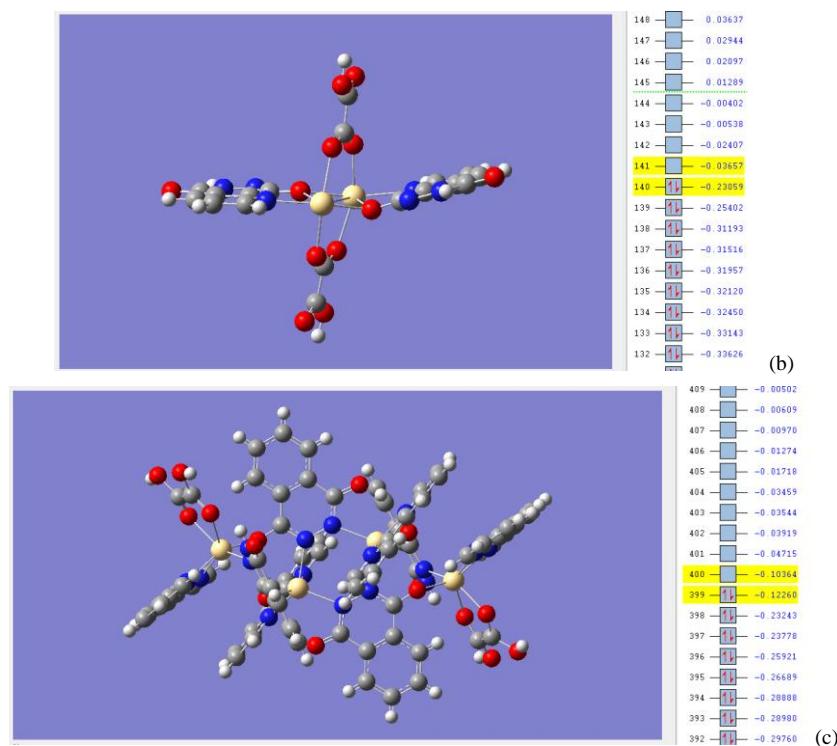


Fig. S6. The first singlet excited state structures for the solid-state compound **2** (a), compounds **2** (b) and **1** (c) in an aqueous solution.

Computational details for DFT calculations

The optimized geometric structure for the excited states of the solid-state compound **2** was performed with the TD-LC-wPBE method.¹ The calculations for non Cd atoms was carried out with 6-31g(d) basis set,² while for Cd atom with LanLDDZ basis set.³ The emission spectra was obtained using the time-dependent density functional theory TD-B3LYP.⁴ All calculations were performed with the GAUSSIAN 09 program package.⁵ The molecular orbitals were plotted with the GaussView program.⁶

References

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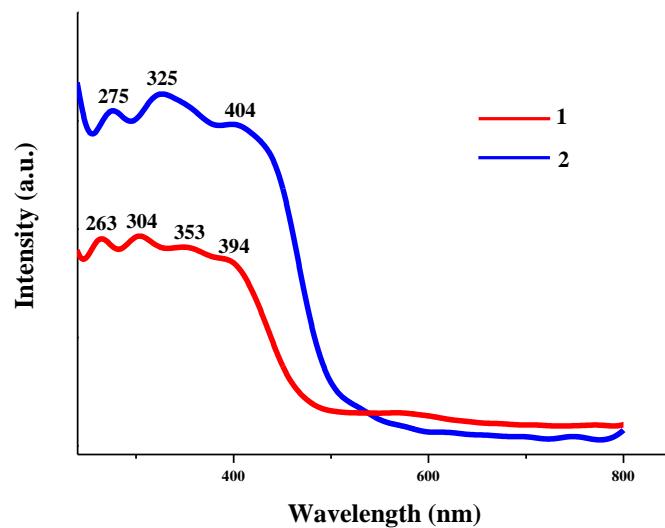


Fig. S7 The electronic absorption spectra of compounds **1** and **2** in the solid state.

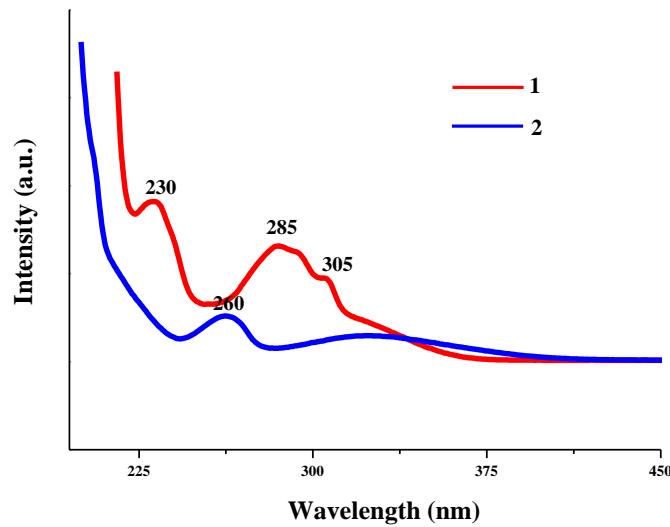


Fig. S8 The electronic absorption spectra of compounds **1** and **2** in an aqueous solution.