Supporting information for

Flexible linkers and dinuclear metallic nodes build up an original metal-

organic framework

Angelica Vlad,^a Mirela-Fernanda Zaltariov,^a Sergiu Shova,^a Ghenadie Novitchi,^b

Cristian-Dragos Varganici,^a Cyrille Train,^{b,c,*} Maria Cazacu^{a,*}

^a"Petru Poni" Institute of Macromolecular Chemistry, Aleea Gr. Ghica Voda 41A, 700487 Iasi, Romania

^bLaboratoire National des Champs Magnétiques Intenses, CNRS UPR 3228, 25 Rue des Martyrs, 38042, Grenoble,

France

^c Université Joseph Fourier, F-38041 Grenoble, France ; Institut Universitaire de France (IUF), 103, bd Saint-Michel, F-75005 Paris, France.

^{*}To whom correspondence will be addressed: e-mail: <u>cyrille.train@lncmi.cnrs.fr</u>, <u>mcazacu@icmpp.ro</u>

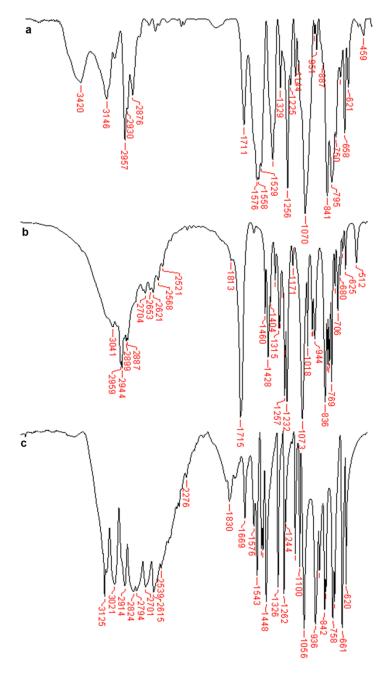


Figure ESI1. IR spectrum of copper complex 1 - a in comparison with that of 1,3bis(carboxypropyl)tetramethyldisiloxane – b and imidazole - c.

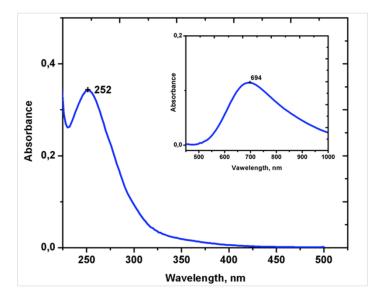
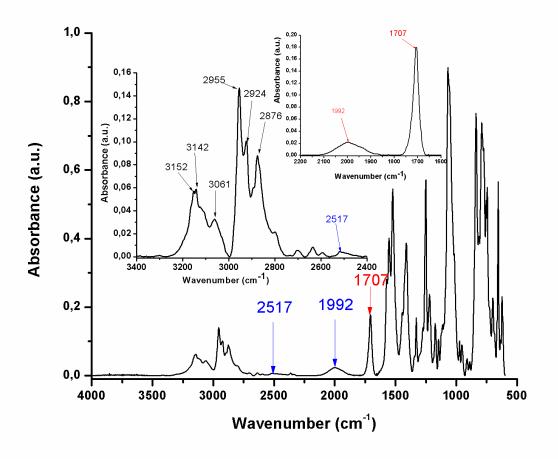


Figure ESI2. UV-Vis spectrum of the complex 1 dissolved in methanol.



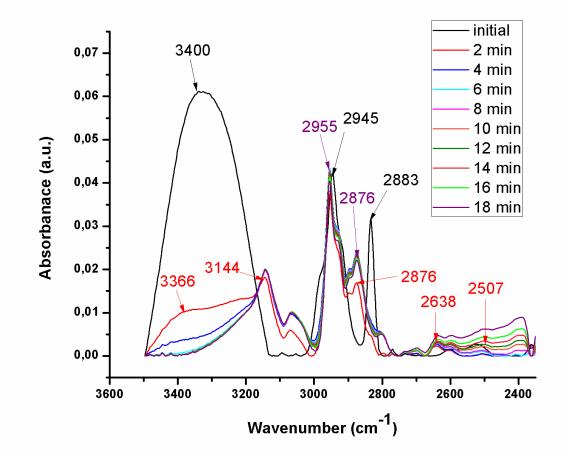


Figure ESI3. ATR-IR spectrum of the complex 1 dissolved in methanol.

Figure ESI4. Details of ATR-IR spectra (2400-3500 cm⁻¹) of the complex **1** dissolved in methanol (initial) and during the evaporation of methanol recorded at two minutes difference between them. Initially only the characteristic bands of methanol at 3400 (vOH), 2945 (vCH₃) and 2883(vCH₃) cm⁻¹ appear. By increase the sample concentration, as a result of the solvent evaporation, the bands characteristic for the complex **1** become visible and also the specific bands of hydrogen bonds at 2600-2500 cm⁻¹ appear.

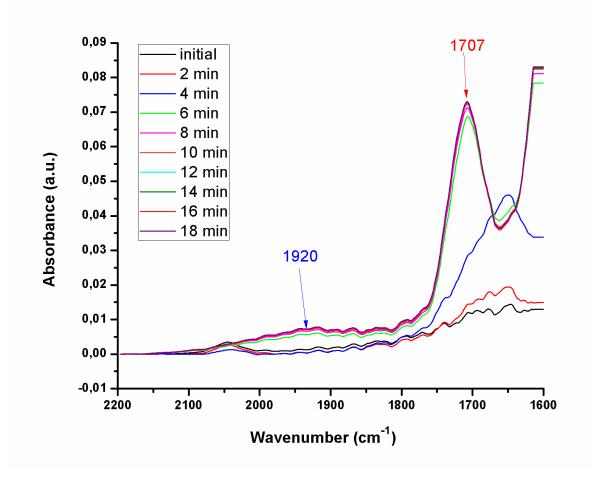


Figure ESI5. Details of ATR-IR spectra (1600-2200 cm⁻¹) of the complex **1** dissolved in methanol (initial) and during the evaporation of methanol recorded at two minutes difference between them. Initially only a broad band attributed to hydroxyl groups appears at 1660 cm⁻¹ (vOH). By increase the sample concentration, by evaporation of the solvent, the carboxylic and hydrogen bonds specific bands at 1707 and 1920 cm⁻¹, respectively develop.

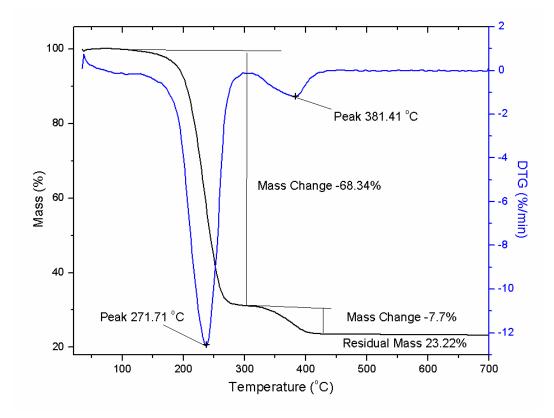


Figure ESI6. TG and DTG curves for complex 1

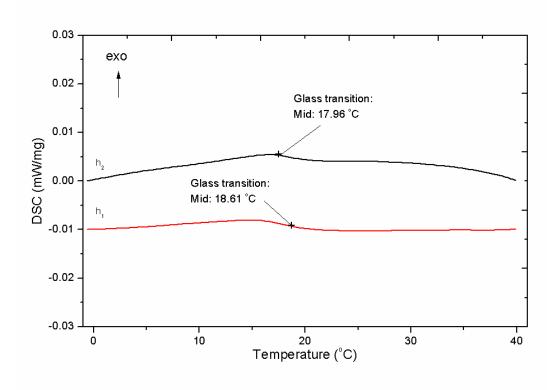


Figure ESI7. DSC curves for the complex (h1-first heating; h2-second heating).

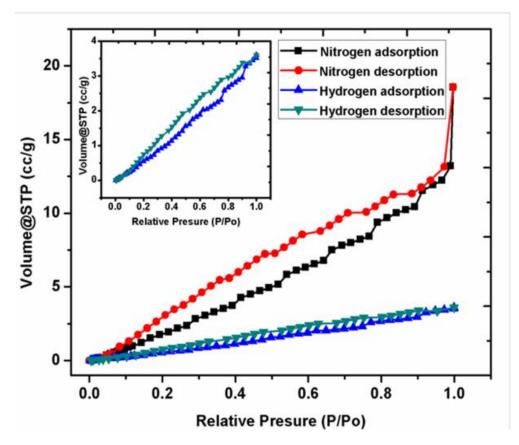


Figure ESI8. H_2 and N_2 sorption-desorption isotherms of 1 at 77 K, P being the relative pressure of the gas and P_0 its saturated vapor pressure.

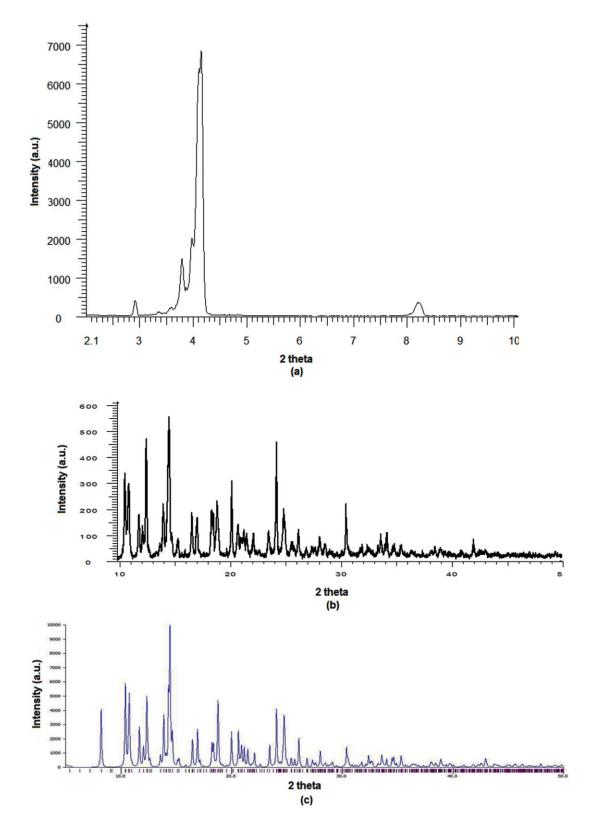


Figure ESI9. The experimental XRPD pattern of the polycrystalline sample in 2-10 2θ range (a); experimental XRPD pattern of the polycrystalline sample in 10-50 2θ range (b); the simulated XRPD on the base of cif.file obtained as the result of single-crystal X-ray study (c).