

Electronic Supplementary Information (ESI)

Structural and magnetic characterizations of the first manganese(III) Schiff base complexes involving hexathiocyanidoplatinate(IV) bridges

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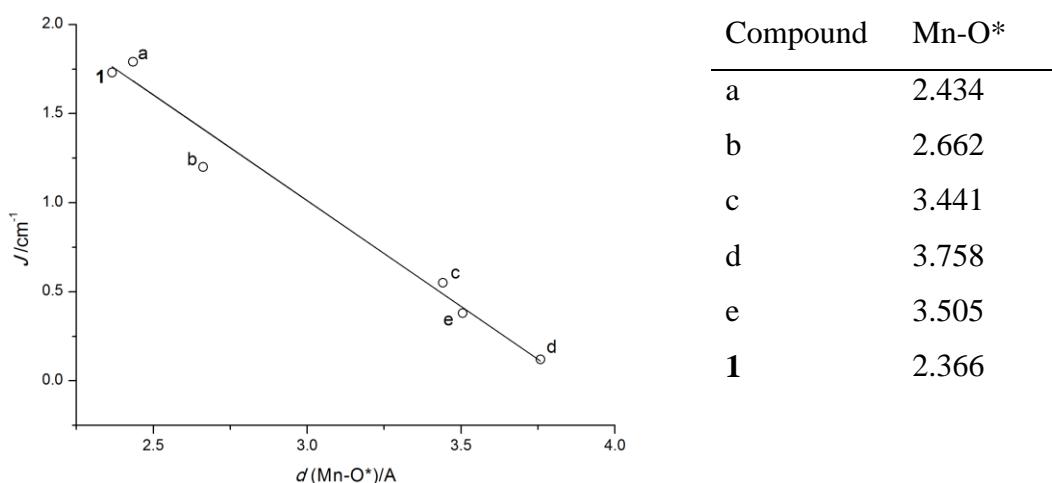


Fig. S1. The correlation between the ferromagnetic exchange (J) and Mn-O* distance. The linear curve was calculated according to ref. [23]. List of compounds: [Mn(saltmen)(H₂O)]ClO₄ (a), [Mn(naphtmen)(H₂O)]ClO₄ (b), [Mn(saltmen)(NCS)] (c), [Mn(naphtmen)(NCS)] (d), [Mn(naphtmen)(Cl)] (e), [{Mn(L4a)}₂{μ-Pt(SCN)₆}]_n (**1**).

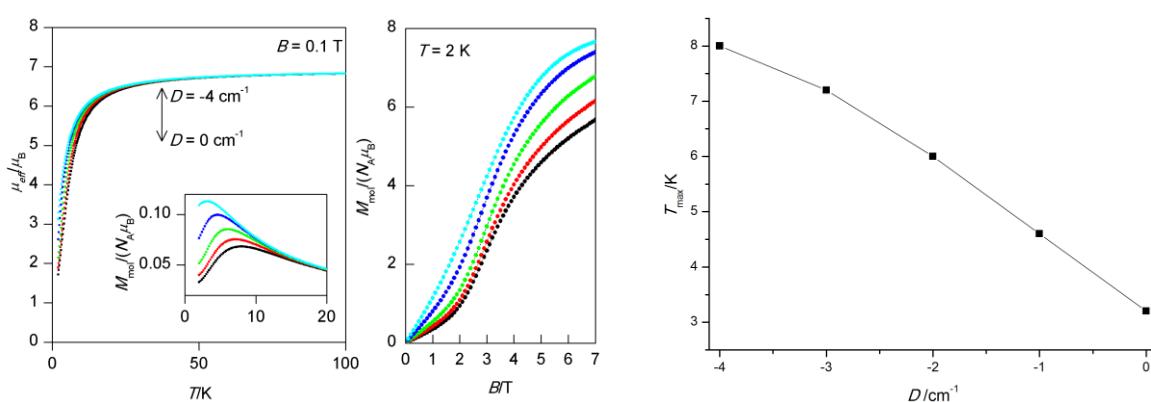


Fig. S2. *Left:* the variation of magnetic properties for $S_1 = S_2 = 2$ dimer, with the fixed parameters $J = -1 \text{ cm}^{-1}$ and $g = 2.0$, while D varies from -4 to 0 cm^{-1} with the step equals 1 cm^{-1} . *Right:* the effect of ZFS on the temperature corresponding to the maximum of magnetization (susceptibility).

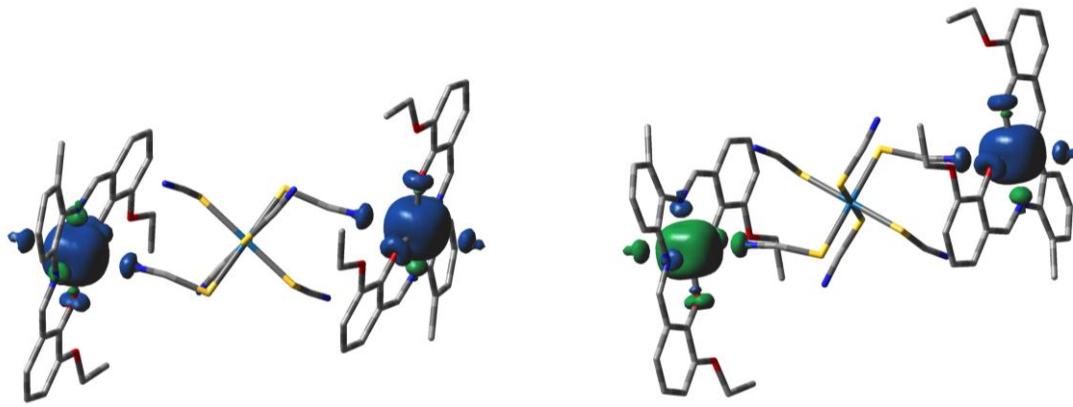


Fig. S3. The calculated spin density distribution using B3LYP/TZVP for $\left[\{\text{Mn}(\text{L4b})(\text{H}_2\text{O})\}_2\{\mu\text{-}\text{Pt}(\text{SCN})_6\}\right]$ (**2**). Positive and negative spin densities are represented by dark blue, and green surfaces, respectively. *Left*: the isodensity surface of high-spin (HS) state with the cutoff values of $0.005 \text{ e}\cdot\text{boh}^{-3}$. *Right*: the isodensity surface of broken symmetry spin (BS) state with the cutoff values of $0.005 \text{ e}\cdot\text{boh}^{-3}$.

Table S1. The DFT calculated net Mulliken spin densities (ρ), the expectation values $\langle S^2 \rangle$ and exchange energy (J) from HS and BS states for the supramolecular dimers $[\{\text{Mn}(\text{L4a})\text{NCS}\}_2]$ and $[\{\text{Mn}(\text{L4b})(\text{H}_2\text{O})(\text{NCS})\}_2]$ extracted from X-ray structures of **1**, and **2**, respectively.

$[\{\text{Mn}(\text{L4a})\text{NCS}\}_2]$ of 1 ^a		$[\{\text{Mn}(\text{L4b})(\text{H}_2\text{O})(\text{NCS})\}_2]$ of 2 ^b		
	HS	BS	HS	BS
$\rho(\text{Mn})$	3.88	-3.88	$\rho(\text{Mn}')$	3.89
$\rho(\text{N}1)$	-0.07	0.07	$\rho(\text{N}1')$	-0.07
$\rho(\text{N}2)$	-0.08	0.09	$\rho(\text{N}2')$	-0.06
$\rho(\text{N}5)$	0.04	-0.04	$\rho(\text{N}3')$	0.04
$\rho(\text{O}1)$	0.02	-0.02	$\rho(\text{O}1')$	0.02
$\rho(\text{O}2)$	0.04	-0.004	$\rho(\text{O}2')$	0.02
$\rho(\text{O}2')$	0.04	0.004	$\rho(\text{O}3')$	0.03
$\rho(\text{O}1')$	0.02	0.02	$\rho(\text{H}1')$	0.002
$\rho(\text{N}5')$	0.04	0.04	$\rho(\text{H}2')$	0.002
$\rho(\text{N}2')$	-0.08	-0.09	$\rho(\text{H}2)$	0.002
$\rho(\text{N}1')$	-0.07	-0.07	$\rho(\text{H}1)$	0.002
$\rho(\text{Mn}')$	3.88	3.88	$\rho(\text{O}3)$	0.03
			$\rho(\text{O}2)$	0.02
			$\rho(\text{O}1)$	0.02
			$\rho(\text{N}3)$	0.04
			$\rho(\text{N}2)$	-0.06
			$\rho(\text{N}1)$	-0.07
			$\rho(\text{Mn})$	3.89
$\langle S^2 \rangle$	20.08	4.08	$\langle S^2 \rangle$	20.09
$J_{\text{Mn}-\text{OPh}-\text{Mn}} = +1.74 \text{ cm}^{-1}$		$J_{\text{Mn}-\text{Owater-H...OPh-Mn}} = -0.45 \text{ cm}^{-1}$		

^a Symmetry code for atoms with the apostrophe: 1-x, 1-y, 1-z. ^b Symmetry code for atoms with the apostrophe: 1-x, -y, 1-z; HS = high-spin state, BS = broken-symmetry spin state.

Table S2. The DFT calculated net Mulliken spin densities (ρ), the expectation values $\langle S^2 \rangle$ and exchange energy (J) from HS and BS states of the trinuclear $[\{\text{Mn}(\text{L4b})(\text{H}_2\text{O})\}_2\{\mu\text{-Pt}(\text{SCN})_6\}]$ (2) based on a geometry following from X-ray structural determination.^a

	HS	BS
$\rho(\text{Mn}')$	3.90	-3.90
$\rho(\text{N1}')$	-0.06	0.06
$\rho(\text{N2}')$	-0.06	0.06
$\rho(\text{O1}')$	0.02	-0.03
$\rho(\text{O2}')$	0.03	-0.03
$\rho(\text{O3}')$	0.02	-0.02
$\rho(\text{N3}')$	0.03	-0.03
$\rho(\text{C25}')$	-0.003	0.004
$\rho(\text{S1A}')$	0.005	-0.006
$\rho(\text{Pt})$	-0.0003	0.0001
$\rho(\text{S1A})$	0.005	0.006
$\rho(\text{C25})$	-0.003	-0.003
$\rho(\text{N3})$	0.03	0.03
$\rho(\text{O3})$	0.02	0.02
$\rho(\text{O2})$	0.03	0.03
$\rho(\text{O1})$	0.02	0.02
$\rho(\text{N2})$	-0.06	-0.06
$\rho(\text{N1})$	-0.06	-0.06
$\rho(\text{Mn})$	3.90	3.90
$\langle S^2 \rangle$	20.07	4.07

$$J = -0.027 \text{ cm}^{-1}$$

^a Symmetry code for atoms with the apostrophe: -x, -y, -z