

## Supporting Information

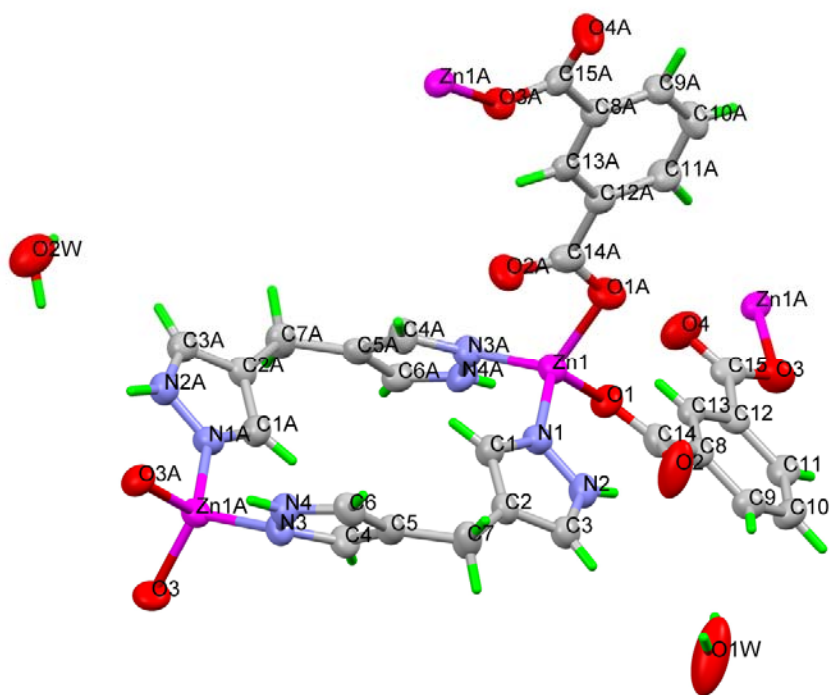
**Title:** Identification of a Robust and Reproducible Noncluster-type SBU: Effect of Coexistent Groups on Network Topologies, Helicity and Properties

**Authors:** Satirtha Sengupta, Sumi Ganguly, Arijit Goswami, Pradip K. Sukul, Raju Mondal\*

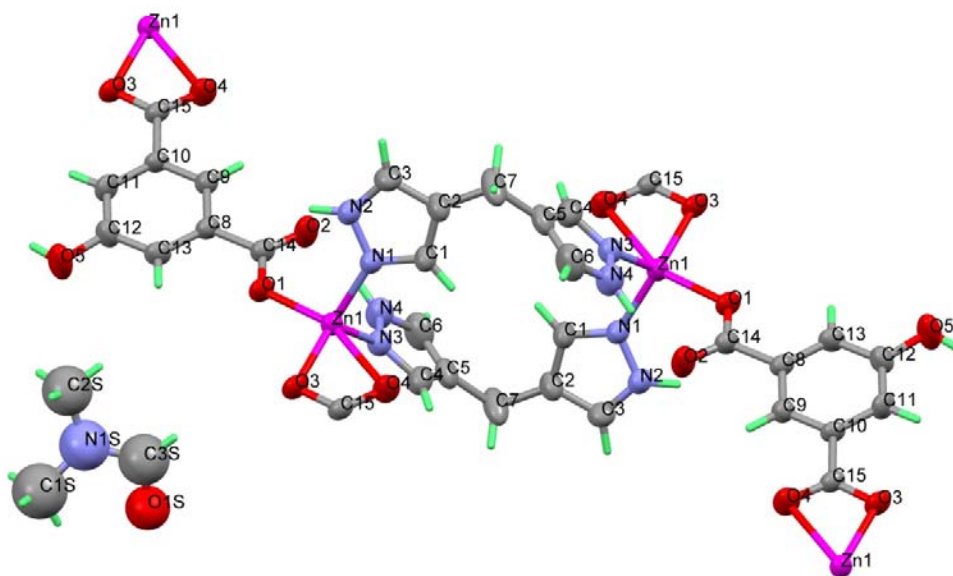
**Journal:** CrystEngComm

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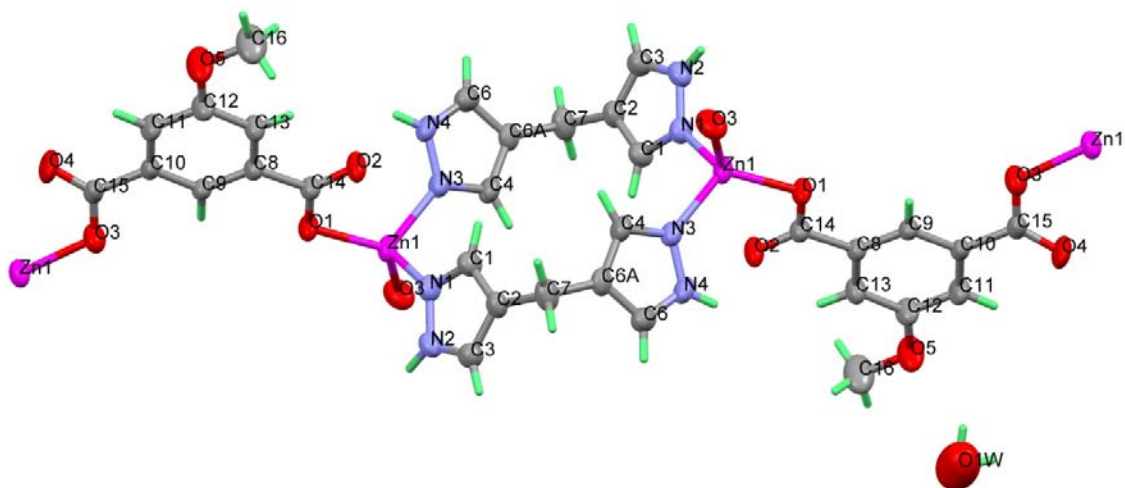
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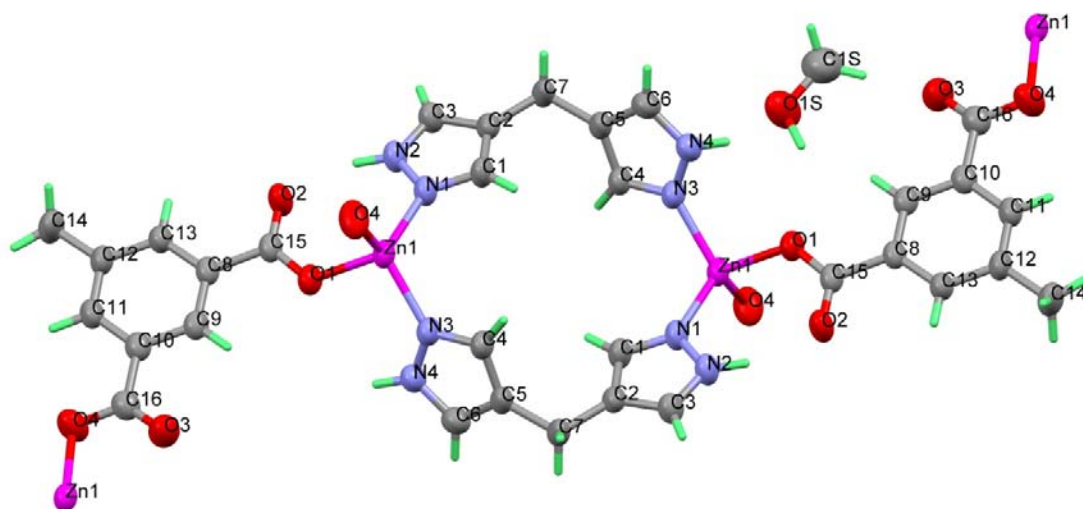
**Figure SI-1.** The coordination environment around the Zn(II) ion of compound **1**.



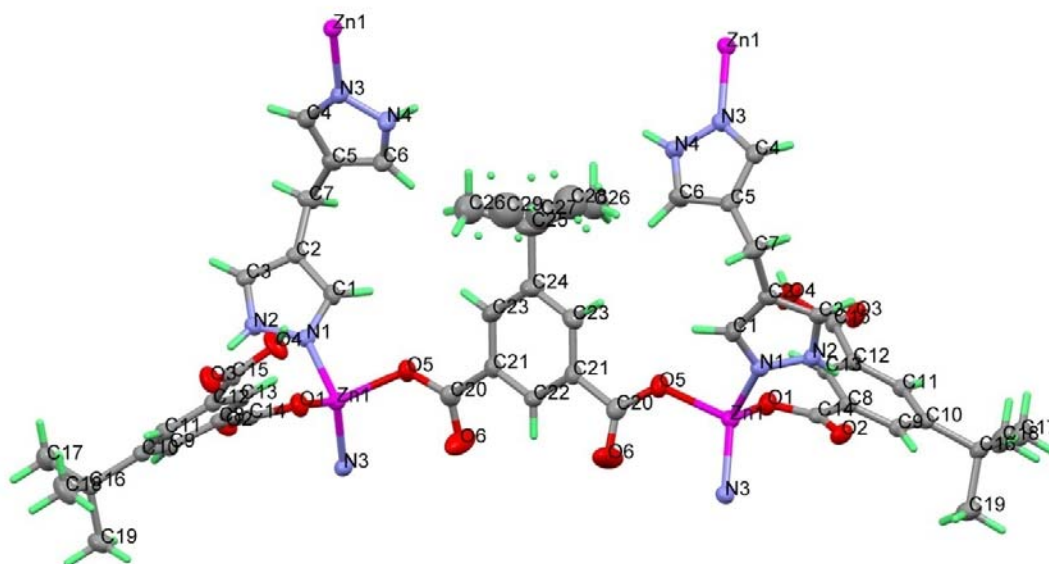
**Figure SI-2.** The coordination environment around the Zn(II) ion of compound **2**.



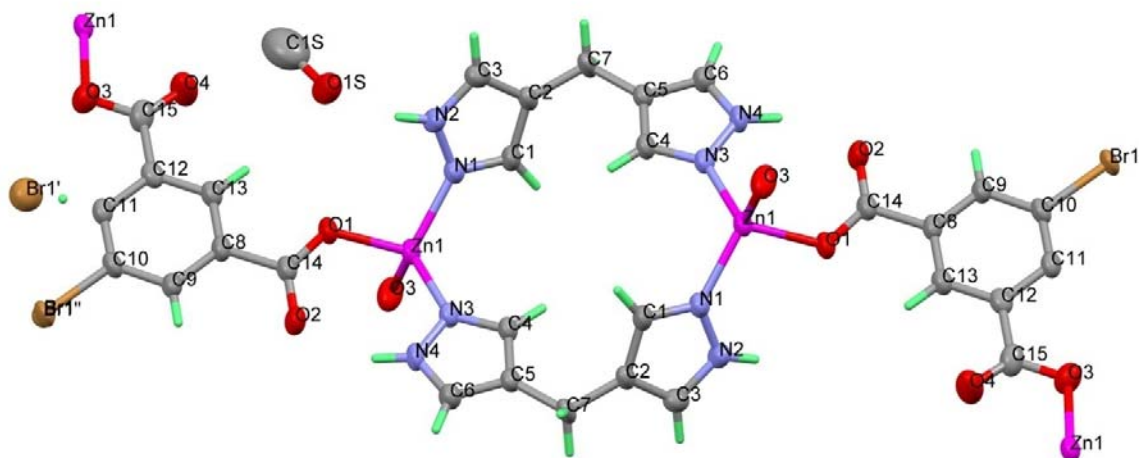
**Figure SI-3.** The coordination environment around the Zn(II) ion of compound **3**.



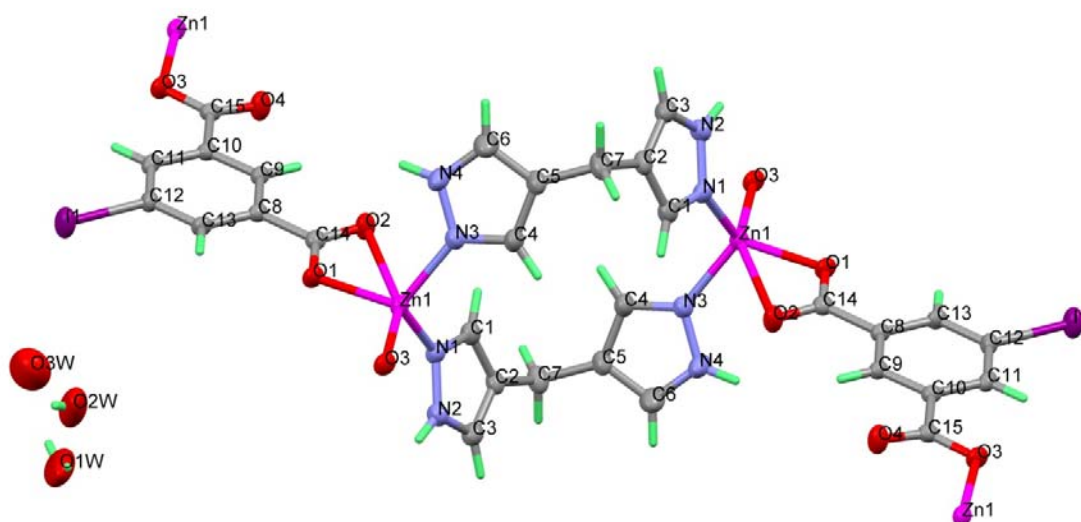
**Figure SI-4.** The coordination environment around the Zn(II) ion of compound **4**.



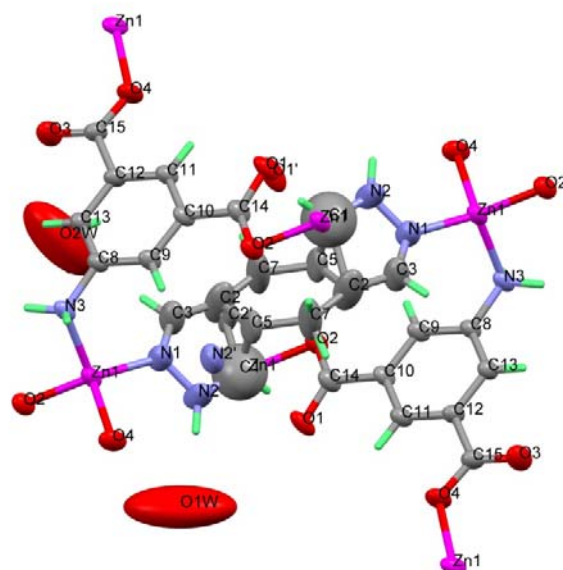
**Figure SI-5.** The coordination environment around the Zn(II) ion of compound **5**.



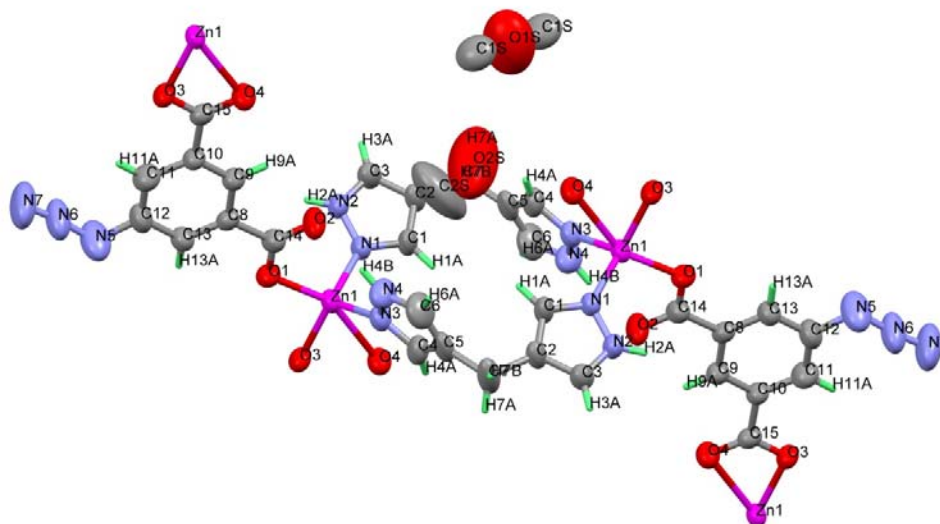
**Figure SI-6.** The coordination environment around the Zn(II) ion of compound **6**.



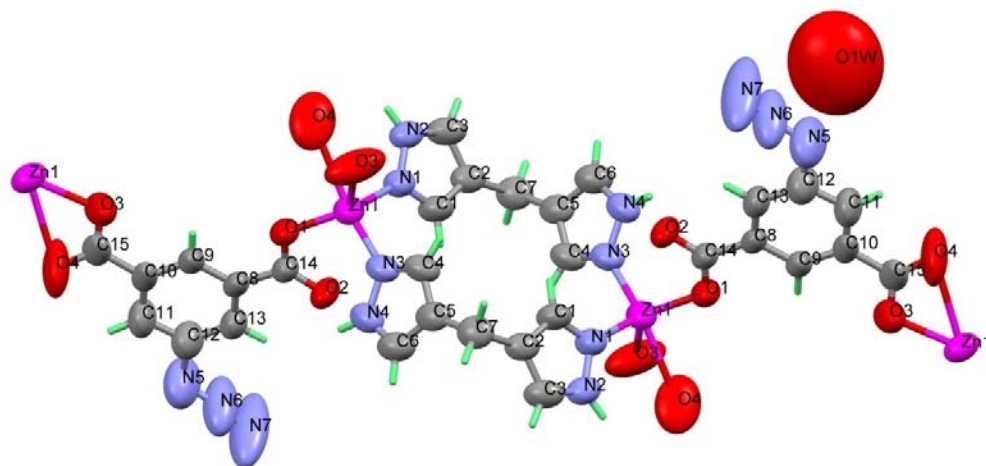
**Figure SI-7.** The coordination environment around the Zn(II) ion of compound **7**.



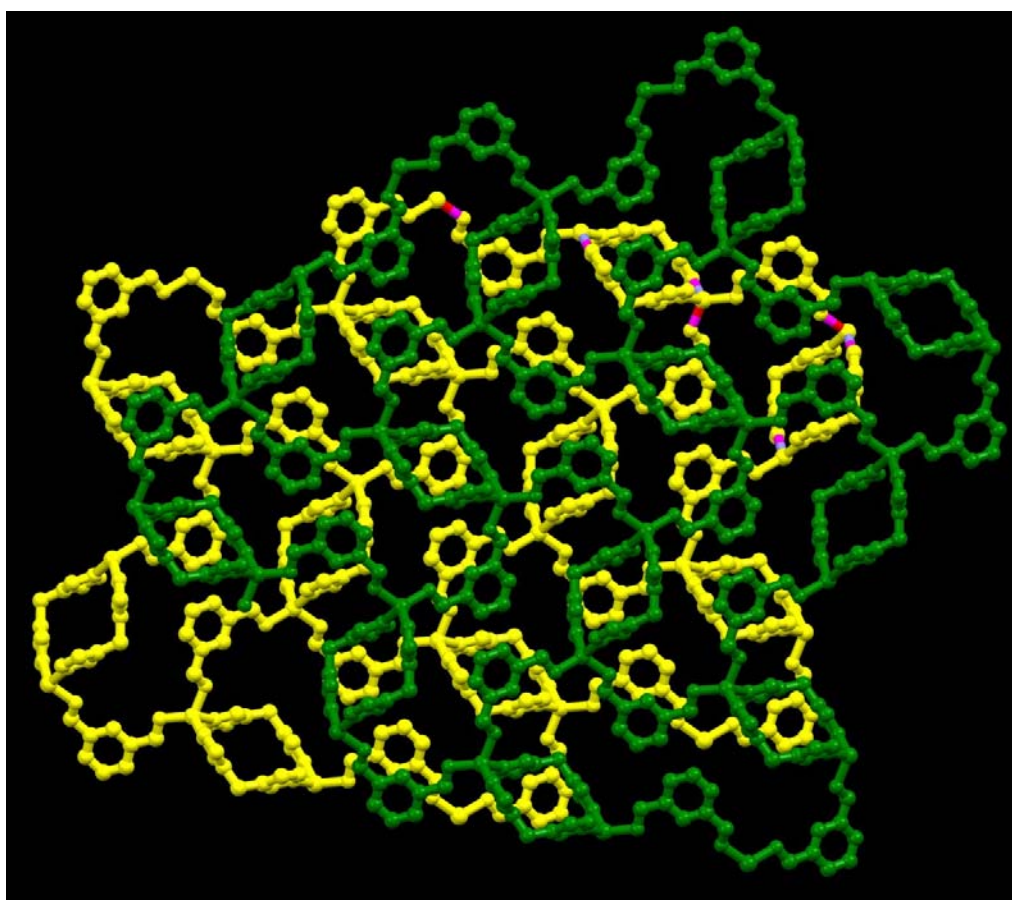
**Figure SI-8.** The coordination environment around the Zn(II) ion of compound **8**.



**Figure SI-9.** The coordination environment around the Zn(II) ion of compound **9**.

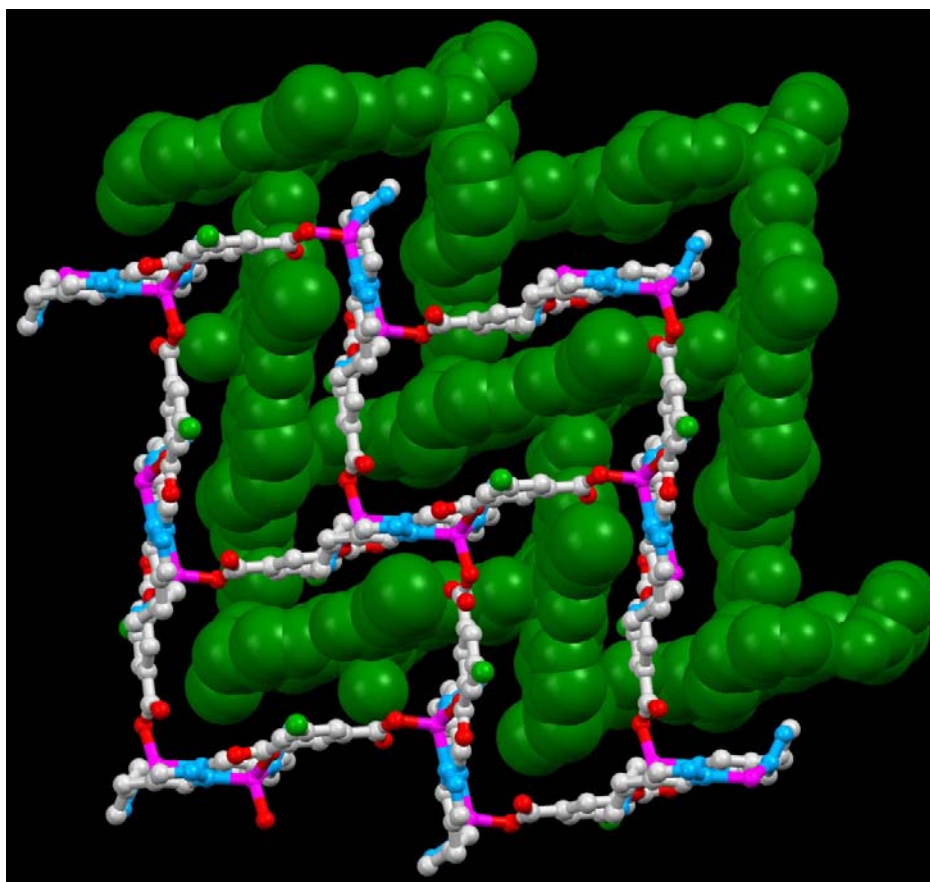


**Figure SI-10.** The coordination environment around the Zn(II) ion of compound **10**.

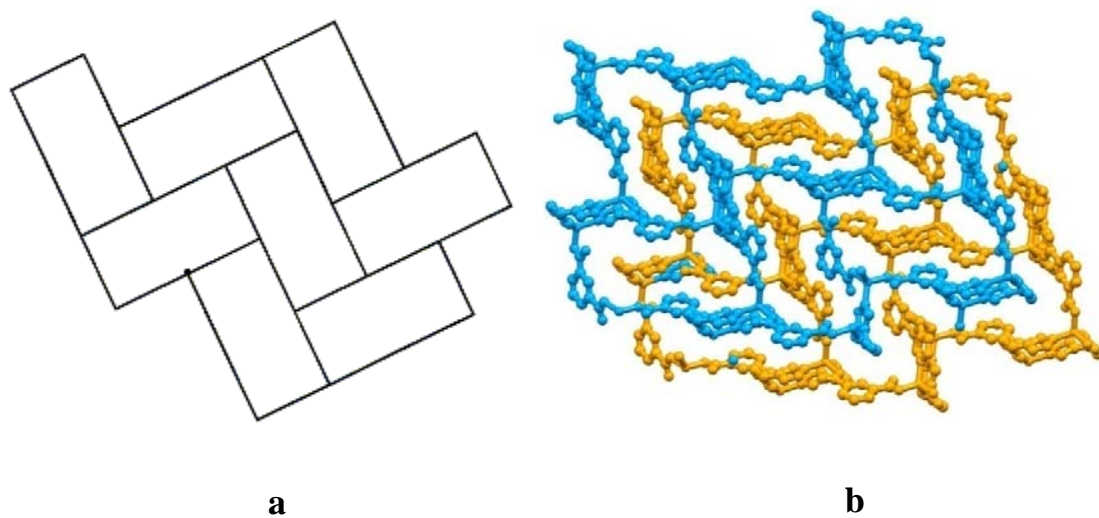


**Figure SI-11.** Crystal structure of **3** showing off-set stacking 2D grid networks.

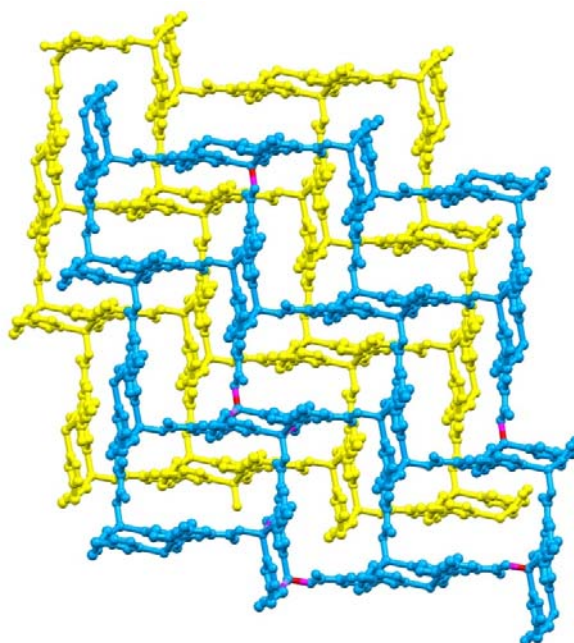




**Figure SI-12.** Crystal structure of **6** showing off-set stacking 2D grid networks.

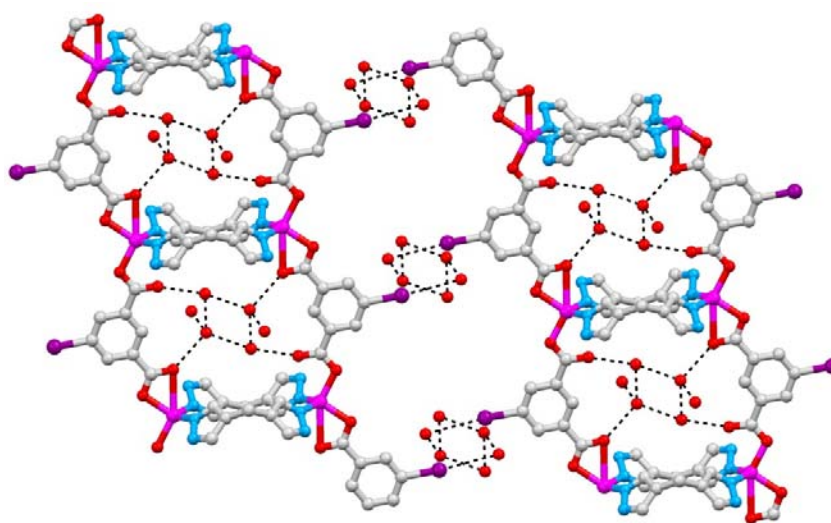


**Figure SI-13.** (a) Schematic diagram of 2D grid network and (b) off-set stacking of 2D grid networks of **1**.

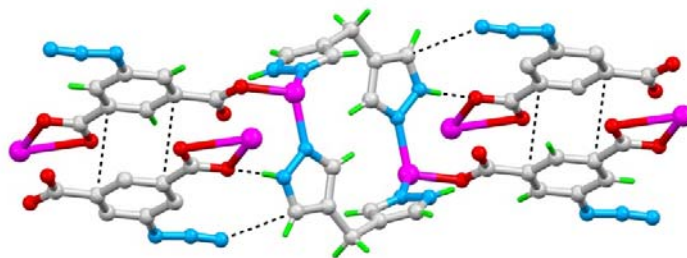


**Figure SI-14.** Off-set stackings of 2D networks of **4**.

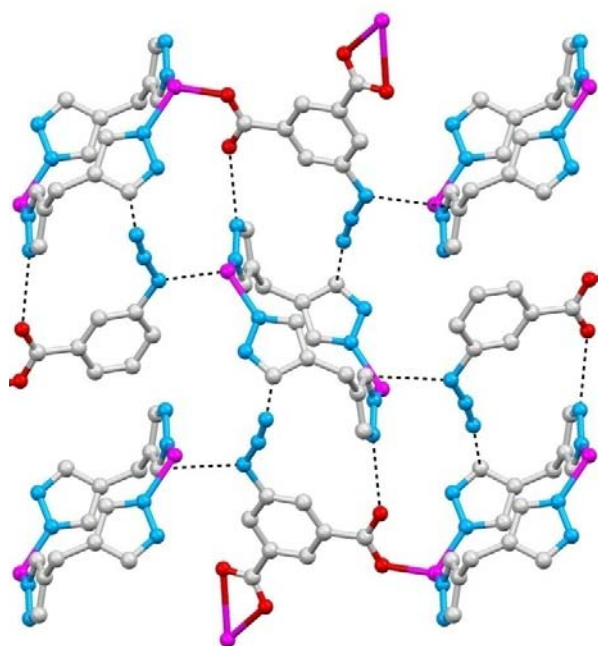




**Figure SI-15.** Crystal structure of **7** illustrating the role of water cluster in linking the adjacent 1D networks.



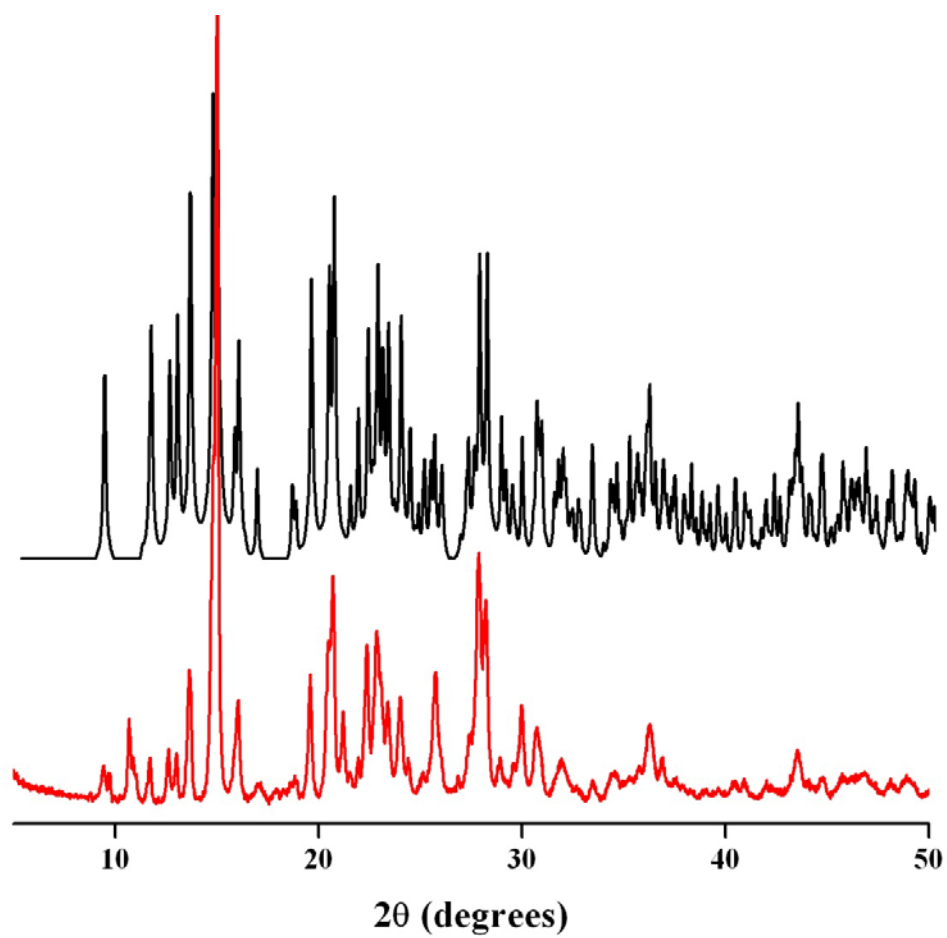
**a**



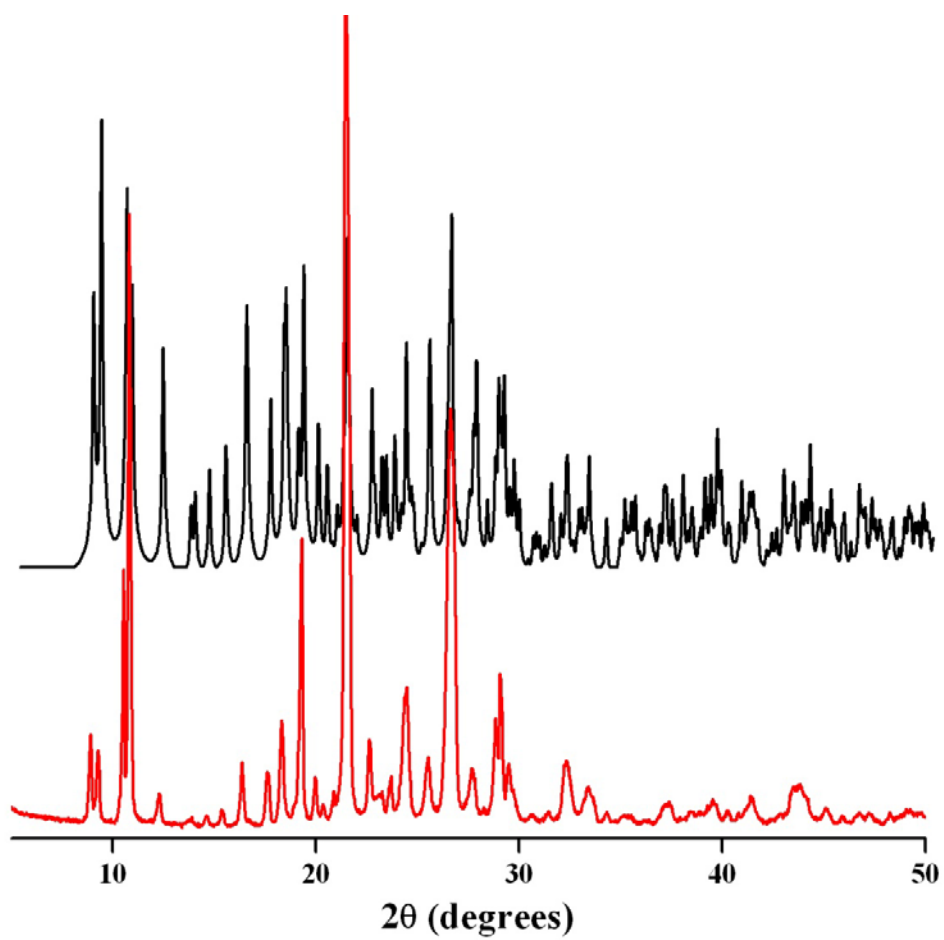
**b**

**Figure SI-16.** a) Crystal structure of **9** showing  $\pi$ - $\pi$  stacked self-assembly of SBU-I based 1D networks. (b) Crystal structure of **10** showing novel azide- $\pi$  interactions.

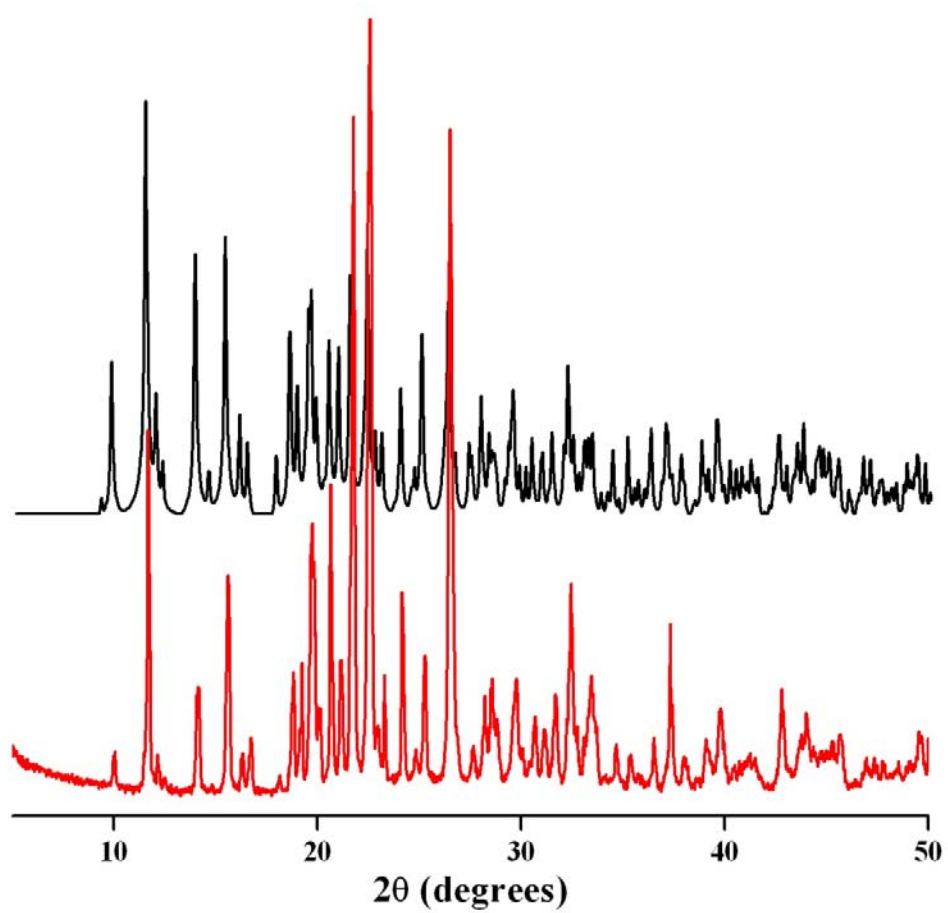
### Powder X-ray diffraction pattern



**Figure SI-17.** Powder XRD pattern of compound 1, simulated (black), experimental (red).

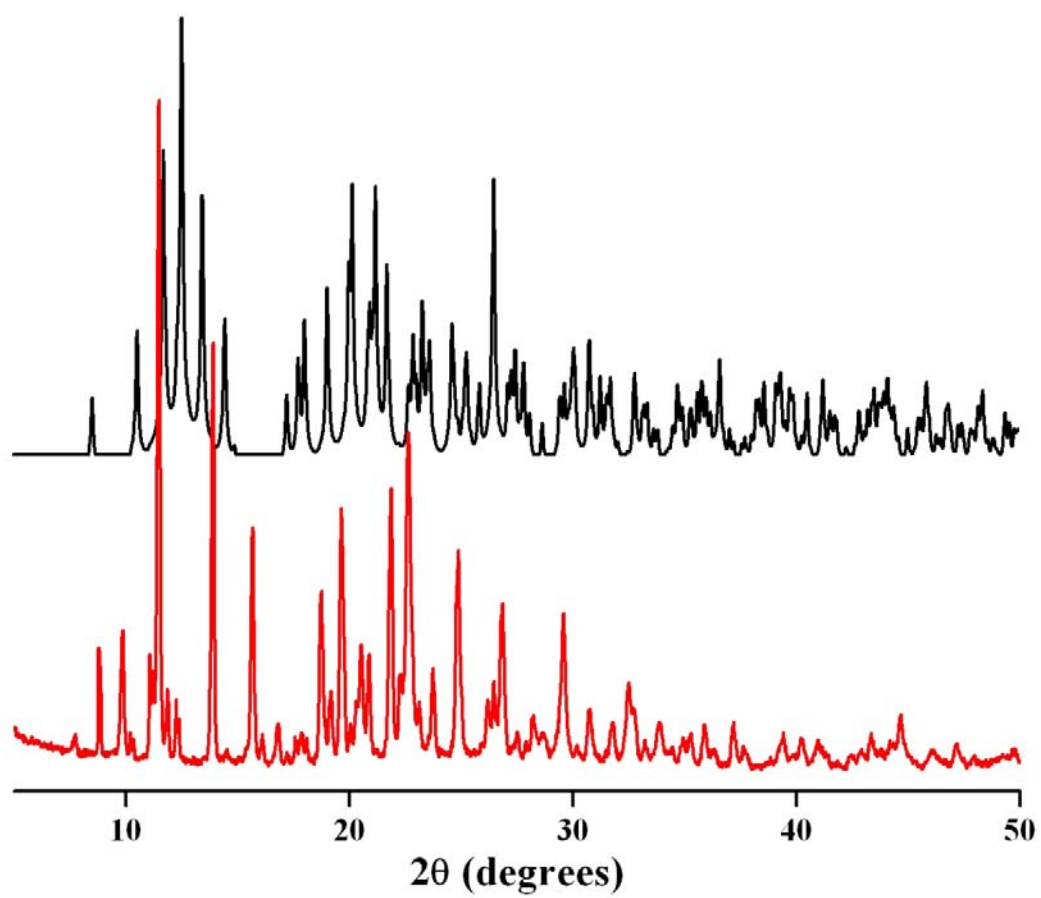


**Figure SI-18.** Powder XRD pattern of compound 2, simulated (black), experimental (red).

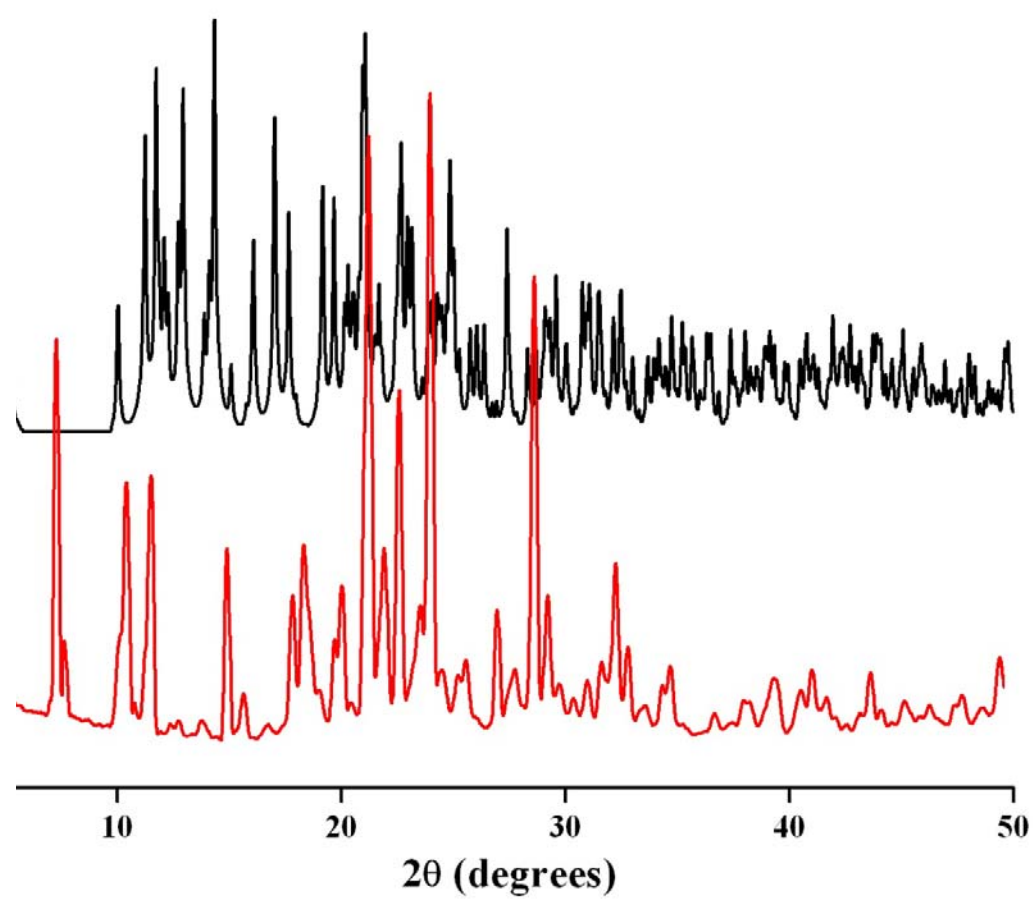


**Figure SI-19.** Powder XRD pattern of compound **3**, simulated (black), experimental (red).

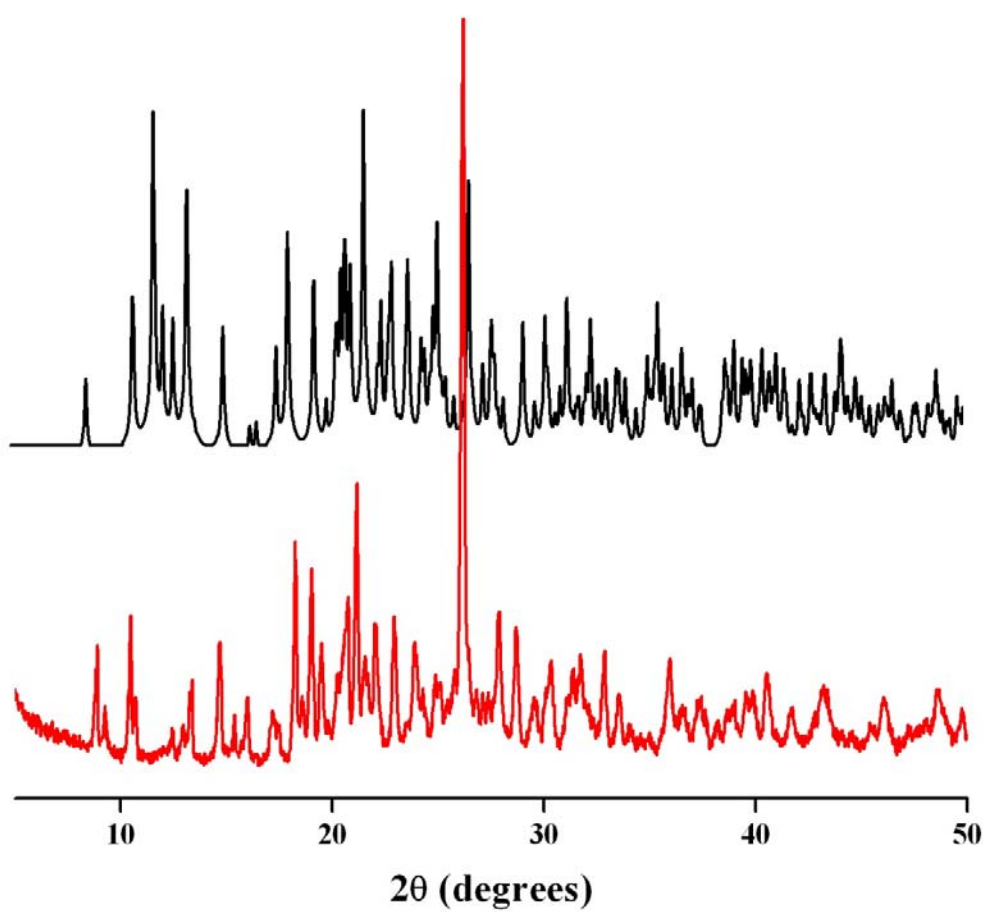




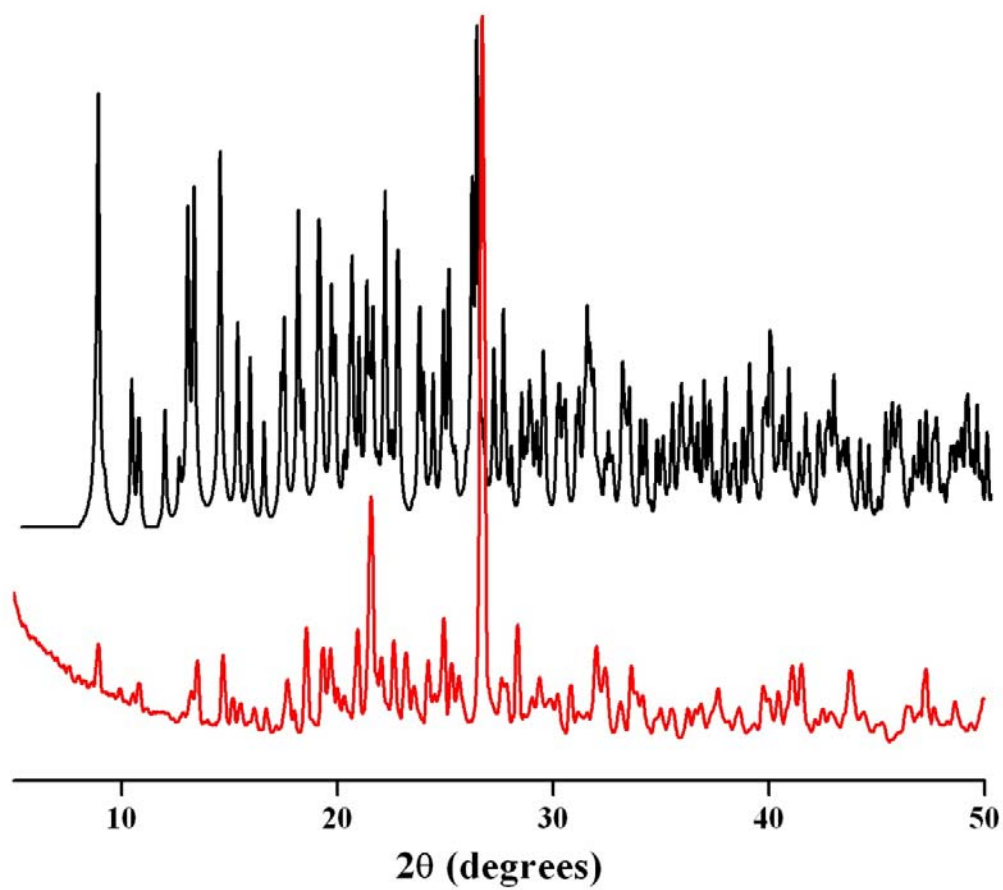
**Figure SI-20.** Powder XRD pattern of compound **4**, simulated (black), experimental (red).



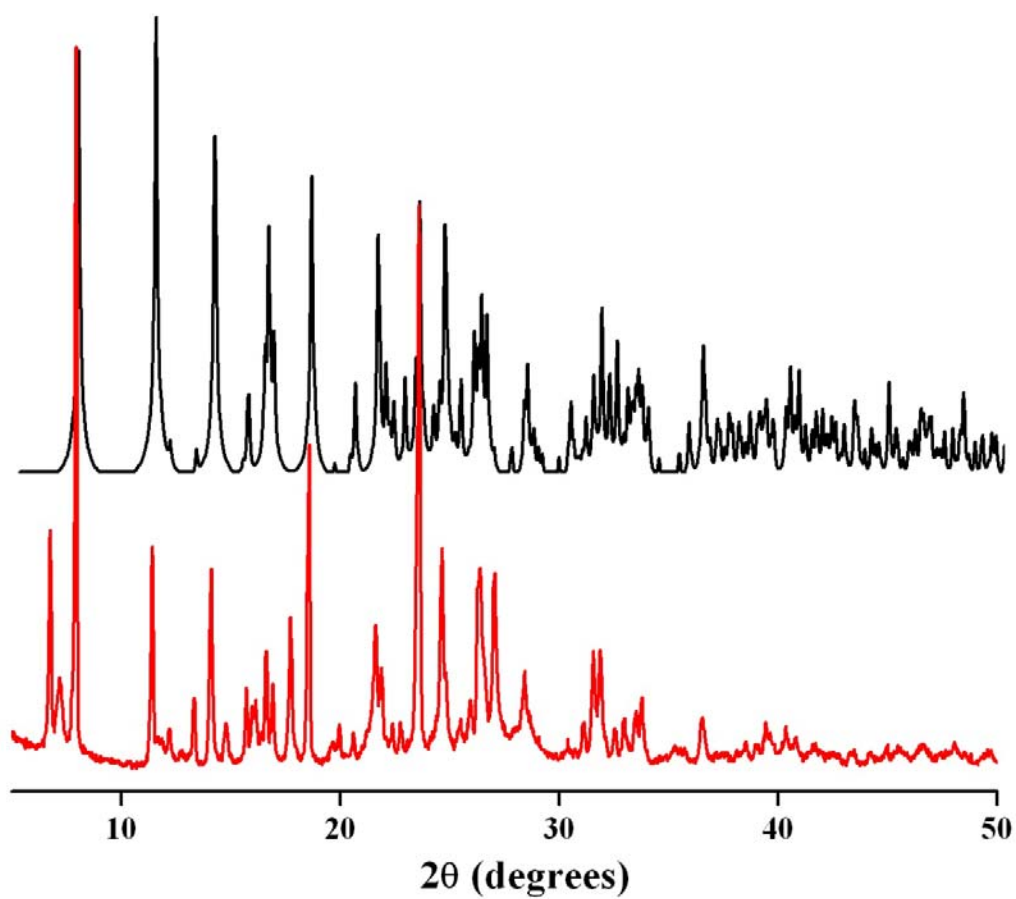
**Figure SI-21.** Powder XRD pattern of compound **5**, simulated (black), experimental (red).



**Figure SI-22.** Powder XRD pattern of compound **6**, simulated (black), experimental (red).

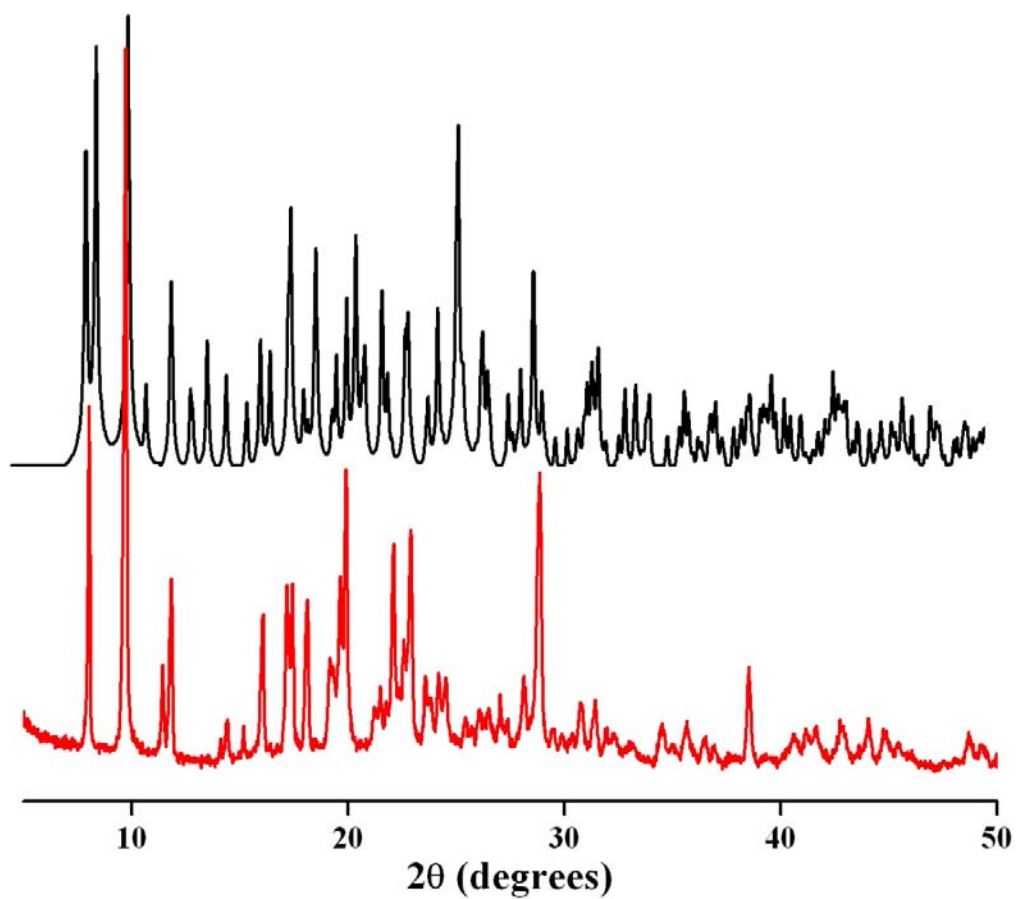


**Figure SI-23.** Powder XRD pattern of compound 7, simulated (black), experimental (red).

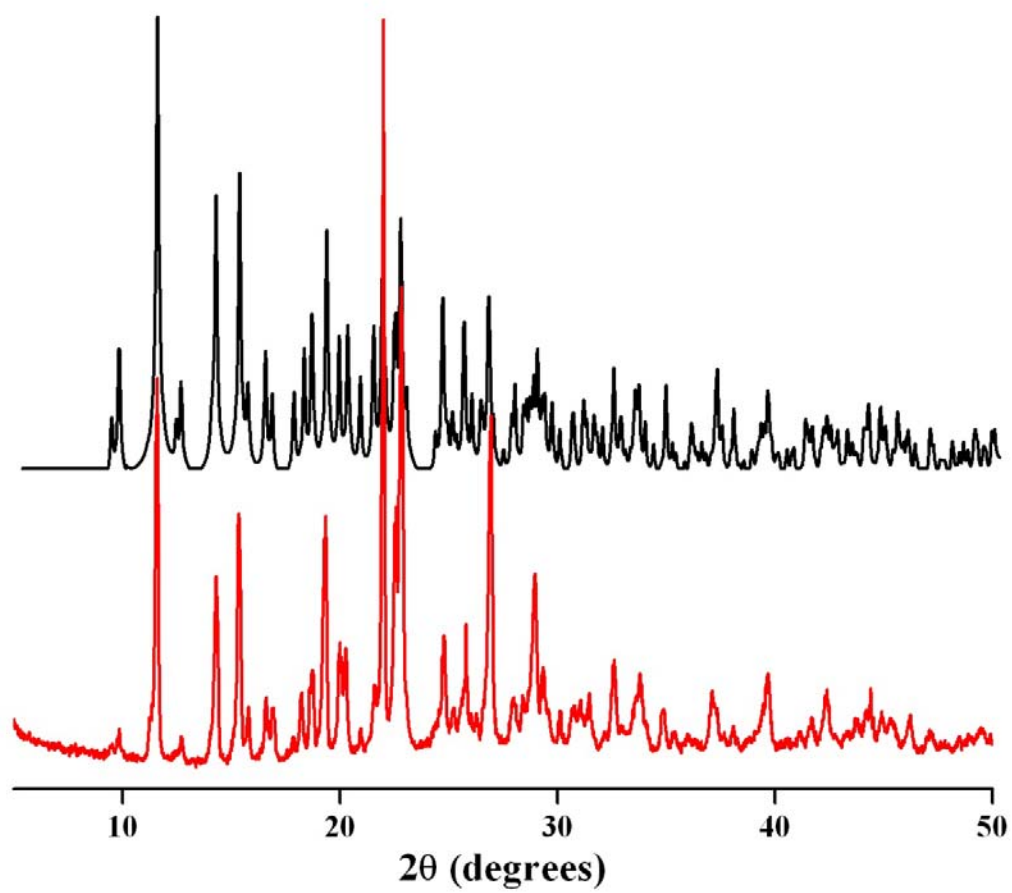


**Figure SI-24.** Powder XRD pattern of compound **8**, simulated (black), experimental (red).

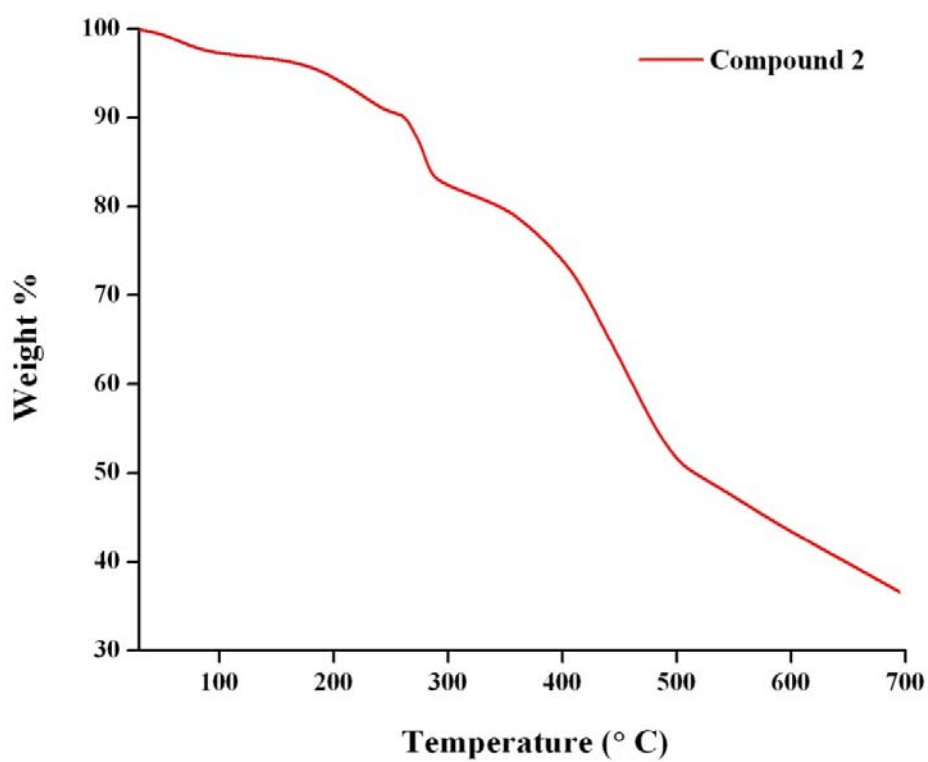
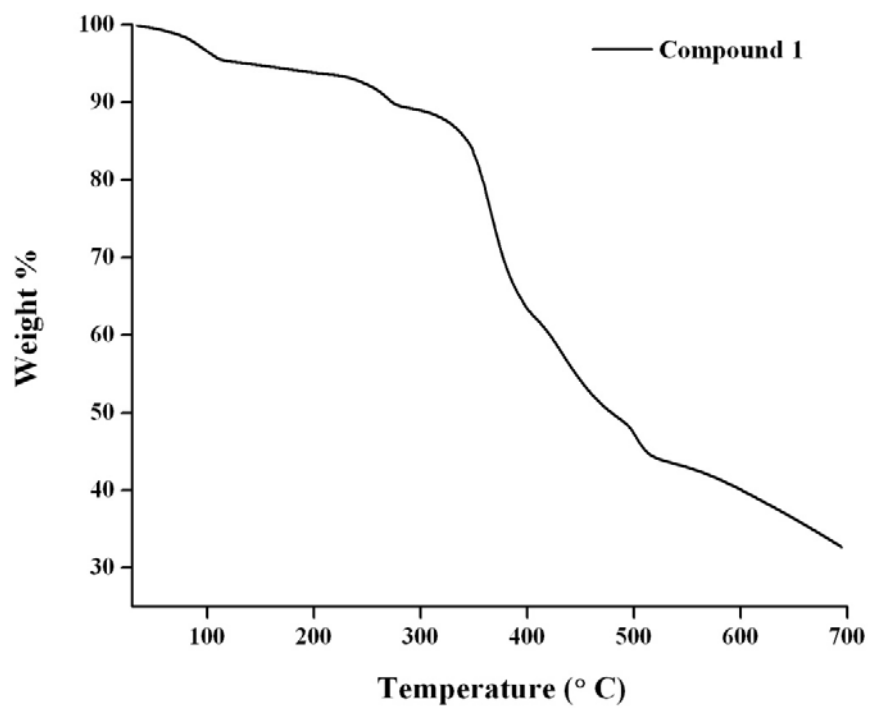


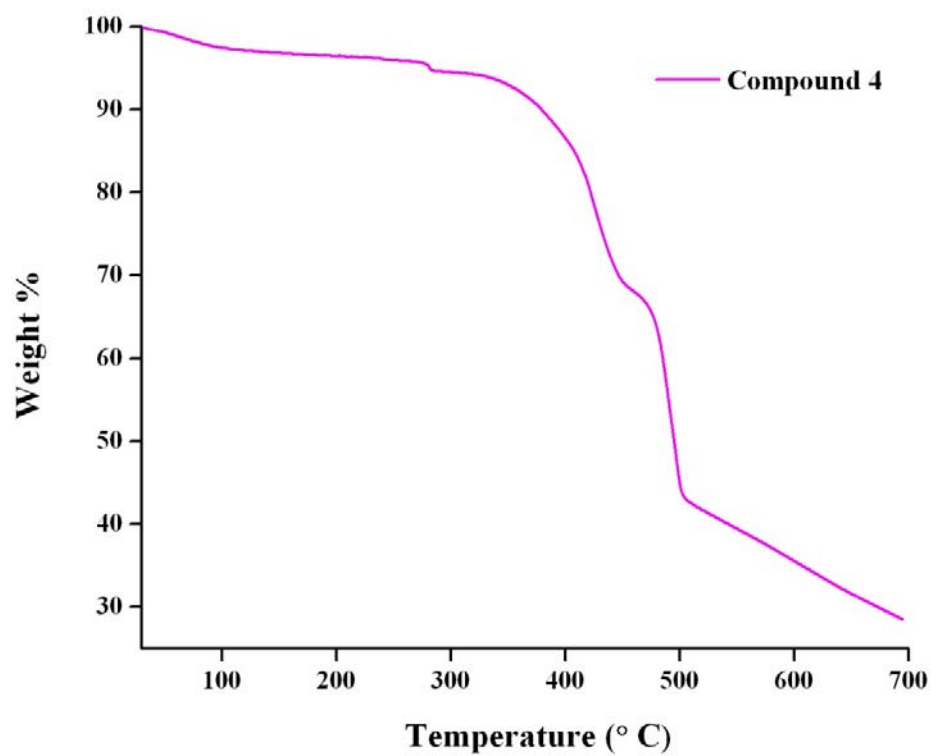
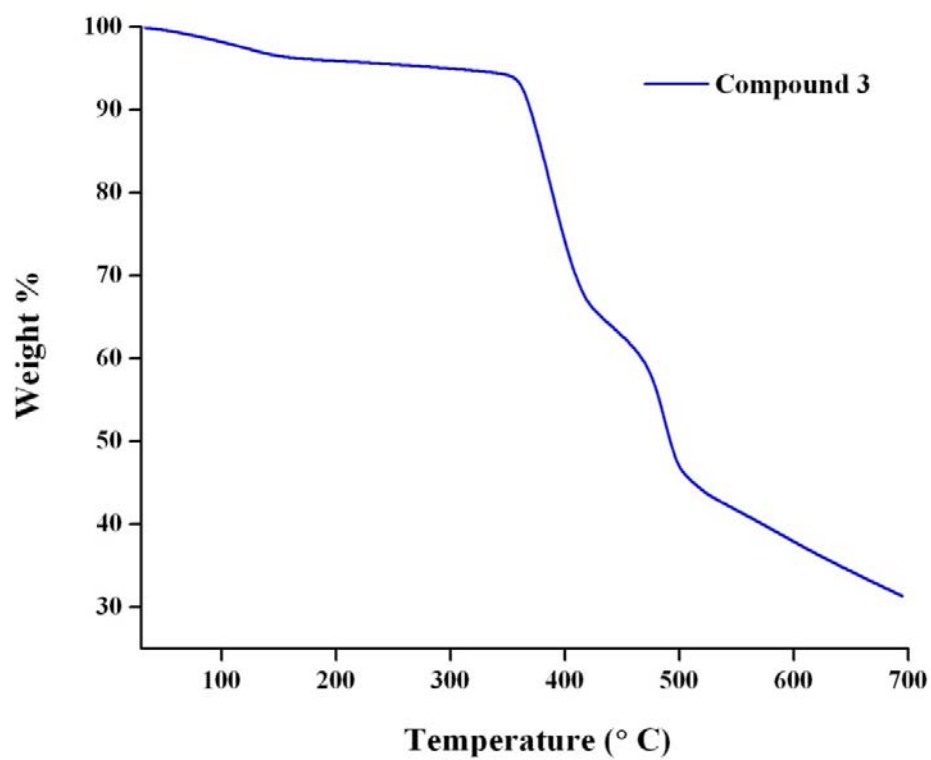


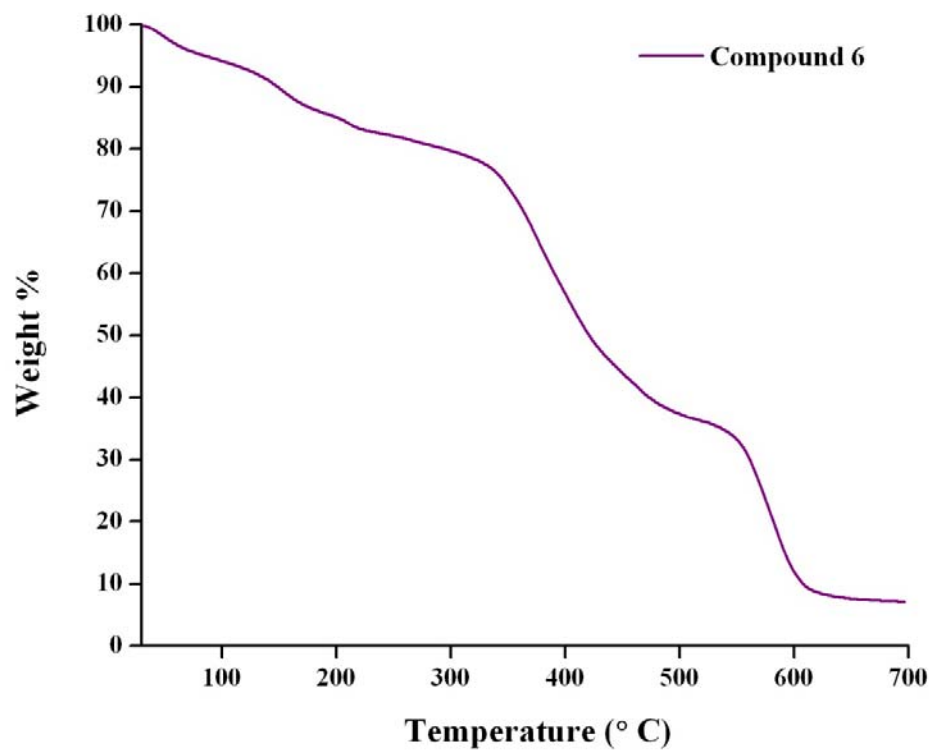
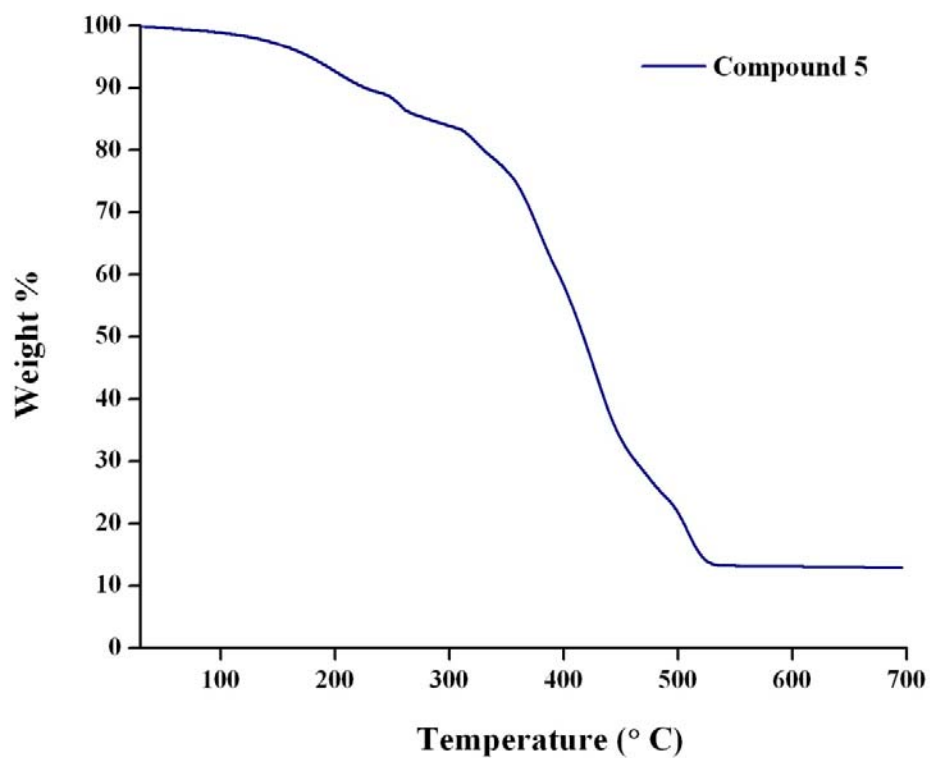
**Figure SI-25.** Powder XRD pattern of compound **9**, simulated (black), experimental (red).



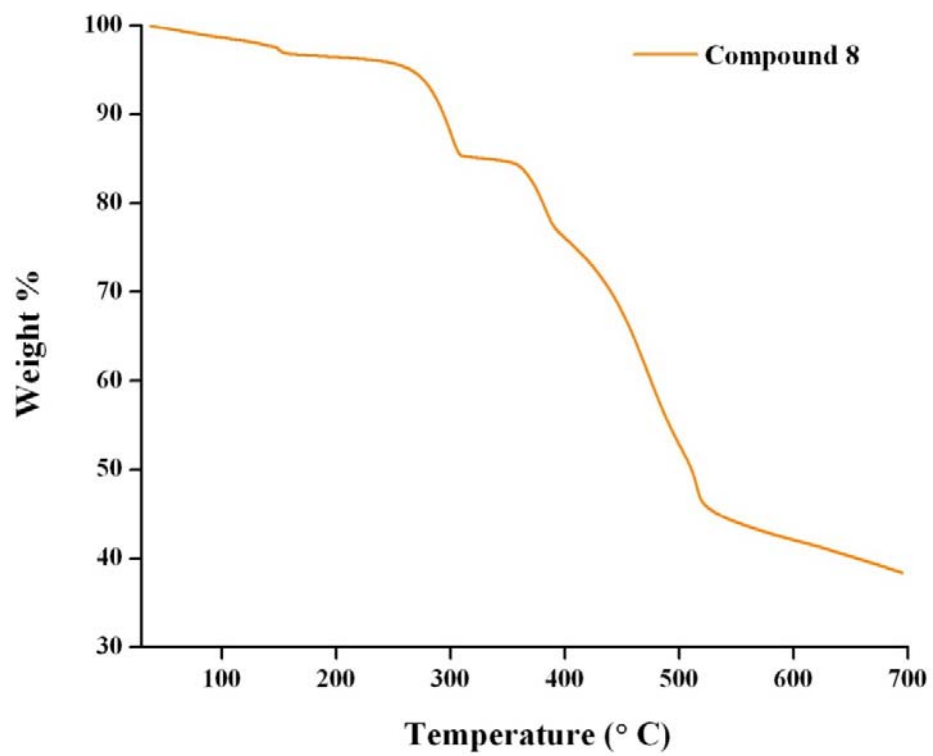
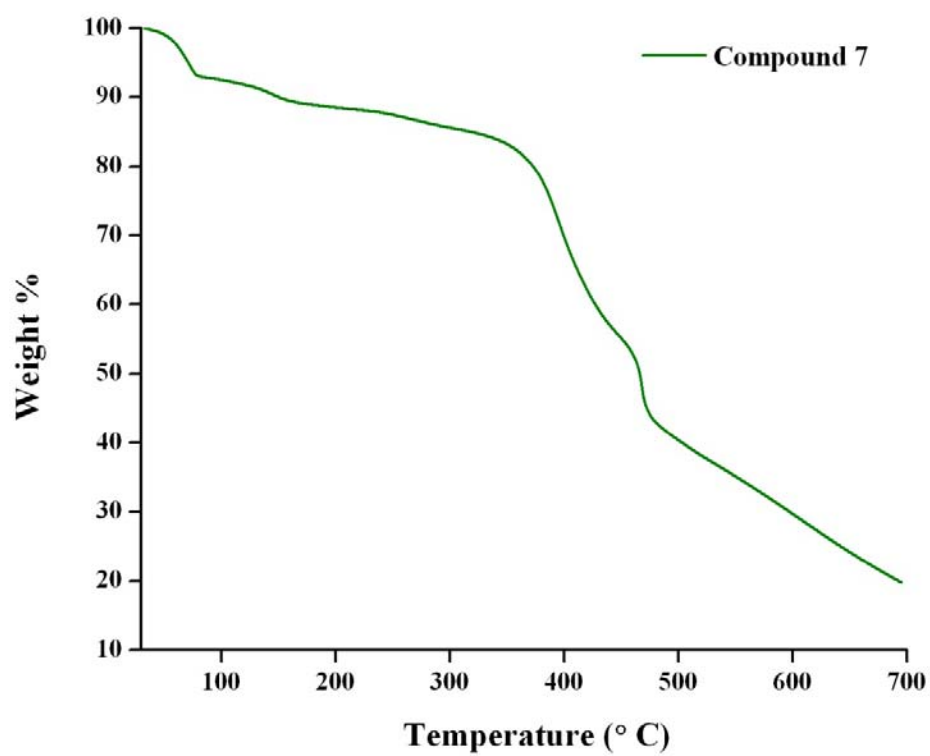
**Figure SI-26.** Powder XRD pattern of compound **10**, simulated (black), experimental (red).











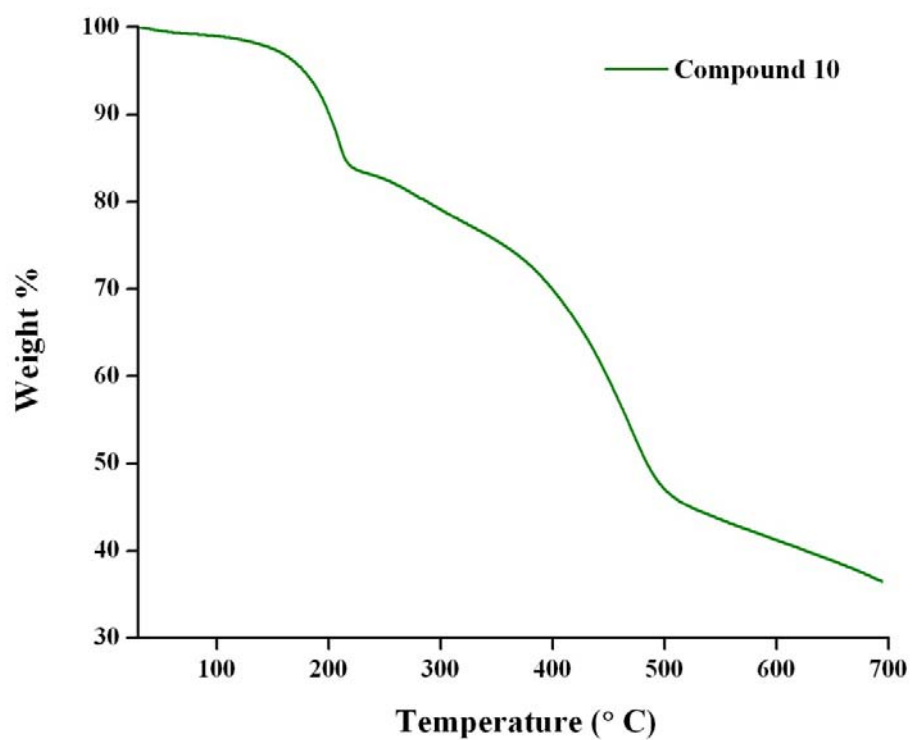
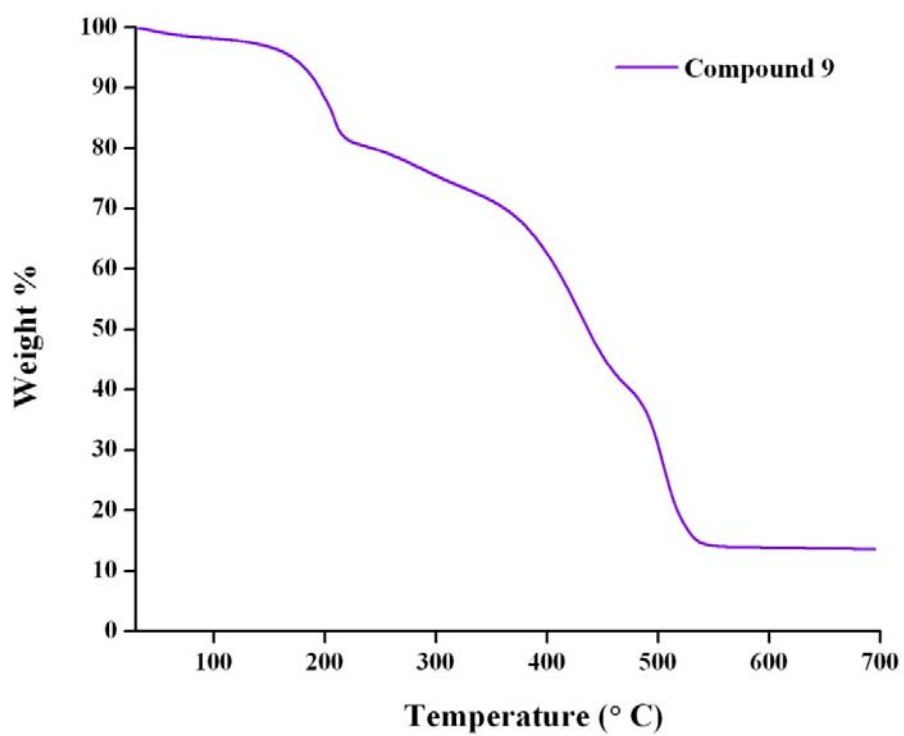


Figure SI-27. TGA plots of compound 1-10 (from top to bottom).

**Table SI-1:** Selected Bond Lengths (Å) and Angles (°) for Compounds **1-6**.

<b>Compound 1</b>				<b>Compound 2</b>			
Zn <sub>1</sub> -O <sub>1A</sub>	1.959(3)	O <sub>1A</sub> -Zn <sub>1</sub> -O <sub>1</sub>	104.17(13)	Zn <sub>1</sub> -O <sub>1</sub>	1.976(2)	N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	107.05(13)
Zn <sub>1</sub> -O <sub>1</sub>	1.965(2)	O <sub>1A</sub> -Zn <sub>1</sub> -N <sub>3A</sub>	121.50(15)	Zn <sub>1</sub> -N <sub>1</sub>	2.022(3)	O <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	96.17(11)
Zn <sub>1</sub> -N <sub>3A</sub>	2.005(3)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3A</sub>	94.31(11)	Zn <sub>1</sub> -N <sub>3</sub>	2.037(3)	N <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	118.99(13)
Zn <sub>1</sub> -N <sub>1</sub>	2.010(4)	O <sub>1A</sub> -Zn <sub>1</sub> -N <sub>1</sub>	109.50(14)	Zn <sub>1</sub> -O <sub>3</sub>	2.062(3)	N <sub>3</sub> -Zn <sub>1</sub> -O <sub>3</sub>	119.61(12)
		O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	113.82(14)	Zn <sub>1</sub> -O <sub>4</sub>	2.326(3)	O <sub>1</sub> -Zn <sub>1</sub> -O <sub>4</sub>	155.38(11)
		N <sub>3A</sub> -Zn <sub>1</sub> -N <sub>1</sub>	112.39(14)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	104.50(12)	N <sub>1</sub> -Zn <sub>1</sub> -O <sub>4</sub>	89.75(12)
				O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	108.36(13)	N <sub>3</sub> -Zn <sub>1</sub> -O <sub>4</sub>	85.78(12)
						O <sub>3</sub> -Zn <sub>1</sub> -O <sub>4</sub>	59.22(10)
<b>Compound 3</b>				<b>Compound 4</b>			
Zn <sub>1</sub> -O <sub>1</sub>	1.9603(9)	O <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	117.98(5)	Zn <sub>1</sub> -O <sub>4</sub>	1.9380(16)	O <sub>4</sub> -Zn <sub>1</sub> -O <sub>1</sub>	101.19(7)
Zn <sub>1</sub> -O <sub>3</sub>	1.9637(11)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	112.53(5)	Zn <sub>1</sub> -O <sub>1</sub>	1.9475(15)	O <sub>4</sub> -Zn <sub>1</sub> -N <sub>3</sub>	121.39(8)
Zn <sub>1</sub> -N <sub>1</sub>	2.0143(14)	O <sub>3</sub> -Zn <sub>1</sub> -N <sub>1</sub>	107.68(5)	Zn <sub>1</sub> -N <sub>3</sub>	2.0037(18)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	99.29(7)
Zn <sub>1</sub> -N <sub>3</sub>	2.0269(13)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	117.02(5)	Zn <sub>1</sub> -N <sub>1</sub>	2.0130(17)	O <sub>4</sub> -Zn <sub>1</sub> -N <sub>1</sub>	109.71(8)
		O <sub>3</sub> -Zn <sub>1</sub> -N <sub>3</sub>	91.93(5)			O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	111.54(7)
		N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	107.59(5)			N <sub>3</sub> -Zn <sub>1</sub> -N <sub>1</sub>	112.36(7)
<b>Compound 6</b>				<b>Compound 5</b>			
Zn <sub>1</sub> -O <sub>3</sub>	1.951(3)	O <sub>3</sub> -Zn <sub>1</sub> -O <sub>1</sub>	99.07(13)	Zn <sub>1</sub> -O <sub>5</sub>	1.955(2)	O <sub>5</sub> -Zn <sub>1</sub> -O <sub>1</sub>	103.34(10)
Zn <sub>1</sub> -O <sub>1</sub>	1.954(3)	O <sub>3</sub> -Zn <sub>1</sub> -N <sub>1</sub>	121.82(16)	Zn <sub>1</sub> -O <sub>1</sub>	1.958(2)	O <sub>5</sub> -Zn <sub>1</sub> -N <sub>1</sub>	98.59(10)
Zn <sub>1</sub> -N <sub>1</sub>	1.992(4)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	103.09(15)	Zn <sub>1</sub> -N <sub>1</sub>	2.008(3)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	107.51(10)
Zn <sub>1</sub> -N <sub>3</sub>	2.003(3)	O <sub>3</sub> -Zn <sub>1</sub> -N <sub>3</sub>	106.84(15)	Zn <sub>1</sub> -N <sub>3</sub>	2.008(3)	O <sub>5</sub> -Zn <sub>1</sub> -N <sub>3</sub>	112.07(11)
		O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	111.47(13)			O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	120.25(10)
		N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	113.39(14)			N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	112.58(11)

<b>Table SI-2:</b> Selected Bond Lengths (Å) and Angles (°) for Compounds <b>7-10</b> .							
<b>Compound 7</b>				<b>Compound 8</b>			
Zn <sub>1</sub> -O <sub>3</sub>	1.976(3)	O <sub>3</sub> -Zn <sub>1</sub> -N <sub>1</sub>	108.29(12)	Zn <sub>1</sub> -O <sub>4</sub>	1.939(2)	O <sub>4</sub> -Zn <sub>1</sub> -O <sub>2</sub>	101.63(8)
Zn <sub>1</sub> -N <sub>1</sub>	2.008(3)	O <sub>3</sub> -Zn <sub>1</sub> -O <sub>1</sub>	96.64(11)	Zn <sub>1</sub> -O <sub>2</sub>	1.9779(17)	O <sub>4</sub> -Zn <sub>1</sub> -N <sub>1</sub>	103.26(11)
Zn <sub>1</sub> -O <sub>1</sub>	2.011(3)	N <sub>1</sub> -Zn <sub>1</sub> -O <sub>1</sub>	117.28(12)	Zn <sub>1</sub> -N <sub>1</sub>	2.002(3)	O <sub>2</sub> -Zn <sub>1</sub> -N <sub>1</sub>	124.25(12)
Zn <sub>1</sub> -N <sub>3</sub>	2.014(3)	O <sub>3</sub> -Zn <sub>1</sub> -N <sub>3</sub>	109.31(12)	Zn <sub>1</sub> -N <sub>3</sub>	2.067(2)	O <sub>4</sub> -Zn <sub>1</sub> -N <sub>3</sub>	114.37(10)
		N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	111.96(13)			O <sub>2</sub> -Zn <sub>1</sub> -N <sub>3</sub>	103.38(9)
		O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	112.02(12)			N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	110.12(11)
<b>Compound 9</b>				<b>Compound 10</b>			
Zn <sub>1</sub> -O <sub>1</sub>	1.973(3)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	107.78(13)				
Zn <sub>1</sub> -N <sub>1</sub>	2.012(3)	O <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	95.29(11)				
Zn <sub>1</sub> -O <sub>3</sub>	2.027(3)	N <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	118.11(13)	Zn <sub>1</sub> -O <sub>1</sub>	1.945(4)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	110.68(19)
Zn <sub>1</sub> -N <sub>3</sub>	2.035(3)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	112.09(13)	Zn <sub>1</sub> -N <sub>3</sub>	1.998(5)	O <sub>1</sub> -Zn <sub>1</sub> -N <sub>1</sub>	114.2(2)
		N <sub>1</sub> -Zn <sub>1</sub> -N <sub>3</sub>	107.49(13)	Zn <sub>1</sub> -N <sub>1</sub>	2.006(5)	N <sub>3</sub> -Zn <sub>1</sub> -N <sub>1</sub>	106.55(19)
		O <sub>3</sub> -Zn <sub>1</sub> -N <sub>3</sub>	115.38(13)	Zn <sub>1</sub> -O <sub>3</sub>	2.056(5)	O <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	124.5(3)
						N <sub>3</sub> -Zn <sub>1</sub> -O <sub>3</sub>	89.5(2)
						N <sub>1</sub> -Zn <sub>1</sub> -O <sub>3</sub>	107.6(2)

**Table SI-3:**  $\pi$ - $\pi$  interactions observed in compounds 1-10

Compound	Cg (I)-Cg (J)	Cg-Cg Distance (Å)	dihedral angle ( $\alpha$ )
Compound 1	Cg1 Cg3 <sup>1</sup>	3.8633	73
	Cg2 Cg4 <sup>2</sup>	3.8749	65
	Cg2 Cg5 <sup>2</sup>	3.9197	47
Compound 2	Cg1 Cg3 <sup>1</sup>	3.9539	58
	Cg1 Cg4 <sup>1</sup>	3.9321	45
	Cg2 Cg3	3.7119	0
	Cg5 Cg5 <sup>4</sup>	3.8459	0
Compound 3	Cg1 Cg2 <sup>1</sup>	3.9035	66
	Cg1 Cg3 <sup>1</sup>	3.8545	58
	Cg3 Cg3 <sup>5</sup>	3.9174	0
	Cg4 Cg4 <sup>4</sup>	3.8374	0
Compound 4	Cg1 Cg3 <sup>1</sup>	3.8277	73
	Cg3 Cg5 <sup>4</sup>	3.8768	13
Compound 5	Cg3 Cg3 <sup>6</sup>	3.5657	0
Compound 6	Cg1 Cg3 <sup>1</sup>	3.9825	41
	Cg1 Cg4 <sup>1</sup>	3.7395	81
	Cg4 Cg5 <sup>7</sup>	3.7246	8
Compound 7	Cg1 Cg2 <sup>1</sup>	3.9452	58
	Cg1 Cg3 <sup>1</sup>	3.7678	62
	Cg3 Cg3 <sup>8</sup>	3.5682	0
	Cg4 Cg4 <sup>9</sup>	3.5692	0
Compound 8	Cg8 Cg8 <sup>10</sup>	3.8388	0
Compound 9	Cg1 Cg3 <sup>1</sup>	3.9358	59
	Cg1 Cg4 <sup>1</sup>	3.8863	51
	Cg4 Cg4 <sup>10</sup>	3.6002	0
	Cg5 Cg5 <sup>5</sup>	3.7359	0
Compound 10	Cg1 Cg3 <sup>1</sup>	3.8597	74
	Cg5 Cg5 <sup>11</sup>	3.6700	0

(1) X,Y,Z; (2) 2-X,-1/2+Y,1-Z (3) 1-X,2-Y,-Z; (4) 1-X,1-Y,1-Z; (5) 1-X,-Y,1-Z; (6) -X,1-Y,-Z; (7) 2-X,-Y,-Z;  
 (8) -X,-Y,-Z; (9) -X,1-Y,1-Z; (10) 1-X,1-Y,-Z; (11) 1-X,-Y,-Z

I and J denote ring numbers between which Cg-Cg distance is calculated

**Table SI-4:** Hydrogen bonding interactions in compounds 1-10

Compound	Donor --- H...Acceptor	D - H	H...A	D...A	D - H...A
Compound 1	O2W --H2WA ..O4A <sup>1</sup>	0.95	2.03	2.9728	173
	N2A --H22A ..O4A <sup>2</sup>	0.90	1.80	2.7043	174
	N4A --H44A ..O1	0.68	2.55	2.9723	122
	N4A --H44A ..O2W <sup>3</sup>	0.68	2.25	2.8100	140
	N2 --H222 ..O2	0.73	2.01	2.6909	156
	N4 --H444 ..O1W <sup>4</sup>	0.78	2.23	2.8567	137
Compound 2	O5 --H5 ..O1S <sup>5</sup>	0.75	1.89	2.6304	173
	N2 --H222 ..O3 <sup>6</sup>	0.89	1.96	2.8400	169
	N4 --H444 ..O2	0.88	1.94	2.7431	152
Compound 3	O1W --H1WA ..O5 <sup>6</sup>	0.93	2.50	3.0107	115
	N2 --H222 ..O2 <sup>7</sup>	0.87	2.03	2.8252	151
	N4 --H444 ..O4 <sup>6</sup>	0.82	1.91	2.6991	162
Compound 4	O1S --H1S ..O2 <sup>6</sup>	1.06	1.74	2.7658	162
	N2 --H222 ..O2	0.88	1.90	2.7381	158
	N4 --H444 ..O1S	0.92	1.88	2.7511	158
Compound 5	O 4 --H44 ..O2 <sup>8</sup>	0.77	1.98	2.7234	161
	N 2 --H222 ..O2	0.82	2.39	3.0601	140
	N 2 --H222 ..O3 <sup>9</sup>	0.82	2.24	2.8620	133
	N 4 --H444 ..O6	0.87	1.80	2.6409	162
Compound 6	N2 --H2A ..O1S	0.86	1.91	2.7562	166
	N4 --H4B ..O2	0.86	1.97	2.7594	151
Compound 7	O1W --H1WA ..O3W <sup>10</sup>	0.66	2.33	2.5790	104
	N2 --H2A ..O1 <sup>11</sup>	0.86	2.03	2.8851	172
	O2W --H2WB ..O3W	0.80	2.09	2.6443	126
	O2W --H2WA ..O3W <sup>10</sup>	0.83	2.23	2.7946	126
	N4 --H4B ..O1W <sup>12</sup>	0.86	2.00	2.8305	162
Compound 8	O2W --H2WB ..O1W <sup>13</sup>	1.01	2.56	2.9843	105
	O2W --H2WA ..O1W <sup>13</sup>	1.02	2.48	2.9843	110
	N2 --H2A ..O1W <sup>13</sup>	0.86	1.89	2.7233	163
	N3 --H3AB ..O3 <sup>14</sup>	0.92	2.14	3.0261	161
	N3 --H3AA ..O2 <sup>15</sup>	0.83	2.27	3.0709	163
Compound 9	N(2) --H(2A) ..O(3) <sup>16</sup>	0.86	2.01	2.8522	166
	N(4) --H(4B) ..O(2)	0.86	2.10	2.7795	135
Compound 10	O1W --H1WA ..O3	1.02	2.52	3.5231	167
	N2 --H2A ..O2 <sup>7</sup>	0.86	2.04	2.8648	160
	N4 --H4B ..O2	0.86	2.60	3.1031	119
	N4 --H4B ..O4 <sup>14</sup>	0.86	1.94	2.7394	155

(1)  $-x, 1/2+y, 1-z$ ; (2)  $1-x, 1/2+y, 2-z$ ; (3)  $1+x, y, z$ ; (4)  $x, y, 1+z$ ; (5)  $2-x, 1-y, 1-z$ ; (6)  $1-x, 1-y, 1-z$ ; (7)  $x, 1/2-y, -1/2+z$ ; (8)  $-1/2+x, 1/2+y, z$ ; (9)  $-x, 1-y, -z$ ; (10)  $1-x, 1-y, -z$ ; (11)  $-x, -y, 1-z$ ; (12)  $-1+x, y, z$ ; (13)  $1/2-x, -1/2+y, 1/2-z$ ; (14)  $1-x, -y, -z$ ; (15)  $3/2-x, -1/2+y, 1/2-z$ ; (16)  $1-x, -y, 1-z$ .