Supporting Information

Effect of coordinating (-CN) Vs non-coordinating (-F) substituents in 3-pyridyl urea receptors toward second sphere sulfate recognition: Selective crystallisation of CuSO₄ from mixtures of competing anions/ cations.

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Atoms	Distance (Å)
7 toms	Distance (11)
Cu1 -O3	2.2469
Cu1 -O5	2.4112
Cu1 -N1	2.0223

Table 1S. Metal-Ligand bond distances in complex 1

 Table 2S. Hydrogen bonding table for complex 1

D-H_A	d(H-A) Å	d(D-A) Å	<dha (°)<="" th=""></dha>
N(2)H(2)O(2)	2.44	3.24	156
N(3)H(3)O(2)	2.00	2.85	177
C(1)H(1)O(4)	2.23	2.99	138

Figure 1S. Scattered plot of complex 1



Atoms	Distance (Å)
Cd1 -O3	2.2699
Cd1 -O4	2.2687
Cd1 -N1	2.3394
Cd1 -N2	2.3812

Table 3S. Metal-Ligand bond distances in complex 2

 Table 4S. Hydrogen bonding table for complex 2

D-HA	d(H-A) Å	d(D-A) Å	<dha (°)<="" th=""></dha>
N(3)H(3)O(1)	2.20	3.02	161
N(4)H(4)O(1)	1.95	2.81	173
N(5)H(5)O(2)	2.16	2.98	160
N(6)H(6)O(2)	1.97	2.83	180
C(3)H(3A)O(1)	2.48	2.83	142

Figure 2S. Scattered plot of complex 2



Atoms	Distance (Å)
Cu1 -N1	2.0178
Cu1 -N2	2.0140
Cu1 -N3	2.0183
Cu1 -N4	2.0350
Cu1 -N13	2.5759
Cu1 -N16	2.6716

Table 5S. Metal-Ligand bond distances in complex 3

Table 6S. Hydrogen bonding table for complex 3

D-HA	d(H-A) Å	d(D-A) Å	<dha (°)<="" th=""></dha>
N(5)H(5)O(4)	1.94	2.79	174
N(6)H(6)O(3)	2.14	2.97	162
N(7)H(7)O(1)	2.03	2.80	149
N(8)H(8A)O(1)	2.14	2.95	156
N(8)H(8A)O(2)	2.42	3.09	136
N(9)H(9)O(1)	2.01	2.85	168
N(10)H(10)O(4)	1.98	2.84	177
N(11)H(11)O(2)	2.01	2.83	160
N(12)H(12)O(3)	2.15	3.00	171
C(8)H(8)O(3)	2.53	3.24	135
C(11)H(11A)O(1)	2.55	3.47	169





 Table 7S. Metal-Ligand bond distances in complex 4

Atoms	Distance (Å)	
Co1 -O1	2.1106	
Co1 -O2	2.1131	
Co1 -N1	2.1467	

D-HA	d(H-A) Å	d(D-A) Å	<dha (°)<="" th=""></dha>
N(2)H(2)O(4A)	2.04	2.90	173
N(3)H(3)O(5A)	2.15	2.98	160
C(8)H(8)O(5A)	2.46	3.24	142
C(11)H(11)O(3)	2.35	3.26	165
C(16)H(16C)N(4)	2.59	3.50	158

Table 8S. Hydrogen bonding table for complex 4





Figure 5S. ESI-MS spectrum of L^1



Figure 6S. ¹H-NMR spectrum of L¹



Figure 7S. ¹³C-NMR spectrum of L¹



Figure 8S. ESI-MS spectrum of L^2



Figure 9S^{\cdot 1}H-NMR spectrum of L^2



Figure 10S. ¹³C-NMR spectrum of L²

Figure 11S. FT-IR spectrum of complex 1

Peak at 1100 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .

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Figure 12S. FT-IR spectrum of complex 2

Peak at 1105 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .

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Figure 13S. FT-IR spectrum of complex 3

Peak at 1116 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .

Figure 14S. FT-IR spectrum of complex 4

Peak at 1110 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .

Figure 15S. Comperison of FT-IR spectra of complexes 1, 1', 1" and 1""