

Supporting Information

Solvents influence on sizes of channels in three fry topological Mn(II)-MOFs based on metal-carboxylate chains: Syntheses, structures and magnetic properties†

Xiang Zhou, Ping Liu, Wen-Huan Huang, Meng Kang, Yao-Yu Wang,* and Qi-Zhen
Shi

Key Laboratory of Synthetic and Natural Functional Molecule Chemistry of the Ministry of Education, Shaanxi;

Key Laboratory of Physico-Inorganic Chemistry, College of Chemistry & Materials Science, Northwest

University, Xi'an 710069, P. R. China. E-mail: wyaoyu@nwu.edu.cn

The Supporting Pictures For Article

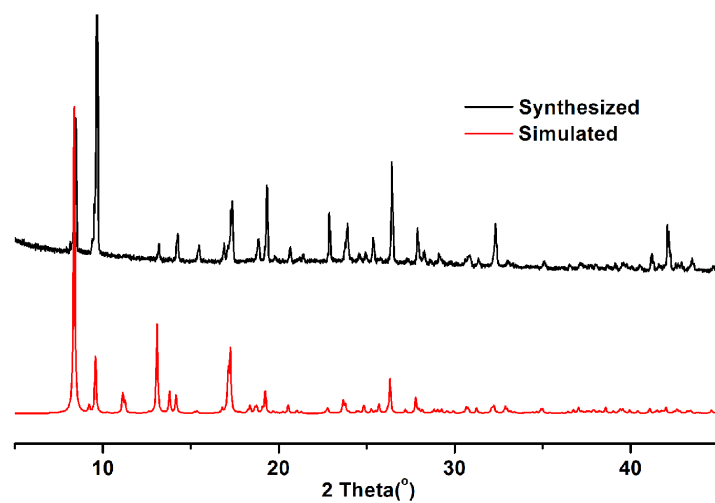


Fig. S1 Powder X-ray diffraction patterns of compound 1.

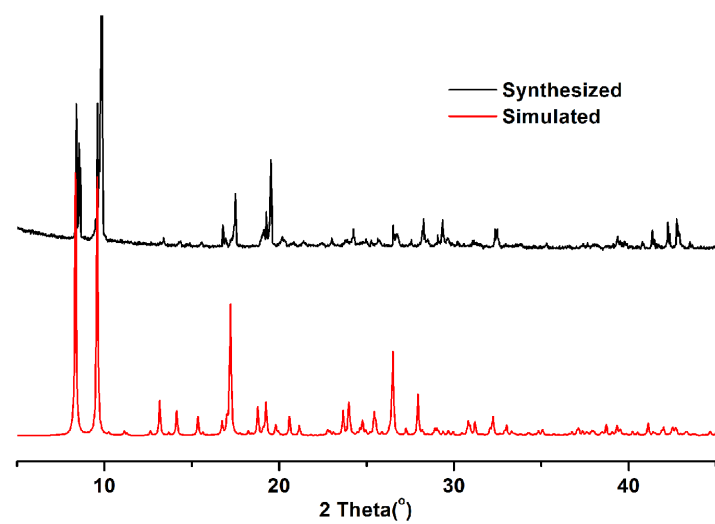


Fig. S2 Powder X-ray diffraction patterns of compound 2.

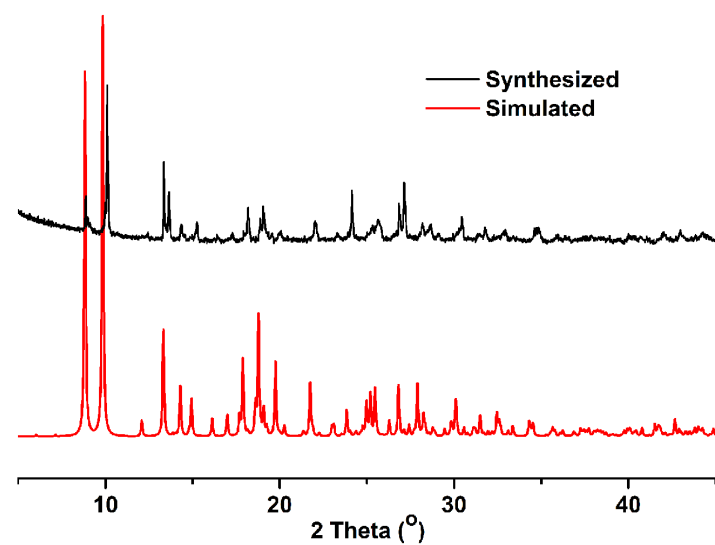


Fig. S3 Powder X-ray diffraction patterns of compound 3.

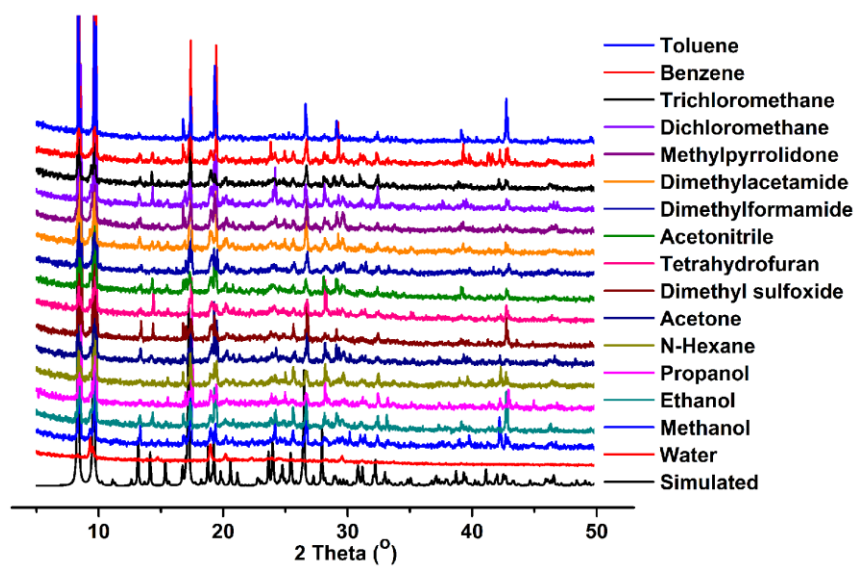


Fig. S4 The PXR D patterns of 2 after soaking in various solvents for 1 week.

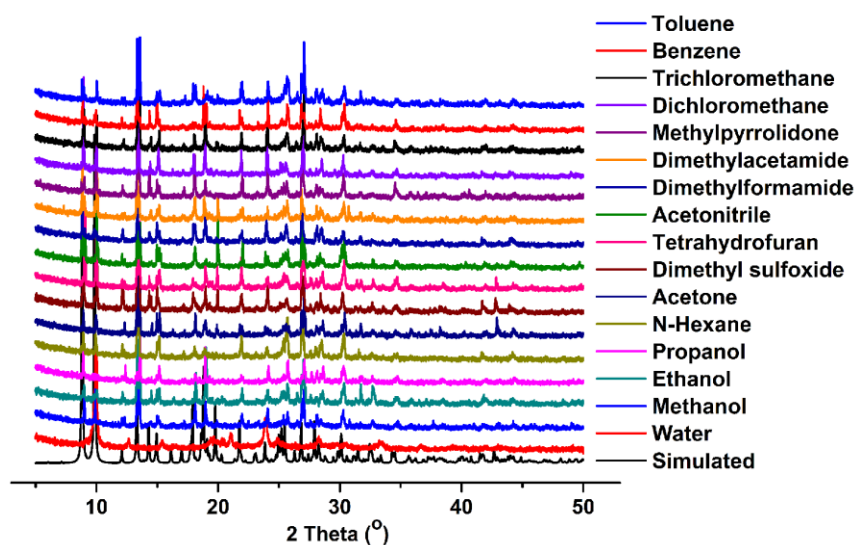


Fig. S5 The PXR D patterns of 3 after soaking in various solvents for 1 week.

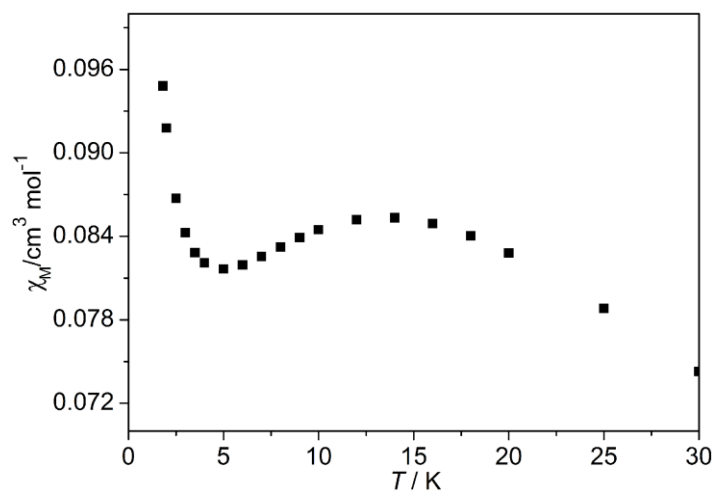


Fig. S6 The χ_M VS. T plot for compound 1.

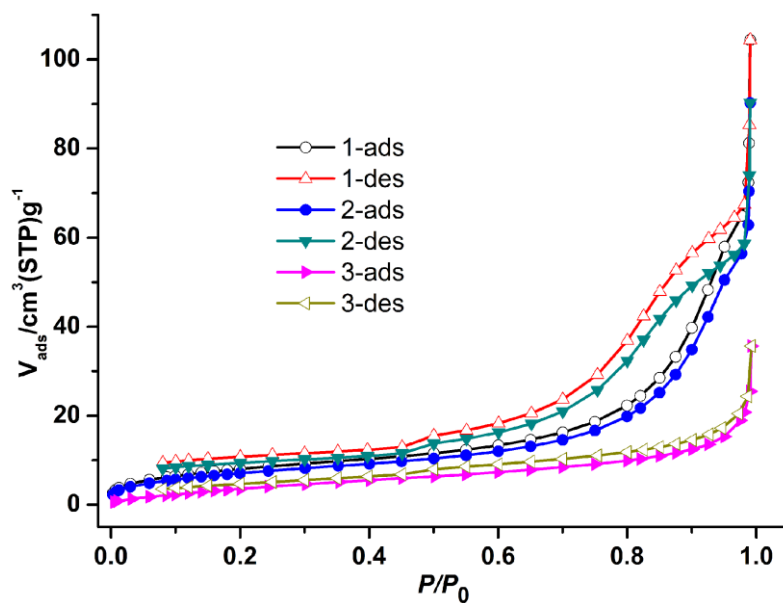


Fig. S7 Nitrogen adsorption–desorption isotherms of 1–3 at 77 K.

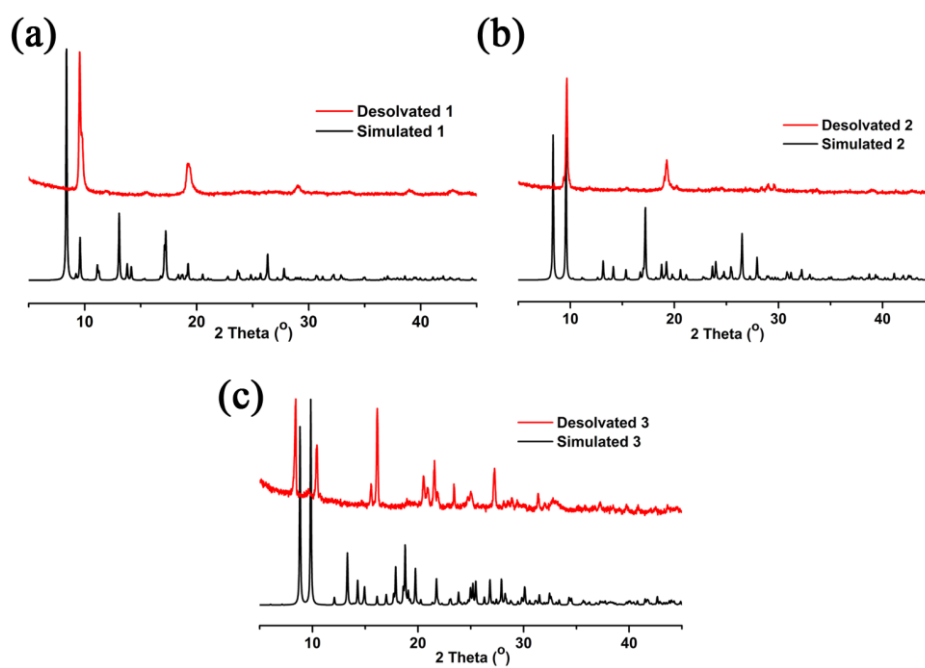
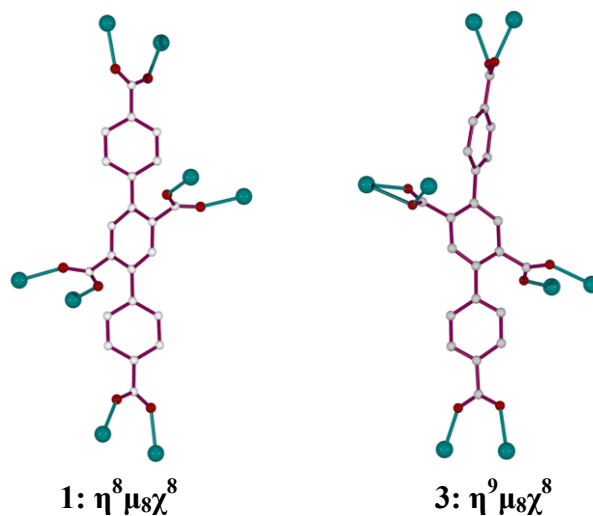


Fig. S8 PXRD patterns of 1 (a), 2 (b), 3 (c), desolvated 1, 2 and 3.

The Supporting Scheme For Article

Scheme S1. Coordination Modes of Ligands in Compounds 1 and 3



Two coordination fashions of ligands H_4tpc in compounds **1** and **3**. The symbol of $\eta^x \mu_y \chi^z$ describes the coordination modes, where: η^x – number of coordination bonds donated from the ligand; μ_y – number of metal centers bonded to the ligand; χ^z – number of atoms donating coordination bonds.

The Supporting Table For Article

Table S1. Comparison of the diverse solvent-controlled Mn-tpc system

Compound	1	2	3
Crystallizing medium	H ₂ O/NMP	H ₂ O/DMA	H ₂ O/DMF
Coordinated solvents	H ₂ O/NMP	H ₂ O/DMA	DMF
Coordination modes of solvent	bridging	bridging	monodentate/bridging
Bridging fashions of carboxyl groups	$\mu_2\text{-}\eta^1\text{:}\eta^1$	$\mu_2\text{-}\eta^1\text{:}\eta^1$	$\mu_2\text{-}\eta^1\text{:}\eta^1/\mu_2\text{-}\eta^1\text{:}\eta^2$
Metal coordination	octahedron (MnO ₆)	octahedron (MnO ₆)	octahedron (MnO ₆)
Intermetallic bridging modes	($\mu\text{-H}_2\text{O/NMP}$)($\mu\text{-COO}$) ₂	($\mu\text{-H}_2\text{O/DMA}$)($\mu\text{-COO}$) ₂	($\mu\text{-DMF}$)($\mu\text{-COO}$) ₂ /($\mu\text{-COO}$) ₂
Interchain Mn...Mn distances	9.494 Å, 12.628 Å	9.490 Å, 12.646 Å	8.880 Å, 12.463 Å
Size of channel	9.494 × 3.725 Å ²	9.490 × 3.706 Å ²	8.880 × 3.729 Å ²
Pore volume	1242.4 Å ³	1205.9 Å ³	1005.3 Å ³
Void ratio	41.4%	40.5%	37.0%

Table S2. Selected Bond Lengths (Å) and Angles (°) for **1-3**.

Complex 1					
Mn(1)-O(2)	2.114(3)	Mn(1)-O(8)#1	2.132(4)	Mn(1)-O(6)#2	2.136(3)
Mn(1)-O(4) #3	2.155(3)	Mn(1)-O(9)	2.240(3)	Mn(1)-O(10)	2.383(3)
Mn(2)-O(7) #3	2.155(3)	Mn(2)-O(5)#6	2.159(3)	Mn(2)-O(9)#6	2.261(3)
Mn(3)-O(3) #3	2.133(3)	Mn(3)-O(1)	2.149(3)	Mn(3)-O(10)	2.253(3)
O(2)-Mn(1)-O(8)#1	162.87(14)	O(4)#3-Mn(1)-O(10)	99.89(12)		
O(2)-Mn(1)-O(6)#2	91.49(14)	O(9)-Mn(1)-O(10)	169.79(11)		
O(8)#1-Mn(1)-O(6)#2	96.44(14)	O(7)#3-Mn(2)-O(7)#4	180.00(15)		
O(2)-Mn(1)-O(4)#3	87.59(14)	O(7)#3-Mn(2)-O(5)#5	91.85(13)		
O(8)#1-Mn(1)-O(4)#3	85.75(14)	O(7)#4-Mn(2)-O(5)#5	88.15(13)		
O(6)#2-Mn(1)-O(4)#3	175.04(13)	O(7)#3-Mn(2)-O(5)	88.15(13)		
O(2)-Mn(1)-O(9)	94.10(12)	O(7)#4-Mn(2)-O(5)	91.85(13)		
O(8)#1-Mn(1)-O(9)	101.67(13)	O(5)#5-Mn(2)-O(5)	180.0		
O(6)#2-Mn(1)-O(9)	85.14(13)	O(7)#3-Mn(2)-O(9)#6	87.63(12)		
O(4)#3-Mn(1)-O(9)	90.07(13)	O(7)#4-Mn(2)-O(9)#6	92.37(12)		
O(2)-Mn(1)-O(10)	88.60(12)	O(5)#5-Mn(2)-O(9)#6	90.07(12)		
O(8)#1-Mn(1)-O(10)	77.05(13)	O(5)-Mn(2)-O(9)#6	89.93(12)		
O(6)#2-Mn(1)-O(10)	84.95(12)	O(7)#3-Mn(2)-O(9)#7	92.37(12)		
O(7)#4-Mn(2)-O(9)#7	87.63(12)	O(1)#9-Mn(3)-O(1)	180.000(1)		
O(5)#5-Mn(2)-O(9)#7	89.93(12)	O(3)#3-Mn(3)-O(10)	86.98(13)		
O(5)-Mn(2)-O(9)#7	90.07(12)	O(3)#8-Mn(3)-O(10)	93.02(13)		
O(9)#6-Mn(2)-O(9)#7	180.000(1)	O(1)#9-Mn(3)-O(10)	86.58(12)		
O(3)#3-Mn(3)-O(3)#8	180.000(1)	O(1)-Mn(3)-O(10)	93.42(12)		
O(3)#3-Mn(3)-O(1)#9	84.39(14)	O(3)#3-Mn(3)-O(10)#9	93.02(13)		
O(3)#8-Mn(3)-O(1)#9	95.61(13)	O(3)#8-Mn(3)-O(10)#9	86.98(13)		
O(3)#3-Mn(3)-O(1)	95.61(13)	O(1)#9-Mn(3)-O(10)#9	93.42(12)		
O(3)#8-Mn(3)-O(1)	84.39(14)	O(1)-Mn(3)-O(10)#9	86.58(12)		
O(10)-Mn(3)-O(10)#9	180.000(1)	Mn(1)-O(9)-Mn(2)#12	111.70(13)		
Mn(3)-O(10)-Mn(1)	109.22(12)				

Symmetry transformations used to generate equivalent atoms: #1 $x-1, -y+3/2, z+1/2$; #2 $x, -y+3/2, z+1/2$; #3 $x-1, y, z$; #4 $-x+1, -y+1, -z+1$; #5 $-x, -y+1, -z+1$; #6 $-x, y-1/2, -z+3/2$; #7 $x, -y+3/2, z-1/2$; #8 $-x+1, -y+1, -z+2$; #9 $-x, -y+1, -z+2$; #12 $-x, y+1/2, -z+3/2$

Complex 2					
Mn(1)-O(8)#1	2.114(4)	Mn(1)-O(5)	2.135(3)	Mn(1)-O(3)#2	2.141(4)
Mn(1)-O(1)#3	2.149(4)	Mn(1)-O(9)	2.246(3)	Mn(1)-O(10)	2.373(3)
Mn(2)-O(6)	2.155(3)	Mn(2)-O(2)#3	2.165(4)	Mn(2)-O(9)	2.276(3)
Mn(3)-O(4)#2	2.118(4)	Mn(3)-O(7)#1	2.175(4)	Mn(3)-O(10)	2.236(3)
O(8)#1-Mn(1)-O(5)	91.58(16)	O(2)#5-Mn(2)-O(9)#4	93.96(14)		
O(8)#1-Mn(1)-O(3)#2	87.90(17)	O(2)#3-Mn(2)-O(9)#4	86.04(14)		
O(5)-Mn(1)-O(3)#2	175.00(15)	O(6)#4-Mn(2)-O(9)	91.10(14)		
O(8)#1-Mn(1)-O(1)#3	162.04(16)	O(6)-Mn(2)-O(9)	88.90(14)		

O(5)-Mn(1)-O(1)#3	96.26(16)	O(2)#5-Mn(2)-O(9)	86.04(14)
O(3)#2-Mn(1)-O(1)#3	85.65(17)	O(2)#3-Mn(2)-O(9)	93.96(14)
O(8)#1-Mn(1)-O(9)	93.14(15)	O(9)#4-Mn(2)-O(9)	180.000(1)
O(5)-Mn(1)-O(9)	84.77(14)	O(4)#2-Mn(3)-O(4)#6	180.000(1)
O(3)#2-Mn(1)-O(9)	90.29(15)	O(4)#2-Mn(3)-O(7)#1	95.99(16)
O(1)#3-Mn(1)-O(9)	103.62(15)	O(4)#6-Mn(3)-O(7)#1	84.01(17)
O(8)#1-Mn(1)-O(10)	89.44(15)	O(4)#2-Mn(3)-O(7)#7	84.01(17)
O(5)-Mn(1)-O(10)	84.45(14)	O(4)#6-Mn(3)-O(7)#7	95.99(16)
O(3)#2-Mn(1)-O(10)	100.51(14)	O(7)#1-Mn(3)-O(7)#7	180.000(2)
O(1)#3-Mn(1)-O(10)	75.33(15)	O(4)#2-Mn(3)-O(10)#8	92.28(15)
O(9)-Mn(1)-O(10)	168.98(13)	O(4)#6-Mn(3)-O(10)#8	87.72(15)
O(6)#4-Mn(2)-O(6)	180.000(1)	O(7)#1-Mn(3)-O(10)#8	86.07(14)
O(6)#4-Mn(2)-O(2)#5	88.58(16)	O(7)#7-Mn(3)-O(10)#8	93.93(14)
O(6)-Mn(2)-O(2)#5	91.42(16)	O(4)#2-Mn(3)-O(10)	87.72(15)
O(6)#4-Mn(2)-O(2)#3	91.42(16)	O(4)#6-Mn(3)-O(10)	92.28(15)
O(6)-Mn(2)-O(2)#3	88.58(16)	O(7)#1-Mn(3)-O(10)	93.93(14)
O(2)#5-Mn(2)-O(2)#3	180.000(2)	O(7)#7-Mn(3)-O(10)	86.07(14)
O(6)#4-Mn(2)-O(9)#4	88.90(14)	O(10)#8-Mn(3)-O(10)	180.000(1)
O(6)-Mn(2)-O(9)#4	91.10(14)	Mn(1)-O(9)-Mn(2)	110.05(15)
Mn(3)-O(10)-Mn(1)	108.61(15)		

Symmetry transformations used to generate equivalent atoms: #1 $x, -y+3/2, z+1/2$; #2 $x+1, -y+3/2, +1/2$; #3 $x+1, y, z$; #4 $-x+2, -y+2, -z+2$; #5 $-x+1, -y+2, -z+2$; #6 $-x+1, y-1/2, -z+3/2$; #7 $-x+2, y-1/2, -z+3/2$; #8 $-x+2, -y+1, -z+2$

Complex 3					
Mn(1)-O(1)#1	2.071(2)	Mn(1)-O(3)	2.195(2)	Mn(1)-O(9)	2.327(3)
Mn(2)-O(4)	2.105(2)	Mn(2)-O(2)#2	2.146(2)	Mn(2)-O(9)#3	2.396(3)
Mn(3)-O(8)#7	2.099(3)	Mn(3)-O(5)#8	2.166(2)	Mn(3)-O(7)#9	2.178(2)
Mn(3)-O(10)	2.180(3)	Mn(3)-O(6)	2.214(3)	Mn(3)-O(5)	2.408(2)
O(1)#1-Mn(1)-O(1)#2	180.000(1)	O(1)#1-Mn(1)-O(3)	84.33(10)		
O(1)#2-Mn(1)-O(3)	95.67(10)	O(3)#3-Mn(1)-O(3)	180.0		
O(1)#1-Mn(1)-O(9)	92.69(10)	O(1)#2-Mn(1)-O(9)	87.31(10)		
O(3)#3-Mn(1)-O(9)	89.82(10)	O(3)-Mn(1)-O(9)	90.18(9)		
O(9)#3-Mn(1)-O(9)	180.0	O(4)-Mn(2)-O(4)#4	180.000(1)		
O(4)-Mn(2)-O(2)#5	87.28(10)	O(4)#4-Mn(2)-O(2)#5	92.72(10)		
O(2)#5-Mn(2)-O(2)#2	180.0	O(4)-Mn(2)-O(9)#3	87.26(10)		
O(4)#4-Mn(2)-O(9)#3	92.74(10)	O(2)#5-Mn(2)-O(9)#3	82.78(9)		
O(2)#2-Mn(2)-O(9)#3	97.22(9)	O(9)#3-Mn(2)-O(9)#6	180.00(11)		
O(8)#7-Mn(3)-O(5)#8	98.36(10)	O(8)#7-Mn(3)-O(7)#9	170.07(10)		
O(5)#8-Mn(3)-O(7)#9	91.54(10)	O(8)#7-Mn(3)-O(10)	86.47(12)		
O(5)#8-Mn(3)-O(10)	96.58(11)	O(7)#9-Mn(3)-O(10)	93.17(11)		
O(8)#7-Mn(3)-O(6)	91.70(11)	O(5)#8-Mn(3)-O(6)	94.67(9)		
O(7)#9-Mn(3)-O(6)	86.71(10)	O(10)-Mn(3)-O(6)	168.75(11)		
O(8)#7-Mn(3)-O(5)	87.30(9)	O(5)#8-Mn(3)-O(5)	151.07(6)		
O(7)#9-Mn(3)-O(5)	83.64(9)	O(10)-Mn(3)-O(5)	112.12(10)		

O(6)-Mn(3)-O(5)	56.67(9)	Mn(3)#10-O(5)-Mn(3)	116.74(10)
Mn(1)-O(9)-Mn(2)#12	104.28(10)		

Symmetry transformations used to generate equivalent atoms: #1 $-x+1, -y+1, -z+1$; #2 $x-1, y, z$; #3 $-x, -y+1, -z+1$; #4 $-x, -y+2, -z+1$; #5 $-x+1, -y+2, -z+1$; #6 $x, y+1, z$; #7 $x+1, y, z$; #8 $-x+1, y-1/2, -z+1/2$; #9 $-x, y-1/2, -z+1/2$; #10 $-x+1, y+1/2, -z+1/2$; #12 $x, y-1, z$