

Supporting Information

Battlement shaped 1D coordination polymer, based on a bis(*N*-methylimidazol-2-yl)butadiyne ligand

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Experimental Section

The synthesis of 2-iodo-1-methylimidazole (**4**) and 1-methyl-2-trimethylsilyethynylimidazole (**5**) were performed following literature procedures. ¹H-NMR spectra were recorded on a Bruker DPX 300 AVANCE, δ are given values relative to the residual solvent signal (1H: CHCl₃, 7.26 ppm; ¹³C{¹H}: CDCl₃, 77.2 ppm). Elemental analyses were carried out on a Euro EA 3000 (Euro Vector). Infrared spectra were recorded with a EXCALIBUR FTS-3500 FT-IR spectrometer. X-Ray structure analyses were carried out on a Bruker Kappa CCD (SADABS-2008/1 (Bruker, 2008)). All calculations were carried out with the SHELX97 software package.

2-(2-Chloroethenyl)-*N*-methylimidazole-2-yl (3a,b): To a solution of (chloromethyl)triphenylphosphonium chloride (25.0 g, 72.0 mmol) in THF (100 mL) *n*-butyllithium (45.0 mL, 72.0 mmol, 1.60 M in hexanes) was added drop wise at 0 °C, and stirred for 1 h. Subsequently 1-methylimidazole-2-carbaldehyde (**2**) (5.00 g, 45.0 mmol) was added and stirred for 12 hours at room temperature. The yellow-brown product containing a mixture of *E/Z*-isomers was purified by column chromatography (silica, 6 × 20 cm, hexane/EtOAc 5/1 and silica 6 × 40 cm, CHCl₃/MeOH 12/1) yielding a brown oil. (1.60 g, 11.2 mmol, 25 %). ¹H-NMR (300 MHz, CDCl₃): (*E*)-2-(2-Chloroethenyl)-*N*-methylimidazole-2-yl: δ = 3.62 (s, 3H, CH₃), 6.66 (d, 1H, ²*J*_{HH} = 13.2 Hz, Cl-HC=CH-), 6.84 (s, 1H, C_{im}), 7.00 (d, 1H, ²*J*_{HH} = 5.94 Hz, C_{im}), 7.08 (d, 1H, ²*J*_{HH} = 13.2 Hz, Cl-HC=CH-) ppm. ¹H-NMR (300 MHz, CDCl₃): (*Z*)-2-(2-Chloroethenyl)-*N*-methylimidazole-2-yl: δ = 3.62 (s, 3H, CH₃), 6.43 (d, 1H, ²*J*_{HH} = 8.19 Hz, Cl-HC=CH-), 6.53 (d, 1H, ²*J*_{HH} = 8.31 Hz, C_{im}), 6.89 (s, 1H, C_{im}), 7.20 (s, 1H, Cl-HC=CH-) ppm. IR : (KBr) $\tilde{\nu}$ = 3047 (m), 2956 (w), 2857 (w), 1720 (s), 1615 (m), 1485 (m), 1438 (s), 1418 (m), 1363 (w), 1290 (s), 1189 (s), 1123 (m), 1120 (m) cm⁻¹.

2-Ethynyl-*N*-methylimidazole-2-yl (6): Route A - A Schlenk flask was charged with 1-methyl-2-trimethylsilyethynylimidazole (**5**) (490 mg, 2.75 mmol) and MeOH (25 mL). After the addition of KOH (13.4 M, 2 mL) the reaction mixture was stirred 2 h. The mixture was diluted with water (100 mL), extracted with CHCl₃ (4 × 50 mL) and dried over Na₂SO₄. The solvent was removed by rotary evaporator and the crude product was purified by column chromatography (silica, 4 × 5 cm, EtOAc/hexane 3/1) to give yellow oil (254 mg, 2.39 mmol, 87 %). Characterization see below. **Route B** - A round bottom flask was charged with 2-(2-chloroethenyl)-*N*-methylimidazol-2-yl (**3a,b**) (400 mg, 2.80 mmol) in dried THF (10 mL), and potassium-*tert*-butoxide (670 mg, 59.0 mmol) in THF (10 mL) was added at 0 °C. The mixture was stirred for 12 hours at room temperature. Afterwards the mixture was poured on ice water and neutralized with a solution of ammonium chloride (20%). The brown mixture was extracted with CH₂Cl₂ (4 × 50 mL), dried over Na₂SO₄. The solvent was removed by rotary evaporator. The product was purified by column chromatography (silica 3 × 30cm, CHCl₃/MeOH/Et₃N, 12/1/1) to yield a yellow-brown oil (90.0 mg 0.80 mmol, 28.5%). ¹H-NMR (300 MHz, CDCl₃): δ = 3.31 (s, 1H, HC≡C-), 3.73 (s, 3H, CH₃), 6.89 (s, 1H, C_{im}) 7.02 (s, 1H, C_{im}) ppm. ¹³C{¹H}-NMR (300 MHz, CDCl₃): δ = 33.7 (CH₃), 72.2 (HC≡C-), 73.1 (HC≡C-), 128.4 (C_{im}), 129.5 (C_{im}), 130.5 (C_{im}) ppm.

Bis(*N*-methylimidazol-2-yl)butadiyne (bmib) (7): CuCl (0.049 g, 0.500 mmol) in pyridine (3 mL) was reacted with 2-ethynyl-*N*-methylimidazole (**6**) (0.530 g, 5.00 mmol) under oxygen atmosphere for 1.5 h at 45 °C. Pyridine was removed in vacuum, the brown residue was washed with aqueous NH₄Cl solution (10%, 70 mL), and extracted with CH₂Cl₂ (3 × 100 mL). The combined organic phases were washed (sat. NH₄Cl solution), and dried (MgSO₄). Solvent was removed and the product was purified by column chromatography (silica, \emptyset = 3 cm, 30 cm, CHCl₃/MeOH/Et₃N, 12/1/1, v/v/v) to yield a yellow-brown powder. Crystals suitable for X-Ray diffraction analysis were obtained by slow evaporation from a solution of **7** in acetone. Yield (0.490 g, 2.33 mmol, 93%). ¹H-NMR (300 MHz, CDCl₃): δ = 3.70 (s, 3H, CH₃), 6.89 (s, 1H, C_{im}), 7.03 (s, 1H, C_{im}) ppm. ¹³C{¹H}-NMR (300 MHz, acetone-d₆): δ = 33.9 (CH₃), 73.7 (-C≡C-C≡C-), 76.5 ((-C≡C-C≡C-), 124.6 (C_{im}), 131.0 (C_{im}), 132.2 (C_{im}) ppm. C₁₂H₁₀N₄ (210.09 g/mol): calculated C: 68.56; H: 4.79; N: 26.65, found C: 68.96; H: 4.43; N: 27.21. ESI/TOF MS (MeOH): m/z (%) = 443.1702 (100) [2 × M + Na]⁺. IR: (KBr) $\tilde{\nu}$ = 2153 (w), 1709 (w), 1619 (w), 1509 (s), 1477 (s), 1454 (m), 1445 (m), 1415 (m), 1401 (m), 1388 (m), 1355 (m),

1287 (s), 1192 (m), 1150 (m), 1139 (m), 1083 (w), 1046 (w), 921 (w), 911 (w), 863 (w), 771(w), 754 (s), 697(w), 691(w), 623 (w), 598 (w), 537 (w) cm^{-1} .

[Zn₅(OAc)₁₀(bmib)₂]_n (8): A solution of bmib (50.0 mg, 0.236 mmol) in MeCN was layered with a solution of Zn(OAc)₂ in THF (35 mL). By slow evaporation of the solvents, single crystals suitable for X-Ray diffraction analysis were obtained after three days.

Steady-state absorption: Absorption spectra of all samples were recorded with a Lambda 2 UV/Vis-spectrometer from Perkin Elmer (190 to 1100 nm; double-beam-instrument) using a quartz cell and 0.5 nm resolution.

Steady-state emission: The spectra were recorded on a FluoroMax 3 fluorometer built by HORIBA JobinYvon.

Time resolved absorption: Femtosecond transient absorption studies were performed with 258 nm laser pulses (1 kHz, 150 fs pulse width) from an amplified Ti:Sapphire laser system (model CPA 2101, Clark-MXR Inc.) in the TAPPS, transient absorption pump/probe system, Helios from Ultrafast Systems with 1 μJ laser energy. Nanosecond laser flash photolysis experiments were performed with 266 nm laser pulses from a Brilliant CDR Nd:YAG system (4 ns pulse width, Quantel) in a front face excitation geometry with 5 mJ laser energy.

Crystallographic data for **7**. C₁₂H₁₀N₄, M = 210.24, monoclinic, *P*2₁/*c*, *a* = 7.0586(4), *b* = 12.6096(4), *c* = 12.8621(7) Å, β = 92.413(5)°, *V* = 1053.08(9) Å³, *T* = 153(2) K, *Z* = 4, $\mu(\text{Mo-K}\alpha)$ = 0.085 mm⁻¹, 14414 data collected, 2146 independent reflections (*R*_{int} = 0.0230). Final *R*₁ [*I* > 2 σ (*I*)] = 0.0387, *wR*₂[all data] = 0.0993. CCDC 949213. Crystallographic data for **8**. C₂₂H₂₅N₄O₁₀Zn_{2.5}, M = 668.89, triclinic, *P*-1, *a* = 8.0568(10), *b* = 10.3125(10), *c* = 16.0151(16) Å, α = 98.594(8), β = 94.875(9), γ = 93.583(9)°, *V* = 1307.1(2) Å³, *T* = 153(2) K, *Z* = 2, $\mu(\text{Mo-K}\alpha)$ = 2.344 mm⁻¹, 13144 data collected, 5321 independent reflections (*R*_{int} = 0.0743). Final *R*₁ [*I* > 2 σ (*I*)] = 0.0530, *wR*₂[all data] = 0.1285. CCDC 949214.

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#SUBMISSION INFORMATION

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1. SUBMISSION DETAILS

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Please consider this CIF submission (2 compounds) for structural data
deposition. We are awaiting a notice about the assigned CSD numbers.
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2. TITLE AND AUTHOR LIST

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Battlement shaped 1D coordination polymer, based on a
bis(N-methyl-imidazol-2-yl)butadiyne ligand
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UNIT CELL INFORMATION

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'-x, -y, -z'

'x, -y-1/2, z-1/2'

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Refinement of F2 against ALL reflections. The weighted R-factor wR and
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#-----#
MOLECULAR GEOMETRY #
#-----#

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are estimated using the full covariance matrix. The cell esds are taken
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used when they are defined by crystal symmetry. An approximate (isotropic)
treatment of cell esds is used for estimating esds involving l.s. planes.

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C1 C2 C3 178.43(14) . . ?
C4 C3 C2 178.38(14) . . ?
C3 C4 C21 179.06(13) . . ?
N11 C11 N12 111.86(11) . . ?
N11 C11 C1 125.05(11) . . ?
N12 C11 C1 123.08(11) . . ?
C13 C12 N12 106.35(11) . . ?
C12 C13 N11 110.78(11) . . ?
N21 C21 N22 112.04(11) . . ?
N21 C21 C4 125.02(12) . . ?
N22 C21 C4 122.95(11) . . ?
C23 C22 N22 106.21(11) . . ?
C22 C23 N21 111.11(12) . . ?
C11 N11 C13 104.53(10) . . ?
C11 N12 C12 106.47(10) . . ?

C11 N12 C14 127.15(11) . . ?
C12 N12 C14 126.36(11) . . ?
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C22 N22 C21 106.43(11) . . ?
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C21 N22 C24 126.45(11) . . ?

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data_compound8

_audit_creation_date 2013-06-28T11:26:43-00:00
_audit_creation_method 'WinGX routine CIF_UPDATE'

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CHEMICAL INFORMATION

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_chemical_formula_moiety 'C44 H50 N8 O20 Zn5'
_chemical_formula_sum 'C44 H50 N8 O20 Zn5'
_chemical_formula_weight 1337.87
_chemical_compound_source 'synthesis as described'

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UNIT CELL INFORMATION

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_symmetry_space_group_name_H-M 'P -1'
_symmetry_space_group_name_Hall '-P 1'
_symmetry_Int_Tables_number 2
loop_
_symmetry_equiv_pos_as_xyz
'x, y, z'
'-x, -y, -z'

_cell_length_a 8.0568(10)
_cell_length_b 10.3125(10)
_cell_length_c 16.0151(16)
_cell_angle_alpha 98.594(8)
_cell_angle_beta 94.875(9)
_cell_angle_gamma 93.583(9)
_cell_volume 1307.1(2)
_cell_formula_units_Z 1
_cell_measurement_temperature 150(2)
_cell_measurement_reflns_used 51
_cell_measurement_theta_min 6
_cell_measurement_theta_max 20
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CRYSTAL INFORMATION

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_exptl_crystal_size_mid 0.103
_exptl_crystal_size_min 0.056
_exptl_crystal_density_diffn 1.700
_exptl_crystal_density_method 'not measured'
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_exptl_absorpt_coefficient_mu      2.344  
_exptl_absorpt_correction_type     multi-scan  
_exptl_absorpt_process_details     'SADABS, Bruker-AXS, 2002'  
_exptl_absorpt_correction_T_min    0.5815  
_exptl_absorpt_correction_T_max    0.7454  
  
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#           DATA COLLECTION                 #  
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_diffrn_radiation_wavelength        0.71073  
_diffrn_radiation_type              MoK\alpha  
_diffrn_radiation_monochromator      graphite  
_diffrn_radiation_probe              x-ray  
_diffrn_detector_area_resol_mean     9  
_diffrn_measurement_device_type      'Bruker-Nonius KappaCCD'  
_diffrn_measurement_method  
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\f- and \w-rotations with 1.60 \% and 192 sec per frame  
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_diffrn_reflns_av_R_equivalents      0.0743  
_diffrn_reflns_av_unetl/netl         0.1017  
_diffrn_reflns_number                 13144  
_diffrn_reflns_limit_h_min            -10  
_diffrn_reflns_limit_h_max            10  
_diffrn_reflns_limit_k_min            -12  
_diffrn_reflns_limit_k_max            11  
_diffrn_reflns_limit_l_min            -20  
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_diffrn_reflns_theta_full              26.5  
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_reflns_number_gt                     3632  
_reflns_threshold_expression           >2sigma(I)  
  
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#           COMPUTER PROGRAMS USED          #  
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_computing_data_reduction             'EVALCCD (Duisenberg, 2003)'  
_computing_structure_solution          'SHELXS-86 (Sheldrick, 1986)'  
_computing_structure_refinement        'SHELXL-97 (Sheldrick, 1997)'  
_computing_molecular_graphics          'Ortep-3 for Windows (Farrugia, 1997)'  
_computing_publication_material  
  
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#           REFINEMENT INFORMATION         #  
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Refinement of F2 against ALL reflections. The weighted R-factor wR and
```

goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

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_atom_sites_solution_secondary        difmap
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_refine_ls_extinction_method          none
_refine_ls_number_reflns              5321
_refine_ls_number_parameters          356
_refine_ls_number_restraints          0
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_refine_ls_R_factor_gt               0.053
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_refine_ls_wR_factor_gt              0.1122
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_refine_ls_restrained_S_all          1.015
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_refine_diff_density_min              -0.711
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#          ATOMIC TYPES, COORDINATES AND THERMAL PARAMETERS          #
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H H 0 0 'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
N N 0.0061 0.0033 'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
O O 0.0106 0.006 'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
Zn Zn 0.2839 1.4301 'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'

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  _atom_site_adp_type
  _atom_site_occupancy
  _atom_site_symmetry_multiplicity
  _atom_site_calc_flag
  _atom_site_refinement_flags
  _atom_site_disorder_assembly
  _atom_site_disorder_group
C1 C -0.2924(7) 0.0679(4) -0.0085(3) 0.0273(12) Uani 1 1 d ...
C2 C -0.2285(7) 0.0445(5) 0.0570(3) 0.0289(12) Uani 1 1 d ...
C3 C -0.1527(7) 0.0174(5) 0.1313(3) 0.0301(13) Uani 1 1 d ...
C4 C -0.0839(7) -0.0095(5) 0.1954(3) 0.0290(13) Uani 1 1 d ...
C11 C -0.3618(7) 0.0880(5) -0.0896(3) 0.0257(12) Uani 1 1 d ...
C12 C -0.4466(7) 0.0376(5) -0.2248(3) 0.0350(14) Uani 1 1 d ...
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H12 H -0.4729 -0.0079 -0.2794 0.042 Uiso 1 1 calc R . .
C13 C -0.4657(7) 0.1675(5) -0.1979(3) 0.0331(14) Uani 1 1 d . . .
H13 H -0.5082 0.226 -0.2316 0.04 Uiso 1 1 calc R . .
C14 C -0.3425(9) -0.1474(5) -0.1560(4) 0.0456(17) Uani 1 1 d . . .
H14A H -0.3336 -0.1896 -0.2129 0.068 Uiso 1 1 calc R . .
H14B H -0.4291 -0.1935 -0.1319 0.068 Uiso 1 1 calc R . .
H14C H -0.2381 -0.1483 -0.1224 0.068 Uiso 1 1 calc R . .
C21 C -0.0062(6) -0.0549(4) 0.2665(3) 0.0247(12) Uani 1 1 d . . .
C22 C 0.0972(7) -0.1947(5) 0.3456(3) 0.0303(13) Uani 1 1 d . . .
H22 H 0.1315 -0.2709 0.3642 0.036 Uiso 1 1 calc R . .
C23 C 0.1109(7) -0.0718(5) 0.3911(3) 0.0305(13) Uani 1 1 d . . .
H23 H 0.1577 -0.0497 0.447 0.037 Uiso 1 1 calc R . .
C24 C -0.0127(8) -0.2894(5) 0.1940(3) 0.0388(15) Uani 1 1 d . . .
H24A H 0.086 -0.3008 0.1649 0.058 Uiso 1 1 calc R . .
H24B H -0.1007 -0.2661 0.1561 0.058 Uiso 1 1 calc R . .
H24C H -0.0466 -0.37 0.2132 0.058 Uiso 1 1 calc R . .
C31 C -0.2753(7) 0.2680(5) 0.4119(3) 0.0280(12) Uani 1 1 d . . .
C32 C -0.4198(7) 0.3415(5) 0.4422(4) 0.0385(14) Uani 1 1 d . . .
H32A H -0.4247 0.4206 0.4175 0.058 Uiso 1 1 calc R . .
H32B H -0.4052 0.3635 0.5029 0.058 Uiso 1 1 calc R . .
H32C H -0.5219 0.2872 0.4255 0.058 Uiso 1 1 calc R . .
C41 C 0.1878(6) 0.4301(5) 0.3324(3) 0.0232(11) Uani 1 1 d . . .
C42 C 0.2989(8) 0.4903(5) 0.2755(3) 0.0337(13) Uani 1 1 d . . .
H42A H 0.3228 0.5823 0.2966 0.051 Uiso 1 1 calc R . .
H42B H 0.2435 0.4795 0.2192 0.051 Uiso 1 1 calc R . .
H42C H 0.4013 0.4475 0.2744 0.051 Uiso 1 1 calc R . .
C51 C 0.2205(6) 0.2859(5) 0.5539(3) 0.0253(12) Uani 1 1 d . . .
C52 C 0.3286(7) 0.2456(5) 0.6245(3) 0.0376(14) Uani 1 1 d . . .
H52A H 0.4426 0.2483 0.6111 0.056 Uiso 1 1 calc R . .
H52B H 0.2927 0.1578 0.6318 0.056 Uiso 1 1 calc R . .
H52C H 0.3201 0.3048 0.6759 0.056 Uiso 1 1 calc R . .
C61 C -0.2172(6) 0.6081(5) -0.0299(3) 0.0240(11) Uani 1 1 d . . .
C62 C -0.0508(7) 0.6690(5) -0.0461(4) 0.0408(15) Uani 1 1 d . . .
H62A H -0.0411 0.7607 -0.0227 0.061 Uiso 1 1 calc R . .
H62B H -0.0423 0.6592 -0.1061 0.061 Uiso 1 1 calc R . .
H62C H 0.037 0.6258 -0.0198 0.061 Uiso 1 1 calc R . .
C71 C -0.6428(6) 0.5574(5) -0.1406(3) 0.0254(12) Uani 1 1 d . . .
C72 C -0.7324(7) 0.6002(5) -0.2173(3) 0.0339(13) Uani 1 1 d . . .
H72A H -0.8472 0.6107 -0.2079 0.051 Uiso 1 1 calc R . .
H72B H -0.7267 0.5348 -0.2662 0.051 Uiso 1 1 calc R . .
H72C H -0.6803 0.6824 -0.2268 0.051 Uiso 1 1 calc R . .
N11 N -0.4123(5) 0.1978(4) -0.1133(2) 0.0233(9) Uani 1 1 d . . .
N12 N -0.3829(6) -0.0115(4) -0.1571(3) 0.0284(10) Uani 1 1 d . . .
N21 N 0.0454(6) 0.0156(4) 0.3423(3) 0.0269(10) Uani 1 1 d . . .
N22 N 0.0228(5) -0.1844(4) 0.2672(3) 0.0263(10) Uani 1 1 d . . .
O31 O -0.1287(4) 0.3186(3) 0.4409(2) 0.0280(8) Uani 1 1 d . . .
O32 O -0.2938(5) 0.1667(4) 0.3607(3) 0.0561(13) Uani 1 1 d . . .
O41 O 0.1475(4) 0.5001(3) 0.3967(2) 0.0271(8) Uani 1 1 d . . .
O42 O 0.1411(5) 0.3074(3) 0.3098(2) 0.0295(9) Uani 1 1 d . . .
O51 O 0.1637(4) 0.3980(3) 0.5654(2) 0.0273(8) Uani 1 1 d . . .
O52 O 0.1937(5) 0.2030(3) 0.4867(2) 0.0297(9) Uani 1 1 d . . .
O61 O -0.2545(4) 0.4906(3) -0.0621(2) 0.0308(9) Uani 1 1 d . . .
O62 O -0.3080(5) 0.6768(3) 0.0159(2) 0.0348(9) Uani 1 1 d . . .
O71 O -0.5851(5) 0.4467(3) -0.1482(2) 0.0313(9) Uani 1 1 d . . .
O72 O -0.6335(5) 0.6366(3) -0.0725(2) 0.0329(9) Uani 1 1 d . . .
Zn1 Zn 0 0.5 0.5 0.0213(2) Uani 1 2 d S . .
Zn2 Zn -0.46055(7) 0.37417(5) -0.04817(3) 0.02167(17) Uani 1 1 d . . .
Zn3 Zn 0.03528(8) 0.20684(5) 0.38644(4) 0.02283(17) Uani 1 1 d . . .

loop_

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_atom_site_aniso_U_22
_atom_site_aniso_U_33
_atom_site_aniso_U_23
_atom_site_aniso_U_13
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C2 0.033(3) 0.023(2) 0.031(3) 0.004(2) 0.003(3) 0.001(2)
C3 0.036(3) 0.024(3) 0.028(3) 0.000(2) 0.002(3) 0.002(2)
C4 0.035(3) 0.022(3) 0.027(3) -0.003(2) 0.001(3) -0.002(2)
C11 0.025(3) 0.027(3) 0.027(3) 0.008(2) 0.009(2) 0.002(2)
C12 0.045(4) 0.035(3) 0.022(3) -0.006(2) 0.006(3) 0.001(3)
C13 0.051(4) 0.029(3) 0.020(3) 0.005(2) -0.001(3) 0.008(3)
C14 0.065(5) 0.017(3) 0.053(4) -0.002(3) 0.014(3) 0.003(3)
C21 0.025(3) 0.023(2) 0.026(3) 0.004(2) 0.003(2) 0.001(2)
C22 0.036(3) 0.019(2) 0.037(3) 0.008(2) 0.004(3) 0.003(2)
C23 0.040(4) 0.027(3) 0.024(3) 0.006(2) 0.000(2) 0.002(2)
C24 0.047(4) 0.025(3) 0.037(3) -0.009(3) -0.011(3) 0.001(3)
C31 0.029(3) 0.029(3) 0.025(3) 0.002(2) 0.002(2) -0.002(2)
C32 0.021(3) 0.046(3) 0.047(4) 0.001(3) 0.006(3) 0.000(2)
C41 0.019(3) 0.028(3) 0.022(3) 0.007(2) -0.002(2) 0.000(2)
C42 0.046(4) 0.030(3) 0.028(3) 0.010(2) 0.019(3) 0.000(2)
C51 0.022(3) 0.027(3) 0.027(3) 0.008(2) 0.002(2) 0.001(2)
C52 0.041(4) 0.039(3) 0.031(3) 0.003(3) -0.005(3) 0.008(3)
C61 0.022(3) 0.030(3) 0.021(3) 0.010(2) -0.002(2) -0.004(2)
C62 0.031(3) 0.042(3) 0.051(4) 0.011(3) 0.013(3) -0.004(3)
C71 0.019(3) 0.029(3) 0.029(3) 0.008(2) 0.001(2) 0.001(2)
C72 0.036(3) 0.041(3) 0.025(3) 0.004(2) 0.000(3) 0.011(3)
N11 0.029(3) 0.023(2) 0.018(2) 0.0051(17) 0.0028(18) 0.0022(18)
N12 0.040(3) 0.022(2) 0.023(2) 0.0014(18) 0.006(2) 0.0017(19)
N21 0.036(3) 0.022(2) 0.022(2) 0.0011(18) 0.001(2) 0.0025(19)
N22 0.031(3) 0.022(2) 0.024(2) 0.0021(18) 0.000(2) 0.0009(18)
O31 0.023(2) 0.0239(17) 0.034(2) -0.0030(16) 0.0015(16) -0.0004(15)
O32 0.045(3) 0.044(2) 0.067(3) -0.023(2) -0.005(2) -0.002(2)
O41 0.037(2) 0.0251(17) 0.0188(18) -0.0014(15) 0.0100(16) -0.0003(15)
O42 0.041(2) 0.0217(17) 0.0263(19) 0.0033(15) 0.0088(17) 0.0001(15)
O51 0.029(2) 0.0248(17) 0.0259(19) 0.0000(15) -0.0037(16) 0.0039(15)
O52 0.038(2) 0.0240(17) 0.0246(19) -0.0025(15) -0.0035(17) 0.0039(16)
O61 0.029(2) 0.0292(19) 0.035(2) 0.0055(17) 0.0077(17) 0.0000(16)
O62 0.031(2) 0.0330(19) 0.041(2) 0.0039(18) 0.0125(18) -0.0001(17)
O71 0.037(2) 0.0308(19) 0.027(2) 0.0057(16) 0.0020(17) 0.0054(17)
O72 0.043(2) 0.0347(19) 0.0218(19) 0.0040(16) 0.0021(17) 0.0112(17)
Zn1 0.0255(5) 0.0185(4) 0.0194(4) 0.0002(3) 0.0043(3) 0.0022(3)
Zn2 0.0249(3) 0.0189(3) 0.0213(3) 0.0020(2) 0.0039(2) 0.0033(2)
Zn3 0.0293(4) 0.0183(3) 0.0206(3) 0.0010(2) 0.0038(3) 0.0030(2)

#-----#
MOLECULAR GEOMETRY #
#-----#

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All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

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C3 C4 1.202(7) . ?
C4 C21 1.407(7) . ?
C11 N11 1.323(6) . ?
C11 N12 1.366(6) . ?
C12 N12 1.340(7) . ?
C12 C13 1.366(7) . ?

C13 N11 1.369(6) . ?
C14 N12 1.460(6) . ?
C21 N21 1.338(6) . ?
C21 N22 1.371(6) . ?
C22 C23 1.357(6) . ?
C22 N22 1.367(6) . ?
C23 N21 1.379(6) . ?
C24 N22 1.467(6) . ?
C31 O32 1.221(5) . ?
C31 O31 1.281(6) . ?
C31 C32 1.503(7) . ?
C41 O41 1.245(5) . ?
C41 O42 1.286(5) . ?
C41 C42 1.501(7) . ?
C51 O51 1.263(6) . ?
C51 O52 1.264(5) . ?
C51 C52 1.493(7) . ?
C61 O61 1.251(5) . ?
C61 O62 1.255(6) . ?
C61 C62 1.503(7) . ?
C71 O71 1.252(6) . ?
C71 O72 1.255(5) . ?
C71 C72 1.510(7) . ?
N11 Zn2 2.034(4) . ?
N21 Zn3 2.002(4) . ?
O31 Zn3 1.980(3) . ?
O31 Zn1 2.129(3) . ?
O32 Zn3 2.645(4) . ?
O41 Zn1 2.118(3) . ?
O42 Zn3 1.938(4) . ?
O51 Zn1 2.052(3) . ?
O52 Zn3 1.971(3) . ?
O61 Zn2 2.035(3) . ?
O62 Zn2 2.036(4) 2_465 ?
O71 Zn2 2.077(4) . ?
O72 Zn2 2.036(3) 2_465 ?
Zn1 O51 2.052(3) 2_566 ?
Zn1 O41 2.118(3) 2_566 ?
Zn1 O31 2.129(3) 2_566 ?
Zn1 Zn3 3.3287(6) . ?
Zn2 O62 2.036(4) 2_465 ?
Zn2 O72 2.036(3) 2_465 ?
Zn2 Zn2 2.9489(10) 2_465 ?

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C1 C2 C3 178.9(6) . . ?
C4 C3 C2 178.1(6) . . ?
C3 C4 C21 174.0(5) . . ?
N11 C11 N12 109.9(4) . . ?
N11 C11 C1 128.4(4) . . ?
N12 C11 C1 121.6(4) . . ?
N12 C12 C13 106.8(4) . . ?
C12 C13 N11 109.2(5) . . ?
N21 C21 N22 109.7(4) . . ?
N21 C21 C4 127.3(4) . . ?
N22 C21 C4 122.9(4) . . ?
C23 C22 N22 106.4(4) . . ?
C22 C23 N21 110.0(4) . . ?
O32 C31 O31 120.5(5) . . ?
O32 C31 C32 122.7(5) . . ?

O31 C31 C32 116.8(4) . . ?
O41 C41 O42 124.9(5) . . ?
O41 C41 C42 119.1(4) . . ?
O42 C41 C42 116.0(4) . . ?
O51 C51 O52 125.4(5) . . ?
O51 C51 C52 118.4(4) . . ?
O52 C51 C52 116.1(4) . . ?
O61 C61 O62 124.3(5) . . ?
O61 C61 C62 117.3(5) . . ?
O62 C61 C62 118.4(4) . . ?
O71 C71 O72 124.6(5) . . ?
O71 C71 C72 118.9(4) . . ?
O72 C71 C72 116.5(4) . . ?
C11 N11 C13 106.2(4) . . ?
C11 N11 Zn2 133.3(3) . . ?
C13 N11 Zn2 119.3(3) . . ?
C12 N12 C11 107.9(4) . . ?
C12 N12 C14 126.0(4) . . ?
C11 N12 C14 126.1(5) . . ?
C21 N21 C23 106.0(4) . . ?
C21 N21 Zn3 131.1(3) . . ?
C23 N21 Zn3 122.9(3) . . ?
C22 N22 C21 107.8(4) . . ?
C22 N22 C24 127.3(4) . . ?
C21 N22 C24 124.8(4) . . ?
C31 O31 Zn3 108.0(3) . . ?
C31 O31 Zn1 141.3(3) . . ?
Zn3 O31 Zn1 108.16(16) . . ?
C31 O32 Zn3 77.8(3) . . ?
C41 O41 Zn1 143.2(3) . . ?
C41 O42 Zn3 121.1(3) . . ?
C51 O51 Zn1 136.6(3) . . ?
C51 O52 Zn3 128.7(3) . . ?
C61 O61 Zn2 128.1(3) . . ?
C61 O62 Zn2 128.2(3) . 2_465 ?
C71 O71 Zn2 123.2(3) . . ?
C71 O72 Zn2 132.8(3) . 2_465 ?
O51 Zn1 O51 180.00(16) 2_566 . ?
O51 Zn1 O41 86.43(14) 2_566 . ?
O51 Zn1 O41 93.57(14) . . ?
O51 Zn1 O41 93.57(14) 2_566 2_566 ?
O51 Zn1 O41 86.43(14) . 2_566 ?
O41 Zn1 O41 180.0000(10) . 2_566 ?
O51 Zn1 O31 90.50(13) 2_566 . ?
O51 Zn1 O31 89.50(13) . . ?
O41 Zn1 O31 90.54(13) . . ?
O41 Zn1 O31 89.46(13) 2_566 . ?
O51 Zn1 O31 89.50(13) 2_566 2_566 ?
O51 Zn1 O31 90.50(13) . 2_566 ?
O41 Zn1 O31 89.46(13) . 2_566 ?
O41 Zn1 O31 90.54(13) 2_566 2_566 ?
O31 Zn1 O31 180.00(18) . 2_566 ?
O51 Zn1 Zn3 109.54(8) 2_566 . ?
O51 Zn1 Zn3 70.46(8) . . ?
O41 Zn1 Zn3 64.42(8) . . ?
O41 Zn1 Zn3 115.58(8) 2_566 . ?
O31 Zn1 Zn3 34.41(9) . . ?
O31 Zn1 Zn3 145.59(9) 2_566 . ?
N11 Zn2 O61 102.06(15) . . ?
N11 Zn2 O62 98.77(15) . 2_465 ?
O61 Zn2 O62 158.97(14) . 2_465 ?
N11 Zn2 O72 102.81(15) . 2_465 ?
O61 Zn2 O72 89.47(15) . 2_465 ?
O62 Zn2 O72 88.74(16) 2_465 2_465 ?
N11 Zn2 O71 97.71(14) . . ?
O61 Zn2 O71 88.53(15) . . ?
O62 Zn2 O71 85.82(15) 2_465 . ?
O72 Zn2 O71 159.35(14) 2_465 . ?

N11 Zn2 Zn2 178.24(12) . 2_465 ?
O61 Zn2 Zn2 79.68(10) . 2_465 ?
O62 Zn2 Zn2 79.51(10) 2_465 2_465 ?
O72 Zn2 Zn2 76.86(10) 2_465 2_465 ?
O71 Zn2 Zn2 82.56(9) . 2_465 ?
O42 Zn3 O52 108.98(16) . . ?
O42 Zn3 O31 106.70(15) . . ?
O52 Zn3 O31 99.03(14) . . ?
O42 Zn3 N21 108.03(16) . . ?
O52 Zn3 N21 93.80(15) . . ?
O31 Zn3 N21 136.42(16) . . ?
O42 Zn3 O32 116.02(16) . . ?
O52 Zn3 O32 132.16(16) . . ?
O31 Zn3 O32 53.72(12) . . ?
N21 Zn3 O32 87.35(15) . . ?
O42 Zn3 Zn1 84.41(9) . . ?
O52 Zn3 Zn1 78.36(9) . . ?
O31 Zn3 Zn1 37.43(9) . . ?
N21 Zn3 Zn1 167.06(12) . . ?
O32 Zn3 Zn1 90.35(8) . . ?