

Supporting information

Hydrothermal synthesis of ordered β -NaYF₄ nanorods self-assemblies with multicolor up- and down-conversions

Mingye Ding, Chunhua Lu*, Yan Song, Yaru Ni, Zhongzi Xu*

State Key Laboratory of Materials-orient Chemical Engineering, College of Materials Science and Engineering, Nanjing University of Technology, Nanjing 210009, PR China

E-mail: chhlu@njut.edu.cn, xzz@njut.edu.cn

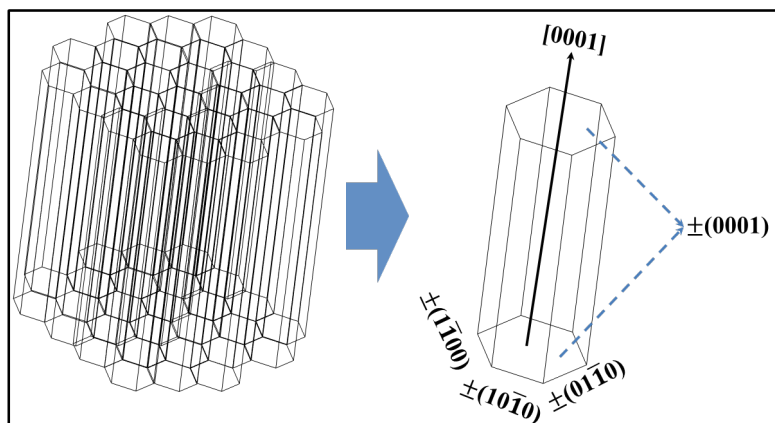


Fig. S1 Schematic diagram showing the anisotropy of the β -NaYF₄ nanorods self-assemblies.

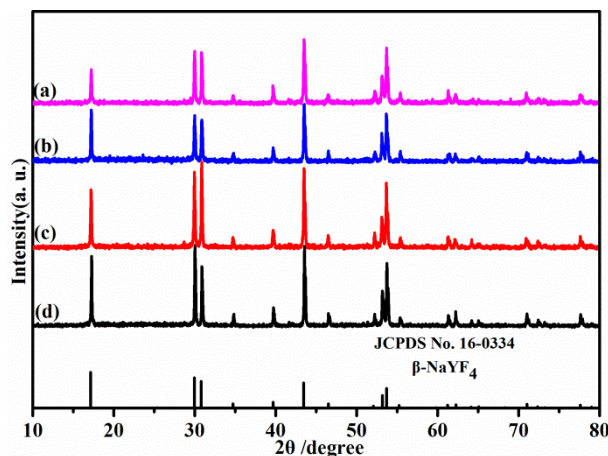


Fig. S2 The XRD patterns of as-prepared β -NaYF₄:20%Yb³⁺, 2%Er³⁺ samples prepared with different molar ratios of sodium oleate:Re³⁺. (a) 0 (without sodium oleate), (b) 1:1, (c) 4:1, (d) 8:1. All of the samples were hydrothermally treated at 260 °C for 24 h (molar ratio for NaF:Re³⁺ = 12:1).

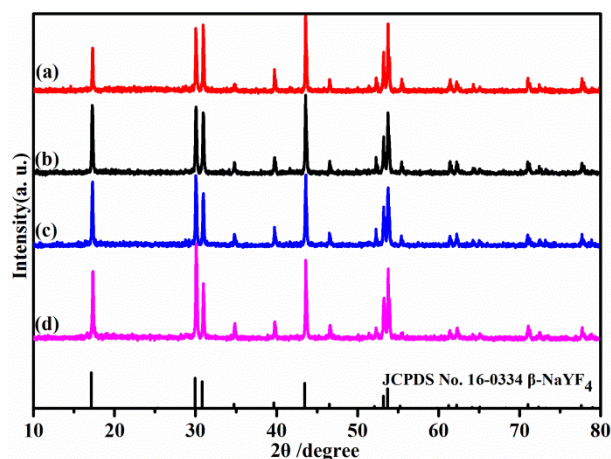


Fig. S3 The XRD patterns of β -NaYF₄:20%Yb³⁺, 2%Er³⁺ samples prepared with different molar ratios of NaF:Re³⁺. (a) 8:1, (b) 12:1, (c) 14:1, (d) 18:1. All of the samples were hydrothermally treated at 260 °C for 24 h (molar ratio for NaOA:Re³⁺ = 4:1).

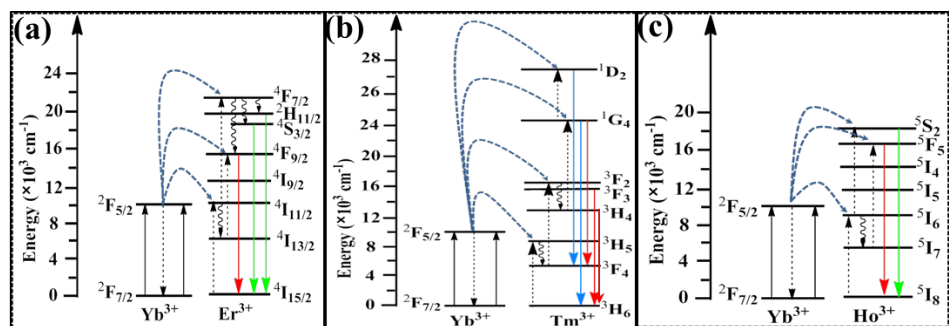


Fig. S4 The proposed UC energy transfer mechanism in Yb³⁺/Er³⁺ (a), Yb³⁺/Tm³⁺ (b), Yb³⁺/Ho³⁺ (c) -doped β -NaYF₄ samples.

Table S1. The summary of compositions, emission color and chromaticity coordinates (x, y) of the as-prepared porous microsheets.

Samples	Excitation	Emission colors	Chromaticity Coordinates
β -NaYF ₄ :5%Eu ³⁺	396 nm	red	(0.5588, 0.3503)
β -NaYF ₄ :5%Tb ³⁺	378 nm	green	(0.3229, 0.4765)
β -NaYF ₄ :5%Sm ³⁺	402 nm	orange	(0.5296, 0.4603)
β -NaYF ₄ :20%Yb ³⁺ /2%Er ³⁺	980 nm	yellow	(0.3705, 0.6065)
β -NaYF ₄ :20%Yb ³⁺ /1%Tm ³⁺	980 nm	blue	(0.1469, 0.1294)
β -NaYF ₄ :20%Yb ³⁺ /1%Ho ³⁺	980 nm	green	(0.2523, 0.7283)