

**Supporting Information**

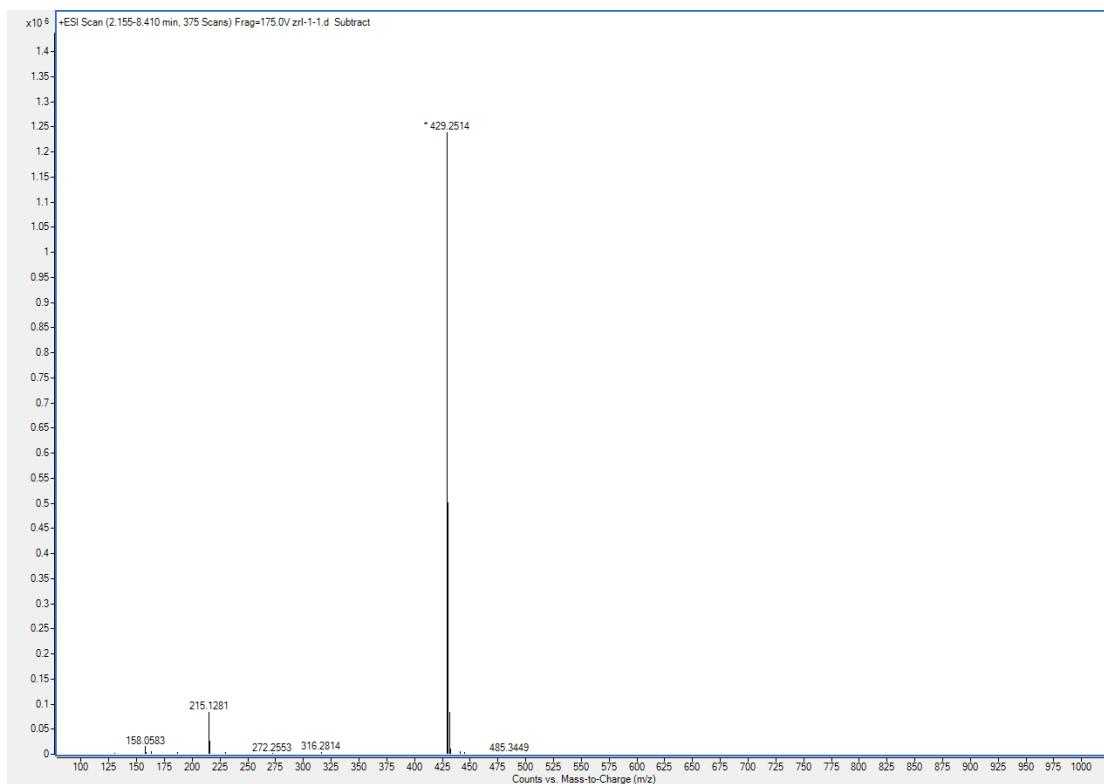
**Two Polymorphs and Cocrystal of Styryl-pyridine Derivatives with Tuned Emission Induced by Co<sup>2+</sup> and Zn(phen)<sub>3</sub><sup>2+</sup>**

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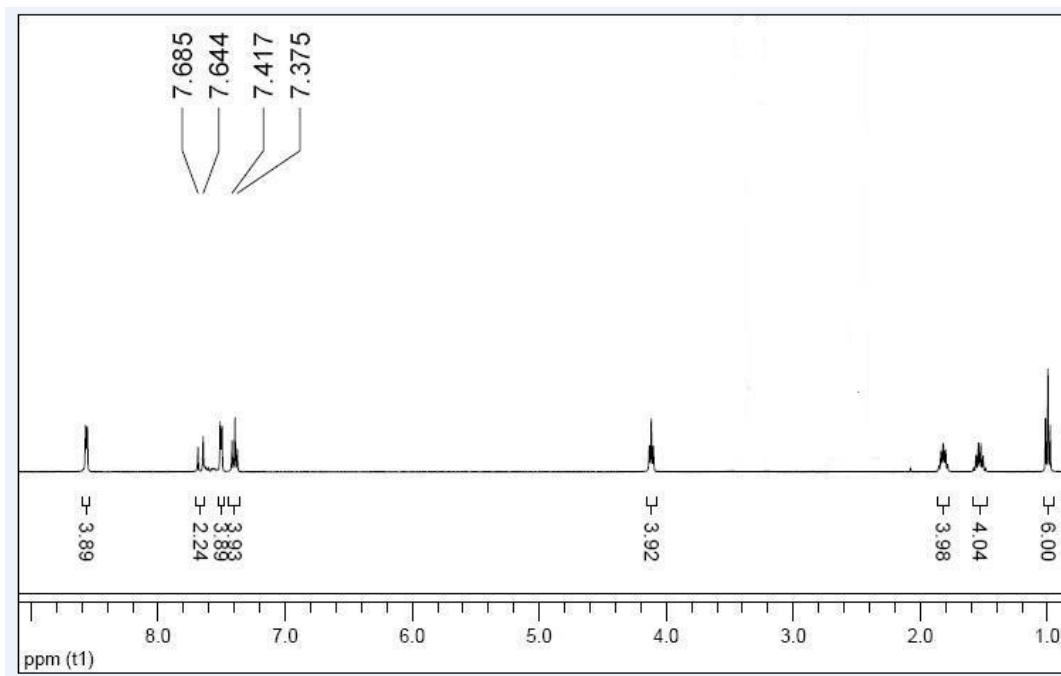
*Received (in XXX, XXX) Xth XXXXXXXXXX 201X, Accepted Xth XXXXXXXXXX 201X*  
DOI: 10.1039/b000000x

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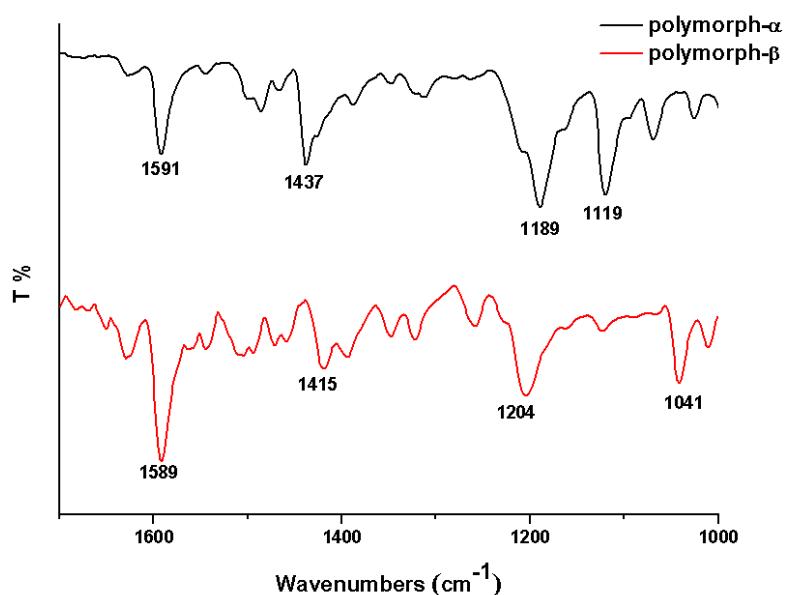
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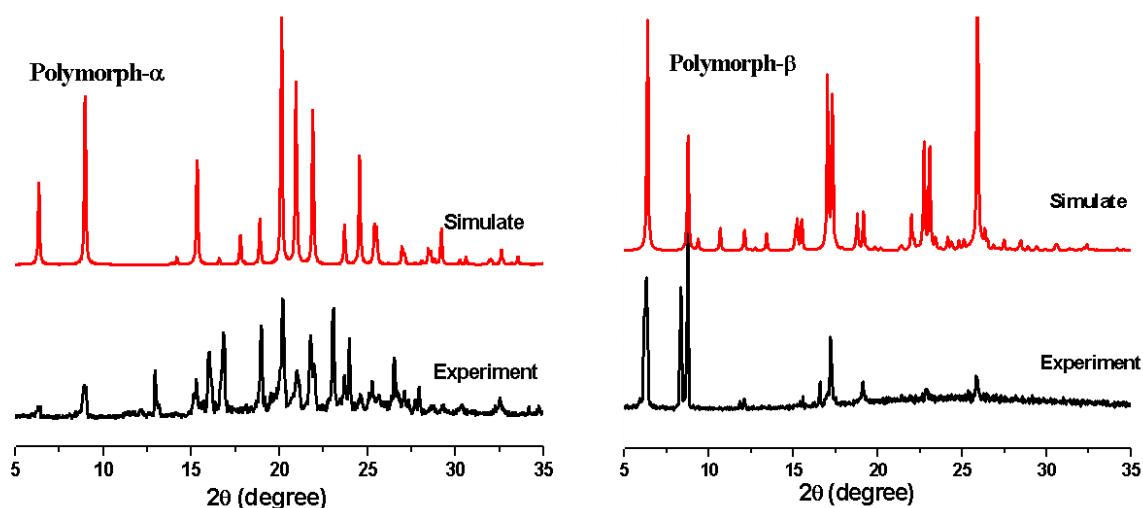
**Fig. S1** HRMS spectrum of compound **L**



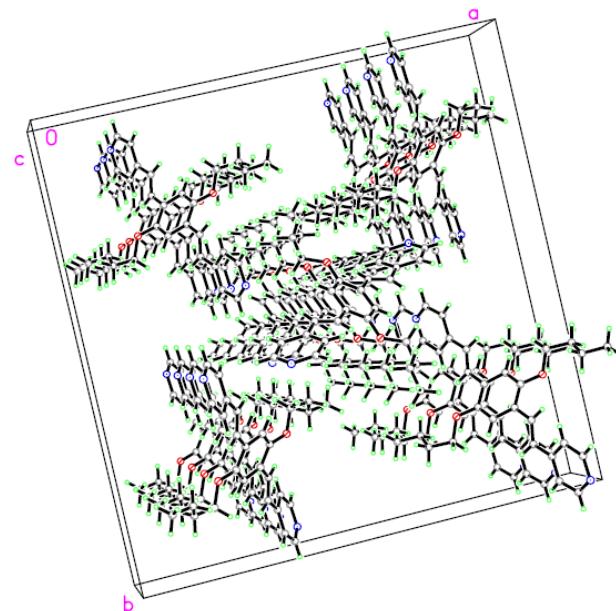
**Fig. S2** <sup>1</sup>H NMR (<sup>6</sup>-DMSO, 400 MHz) of compound L



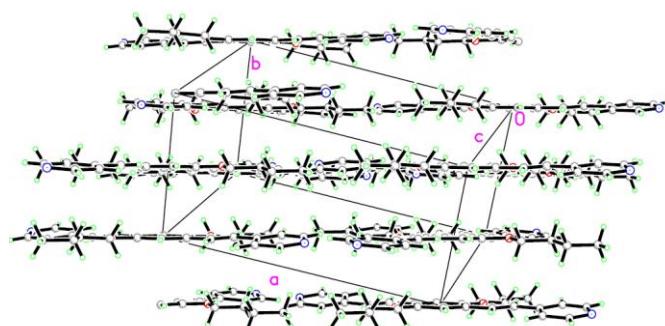
**Fig. 3S** Comparing IR pattern of polymorph- $\alpha$  and polymorph- $\beta$ .



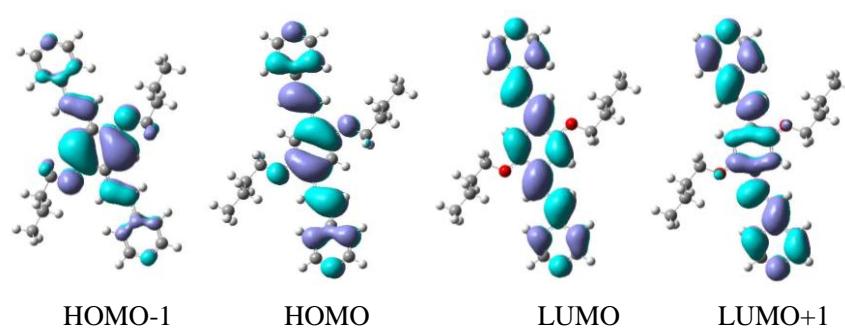
**Fig. 4S** As-experimental and as-simulated XRD patterns for polymorph- $\alpha$  and polymorph- $\beta$ .



**Fig. S5** Packing diagrams of polymorph- $\alpha$



**Fig. S6** Packing diagrams of polymorph- $\beta$



**Fig. S7** Molecular orbital plots of L (The molecular structures of the species studied were calculated at TDDFT level of theory).

**Table 1.** Crystallographic data for polymorph- $\alpha$ , polymorph- $\beta$  and  $[\text{Zn}(\text{phen})_3]_2\mathbf{L}(\text{ClO}_4)_4$

Compound	Polymorph- $\alpha$	Polymorph- $\beta$	$[\text{Zn}(\text{phen})_3]_2\mathbf{L}(\text{ClO}_4)_4$
Chemical Formula	$\text{C}_{28}\text{H}_{32}\text{N}_2\text{O}_2$	$\text{C}_{56}\text{H}_{64}\text{N}_4\text{O}_4$	$\text{C}_{100}\text{H}_{80}\text{Cl}_4\text{N}_{14}\text{O}_{18}\text{Zn}_2$
Formula weight	428.56	428.56	2038.32
Crystal system	Tetragonal	Triclinic	Triclinic
Space group	$I4_1/a$	$P\bar{1}$	$P\bar{1}$
$a(\text{\AA})$	27.908(5)	8.5234	10.976(5)
$b(\text{\AA})$	27.908(5)	10.7419	12.928(5)
$c(\text{\AA})$	6.521(5)	14.505(3)	18.049(5)
$\alpha$	90.000(5)	73.507(2)	90.231(5)
$\beta$	90.000(5)	83.091(2)	92.082(5)
$\gamma$	90.000(5)	76.901(2)	113.369(5)
$V(\text{\AA}^3)$	5079(4)	1238.0(4)	2349.0(15)
$Z$	8	2	2
$D_{\text{calcd.}} [\text{g}\cdot\text{cm}^{-3}]$	1.121	1.150	1.441
$R_I, wR_2 [I \geq 2\sigma(I)]$	0.0749, 0.1921	0.0841, 0.2463	0.0731, 0.2088
$R_I, wR_2 [\text{all data}]$	0.1679, 0.2462	0.1736, 0.3245	0.1143, 0.2326

**Table 2.** Calculated standard molar enthalpies of formation ( $\Delta H_m$ ) of each conformation (kJ/mol )

	Conformation 1	Conformation 2	Conformation 3	Conformation 4
<b>B3LYP</b>	-12.561037	-12.608192	-12.484656	-12.619073
<b>BLYP</b>	-11.912252	-11.950629	-11.826337	-11.962908
<b>B3P86</b>	-16.716701	-16.761831	-16.639083	-16.772874

**Table 3.** Solid-state photophysical properties for the crystals

Crystal	$\lambda_{\text{ab}}$ (nm)	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	$\tau_F$ (ns)
<b>Polymorph-<math>\alpha</math></b>	392	400	505	3.46
<b>Polymorph-<math>\beta</math></b>	403	400	525	4.82
<b><math>[\text{Zn}(\text{phen})_3]_2\mathbf{L}(\text{ClO}_4)_4</math></b>	412	400	488	3.21